

**Intramolecular Hydrofunctionalization Reactions of  
Alkenyl Amines Catalyzed by Disulfonimides**

(ジスルホンイミドを用いたアルケニルアミンのヒドロアミノ化反応)

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Intramolecular Hydrofunctionalization Reactions of  
Alkenyl Amines Catalyzed by Disulfonimides  
(ジスルホンイミドを用いたアルケニルアミンのヒドロアミノ化反応)

## 2. Articles

(1) Bis(trifluoromethanesulfonimide) (BSI): Acidity and application to hydrofunctionalization as a Brønsted acid catalyst

Ryukichi Takagi, Yuichiro Sakai, Duyen Thi Duong

*Tetrahedron*. **2021**,132037.

(2) Disulfonimide catalyzed asymmetric intramolecular hydroamination of alkenyl thioureas: Concentration effect in the hydroamination

Ryukichi Takagi, Duyen Thi Duong, Toshiya Ichiki

*Tetrahedron*. **2021**,132332.

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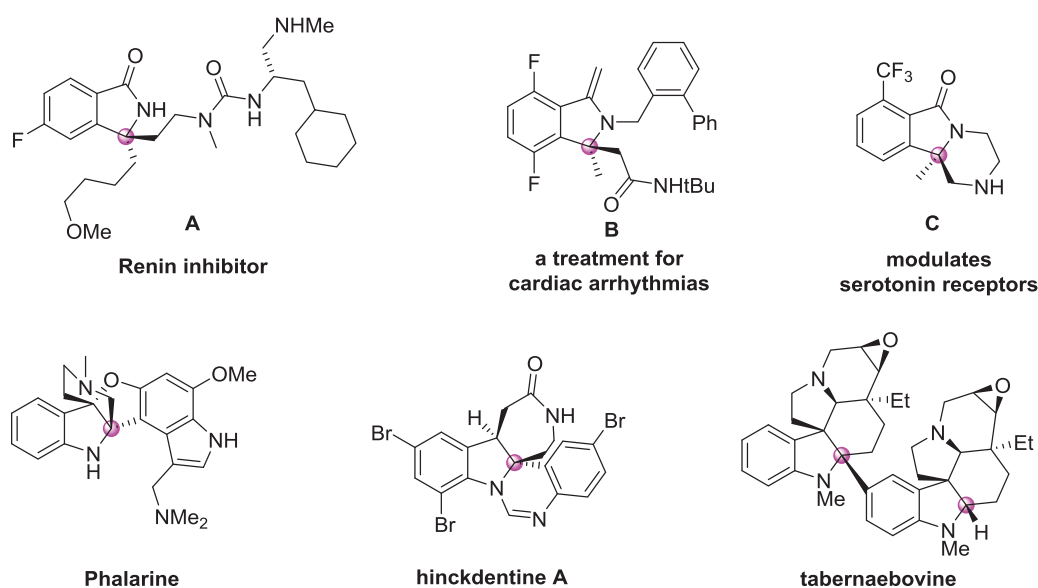


## **Chapter 1. General Introduction**

## 1-1. Introduction

Chirality is a geometric property of many organic molecules which are nonsuperimposable with its mirror image in the same way as a right hand can not be superimposed on a left hand. A chiral molecule and its mirror image are defined as being a pair of enantiomers. Chirality is at the base of life on Earth because all living beings are composed of chiral molecules in a pure enantiomeric form. That has significant effects on the production of medicine, which interacts specifically with receptors in living organisms like cellular receptors, or proteins, or nucleic acids. Thus, just one enantiomer of a drug has a desired pharmacological activity, whereas the other enantiomer can cause serious effects. Therefore, it is very essential to create bioactive compounds in a single and pure enantiomeric form.

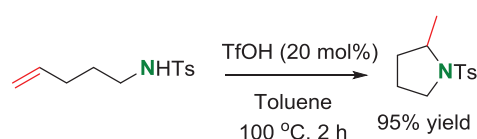
Nitrogen-atom containing heterocycles are the structural cores of enormous drugs and biologically active natural products<sup>1</sup>. Indeed, it is estimated that currently about 40% of the chiral drugs sold on the market having a chiral cyclic amine as the structural motif. Especially, cyclic amines with an alpha-tetrasubstituted carbon stereocenter as a core structure are often found in many drugs and bioactive natural products such as chiral isoindolinones **A-C** (**A**: renin inhibitor<sup>2</sup>, **B**: drug for the treatment of cardiac arrhythmias<sup>3</sup>, **C**: modulator of serotonin receptors<sup>4</sup>). Some more complex C<sub>2</sub>-arylidoline, such as phalarine<sup>5</sup>, hinckdentine **A**<sup>6</sup>, and tabernaebovine<sup>7</sup>, have been also isolated from natural products (Figure 1).



**Figure 1.** Several cyclic amines with an alpha-tetrasubstituted carbon stereocenter to the nitrogen atom.

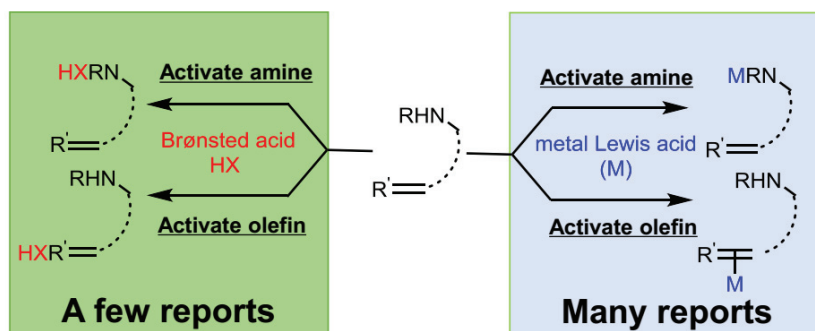
Among the various approaches to the synthesis of nitrogen-containing heterocycles, the Brønsted acid-catalyzed hydroamination reaction of alkenyl amines is one of the atom-economical and effective methods. Because the addition of a nucleophilic amine to the

protonated olefin with Brønsted acid provides directly highly substituted nitrogen-containing heterocycles in a single step<sup>8-11</sup> (Scheme 1). In 2002, Hartwig et al. performed the hydroamination of alkenyl amine (RNHTs) in the presence of strong acids such as triflic acid, they found that cyclic amines were formed in excellent yields<sup>4</sup>. After the pioneering work, other Brønsted acids such as perfluorous superacid (C<sub>8</sub>F<sub>17</sub>SO<sub>3</sub>H)<sup>9</sup> and triflic acid (TfOH)<sup>12,13</sup> have been also reported to catalyze the hydroamination. Moreover, Lewis acid-mediated hydroamination using BF<sub>3</sub>·Et<sub>2</sub>O<sup>10</sup> and Ca(NTf<sub>2</sub>)<sub>2</sub><sup>11</sup> have been also developed. However, these Brønsted acids and Lewis acids have some drawbacks as corrosivity and moisture sensitivity, and so on.



**Scheme 1.** Brønsted acid-catalyzed hydroamination of alkenyl amine.

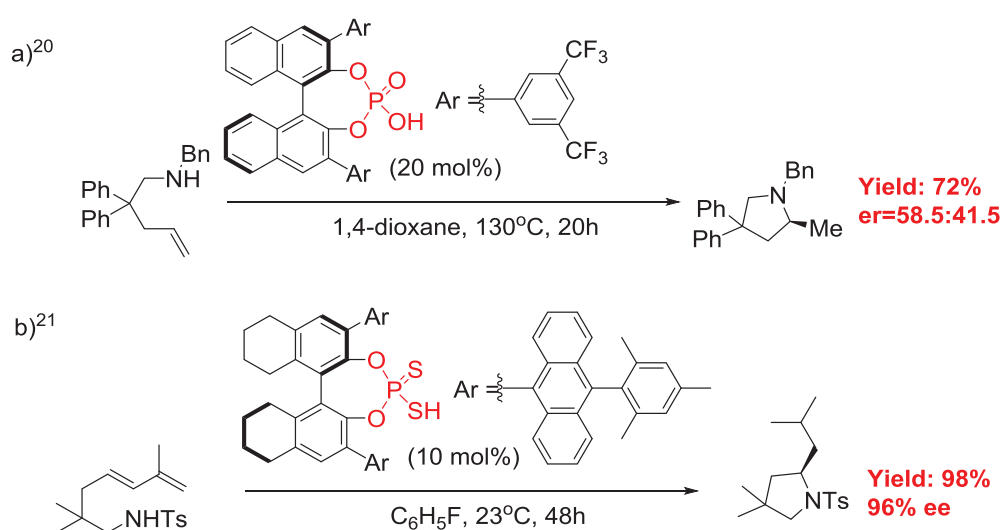
On asymmetric hydroamination, there have been many reports using metal Lewis acids as catalysts. Because Lewis acidic metals can directly activate the olefin or amine moiety of alkenyl amine<sup>14</sup>. For these Lewis acidic metals, rare and/or toxic elements have been often used. In addition, some chiral metal catalysts used on hydroamination have generally a lower activity, and higher temperatures and high catalyst loadings are needed for good results. As a result, asymmetric hydroamination catalyzed by metal Lewis acid might be non-environmentally friendly.



**Figure 2.** Activation modes in asymmetric hydroamination of alkenyl amines catalyzed by metal Lewis acid or Brønsted acid.

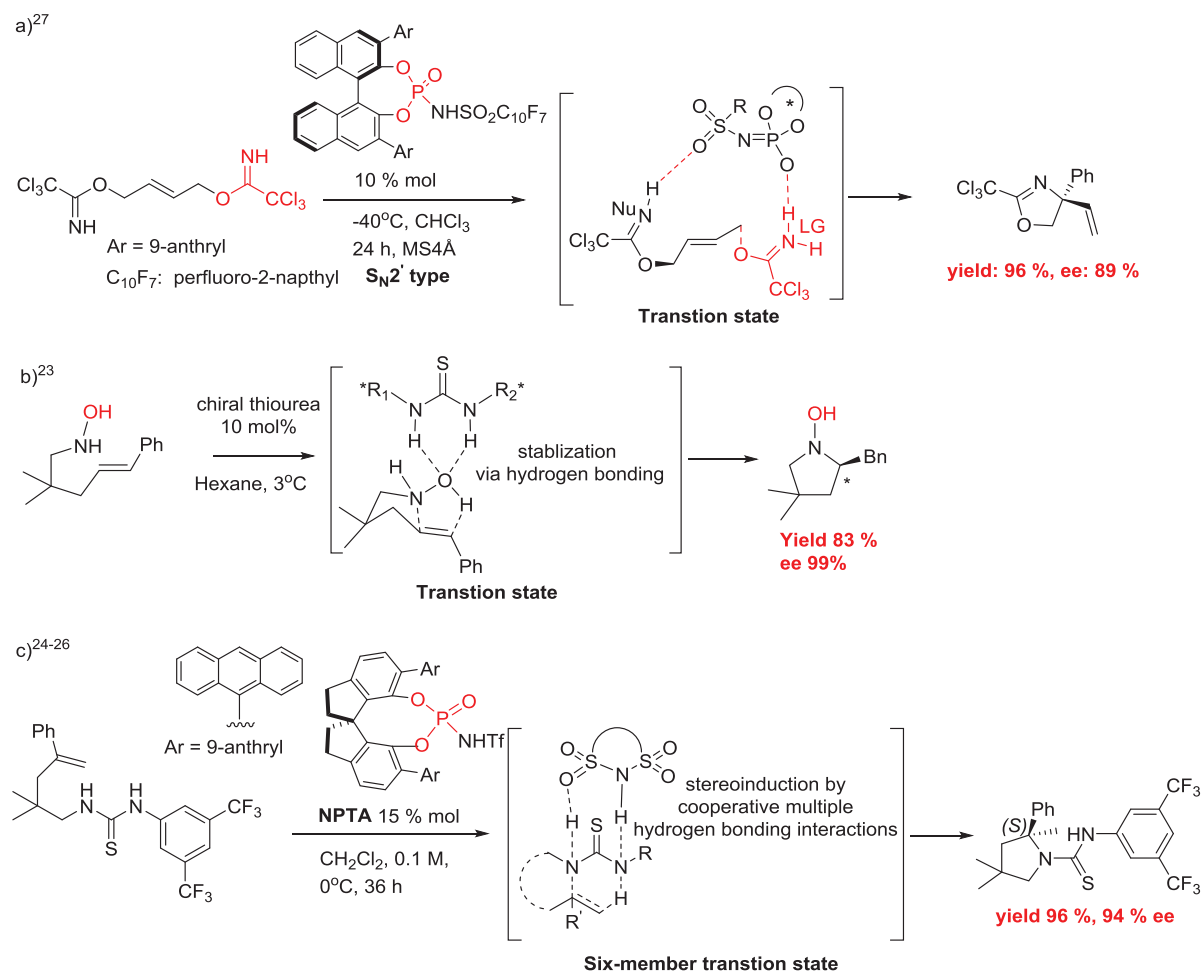
In contrast to these drawbacks of metal Lewis catalysts, organic Brønsted acids are consisted of organic elements, environmentally friendly, and not sensitive to moisture and oxygen<sup>15-18</sup>. Their tunability and their tolerance toward diverse functional groups are also advantageous points. Despite their superior advantages, the use of chiral organic Brønsted acids

on hydroamination reaction of alkenyl amines has been less studied because Brønsted acids are not sufficiently acidic to activate the olefin moiety. So far, only a limited number of examples have been reported for chiral organic Brønsted acid-catalyzed asymmetric hydroamination of alkenyl amine via direct activation of the olefin moiety (Scheme 2)<sup>19–22</sup>. The hydroamination of terminal alkenyl amine catalyzed by chiral phosphoric acid needed high temperature and the hydroamination product was obtained with low enantioselectivity (Scheme 2a)<sup>20</sup>. Toste et al. have reported that the hydroamination of dienyl amine catalyzed by dithiophosphoric acid affords hydroamination product with good yield and high enantioselectivity under mild reaction conditions (Scheme 2b)<sup>21</sup>. In this reaction, more nucleophilic diene and more acidic dithiophosphoric acid were used as a substrate and Brønsted acid, respectively, in order to accelerate the hydroamination.



**Scheme 2.** Chiral Brønsted acid-catalyzed hydroamination of alkenyl amine via the direct activation of the olefin moiety.

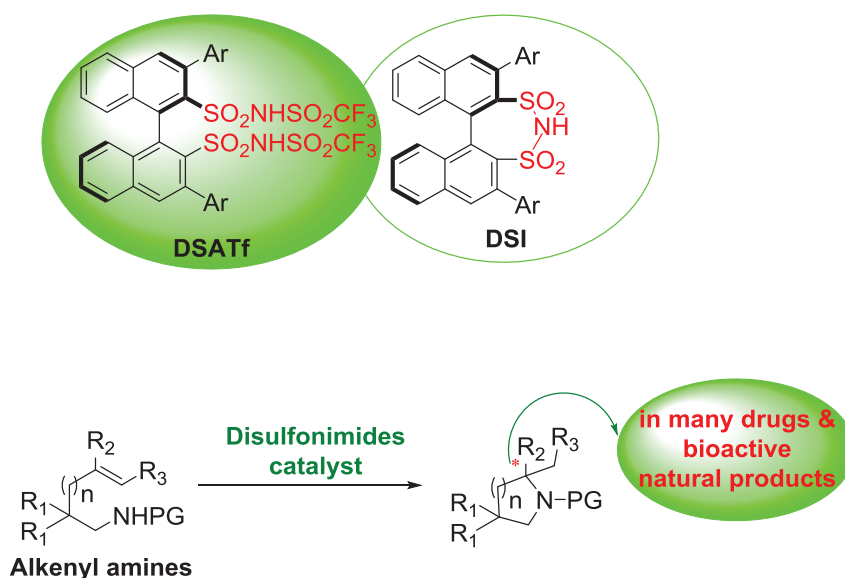
In order to avoid the acidity limitation in chiral Brønsted acid catalyzed hydroamination of alkenyl amines, a few asymmetric hydroamination of alkenyl amines with specific protecting groups, which are activated by Brønsted acid, have been developed (Scheme 3)<sup>23–27</sup>. The introduction of a leaving group at the allylic position of an alkenyl amine enables the activation of the leaving group and the nitrogen atom via hydrogen bonding with the chiral phosphoric acid, leading to formal enantioselective hydroamination via an intramolecular S<sub>N</sub>2' reaction (Scheme 3a)<sup>27</sup>. The enantioselective hydroamination of alkenyl amines *N*-substituted with basic functional groups, such as hydroxyl amine<sup>23</sup> or thiourea group<sup>24–26</sup>, have been developed (Scheme 3b and c). In these reactions, chiral Brønsted acid coordinates to the functional groups on the amine moiety to create an asymmetric environment and promote the protonation of alkene via the activated functional group.



**Scheme 3.** Chiral Brønsted acid-catalyzed hydroamination via the activation of amine moiety.

## 1-2. Purpose of this Study

In this study, in order to overcome the acidity limitations of Brønsted acids for the direct olefine activation on hydroamination, a new chiral BINOL-derived disulfonamide trifluoromethanesulfonate (**DSATf**) was designed, based on the strong acidity of Tf<sub>2</sub>NH. **DSATf**-catalyzed hydroamination of alkenyl amines was examined. Hydroamination of alkenyl thiourea catalyzed by a cyclic **DSI** catalyst<sup>28</sup>, which was developed by List et al., was also examined. The mechanism of the hydroamination reaction was also investigated using density functional theory calculations to understand the behavior of the reactions and the origin of enantioselectivity.

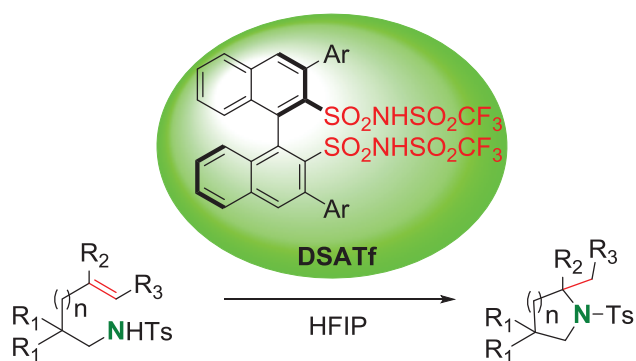


## Chapter 2

### Synthesis of New Strong Brønsted Acids

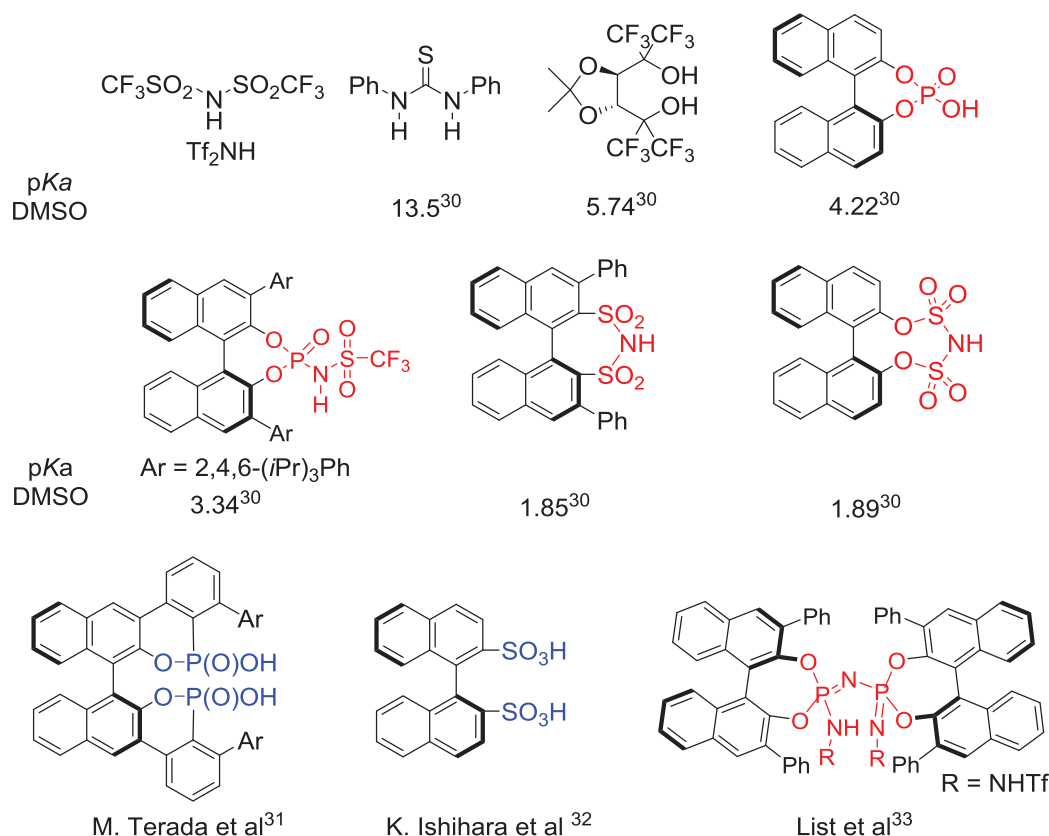
and

### its Application to Hydroamination



## 2-1. Design of a New Strong Brønsted Acid

Brønsted acid catalysts with high acidity had the possibility to give the product with good yield and enantioselectivity in some reported asymmetric reactions<sup>29</sup>. Thus, many chiral organic Brønsted acids have been developed in order to create a strong chiral Brønsted acid (Figure 1)<sup>30–33</sup>.

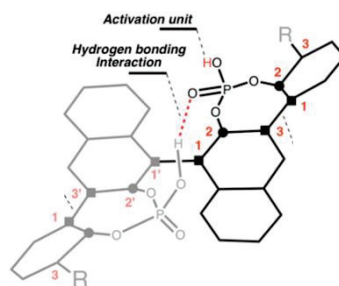


**Figure 1.** Some chiral strong Brønsted acids and experimental  $pK_a$  values

So far, there are two strategies for the design of Brønsted acids. The first one is the introduction of strongly electron-withdrawing groups, such as *N*-triflyl phosphoramidate<sup>10</sup>, disulfonimide<sup>34–36</sup>, imidodiphosphate<sup>37–41</sup>, and imidodiphosphorimidate<sup>22,33</sup> onto the chiral binaphthol (BINOL) skeleton to increase the acidity of the Brønsted acids. The second strategy is the stabilization of the conjugated base of Brønsted acids by intramolecular hydrogen bonds from the adjacent functional groups. Based on the second strategy, chiral Brønsted acids bearing two strongly electron-withdrawing groups, such as dicarboxylic acid<sup>42</sup>, disulfonic acid<sup>32,43–45</sup>, diphosphoric acid<sup>31,46,47</sup>, tether-linked diphosphoric acid<sup>48–50</sup> and conjugate-base-stabilized carboxylic acids<sup>51–54</sup> have been developed. Terada and co-workers used X-ray diffraction analysis to reveal an intramolecular hydrogen bond between the two phosphoric acid moieties in diphosphoric acid (Figure 2)<sup>31</sup>. This suggests that intramolecular hydrogen bonds in a

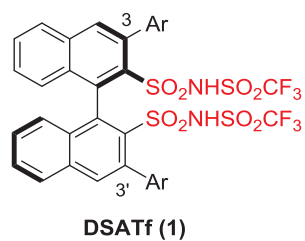


Brønsted acid bearing two adjacent strongly electron-withdrawing groups possibly play an important role in attaining high reactivity and stereoselectivity.



**Figure 2.** X-ray analysis bis-phosphoric acid

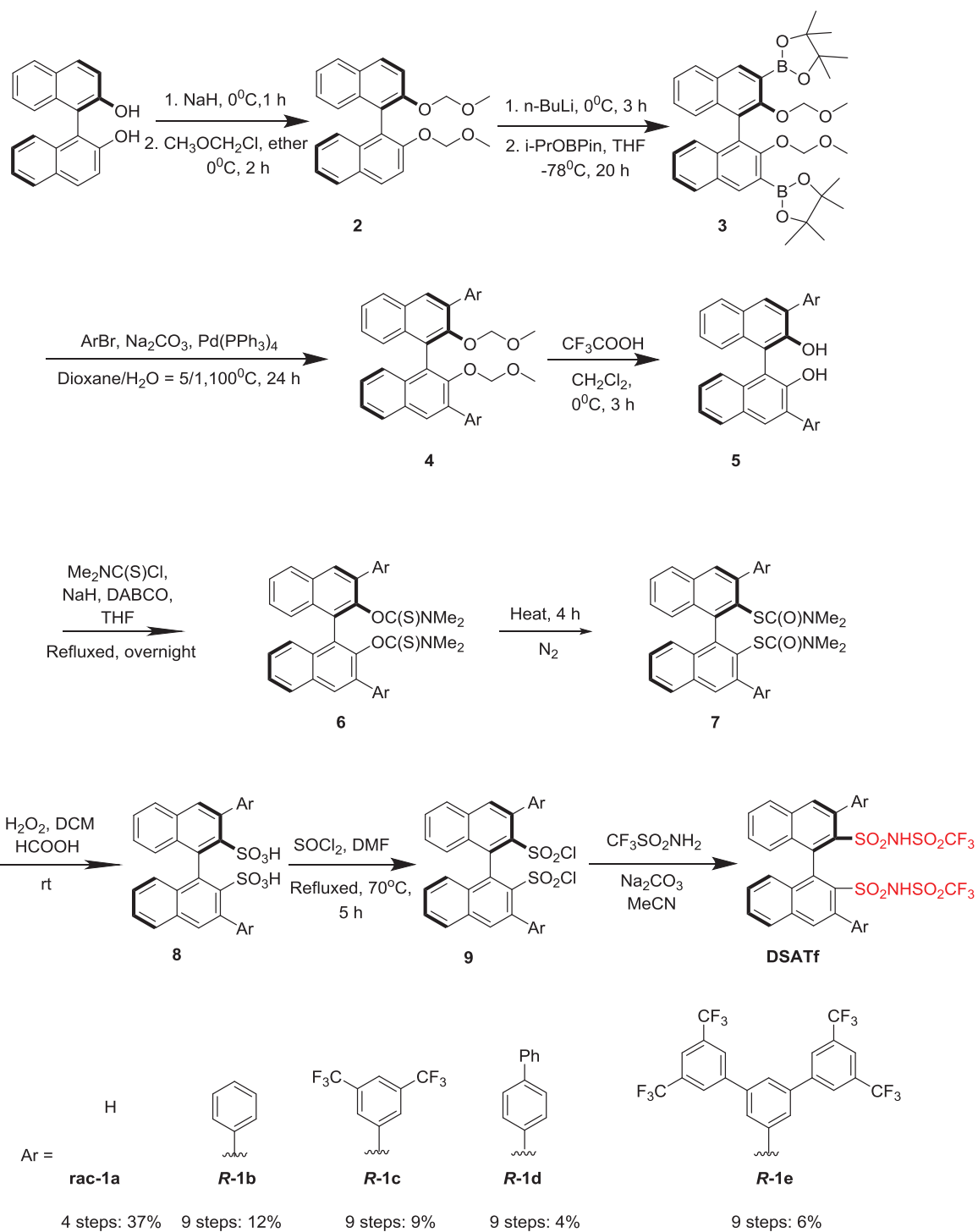
On the basis of these hypotheses, a new BINOL-derived disulfonamide trifluoromethanesulfonate (**DSATf**) was designed, based on the strong acidity of a superacid  $\text{Tf}_2\text{NH}^{55}$  (Figure 3).



**Figure 3.** DSATf (1)

## 2-2. Synthesis of New Strong Brønsted Acids DSATf

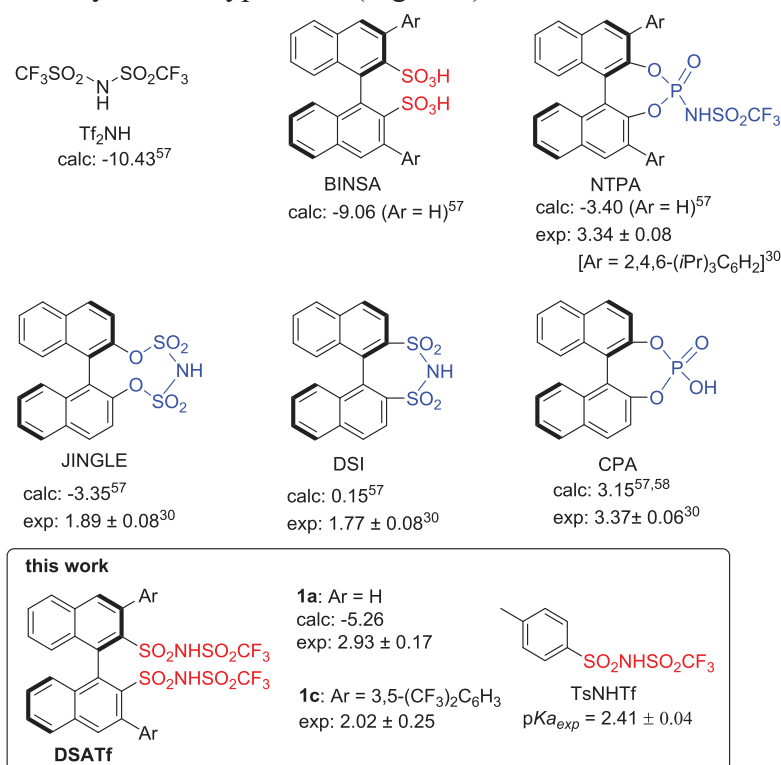
**DSATf 1** was prepared according to the literature method (Scheme 1)<sup>36,56</sup>. Two hydroxy groups of *R*-binaphthanol were protected with methoxymethyl group in diethyl ether to produce compound **2** and then was coupled with dioxaborolane to give compound **3**. The compound **3** coupled with ArBr using Na<sub>2</sub>CO<sub>3</sub> and Pd[(Ph)<sub>3</sub>P]<sub>4</sub> in dioxane: H<sub>2</sub>O to produce the desired product **4**. Next, compound **4** went through the deprotection to get compound **5**. Compound **5** was carbamothioated to produce compound **6** and S-rearranged via Newman–Kwart rearrangement by heating at 60°C to get compound **7**. The compound **7** was easily oxidized in dichloromethane by a performic acid solution to give sulfonic acid **8**. After chlorination of sulfonic acid, compound **9** was formed. Lastly, the introduction of trifluoromethanesulfonamide group to **9** was conducted via substitution reaction with TfNH<sub>2</sub> in MeCN to give **DSATf (1)**. An observation was that non-substituted disulfonamide **1a** absorbed moisture from the air so much like Tf<sub>2</sub>NH, but disulfonamides **1b-d** were not so.



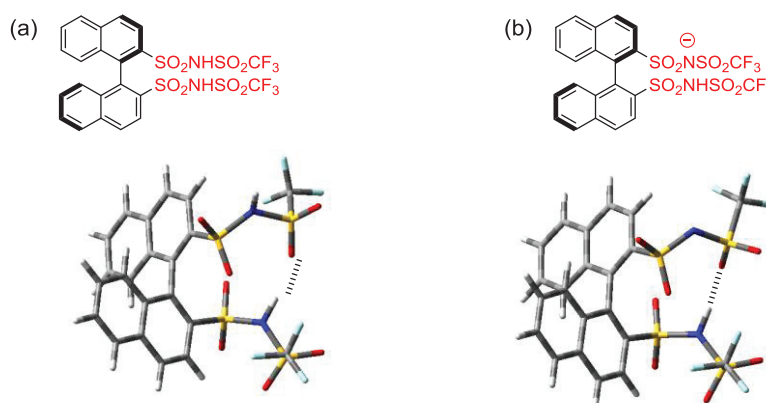
**Scheme 1.** Synthesis of **DSATf (1a-e)**

## 2-3. p*K*<sub>a</sub> value of DSATf

In order to evaluate the acidic strength of **DSATf**, the computational prediction of p*K*<sub>a</sub> value using the direct method at the SMD/M06-2x/6-311++G(2df,2p)//B3LYP/6-31+G(d) level of theory<sup>57,58</sup> and the experimental measurement of the p*K*<sub>a</sub> value by an overlapping indicator method via UV-Vis spectrophotometric titrations in DMSO<sup>30</sup> were performed (Figure 4). The predicted p*K*<sub>a</sub> value (-5.26) of **1a** suggested that the acidity of **DSATf 1a** was stronger than that of **NPTA** and weaker than that of **BINSA**. The optimized structure of **1a** and its conjugated base indicated that the intramolecular hydrogen bond between the two sulfonimides plays an important role in high acidity, as our hypothesis (Figure 5).

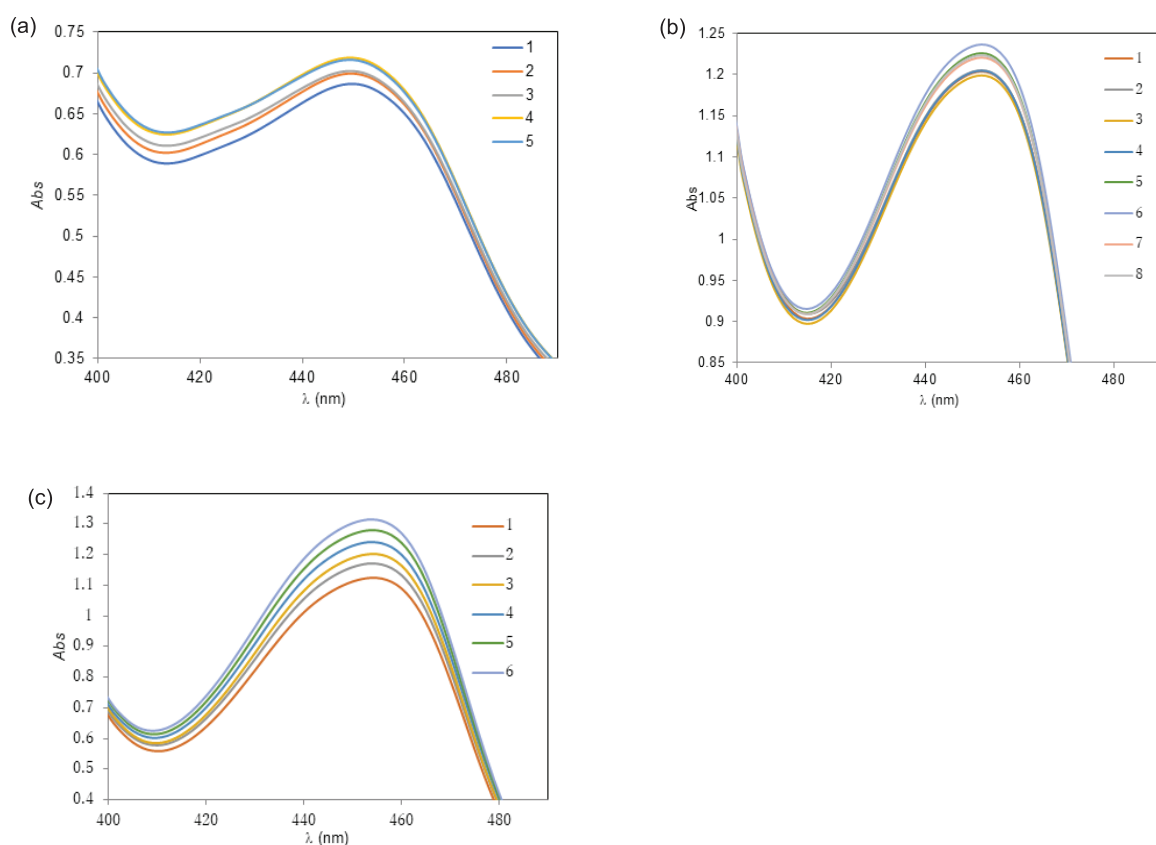


**Figure 4.** Computational and experimental p*K*<sub>a</sub> values of **DSATf** and Brønsted Acids in DMSO.



**Figure 5.** Optimized structure of (a) **1a** and (b) anion of **1a** at the B3LYP/6-31+G(d) level of theory.

The experimental  $pK_a$  values of **DSATf 1a** and **1c** were measured with an overlapping indicator method via UV-Vis spectrophotometric titrations reported by O'Donoghue and Berkessel using 2,4-dinitronaphthol as an indicator (Figure 6)<sup>30</sup>. The experimental  $pK_a$  value (2.02) of **1c** suggested that the acidity of **DSATf 1c** was stronger than that of **NPTA** and was similar to that of **DSI**. The experimental  $pK_a$  value (2.41) of monosulfonamide TsNHTf is larger than that (2.02) of **DSATf 1c**. That indicated the adjacent sulfonamide groups contributed considerably to enhance the acidity of **DSATf 1c**. From the experimental measurement and the computational calculation, it showed that the acidity of **DSATf 1** was one of the strong Brønsted acids.






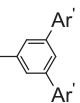
**Figure 6.** UV/Vis spectra of 2,4-dinitronaphthol in DMSO containing **DSATf** (a: **1a**, b: **1c**, c: TsNHTf) and conjugate base.

## 2-4. DSATf Catalyzed Hydroamination of Alkenyl Amines

### 2-4.1. Optimization of 3,3'-substituents (Ar) of BINOL Backbone of DSATf

After the estimation of the acidity of **DSATf**, the efficiency of **DSATf** as a catalyst in the hydroamination of alkenyl amines was evaluated. First, the catalyst was optimized by increasing the size of the 3,3'-substituents of the BINOL backbone (Table 1). The hydroamination reaction using 20 mol% catalysts was performed in DCM at 40°C for 7 days. **DSATf 1a** and **1b** provided the product in low yield (23% and 26%) and enantioselectivity (0% ee and 15% ee) (entries 1-2). The results showed that catalysts **1a** and **1b** were not sufficiently acidic for this reaction. Increasing the size of 3,3'-substituents (Ar) slightly affected the enantioselectivity (entries 1-4) but too large substituents decreased the enantioselectivity (entry 5). Introducing two CF<sub>3</sub> groups to the phenyl ring of catalyst could give the product with the best yield (81%) because of increased acidity of the catalyst by introducing two electron-withdrawing groups CF<sub>3</sub> (entry 3).

**Table 1.** Asymmetric hydroamination using **DSATf 1a-e**

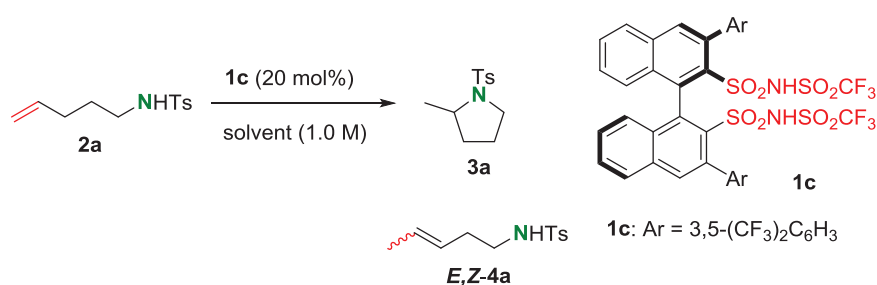
entry	catalyst	yield (%) <sup>a)</sup>	e.e. (%) <sup>b)</sup>
1	<i>rac</i> - <b>1a</b> : Ar = H	23	-
2	<i>R</i> - <b>1b</b> : Ar = 	26	15
3	<i>R</i> - <b>1c</b> : Ar = 	<b>81</b>	3
4	<i>R</i> - <b>1d</b> : Ar = 	23	19
5	<i>R</i> - <b>1e</b> : Ar =  Ar' = 3,5-(CF <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	37	10

[a] Isolated yield. [b] Determined by chiral HPLC analysis. Ts = SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>*p*-Me.

## 2-4.2. Optimization of Solvents

The reaction solvent in the hydroamination of alkenyl amine **2a** catalyzed **1c** was next evaluated (Table 2). In DCM and benzene, the reaction needed 7 days at 40°C to consume **2a** (entries 1 and 2). Under these conditions, the product was obtained with a good yield (81% and 70%) but no enantioselectivity was observed. The hydroamination in HFIP solvent proceeded smoothly and effectively even at 20°C (entry 3). Because polyfluorinated alcohols have unique properties, including their protic and non-nucleophilic character, excellent hydrogen bonding donor ability, high ionizability beneficial for stabilizing cationic species, and high polarity<sup>11,59,60</sup>. Unfortunately, TFE is not effective in the reaction with the chemical yield 25% (entry 4). At a higher temperature 60°C, the reaction in HFIP was accelerated and gave the product **3a** in 73% for 12 h (entry 5). In the reaction, the olefin-isomerized product **4a** was also obtained as a byproduct. Thus, performing the reaction at 60°C in HFIP is the best synthetic condition. No enantioselectivity was observed in these reactions because two functional groups SO<sub>2</sub>NHSO<sub>2</sub>CF<sub>3</sub> were so flexible that led to being difficult to control conformation. Therefore, only the catalytic ability of catalyst **1c** was focused on further examinations.

**Table 2.** Optimization of reaction conditions of hydroamination.



entry	conditions	yield (%) <sup>a</sup>	e.e.(%) <sup>b</sup>
1	DCM, 40°C, 7 days	81	3
2	Benzene 40°C, 7 days	70	0
3	HFIP, 20°C, 4 days	82 <sup>c</sup>	N.D.
4	TFE, 20°C, 4 days	25	0
5	HFIP, 60°C, 12 h	73	N.D.

[a] Isolated yield. [b] Determined by chiral stationary phase HPLC analysis.

[c] Alkene **4a**, which was generated from olefin isomerization of **2a**, was also obtained: *E*-**4a**: 2.5% yield, *Z*-**4a**: 1.8% yield. Ts = *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>. DCM = CH<sub>2</sub>Cl<sub>2</sub>. HFIP = (CF<sub>3</sub>)<sub>2</sub>CHOH. TFE = CF<sub>3</sub>CH<sub>2</sub>OH, N.D. = not determined.

### 2-4.3. Comparison of catalytic efficiency of some Brønsted acids with DSATf

The hydroamination reaction of alkenyl amine **2a** using 20 mol% of **DSATf 1c** and several common Brønsted acids was performed in HFIP at 20°C for 4 days in order to compare the catalytic efficiencies (Table 3). The hydroamination catalyzed by TsOH.H<sub>2</sub>O, Tf<sub>2</sub>NH, TfOH, and **1c** produced pyrrolidine **3a** in good chemical with the small amount of the isomerized product **4** (entries 1-4). When using phosphoric acid **5**, the olefin-isomerized product **4a** became the major product (entry 5). Because the nucleophilicity of the conjugated base of **5** was stronger than that of **1c**, the isomerization reaction was promoted to form the olefin-isomerized product **4a**.

**Table 3.** Comparison of the catalytic efficiency of some Brønsted acids with **DSATf**.

entry	catalyst	con(%) <sup>a</sup>	Yield (%) <sup>b</sup>		
			<b>3a</b>	<b>E-4a</b>	<b>Z-4a</b>
1	TsOH.H <sub>2</sub> O	96	89	3.9	3.2
2	Tf <sub>2</sub> NH	94	88	3.6	2.3
3	TfOH	97	85	1	0.6
4	<b>1c</b>	86	82	2.5	1.8
5	<b>5</b>	81	7	61	6
6	<b>6</b>	48	33	9.4	3.3

[a] Determined by <sup>1</sup>H NMR measurement of the crude product using 1,3,5-trimethoxybenzene as an internal standard. [b] Isolated yield. HFIP = (CF<sub>3</sub>)<sub>2</sub>CHOH. Tf<sub>2</sub>NH = (CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>NH. TfOH = CF<sub>3</sub>SO<sub>3</sub>H

The time profile of the chemical yield of pyrrolidine **3a** in the several Brønsted acid catalyzed hydroamination reactions of **2a** was investigated (Figure 7). The result showed that the order of the catalytic efficiency of Brønsted acid was TsOH·H<sub>2</sub>O, Tf<sub>2</sub>NH > TfOH ≈ **DSATf** >> TsNHTf (**6**). The poor hydroamination result catalyzed by monosulfonamide **6** showed the importance of adjacent sulfonamide groups for the high acidity of **DSATf**. This result is consistent with the experimental pK<sub>a</sub> of **DSATf** and TsNHTf (**6**) (Figure 4).





## 2-5. Scope of Hydroamination Catalyzed by DSATf

The scope of **1c**-catalyzed hydroamination was next examined under the optimized conditions at 60°C in 1M HFIP (Table 4). With substrate substituted with a *gem*-dimethyl group (**2b**), the reaction proceeded quickly only 3h (entry 1). Because of the conformational effect by the *gem*-dimethyl group, the product was obtained with a good yield (92%). In entries 2 and 3, the hydroamination of alkenyl amine **2c** containing an internal olefin required a longer reaction time to consume the starting material. In entry 2, the hydroamination of *E*-**2c** produced pyrrolidine **3c** as a single product with a high yield (91%). Unlike *E*-**2c** substrate, the formation of the two products (**3c** and **4c**) was observed during the hydroamination of *Z*-**2c** alkenyl amine (entry 3). In entry 4, pyrrolidine **3c** was formed as a major product with a chemical yield 50%. Because the olefin-isomerization of alkenyl amine was a competing reaction under hydroamination. On the hydroamination of **2e** with the phenyl substituent at the olefin moiety, the six-membered ring piperidine **4e** was formed with the chemical yield 61% as a single product through an intermediate with the benzylic stabilization (entry 5). These data suggested that the reaction rate of the cyclization at 60°C followed the order of 5-*exo* > 5-*endo* > olefin-isomerization > 6-*exo* (entries 0-4) and the benzylic cation intermediate was generated from **2e** and **2f** as a stable cation intermediate at that temperature (entries 5-6).

**Table 4.** Scope of hydroamination catalyzed by DSATf **1c**.

**1c:** Ar = 3,5-(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>

entry	substrate	time	product [a]	ee [b]		entry	substrate	time	product [a]
0		12 h	 <b>3a: 73%</b>			4[c]		7 h	 <b>3c: 50%</b> <b>4c: 14%</b>
1		3 h	 <b>3b: 92%</b>	3%		5[d]		1 h	 <b>3e: 0%</b> <b>4e: 61%</b>
2		24 h	 <b>3c: 91%</b>	2%		6		9 h	 <b>3f: 85%</b>
3		96 h	 <b>3c: 83%</b> <b>4c: 2%</b>						

[a] Isolated yield. [b] Determined by chiral stationary phase HPLC analysis.

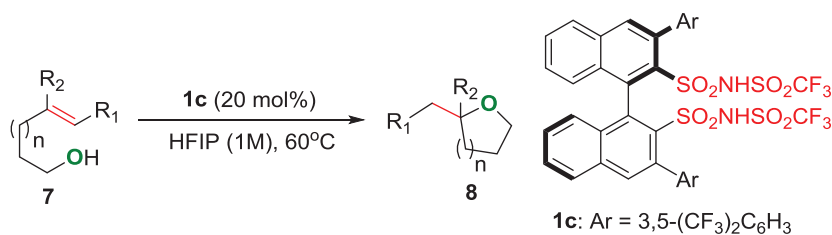
[c] A mixture of *E*- and *Z*-N-toluenesulfonyl 4-hexen-1-amine was also obtained in 8% yield.

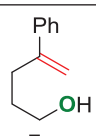
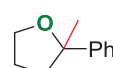
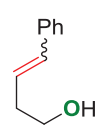
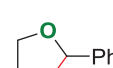
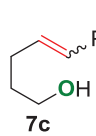
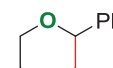
[d] When the reaction was carried out at 20°C for 4 h, a mixture of pyrrolidine derivative and piperidine derivative was obtained (pyrrolidine derivative **3e**: 20% yield, piperidine derivative **4e**: 5% yield).

## 2-6. Scope of Hydroalkoxylation Catalyzed by DSATf

The scope of hydroalkoxylation of alkenyl alcohol catalyzed by **DSATf 1c** was conducted (Table 5). The results showed that the hydroalkoxylation of alkenyl alcohol containing an internal olefin needed a longer reaction time to completely consume the starting material than that of terminal olefin derivatives. The cyclization occurred at the benzylic position, in which the most stable benzylic cation was generated. These results indicate **DSATf** can also catalyze the hydroalkoxylation of alkenyl alcohols in HFIP with high efficiency.

**Table 5.** Scope of hydroalkoxylation catalyzed by **DSATf 1c**.

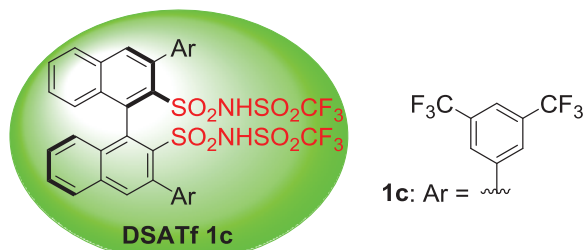


entry	substrate	time	product [a]
1	 <b>7a</b>	1 h	 <b>8a:</b> 78%
2 [b]	 <b>7b</b>	4 h	 <b>8b:</b> 47%
3 [c]	 <b>7c</b>	4 h	 <b>9c:</b> 62 %

[a] Isolated yield. [b] A mixture of *E*- and *Z*-isomer (*E*:*Z* = 5.3:1) was used as the substrate.

[c] A mixture of *E*- and *Z*-isomer (*E*:*Z* = 22:1) was used as the substrate.

## 2-7. Conclusion



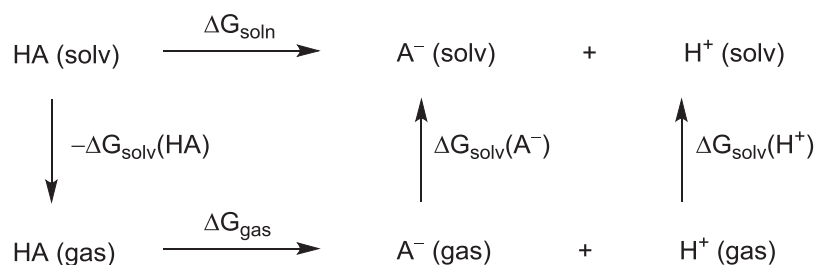
Our new designed disulfonamide

In summary, a new Brønsted acid containing disulfonamide trifluoromethanesulfonate (**DSATf**) functionality has been developed. The computational prediction and experimental measurement of the  $pK_a$  value of **DSATf** categorized it as a strong Brønsted acid. The **DSATf** catalyzed hydroamination reaction proceeded smoothly in HFIP solvent. The catalytic ability order of Brønsted acid in hydroamination: TsOH·H<sub>2</sub>O, Tf<sub>2</sub>NH > TfOH ≈ **DSATf** >> TsNHTf. However, the enantioselectivity in the hydroamination was not observed. A comparison of the catalytic efficiency of **DSATf** and TsNHTf indicated the importance of adjacent sulfonamide groups for enhancing the acidity of **DSATf**. **DSATf** also catalyzed the hydroalkoxylation of alkenyl alcohols in HFIP with high efficiency. The advantages of **DSATf** is including its strong Brønsted acidity, reusability, and ease of handling than TsOH·H<sub>2</sub>O, Tf<sub>2</sub>NH, and TfOH.

## 2-8. Experimental Section

### 2-8-1. Details of computational methods

The structures of all species were carried out with the Gaussian 09 package<sup>61</sup>. Geometry optimizations were conducted at the B3LYP/6-31+G(d) level. The nature of the stationary points was confirmed by frequency calculations at the same level of theory. The solution phase free energy calculations were performed by virtue of SMD model at the M06-2x/6-311++G(2df,2p) level. The direct method was applied to predict  $pK_a$  values (Scheme S1)<sup>58,62-69</sup>. As a proton solvation free energy ( $\Delta G_{\text{solv}}(\text{H}^+)$ ) in DMSO,  $-268.34$  kcal/mol was used<sup>30</sup>. The free energy of acid dissociation in DMSO ( $\Delta G_{\text{soln}}$ ) can be obtained through eq. 1. Then,  $pK_a$  can be obtained through the thermodynamic relationship (eq. 2). In the equations below, asterisks (\*) indicate a standard state of 1 mol/L in any phase<sup>70</sup>.



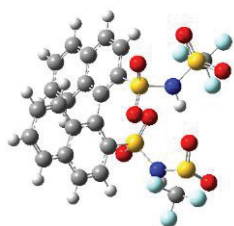
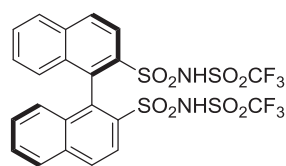
$$\Delta G_{\text{solv}}^* = \Delta G_{\text{gas}}^* + \Delta G_{\text{solv}}^*(\text{A}^-) + \Delta G_{\text{solv}}^*(\text{H}^+) - \Delta G_{\text{solv}}^*(\text{HA}) \quad (\text{eq 1})$$

$$\Delta G_{\text{solv}}^*(\text{H}^+) = -268.34 \text{ kcal/mol}$$

$$pK_a = \frac{\Delta G_{\text{soln}}^*}{RT \ln(10)} \quad (\text{eq 2})$$

**Scheme S1.**  $pK_a$  Calculation via the direct method.

## B3LYP/6-31+G(d) Optimized cartesian coordinates and energies for DSATf 1a



Zero-point correction = 0.355716 (Hartree/Particle)

Thermal correction to Energy = 0.393704

Thermal correction to Enthalpy = 0.394648

Thermal correction to Gibbs Free Energy = 0.281805

Sum of electronic and zero-point Energies = -3749.303256

Sum of electronic and thermal Energies = -3749.265268

Sum of electronic and thermal Enthalpies = -3749.264324

Sum of electronic and thermal Free Energies = -3749.377167

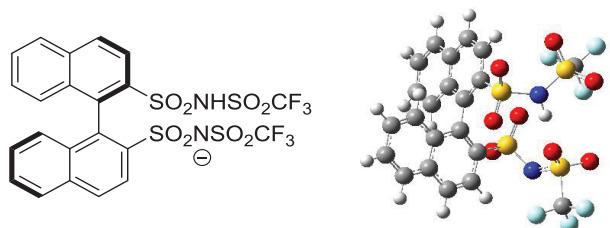
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Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.882229	-1.736623	0.710540
2	6	0	-1.727950	-0.974390	-0.083854
3	6	0	-1.566025	-3.803378	-0.328070
4	6	0	-2.537337	-1.659790	-1.057833
5	6	0	-0.804562	-3.148660	0.605713
6	6	0	-2.440056	-3.085766	-1.185058
7	6	0	-3.447940	-0.971199	-1.910776
8	1	0	-3.146576	-4.836628	-2.239523
9	1	0	-1.506212	-4.885142	-0.415754
10	6	0	-4.204158	-1.651273	-2.840731
11	1	0	-3.544548	0.106176	-1.835544
12	1	0	-4.885478	-1.101961	-3.484582
13	6	0	-4.097906	-3.057118	-2.967379
14	1	0	-4.698710	-3.580523	-3.706165
15	6	0	-3.234512	-3.756294	-2.153761
16	6	0	-1.700909	2.859725	-0.580501
17	1	0	-2.933424	4.302321	0.393750
18	6	0	-2.663037	3.252789	0.312179

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19	6	0	-1.957336	0.502906	0.087806
20	6	0	-3.318003	2.301117	1.137502
21	6	0	-1.341778	1.489888	-0.671092
22	6	0	-2.976931	0.913851	1.018109
23	6	0	-4.314182	2.698512	2.068588
24	1	0	-3.433631	-1.076386	1.765055
25	6	0	-4.958490	1.767581	2.853101
26	1	0	-4.558175	3.754861	2.151040
27	1	0	-5.717339	2.082706	3.564081
28	6	0	-4.632272	0.395776	2.732927
29	1	0	-5.142125	-0.335934	3.353375
30	6	0	-3.670215	-0.020968	1.838158
31	16	0	-0.139009	1.082307	-1.941999
32	8	0	0.544345	-0.165554	-1.640772
33	8	0	-0.710741	1.351767	-3.262101
34	16	0	0.206230	-1.022227	1.956943
35	8	0	0.018478	0.424214	2.031570
36	8	0	0.159580	-1.872249	3.137257
37	7	0	1.785120	-1.147880	1.258997
38	1	0	2.140745	-0.250729	0.912944
39	7	0	1.015955	2.414267	-1.837829
40	1	0	1.125324	2.837642	-2.761330
41	16	0	2.907492	-2.386765	1.341430
42	8	0	4.184396	-1.845845	1.782820
43	8	0	2.282648	-3.561458	1.930762
44	16	0	2.431009	2.536631	-0.933107
45	8	0	2.575124	1.391580	-0.046995
46	8	0	3.483561	3.001032	-1.824594
47	6	0	3.150915	-2.776652	-0.494075
48	6	0	2.021228	4.003384	0.186013
49	9	0	4.076930	-3.736305	-0.598186
50	9	0	3.567551	-1.686009	-1.142868
51	9	0	2.008939	-3.208456	-1.036384
52	9	0	3.114027	4.318735	0.880626
53	9	0	1.653242	5.048659	-0.567261
54	9	0	1.031828	3.687731	1.019426
55	1	0	-0.142279	-3.697363	1.266276
56	1	0	-1.206328	3.589532	-1.210885

## Anion of DSATf 1a



Zero-point correction = 0.343323 (Hartree/Particle)

Thermal correction to Energy = 0.380463

Thermal correction to Enthalpy = 0.381407

Thermal correction to Gibbs Free Energy = 0.271869

Sum of electronic and zero-point Energies = -3748.839328

Sum of electronic and thermal Energies = -3748.802188

Sum of electronic and thermal Enthalpies = -3748.801243

Sum of electronic and thermal Free Energies = -3748.910782

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Center Number	Atomic Number	Forces (Hartrees/Bohr)		
		X	Y	Z
1	6	-0.000001607	0.000003464	0.000002238
2	6	-0.000003130	-0.000001905	-0.000004954
3	6	-0.000003506	-0.000000843	-0.000003442
4	6	0.000001143	-0.000001398	0.000001519
5	6	-0.000002705	-0.000003143	-0.000000203
6	6	0.000000111	-0.000001709	-0.000000180
7	6	0.000001536	-0.000001961	-0.000001373
8	1	-0.000001592	-0.000003093	-0.000002338
9	1	-0.000002826	-0.000001592	-0.000001883
10	6	0.000001291	-0.000004035	-0.000001428
11	1	0.000001808	-0.000002795	-0.000000049
12	1	0.000001702	-0.000003806	-0.000001731
13	6	0.000000184	-0.000002941	-0.000002141
14	1	0.000000350	-0.000004310	-0.000002662
15	6	-0.000001556	-0.000003451	-0.000002781
16	6	0.000001905	-0.000002542	0.000003292
17	1	0.000002943	-0.000000389	0.000000725

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18	6	0.000002385	0.000000911	0.000000334
19	6	0.000000551	-0.000001771	0.000000608
20	6	0.000001492	-0.000000620	-0.000000470
21	6	0.000002445	-0.000000426	-0.000000070
22	6	-0.000000004	0.000000035	-0.000002050
23	6	0.000001719	-0.000001378	0.000000253
24	1	-0.000001181	-0.000001667	-0.000001455
25	6	0.000000621	-0.000002163	-0.000000706
26	1	0.000002688	-0.000001412	-0.000000196
27	1	0.000000947	-0.000002019	-0.000000857
28	6	0.000000076	-0.000002179	-0.000001069
29	1	-0.000000926	-0.000002500	-0.000001773
30	6	-0.000000022	-0.000001985	-0.000001165
31	16	-0.000003285	-0.000000421	-0.000005126
32	8	0.000000636	-0.000003198	0.000002374
33	8	0.000004975	0.000002063	0.000005275
34	16	0.000007917	0.000001921	-0.000011095
35	8	-0.000000321	0.000001949	0.000001331
36	8	-0.000004003	0.000003246	0.000002598
37	7	-0.000014269	-0.000000185	0.000007515
38	1	0.000004508	-0.000004589	0.000002187
39	7	0.000008543	0.000005869	0.000005656
40	16	0.000003425	0.000008426	-0.000002435
41	8	-0.000003561	0.000001618	0.000001158
42	8	-0.000006921	0.000001294	0.000001775
43	16	-0.000006350	-0.000010395	-0.000018917
44	8	-0.000002540	0.000011244	0.000016495
45	8	0.000001758	0.000007604	0.000003175
46	6	-0.000004158	0.000001260	-0.000001611
47	6	0.000007217	0.000011141	0.000010245
48	9	-0.000003204	0.000002976	0.000001589
49	9	-0.000002462	0.000005500	0.000001454
50	9	0.000000643	0.000002888	-0.000000936
51	9	0.000001171	0.000002047	0.000000956
52	9	0.000006744	0.000000032	-0.000002880
53	9	-0.000003616	0.000000607	0.000006142
54	1	-0.000003773	0.000000125	-0.000001649
55	1	0.000004083	0.000000600	0.000000734

## 2-8-2. Details of the determination of $pK_a$ value in DMSO via UV-Vis spectrophotometric titration



$$pK_a(\text{HA}) = \log_{10}K_{\text{eq}} + pK_a(\text{HIn}) \quad (\text{eq 4})$$

$$K_{\text{eq}} = \frac{[\text{HA}]}{[\text{A}^-]} \times \frac{[\text{In}^-]}{[\text{HIn}]} = \frac{[\text{HA}]}{[\text{A}^-]} \times \frac{(\text{Abs} - \text{Abs}_{\text{min}})}{(\text{Abs}_{\text{max}} - \text{Abs})} \quad (\text{eq 5})$$

The overlapping indicator method involves the experimental determination of the equilibrium constant  $K_{\text{eq}}$  for the equilibrium in eq. 3 using UV/Vis spectrophotometry, in which HA and  $\text{A}^-$  are the Brønsted acid and corresponding conjugate base, and HIn and  $\text{In}^-$  are the indicator and indicator anion. The  $pK_a$  values of the Brønsted acids could be calculated from the  $K_{\text{eq}}$  values and known indicator  $pK_a$  values by the application of eq. 4. In this study, 2,4-dinitronaphthol (DNN) was used as an indicator:  $pK_a(\text{HIn}) = 2.11$ .

In a typical experiment, UV/Vis spectra of the indicator ( $1.5 \times 10^{-5}$  M) in DMSO buffered with Brønsted acid and conjugate base were recorded at 25°C. The conjugate base of the Brønsted acid was generated in situ by adding and varying amounts of a solution of sodium hydroxide ( $1.0 \times 10^{-5}$  M) in DMSO. The solution of sodium hydroxide in DMSO was prepared by dilution of aqueous sodium hydroxide solution for volumetric analysis into DMSO so that the final concentration of water was  $< 2$  vol %. The  $[\text{InH}]/[\text{In}^-]$  ratio was determined by comparing the observed absorbance at a given buffer ratio with the maximal and minimal absorbance values for the indicator at 454 nm. The values for  $K_{\text{eq}}$  were determined using eq. 5. For strong Brønsted acids, the natural ionization of HA was significant. O'Donoghue and Berkessel et al. carried out the correction of values of  $[\text{HA}]$  and  $[\text{A}^-]$ .<sup>30</sup> However, in this study, the correction was not carried out, and the  $pK_a$  value of the acid was roughly determined. The standard deviation of  $pK_a$  values ( $^{\circ}pK_a$ ), given in parentheses, was calculated using eq. 6.

$${}^s pK_a = \sqrt{S_a^2 + S_b^2} \quad (\text{eq 6})$$

$S_a$ : standard deviation quoted for  $pK_a$  (HIn) by Bordwell<sup>5</sup>

$$S_a = 0.04 \quad \text{for DNN}$$

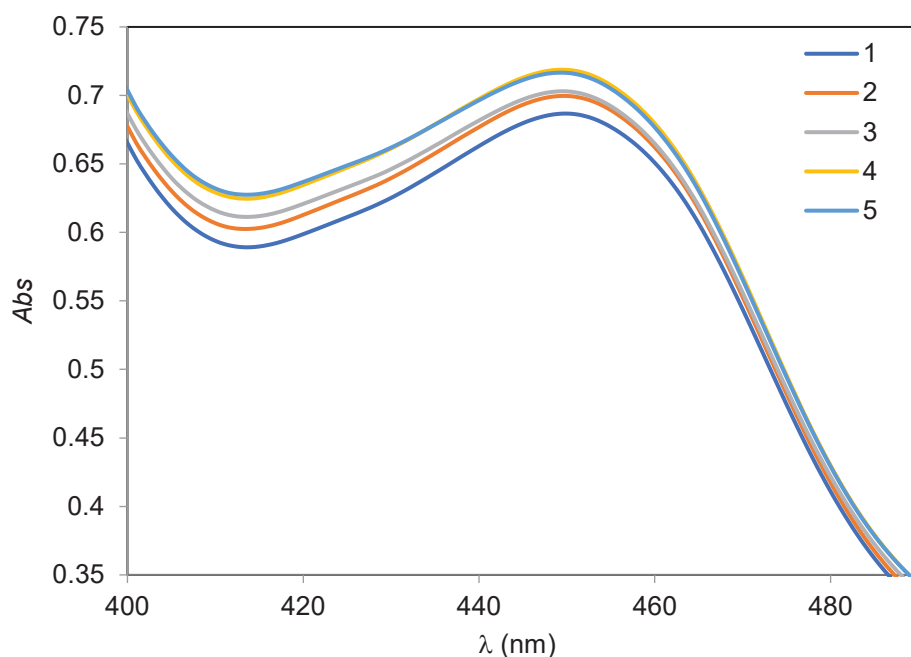
$$S_b = \frac{0.434 \times {}^s K_{eq}}{K_{eq}^{av}}$$

$K_{eq}^{av}$ : mean values of  $K_{eq}$

${}^s K_{eq}$ : standard deviation of  $K_{eq}$

$${}^s K_{eq} = \sqrt{\frac{\sum (K_{eq} - K_{eq}^{av})^2}{n - 1}}$$

## DSATf 1a



**Figure S1.** UV/Vis spectra of 2,4-dinitronaphthol in DMSO containing **DSATf 1a** and conjugate base: spectra 1–5 correspond to the conditions given in Table S1.

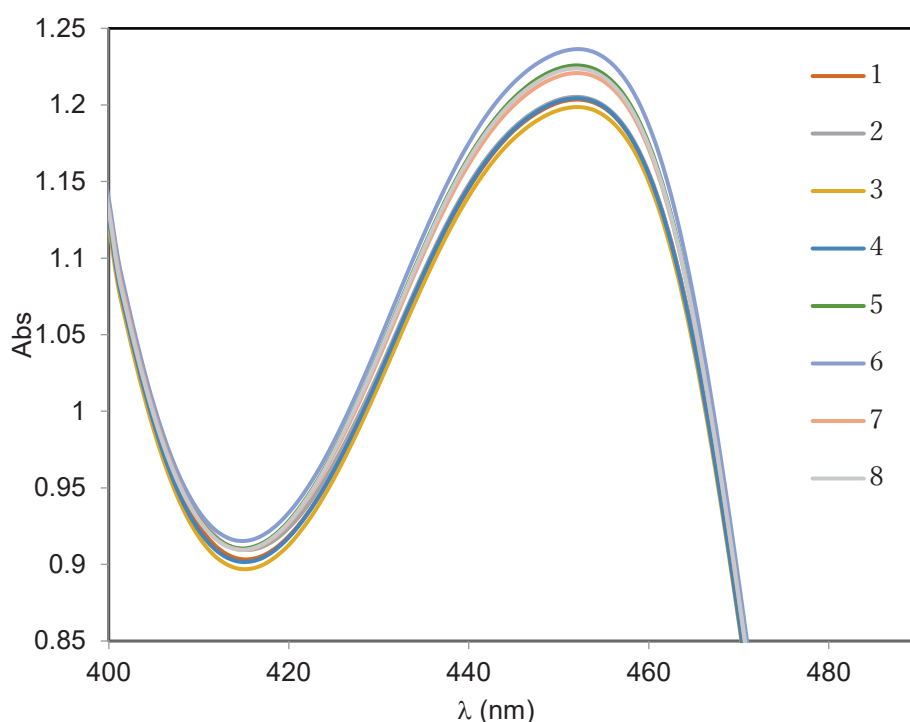
**Table S1.** Absorbance data for 2,4-dinitronaphthol (DNN)<sup>a</sup> in DMSO containing acid (**1a**) and anion<sup>b</sup> (**1a**<sup>-</sup>) at 454 nm and 25°C.

Spectrum	[ <b>1a</b> ] (M)	[ <b>1a</b> <sup>-</sup> ] <sup>b</sup> (M)	<i>Abs</i>	[In <sup>-</sup> ]/[HIn] <sup>c</sup>	<i>K</i> <sub>eq</sub> <sup>d</sup>
1	0.28 × 10 <sup>-3</sup>	0.20 × 10 <sup>-3</sup>	0.680	3.33	4.71
2	0.23 × 10 <sup>-3</sup>	0.25 × 10 <sup>-3</sup>	0.693	6.25	1.10
3	0.18 × 10 <sup>-3</sup>	0.30 × 10 <sup>-3</sup>	0.696	7.63	4.66
4	0.13 × 10 <sup>-3</sup>	0.35 × 10 <sup>-3</sup>	0.705	21.8	8.32
5	0.08 × 10 <sup>-3</sup>	0.40 × 10 <sup>-3</sup>	0.709	52.4	10.9

<sup>a</sup> The concentration of 2,4-dinitronaphthol was  $1.5 \times 10^{-5}$  M. <sup>b</sup> Anion **1a**<sup>-</sup> was generated by addition of NaOH in water solution in DMSO such that the content of water was < 2 vol %. <sup>c</sup> The ratio of phenolate to phenol was obtained using the relationship  $[(Abs - 0.578)/(0.711 - Abs)]$ . <sup>d</sup>  $K_{eq} = ([\mathbf{1a}]/[\mathbf{1a}^-]) \times ((Abs - Abs_{min})/(Abs_{max} - Abs))$ .

The  $pK_a$  value was calculated using equation eq 4:  $pK_a = 2.93 \pm 0.17$ .

## DSATf 1c



**Figure S2.** UV/Vis spectra of 2,4-dinitronaphthol in DMSO containing **DSATf 1c** and conjugate base: spectra 1–8 correspond to the conditions given in Table S2.

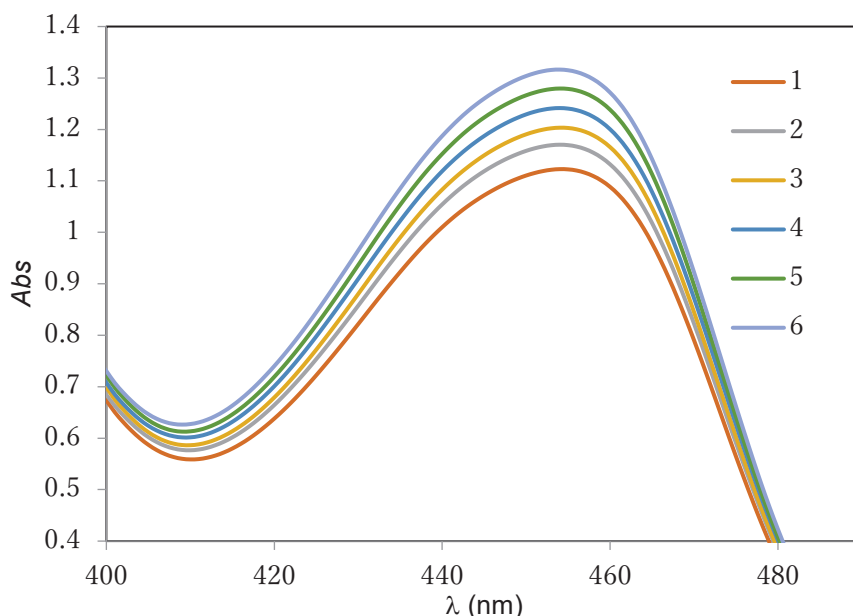
**Table S2.** Absorbance data for 2,4-dinitronaphthol (DNN)<sup>a</sup> in DMSO containing acid (**1c**) and anion<sup>b</sup> (**1c**<sup>-</sup>) at 454 nm and 25°C.

Spectrum	[ <b>1c</b> ] (M)	[ <b>1c</b> <sup>-</sup> ] <sup>b</sup> (M)	<i>Abs</i>	[In <sup>-</sup> ]/[HIn] <sup>c</sup>	<i>K</i> <sub>eq</sub> <sup>d</sup>
1	2.8 x 10 <sup>-3</sup>	6.0 x 10 <sup>-4</sup>	1.203	0.42	1.97
2	2.5 x 10 <sup>-3</sup>	9.0 x 10 <sup>-4</sup>	1.205	0.46	1.29
3	2.2 x 10 <sup>-3</sup>	1.2 x 10 <sup>-3</sup>	1.199	0.31	0.57
4	1.9 x 10 <sup>-3</sup>	1.5 x 10 <sup>-3</sup>	1.204	0.44	0.55
5	1.6 x 10 <sup>-3</sup>	1.8 x 10 <sup>-3</sup>	1.221	1.02	0.90
6	1.3 x 10 <sup>-3</sup>	2.1 x 10 <sup>-3</sup>	1.224	1.18	0.73
7	1.0 x 10 <sup>-3</sup>	2.4 x 10 <sup>-3</sup>	1.226	1.29	0.54
8	7.0 x 10 <sup>-4</sup>	2.7 x 10 <sup>-3</sup>	1.237	2.25	0.58

<sup>a</sup> The concentration of 2,4-dinitronaphthol was  $1.5 \times 10^{-5}$  M. <sup>b</sup> Anion **1c**<sup>-</sup> was generated by addition of NaOH in water solution in DMSO such that the content of water was < 2 vol %. <sup>c</sup> The ratio of phenolate to phenol was obtained using the relationship  $[(Abs - 1.262)/(1.179 - Abs)]$ . <sup>d</sup>  $K_{eq} = ([1c]/[1c^-]) \times ((Abs - Abs_{min})/(Abs_{max} - Abs))$ .

The  $pK_a$  value was calculated using equation eq 4:  $pK_a = 2.02 \pm 0.25$

### Monosulfonamide **6**



**Figure S3.** UV/Vis spectra of 2,4-dinitronaphthol in DMSO containing **6** and conjugate base: spectra 1–8 correspond to the conditions given in Table S3.

**Table S3.** Absorbance data for 2,4-dinitronaphthol (DNN)<sup>a</sup> in DMSO containing acid (**6**) and anion<sup>b</sup> (**6**<sup>-</sup>) at 454 nm and 25°C.

Spectrum	[ <b>6</b> ] (M)	[ <b>6</b> <sup>-</sup> ] <sup>b</sup> (M)	<i>Abs</i>	[In <sup>-</sup> ]/[HIn] <sup>c</sup>	$K_{eq}$ <sup>d</sup>
1	$9.2 \times 10^{-3}$	$1.0 \times 10^{-4}$	1.123	1.44	13.2
2	$8.7 \times 10^{-3}$	$1.5 \times 10^{-4}$	1.170	1.93	11.1
3	$8.2 \times 10^{-3}$	$2.0 \times 10^{-3}$	1.203	2.40	9.78
4	$7.7 \times 10^{-3}$	$2.5 \times 10^{-3}$	1.242	3.18	9.76
5	$7.2 \times 10^{-3}$	$3.0 \times 10^{-3}$	1.280	4.41	10.5
6	$6.7 \times 10^{-4}$	$3.5 \times 10^{-3}$	1.316	6.57	12.5

<sup>a</sup> The concentration of 2,4-dinitronaphthol was  $1.5 \times 10^{-5}$  M. <sup>b</sup> Anion **6**<sup>-</sup> was generated by addition of NaOH in water solution in DMSO such that the content of water was < 2 vol %. <sup>c</sup> The ratio of phenolate to phenol was obtained using the relationship  $[(Abs - 1.408)/(0.712 - Abs)]$ . <sup>d</sup>  $K_{eq} = ([\mathbf{6}]/[\mathbf{6}^-]) \times ((Abs - Abs_{min})/(Abs_{max} - Abs))$ .

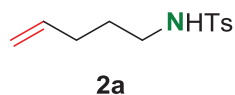
The  $pK_a$  value was calculated using equation eq 4:  $pK_a = 3.16 \pm 0.07$

### 2-8-3. General Information.

Internal references for  $^1\text{H}$  NMR spectra were 0.0 ppm ( $\text{Me}_4\text{Si}$ ) for  $\text{CDCl}_3$  and  $\text{CD}_3\text{OD}$  (3.31 ppm). Chemical shifts for  $^{13}\text{C}$  NMR spectra were referenced to  $\text{CDCl}_3$  (77.0 ppm) and  $\text{CD}_3\text{OD}$  (49.0 ppm). The chemical shift for  $^{19}\text{F}$  NMR spectra was reported on the basis of  $\text{CF}_3\text{CO}_2\text{H}$  (-76.0 ppm) as an external standard. High resolution mass spectral (HRMS) data were recorded with an LTQ Orbitrap trap mass spectrometer using electrospray ionization (ESI) method. Optical rotations were measured on a digital polarimeter with a 0.1 dm cell at room temperature. All reactions involving air- and moisture-sensitive reagents were carried out under  $\text{N}_2$ . All reactions were monitored by analytical thin-layer chromatography (TLC), which was visualized by ultraviolet light (254 nm).

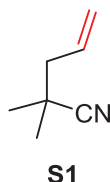
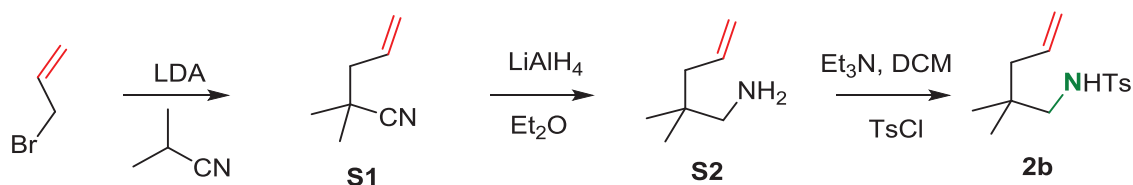
### 2-8-4. Preparation of alkenyl amines and alcohols

#### Preparation of 2a

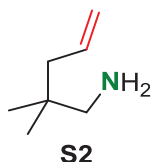


A mixture of 5-bromo-1-pentene (0.80m, 6.75mmol), *p*-toluenesulfonamide (1.20 g, 7.01 mmol) and potassium carbonate (1.84 g, 13.3 mmol) was dissolved in acetone (7.0 ml) and stirred for 1 day at 40°C. The reaction mixture was cooled and passed through a pad of celite. The filtrate was quenched with saturated  $\text{NH}_4\text{Cl}$ , and the mixture was extracted with EtOAc. The organic phase was washed with water and brine, dried over  $\text{Na}_2\text{SO}_4$ , and filtered. The filtrate was concentrated under vacuum. The crude product was purified by column chromatography on silica gel (hexane: EtOAc= 5: 1) to give 4-methyl-*N*-(pent-4-en-1-yl) benzenesulfonamide **2a**<sup>71</sup> (1.20 g, 75%) as a yellow oil.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.94–7.53 (m, 2H), 7.31 (d,  $J$  = 7.9 Hz, 2H), 5.70 (ddt,  $J$  = 17.0, 10.2, 6.7 Hz, 1H), 5.08–4.84 (m, 2H), 4.59 (s, 1H), 2.95 (dd,  $J$  = 13.5, 6.9 Hz, 2H), 2.43 (s, 3H), 2.04 (td,  $J$  = 7.0, 1.2 Hz, 2H), 1.62–1.45 (m, 2H).

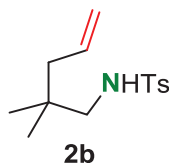
## Preparation of 2b



A mixture of diisopylamine (3.4 ml, 24.0 mmol) in THF (32 mL) was added n-BuLi (13 ml, 16.7 mmol) at 0°C and stirred at 0°C for 30 min and then added isobutyronitrile (1.5 ml, 16.7 mmol) to this mixture at -78°C and stirred for 1 h. Next, allyl bromide (2.5 ml, 28.9 mmol) was also added to this mixture at -78°C and stirred at room temperature overnight. The reaction mixture was added to saturated NH<sub>4</sub>Cl and extracted by diethyl ether. The organic layer was washed by water, brine, and dried over Na<sub>2</sub>SO<sub>4</sub>. The reaction mixture was evaporated to obtain the crude **S1** (2.55 g). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 5.87 (m, 1 H), 5.22 (dt, *J* = 1.0, 10 Hz, 1 H), 5.18 (dt, *J* = 1.0, 17.0 Hz, 1 H), 2.28 (dd, *J* = 1.0, 7.4 Hz, 2 H), 1.34 (s, 6 H).



**S1** (crude, 2.55 g) was dissolved in dry diethyl ether (20 ml), and then added lithium aluminum hydride (1.86 g, 49.1 mmol) at 0°C and refluxed for 2 h. The reaction mixture was added to water and filtered. The filtrate was extracted by diethyl ether. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>. The reaction mixture was evaporated to obtain **S2** (1.62 g, 0.014 mmol, 61% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 5.77-5.86 (m, 1 H), 5.00-5.04 (m, 2 H), 2.45 (s, 2 H), 1.96 (d, *J* = 7.6 Hz, 2 H), 0.88 (s, 6 H).

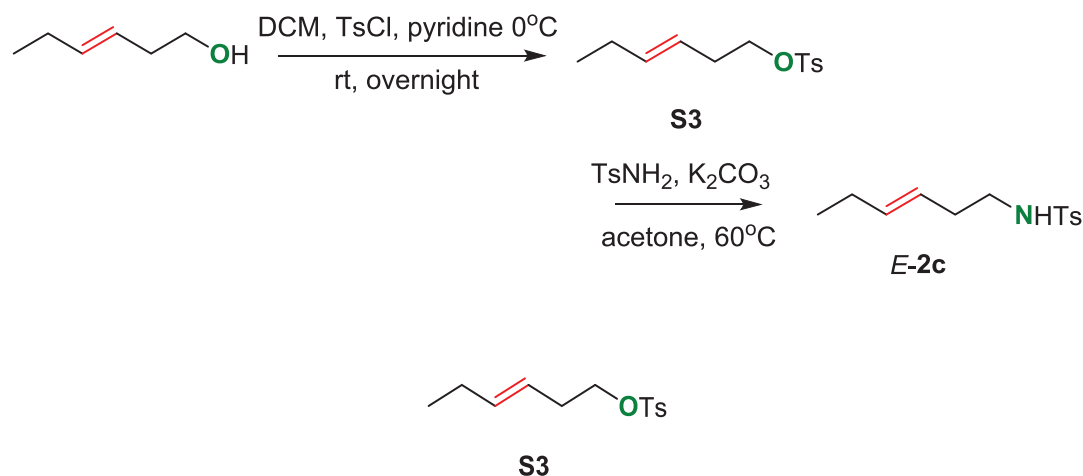


**S2** (1.62 g, 14.3 mmol) and triethylamine (4.0 ml, 28.7 mmol) were dissolved in CH<sub>2</sub>Cl<sub>2</sub> (30 ml). To the reaction mixture was added TsCl (2.74 g, 14.4 mmol) at 0°C, and stirred at room

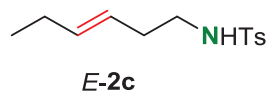


temperature for 7 h. The reaction mixture was washed by saturated NaHCO<sub>3</sub>, saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. The organic layer was concentrated and purified by column chromatography on silica gel (hexane: EtOAc= 5: 1) to give **2b**<sup>72</sup> (2.72 g, 10.2 mmol, 60%) as yellow liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.73 (d, *J* = 8.4 Hz, 2 H), 7.30 (d, *J* = 8.2 Hz, 2 H), 5.68-5.77 (m, 1 H), 4.98-5.04 (m, 2 H), 4.45 (t, *J* = 6.8 Hz, 1 H), 2.68 (d, *J* = 7.1 Hz, 2 H), 2.43 (s, 3 H), 1.96 (d, *J* = 7.6 Hz, 2 H), 0.86 (s, 6 H).

### Preparation of *E*-2c



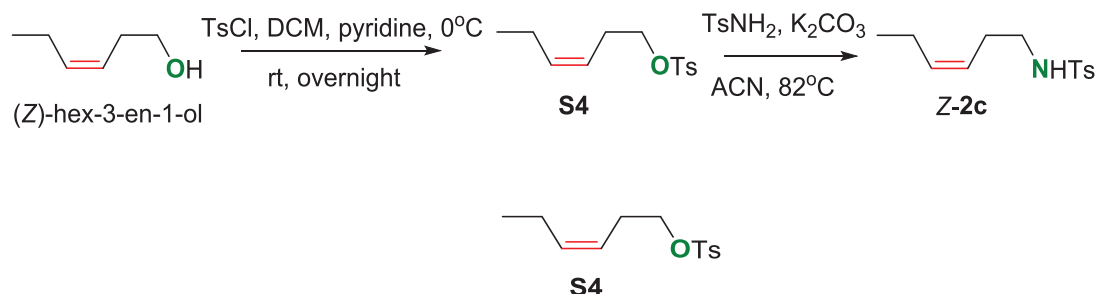
*p*-toluenesulfonyl chloride (4.51 g, 0.024 mmol) and pyridine (2.6 mL) were added to a solution of (*E*)-hex-3-en-1-ol (1.57 g, 0.016 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (26 mL) at 0°C. The reaction was warmed up to room temperature and stirred overnight. The reaction was diluted with ether and washed with water, 1 M HCl, saturated NaHCO<sub>3</sub>, and brine. The organic extracts were combined, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under vacuum to afford a yellow oil. The crude product was purified by column chromatography (Hexane: AcOEt= 5: 1) to give the desired product (3.39 g, 85%) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.82–7.72 (m, 2H), 7.34 (d, *J* = 8.0 Hz, 2H), 5.57 – 5.43 (m, 1H), 5.23 (dtt, *J* = 15.2, 6.8, 1.5 Hz, 1H), 4.01 (t, *J* = 6.9 Hz, 2H), 2.45 (s, 3H), 2.33 (qd, *J* = 6.8, 1.1 Hz, 2H), 1.96 (qdd, *J* = 7.5, 6.4, 1.3 Hz, 2H), 0.93 (t, *J* = 7.5 Hz, 3H).



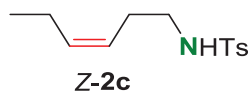
To a solution of *p*-toluenesulfonamide (4.57 g, 0.027 mmol) in acetone (14 mL), (*E*)-hex-3-en-1-yl 4-methylbenzenesulfonate (3.4 g, 0.013 mmol) and K<sub>2</sub>CO<sub>3</sub> (3.7 g, 0.027 mmol) were added. The reaction mixture was refluxed at 60°C for 2 days. The reaction mixture was quenched with saturated NH<sub>4</sub>Cl and the mixture was extracted with Et<sub>2</sub>O. The organic phase was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and filtered. The filtrate was concentrated under vacuum. The crude

residue was purified by column chromatography (Hexane: AcOEt= 2: 1) to afford the desired product *E*-**2c**<sup>73</sup> (2.28 g, 67%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.78–7.70 (m, 2H), 7.31 (d, *J* = 8.0 Hz, 2H), 5.55–5.42 (m, 1H), 5.25–5.06 (m, 1H), 4.37 (s, 1H), 2.97 (q, *J* = 6.5 Hz, 2H), 2.43 (s, 3H), 2.13 (qd, *J* = 6.7, 1.0 Hz, 2H), 2.08–1.89 (m, 2H), 0.94 (t, *J* = 7.5 Hz, 3H).

### Preparation fo *Z*-**2c**

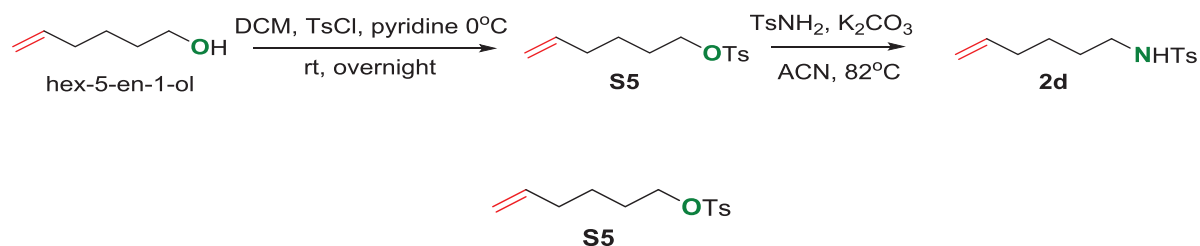


*p*-toluenesulfonyl chloride (4.47 g, 0.023 mol) and pyridine (2.5 mL) were added to a solution of *(Z)*-hex-3-en-1-ol (1.55 g, 0.015 mol) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) at 0°C. The reaction was warmed up to room temperature and stirred overnight. The reaction was diluted with ether and washed with water, 1 M HCl, saturated NaHCO<sub>3</sub>, and brine. The organic extracts were combined and dried over NaSO<sub>4</sub>, filtered, and concentrated under vacuum to afford a yellow oil. The crude product was purified by column chromatography (Hexane: AcOEt= 5: 1) to give the desired product (3.4 g, 86%) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.79 (d, *J* = 8.3 Hz, 2H), 7.46–7.29 (m, 2H), 5.63–5.35 (m, 1H), 5.29–5.04 (m, 1H), 4.00 (t, *J* = 7.0 Hz, 2H), 2.66–2.23 (m, 5H), 2.13–1.84 (m, 2H), 0.93 (t, *J* = 7.5 Hz, 3H).

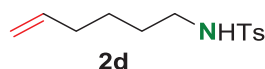


To a solution of *p*-toluenesulfonamide (3.56 g, 0.02 mol) in acetonitrile (14 mL), *(E)*-hex-3-en-1-yl 4-methylbenzenesulfonate (3.4 g, 0.013 mol) and K<sub>2</sub>CO<sub>3</sub> (3.59 mg, 0.026 mmol) were added. The reaction mixture was refluxed at 82°C for 20 h. The reaction mixture was quenched with saturated NH<sub>4</sub>Cl and extracted with Et<sub>2</sub>O. The organic phase was washed with brine, dried over NaSO<sub>4</sub>, and filtered. The filtrate was concentrated under vacuum. The crude residue was purified by column chromatography (Hexane: AcOEt= 2: 1) to afford the desired product *Z*-**2c**<sup>73</sup> (2.7 g, 80%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.79–7.70 (m, 2H), 7.31 (d, *J* = 8.0 Hz, 2H), 5.47 (dt, *J* = 10.7, 7.3 Hz, 1H), 5.21–5.06 (m, 1H), 4.82 (s, 1H), 2.95 (q, *J* = 6.8 Hz, 2H), 2.42 (s, 3H), 2.20 (q, *J* = 7.0 Hz, 2H), 2.03–1.86 (m, 2H), 0.92 (t, *J* = 7.5 Hz, 3H).

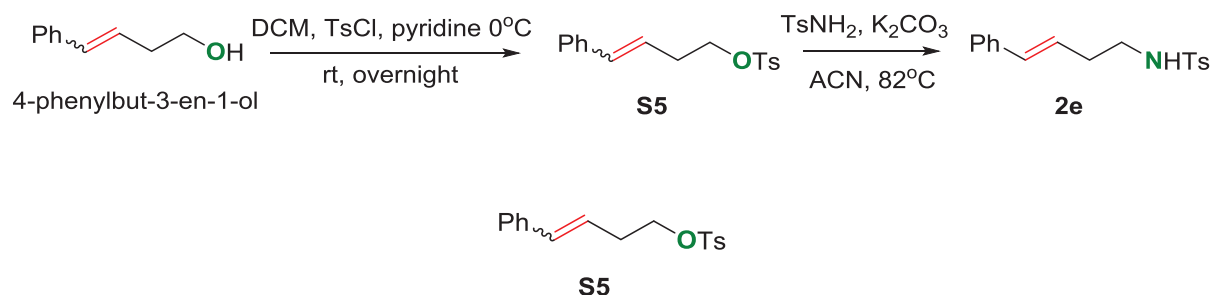
## Preparation fo 2d



*p*-toluenesulfonyl chloride (4.42 g, 0.023 mol) and pyridine (2.5 mL) were added to a solution of hex-5-en-1-ol (1.52 g, 0.015 mol) in  $\text{CH}_2\text{Cl}_2$  (2.5 mL) at  $0^\circ\text{C}$ . The reaction was warmed up to room temperature and stirred overnight. The reaction was diluted with ether and washed with water, 1 M HCl, saturated  $\text{NaHCO}_3$ , and brine. The organic extracts were combined and dried, filtered, and concentrated under vacuum to afford a yellow oil. The crude product was purified by column chromatography (Hexane: AcOEt= 5: 1) to give the desired product (3.2 g, 84%) as a colorless oil.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.85–7.74 (m, 2H), 7.40–7.30 (m, 2H), 5.72 (ddt,  $J$  = 17.0, 10.2, 6.7 Hz, 1H), 5.03–4.86 (m, 2H), 4.03 (t,  $J$  = 6.4 Hz, 2H), 2.45 (s, 3H), 2.09–1.91 (m, 2H), 1.72–1.58 (m, 2H), 1.49–1.30 (m, 2H).

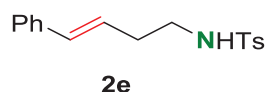


To a solution of *p*-toluenesulfonamide (3.28 g, 0.019 mol) in acetonitrile (13 mL), (*E*)-hex-3-en-1-yl 4-methylbenzenesulfonate (3.24 g, 0.013 mol) and  $\text{K}_2\text{CO}_3$  (3.57 g, 0.026 mol) were added. The reaction mixture was refluxed at  $82^\circ\text{C}$  for 12 h. The reaction mixture was quenched with saturated  $\text{NH}_4\text{Cl}$  and the mixture was extracted with  $\text{Et}_2\text{O}$ . The organic phase was washed with brine, dried over  $\text{NaSO}_4$ , and filtered. The filtrate was concentrated under vacuum. The crude residue was purified by column chromatography (Hexane: AcOEt= 20: 1) to afford the desired product **2d**<sup>71</sup> (2.9 g, 89%).  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.80–7.71 (m, 2H), 7.31 (d,  $J$  = 8.1 Hz, 2H), 5.71 (ddt,  $J$  = 16.9, 10.2, 6.7 Hz, 1H), 4.94 (ddt,  $J$  = 6.0, 2.0, 1.3 Hz, 2H), 4.53 (s, 1H), 2.93 (dd,  $J$  = 13.4, 6.7 Hz, 2H), 2.43 (s, 3H), 2.12–1.86 (m, 2H), 1.57–1.23 (m, 4H).



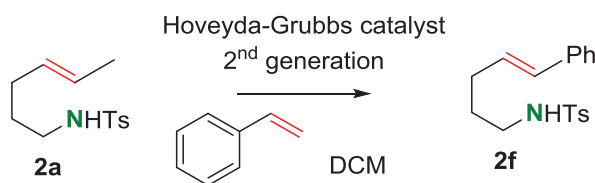
*p*-toluenesulfonyl chloride (0.29 g, 0.0015 mol) and pyridine (0.17 mL, 0.0021 mol) were added to a solution of 4-phenylbut-3-en-1-ol (1.55 g, 0.001 mol) in  $\text{CH}_2\text{Cl}_2$  (1.7 mL) at  $0^\circ\text{C}$ . The

reaction was warmed up to room temperature and stirred overnight. The reaction was diluted with ether and washed with water, 1 M HCl, saturated NaHCO<sub>3</sub>, and brine. The organic extracts were combined and dried, filtered, and concentrated under vacuum to afford a yellow oil. The crude product was purified by column chromatography (Hexane: AcOEt= 10: 1) to give the mixture of inseparable *Z* and *E* isomers (0.23 g, 76%) as a colorless oil **S5** (*E*: *Z*= 5.3: 1) 76% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.84–7.74 (m, 2H), 7.35–7.24 (m, 7H), 6.39 (d, *J* = 15.9 Hz, 1H), 5.99 (dt, *J* = 15.8, 7.0 Hz, 1H), 4.13 (dd, *J* = 10.8, 4.2 Hz, 2H), 2.60–2.40 (m, 5H).



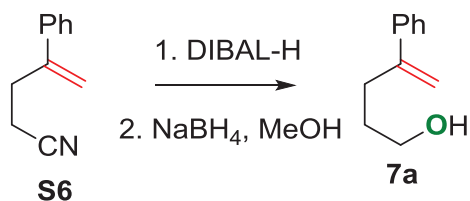
To a solution of *p*-toluenesulfonamide (3.28 g, 0.019 mol) in acetonitrile (0.75 mL), hex-3-en-1-yl 4-methylbenzenesulfonate (230 mg, 0.76 mol) and K<sub>2</sub>CO<sub>3</sub> (210 mg, 1.51 mol) were added. The reaction mixture was refluxed at 82°C for 12 h. The reaction mixture was quenched with saturated NH<sub>4</sub>Cl and extracted with Et<sub>2</sub>O. The organic phase was washed with brine, dried over NaSO<sub>4</sub>, and filtered. The filtrate was concentrated under vacuum. The crude residue was purified by column chromatography (Hexane: AcOEt= 5: 1) to afford the mixture of inseparable *Z* and *E* isomers **2e**<sup>74</sup> (*E*: *Z*= 5.3: 1) (0.15 g, 75%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.72 (dd, *J* = 20.9, 8.3 Hz, 2H), 7.34–7.25 (m, 7H), 6.36 (d, *J* = 15.9 Hz, 1H), 5.97 (dt, *J* = 15.9, 7.1 Hz, 1H), 3.11 (dd, *J* = 12.9, 6.5 Hz, 2H), 2.44–2.31 (m, 5H).

### Preparation of **2f**



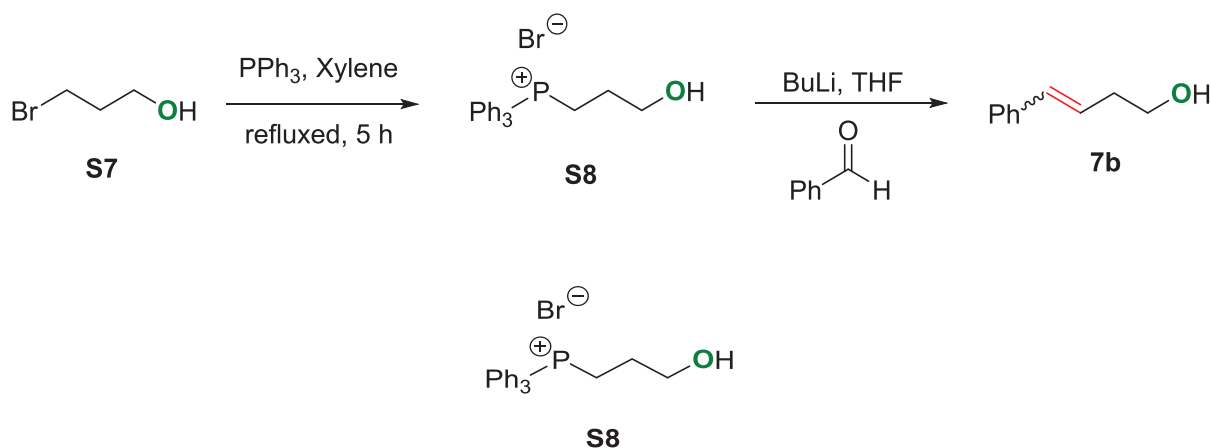
A mixture of **2a** (51.0 mg, 0.23 mmol), Hoveyda-Grubbs catalyst 2<sup>nd</sup> generation (20.0 mg, 24.3 mmol), CH<sub>2</sub>Cl<sub>2</sub> (4.0 ml) and styrene (90 μl, 0.78 mmol) was refluxed for 15 h. The reaction mixture was evaporated to obtain the crude product. The crude product was purified by column chromatography on silica gel (CH<sub>2</sub>Cl<sub>2</sub>) and GPC machine to give **2f**<sup>75</sup> (36.6 mg, 51%) as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.74 (d, *J* = 8.0 Hz, 2H), 7.31 (d, *J* = 8.1 Hz, 2H), 6.26 (dt, *J* = 17.0, 10.3 Hz, 1H), 5.99 (dd, *J* = 15.2, 10.4 Hz, 1H), 5.67–5.45 (m, 1H), 5.09 (d, *J* = 17.0 Hz, 1H), 4.98 (d, *J* = 10.1 Hz, 1H), 4.40 (s, 1H), 2.95 (dd, *J* = 13.5, 6.7 Hz, 2H), 2.43 (s, 3H), 2.08 (dd, *J* = 14.8, 7.7 Hz, 2H), 1.71–1.40 (m, 2H).

## Preparation of 7a

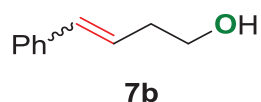


3-Phenyl-3-butenyl-1-cyanide **S6** was prepared according to known procedure<sup>76</sup>. Diisobutylaluminum hydride (DIBAL-H) (1.0 M in hexane, 2.4 ml, 2.4 mmol) was added dropwise to **S6** (240 mg, 1.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (4.4 ml) at -78°C. After the reaction mixture was stirred at the temperature for 2 h, the mixture was quenched by saturated NH<sub>4</sub>Cl at -78°C. The reaction mixture was extracted with EtOAc, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. NaBH<sub>4</sub> (66.0 mg, 1.7 mmol) was added slowly to the obtained crude product in MeOH (4.4 ml) at r.t., and the reaction mixture was stirred overnight. The mixture was quenched by saturated NH<sub>4</sub>Cl, extracted with EtOAc, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The crude product was purified by column chromatography (SiO<sub>2</sub>, EtOAc: hexane= 15: 1) to give **7a**<sup>77</sup> (50 mg, 20 % for 2 steps) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.44–7.26 (m, 5H), 5.30 (d, *J* = 1.4 Hz, 1H), 5.10 (d, *J* = 1.4 Hz, 1H), 3.83–3.55 (m, 2H), 2.67–2.55 (m, 2H), 1.73 (ddd, *J* = 13.9, 10.3, 6.5 Hz, 2H).

## Preparation of 7b

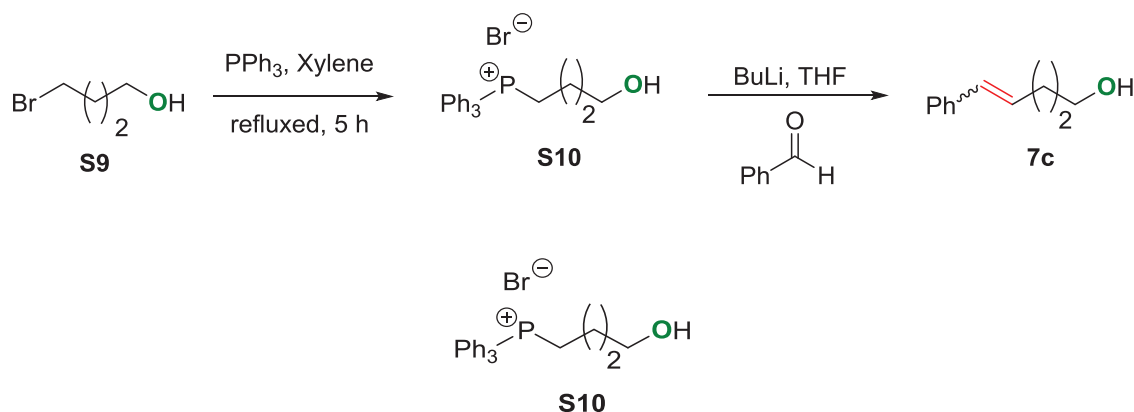


Under nitrogen atmosphere, the solution of triphenylphosphine (6.21 g, 23.67 mmol), 3-bromopropan-1-ol (3.27 g, 23.6 mg) in xylene (21 mL) was stirred under reflux for 5 h. Then the reaction mixture was cooled to room temperature and ether was added. The solids were collected by filtration and evaporated under vacuum to afford the desired product **S8** (8.99 g, 96%) as a white powder which was used directly in the next step without further purification. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.85–7.65 (m, 15H), 3.92–3.71 (m, 4H), 1.90 – 1.79 (m, 2H).

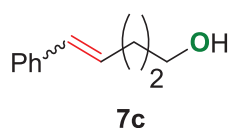


n-BuLi (4.5 mL, 10.35 mmol) was added dropwise to a suspension of (3-propan-1-ol) triphenylphosphonium bromide (1.58 g, 6.71 mmol) in THF (12 mL) at  $-78^{\circ}\text{C}$ . The mixture reaction was stirred at  $-78^{\circ}\text{C}$  for 1 h and then benzaldehyde (0.3 mL, 7 mmol) was added dropwise at  $-78^{\circ}\text{C}$  and stirred at  $-78^{\circ}\text{C}$  for 2 hours. The mixture was warmed up to room temperature and stirred overnight at r.t, and then saturated  $\text{NH}_4\text{Cl}$  solution was added. This reaction mixture was extracted with EtOAc, dried over sodium sulfate, and concentrated. The crude product was purified by column chromatography (Hexene: AcOEt= 5: 1) and GPC machine to give a colorless oil of the mixture of inseparable *Z* and *E* isomers **7b**<sup>78</sup> (220 mg, 34%) (*E*: *Z*= 5.3: 1).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.37–7.21 (m, 5H), 6.59 (d,  $J = 11.7$  Hz, 1H, *cis* isomer), 6.50 (d,  $J = 15.9$  Hz, 1H, *trans* isomer), 6.21 (dt,  $J = 15.8, 7.1$  Hz, 1H, *trans* isomer), 5.69 (dt,  $J = 11.7, 7.4$  Hz, 1H *cis* isomer), 3.85–3.60 (m, 1H *cis* isomer and 2H *trans* isomer), 2.62 (ddd,  $J = 13.9, 6.5, 1.8$  Hz, 1H *cis* isomer), 2.59–2.36 (m, 2H).

### Preparation of 7c



Under nitrogen atmosphere, the solution of triphenylphosphine (3.41 g, 0.013 mol), 3-bromopropan-1-ol (2.08 g, 0.014 mol) in xylene (21 mL) was stirred under reflux for 5 h. Then, the reaction mixture was cooled to room temperature and ether was added. The solids were collected by filtration and dried under vacuum to afford the crude product **S10** (4.5 g, 83%) as a white powder which was used directly in the next step without further purification.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.92–7.37 (m, 15H), 4.14–3.37 (m, 4H), 2.19 (d,  $J = 117.7$  Hz, 4H).

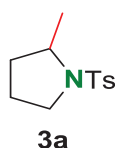


n-BuLi (9 mL, 20.7 mmol) was added dropwise to a suspension of (4-butan-1-ol) triphenylphosphonium bromide (3 g, 7 mmol) in THF at  $-78^{\circ}\text{C}$ . The reaction mixture was stirred

at -78°C for 1 h and benzaldehyde (0.8 mL, 8 mmol) was added dropwise. After 2 h, the mixture was warmed up to room temperature and stirred overnight at r.t, and then saturated NH<sub>4</sub>Cl solution was added. This mixture was extracted with EtOAc, dried over sodium sulfate and concentrated. The crude product was purified by column chromatography (Hexene: AcOEt= 5: 1) and GPC machine to give a colorless oil of the mixture of inseparable *Z* and *E* isomers **7c**<sup>79</sup> (30 mg, 2.6 %) (*E*: *Z* = 22: 1), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.38–7.23 (m, 5H), 6.42 (d, *J* = 15.8 Hz, 1H), 6.23 (dt, *J* = 15.8, 6.9 Hz, 1H), 3.71 (t, *J* = 6.5 Hz, 2H), 2.42–2.19 (m, 2H), 1.82–1.67 (m, 2H).

### 2-8-5. Typical procedure for hydrofunctionalization

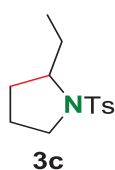
Alkenyl amine 4-methyl-*N*-(pent-4-en-1-yl)benzenesulfonamide **2a** (41.0 mg, 0.17 mmol) was charged in a crew-cap vial equipped with a stir bar. A solution of **DSATf 1c** (37.7 mg, 34 μmol) in HFIP (0.17 ml) was added and the tube was sealed. The reaction mixture was stirred at 60°C and the progress of the reaction was monitored by thin layer chromatography. After the starting material was consumed, the reaction was quenched with solid NaHCO<sub>3</sub> and filtered. The filtrate was evaporated and the crude product was purified by column chromatography (SiO<sub>2</sub>, EtOAc: hexane= 15: 1) to give the hydrofunctionalization product **3a** (30 mg, 73%).



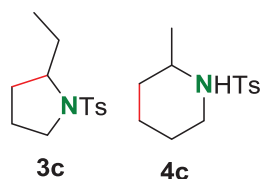
**3a**<sup>8</sup>: 73% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.72 (d, *J* = 8.2 Hz, 2H), 7.31 (d, *J* = 8.0 Hz, 2H), 3.80–3.59 (m, 1H), 3.51–3.35 (m, 1H), 3.15 (dt, *J* = 10.1, 7.2 Hz, 1H), 2.43 (s, 3H), 1.89–1.75 (m, 1H), 1.75–1.63 (m, 1H), 1.58–1.42 (m, 2H), 1.31 (d, *J* = 6.4 Hz, 3H).



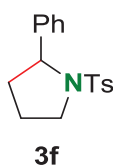
**3b**<sup>80</sup>: 92% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.75–7.69 (m, 2H), 7.30 (d, *J* = 7.9 Hz, 2H), 3.65 (ddq, *J* = 12.4, 8.5, 6.2 Hz, 1H), 3.17 (d, *J* = 10.4 Hz, 1H), 3.07 (dd, *J* = 10.3, 1.2 Hz, 1H), 2.43 (s, 3H), 1.73 (ddd, *J* = 12.5, 7.2, 1.2 Hz, 1H), 1.57 (s, 1H), 1.41 (d, *J* = 6.1 Hz, 3H), 1.04 (s, 3H), 0.55 (s, 3H).



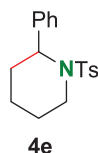
**3c**<sup>81</sup>: 91% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.76–7.69 (m, 2H), 7.31 (d, *J* = 7.9 Hz, 2H), 3.54 (dt, *J* = 10.4, 5.8 Hz, 1H), 3.38 (ddd, *J* = 10.4, 7.0, 5.1 Hz, 1H), 3.19 (dt, *J* = 10.4, 7.2 Hz, 1H), 2.43 (s, 3H), 1.93–1.68 (m, 2H), 1.55–1.43 (m, 4H), 0.91 (t, *J* = 7.5 Hz, 3H).



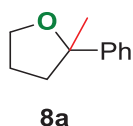
A mixture of **3c** and **4c**<sup>82</sup>: **3c**: 83% yield, **4c**: 2% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.72 (d, *J* = 8.2 Hz, 2H), 7.31 (d, *J* = 8.3 Hz, 2H), 4.23 (s, **4c**-1H), 3.70 (d, *J* = 13.3 Hz, **4c**-1H), 3.54 (td, *J* = 10.2, 5.7 Hz, 1H), 3.38 (ddd, *J* = 10.5, 7.0, 5.2 Hz, 1H), 3.19 (dt, *J* = 10.4, 7.2 Hz, 1H), 3.03–2.92 (m, **4c**-1H), 2.43 (s, 3H), 1.94–1.69 (m, 2H), 1.67–1.46 (m, 4H), 1.07 (d, *J* = 6.9 Hz, **4c**-3H), 0.91 (t, *J* = 7.5 Hz, 3H).



**3f**<sup>83</sup>: 85% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.67 (d, *J* = 8.3 Hz, 2H), 7.40–7.20 (m, 7H), 4.79 (dd, *J* = 7.9, 3.6 Hz, 1H), 3.62 (ddd, *J* = 9.8, 7.1, 4.9 Hz, 1H), 3.43 (dt, *J* = 10.2, 7.2 Hz, 1H), 2.43 (s, 3H), 2.10–1.91 (m, 1H), 1.93–1.74 (m, 2H), 1.73–1.57 (m, 1H).

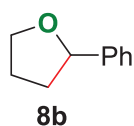


**4e**<sup>8</sup>: 61% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.78 (d, *J* = 8.3 Hz, 2H), 7.40–7.24 (m, 7H), 5.29 (d, *J* = 3.9 Hz, 1H), 3.86 (d, *J* = 14.3 Hz, 1H), 3.13–2.94 (m, 1H), 2.46 (s, 3H), 2.24 (d, *J* = 13.0 Hz, 1H), 1.77–1.24 (m, 5H).

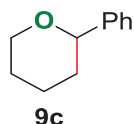


**8a**<sup>84</sup>: 78 % yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.43–7.18 (m, 5H), 4.07–3.96 (m, 1H), 3.96–3.84 (m, 1H), 2.26–2.13 (m, 1H), 1.98 (tdt, *J* = 10.7, 7.9, 6.7 Hz, 2H), 1.87–1.72 (m, 1H), 1.52 (s, 3H).





**8b**<sup>85</sup>: 47% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.43 – 7.22 (m, 5H), 4.92 (t, *J* = 7.2 Hz, 1H), 4.16 (ddd, *J* = 14.9, 14.1, 6.7 Hz, 1H), 3.97 (td, *J* = 7.8, 6.6 Hz, 1H), 2.45–2.25 (m, 1H), 2.12–1.93 (m, 2H), 1.84 (ddd, *J* = 16.0, 12.1, 7.6 Hz, 1H).

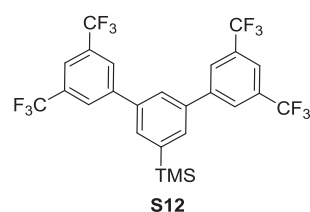
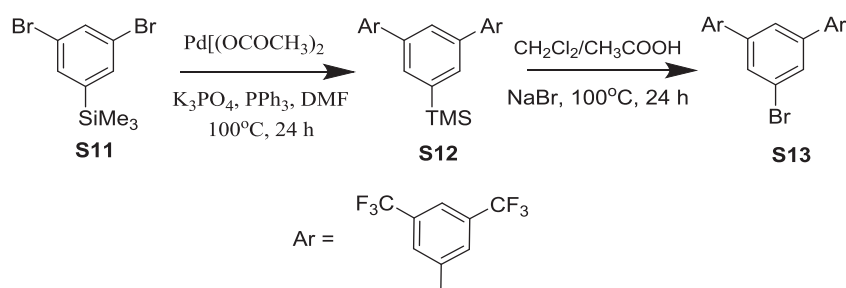


**9c**<sup>86</sup>: 62% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.47–7.16 (m, 5H), 4.40–4.24 (m, 1H), 4.23–4.04 (m, 1H), 3.61 (ddd, *J* = 11.5, 5.3, 2.5 Hz, 1H), 2.03–1.59 (m, 5H).

## 2-8-6. The typical procedure for the synthesis of the catalyst DSATf

According to the literature procedure, the corresponding sulfonyl chloride was prepared by *N*-chlorosuccinimide (NCS) and the corresponding thiocarbamoyl derivative (for **1a**)<sup>56</sup> or thionyl chloride and the corresponding sulfuric acid derivative (for **1b-e**)<sup>36</sup>.

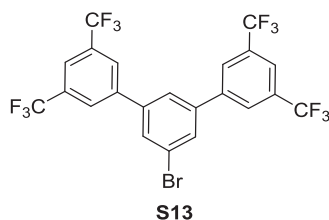
### 2-8-6-1. Preparation of Ar substituent



### 3,5-bis (3,5 - di (3,5 - trifluoromethyl phenyl) -1-trimethyl silyl benzene<sup>87</sup>

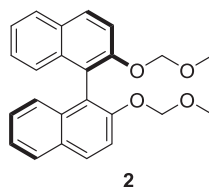
3,5-dibromo trisilyl benzene (3.1151 g, 10.1 mmol), PPh<sub>3</sub> (516.7 mg, 1.97 mmol), K<sub>3</sub>PO<sub>4</sub> (6.8315 g, 32.2 mmol), Ar-OH (6.4712 g, 25.1 mmol), Pd(OCOCH<sub>3</sub>)<sub>2</sub> (197.2 mg, 0.878 mmol) were mixed in DMF. The mixture was stirred at 100°C for 15 h under a nitrogen atmosphere, cooled to room temperature, quenched with saturated NH<sub>4</sub>Cl, and extracted with EtOAc. The organic layer was washed with brine and

dried with Na<sub>2</sub>SO<sub>4</sub>. After the removal of the solvent under reduced pressure, the residue was purified by column chromatography (hexane) to give 3,5-bis (3,5 - di (3,5 - trifluoromethyl phenyl) -1-trimethyl silyl benzene as a white solid (54% yield). R<sub>f</sub> = 0.63 (hexane), <sup>1</sup>H NMR (500MHz, CDCl<sub>3</sub>) δ 8.03 (s, 4H), 7.92 (s, 2H), 7.74 (d, *J* = 1.8 Hz, 2H), 7.69 (t, *J* = 1.8 Hz, 1H), 0.40 (s, 9H)

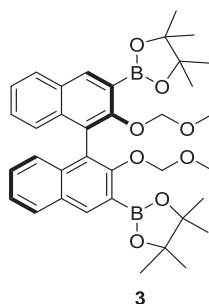


**S12** (999.2 mg, 1.74 mmol) in CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>COOH = 6/1 (4.1 ml) was added NaBr (663.4 mg, 6.45 mmol), *N*-bromosuccinimide (819.1 mg, 4.60 mmol) at room temperature. The mixture was stirred at 100 °C for 25 h, quenched with saturated NaHSO<sub>3</sub> and extracted with Et<sub>2</sub>O. The organic layer was washed with brine and dried with Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent under reduced pressure, the residue was purified by recrystallization (hexane) to give the product (384.3 mg, 38%) as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.02 (s, 4H), 7.94 (s, 2H), 7.81 (d, *J* = 1.6 Hz, 2H), 7.67 (t, *J* = 1.8 Hz, 1H), <sup>13</sup>C NMR (CDCl<sub>3</sub>, 101 MHz) δ 141.4, 141.3, 132.6 (q, *J* = 33.6 Hz), 130.7, 127.5 (q, *J* = 3.5 Hz), 125.0, 124.2, 123.1 (q, *J* = 272.4 Hz), 122.1 (q, *J* = 3.9 Hz), <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz) δ -62.8, HRMS (APCI-) *m/z* calcd for C<sub>22</sub>H<sub>9</sub>F<sub>12</sub>BrCl[M+Cl]<sup>-</sup> 614.93900, found 614.93964, m.p. 98-99°C.

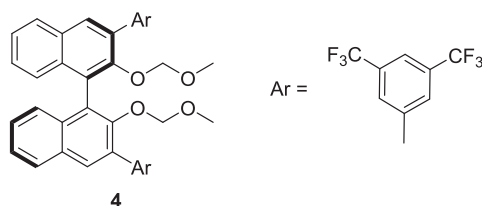
### 2-8-6-2. Preparation of DSATf



(*R*)-Binaphthol (5 g, 17 mmol) in dry THF (70 ml) was added to the mixture of NaH (1.7 g, 68 mmol) was mixed in dry THF (100 ml) at 0°C under an N<sub>2</sub> atmosphere. After the addition, the mixture was warmed up to room temperature for 1 h. After the mixture was cooled to 0°C, chloromethyl methyl ether was slowly added. Then, the mixture was warmed up to room temperature for 3 h. Water was added to the flask, then extracted by EtOAc, washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The crude product was used for the next step (7 g, 107 %). R<sub>f</sub>=0.46 (hexane: EtOAc= 3: 1), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.95 (d, *J*=9.0 Hz, 2H), 7.87 (d, *J*=8.1 Hz, 2H), 7.58 (d, *J* = 9.0 Hz, 2H), 7.34 (t, *J*=8.0Hz, 2H), 7.28-7.18 (m, 4H), 5.08 (dd, *J*=6.8Hz, 4H), 3.14 (s, 6H).

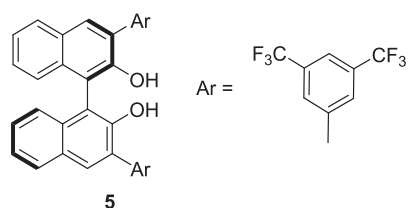


n-BuLi (27.3 ml, 73.4 mmol) was added to **2** in dry THF (150 ml) at 0°C under nitrogen. This mixture was kept at room temperature for 3h, then the mixture was cooled to -78°C, and i-PrOBPin (20 ml, 98 mmol) was added slowly to the mixture at -78°C and then kept at room temperature overnight. The reaction mixture was cooled to 0°C, and then 1N HCl was added for 1 h, extracted with EtOAc, washed with brine, and dried over MgSO<sub>4</sub>. The crude product was recrystallized by hexane to give the product (9.5 g, 93%). R<sub>f</sub> = 0.3 (hexane: EtOAc = 3: 1), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.45 (s, 2H), 7.89 (d, *J* = 8.1 Hz, 2H), 7.36 (t, *J* = 7.4 Hz, 2H), 7.25 (d, *J* = 9.1 Hz, 1H), 7.18 (d, *J* = 8.5 Hz, 2H), 4.39-4.8 (m, 4H), 2.28 (s, 6H), 1.38 (s, 24H).

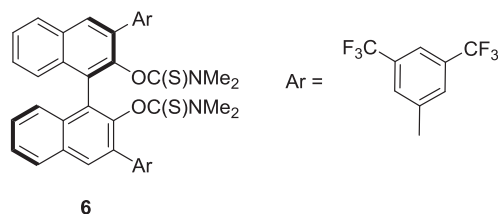


(*R*)-2,2'-bis(methoxymethoxy)-1,1'-binaphthyl-3,3'-bis(boronic acid pinacol ester) (161.2 mg, 0.257 mmol), 3,5 - bis (3,5 - di (3,5 - trifluoromethyl phenyl) -1-bromo benzene **3** (308.5 mg, 0.53 mmol), barium hydroxide deoctahydrate (285.4 mg, 0.91 mmol), Tetrakis(triphenylphosphine)palladium(0) (23.8 mg, 0.0201 mmol) were mixed in dioxane/H<sub>2</sub>O = 5/1 ( 4.2 ml). The mixture was stirred at 100°C for 22 h under nitrogen atmosphere, cooled to room temperature, quenched with 1M HCl. The reaction mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>, washed with brine, and dried over Na<sub>2</sub>SO<sub>4</sub>. After the removal of the solvent under reduced pressure, the residue was purified by column chromatography (hexane: benzene = 4: 1) to give (*R*)-3,3'-bis(3,5-di(3,5-trifluoromethylphenyl)phenyl)-2,2'-dimethoxymethoxy-1,1'-binaphthalene **4** (75% yield) as a white solid. R<sub>f</sub> = 0.30 (hexane: benzene = 3: 1), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.14 (s, 10H), 8.09 (d, *J* = 1.7 Hz, 4H), 8.01 (d, *J* = 8.3 Hz, 2H), 7.96 (s, 4H), 7.79 (t, *J* = 1.7 Hz, 2H), 7.50 – 7.54 (m, 2H), 7.38 – 7.39 (m, 4H), 4.48 4.59 (d, AB, *J*<sub>AB</sub> = 5.6 Hz, 4H), 2.58 (s, 6H), <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz) δ 151, 143, 141, 140, 134, 133 x 2, 132 x 2, 131 x 2, 129, 128 x 2, 127 x 3, 126 x 2, 125 x 2, 122 x 3, 120, 99, <sup>19</sup>F NMR(CDCl<sub>3</sub>, 376 MHz) δ -62.78, HRMS(ESI-MS(FTMS, Positive)), *m/z* calcd for

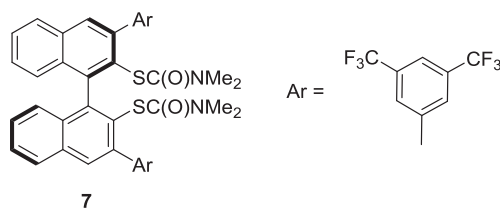
$C_{68}H_{38}F_{24}O_4[M+Na]^+$  1397.24 Found 1397.22754,  $[\alpha]_D^{25}$  ( $c$  7.01,  $CHCl_3$ ) -5.4, m.p 132-133°C.



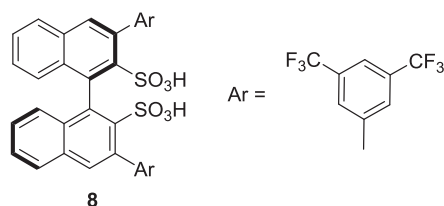
To a side-arm flask containing (*R*)-3,3'-bis(3,5-di(3,5-trifluoromethylphenyl)phenyl)-2,2'-dimethoxymethoxy-1,1'-binaphthalene **4** (257.6 mg, 0.19 mmol) in  $CH_2Cl_2$  (0.40 ml) was added trifluoroacetate (0.15 ml, 1.94 mmol) at 0°C. After stirred 3 h, saturated  $NaHCO_3$  was added and then extracted with  $CH_2Cl_2$ . The organic layer was washed with brine and dried with  $Na_2SO_4$ . After removal of the solvent under reduced pressure, the residue was purified by column chromatography (hexane: benzene= 3: 1) to give (*R*)-3,3'-bis(3,5-di(3,5-trifluoromethylphenyl)phenyl)-1,1'-binaphthalene -2,2'-diol (83% yield) as a white solid.  $R_f$ = 0.30 (hexane: benzene= 3: 1),  $^1H$  NMR (400MHz,  $CDCl_3$ )  $\delta$  8.19 (s, 2H), 8.11 (s, 8H), 8.01-8.05 (m, 6H), 7.93 (s, 4H), 7.78 (t,  $J$  = 1.8 Hz, 2H), 7.30 - 7.49 (m, 4H) 7.26 – 7.30 (m, 2H), 5.45 (s, 2H),  $^{13}C$  NMR ( $CDCl_3$ , 100 MHz)  $\delta$  150, 143, 140, 133 x 3, 132 x 3, 130, 129 x 2, 128 x 3, 126, 125 x 2, 124, 122 x 3,  $^{19}F$  NMR ( $CDCl_3$ , 376 MHz)  $\delta$  -62.70, HRMS(ESI-MS(FTMS, Negative)),  $m/z$  calcd for  $C_{64}H_{30}F_{24}O_2[M-H]^-$  1258.19 Found 1285.17700,  $[\alpha]_D^{25}$  ( $c$  9.23,  $CHCl_3$ ) + 0.35, m.p 158-159°C.



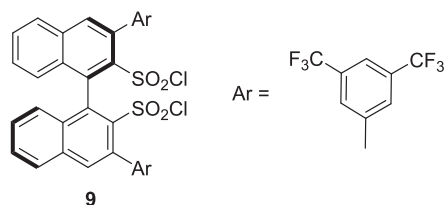
$NaH$  (198.1 mg, 4.95 mmol) was added to a solution of (*R*)-3,3'-bis(3,5-di(3,5-trifluoromethylphenyl)phenyl)-1,1'-binaphthalene -2,2'-diol **5** (1.23 g, 0.96 mmol) in THF (10.6 mL) at 0°C and stirred for 1 h at the temperature. To this solution, was added *N,N*-dimethylthiocarbonylchloride (482.2 mg, 3.90 mmol) and 1,4-diazabicyclo[2.2.2]octane (DABCO) (137.7 mg, 1.23 mmol). The reaction mixture was heated at 60°C for 15 h, then cooled to room temperature, quenched with 2%  $KOH$  solution, and extracted by EtOAc. The combined organic layer was washed with brine, and dried with  $Na_2SO_4$ . After removal of the solvent under reduced pressure, the residue was purified by column chromatography ( $SiO_2$ , hexane: EtOAc= 8: 1) to give the product **6** (1.40 g) quantitatively as a white solid.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.18-7.31 (m, 28H), 2.99-1.99 (m, 12H),  $^{19}F$  NMR (376 MHz,  $CDCl_3$ )  $\delta$  -62.8; HRMS (ESI+)  $m/z$  calcd for  $C_{70}H_{40}F_{25}N_2O_2S_2 [M+H]^+$  1461.22207, found 1461.22314,  $[\alpha]_D^{25}$  ( $c$  0.31,  $CHCl_3$ ) -8.5, m.p. 129-130°C.



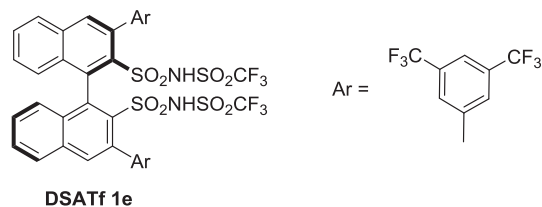
(*R*)-*O,O'*-(3,3'-bis(3,3'',5,5''-tetrakis(trifluoromethyl)-[1,1':3',1''-terphenyl]-5'-yl)-[1,1'-binaphthalene]-2,2'-diyl) bis(dimethylcarbamothioate) **6** (1.37 g, 0.94 mmol) was placed into a round flask and heated to 230°C for 8 h under N<sub>2</sub> atmosphere. The resulting crude product was purified by column chromatography (SiO<sub>2</sub>, hexane: CH<sub>2</sub>Cl<sub>2</sub>= 1: 1) to give the product **7** (875.3 mg, 64% yield) as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.11 (s, 10H), 7.99-7.91 (m, 10H), 7.76 (s, 2H), 7.58-7.52 (m, 2H), 7.33 (s, 4H), 2.45 (s, 12H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 165.7, 145.0, 144.1, 143.0, 142.8, 138.6, 133.6, 132.8, 132.4 (q, *J* = 33.4 Hz), 129.8, 129.6, 127.8, 127.7, 127.5, 126.9, 124.6, 123.3 (q, *J* = 272.7 Hz), 121.4, 36.6, <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -62.7, HRMS (ESI+) *m/z* calcd for C<sub>70</sub>H<sub>40</sub>F<sub>24</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>Na [M+Na]<sup>+</sup> 1483.20402, found 1483.20288, [α]<sub>D</sub><sup>25</sup> (*c* 6.53, CHCl<sub>3</sub>) -5.6, m.p. 127-128°C.



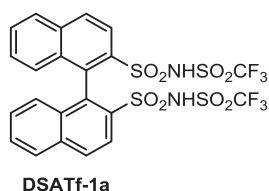
A performic acid solution was prepared by mixing and stirring 30 % H<sub>2</sub>O<sub>2</sub> (3.0 mL, 32.0 mmol) and formic acid (17 ml, 447.0 mmol) at 0°C for 1 h. To this solution was added dropwise a solution of **7** (736.4 mg, 0.51 mmol) in (8.0 ml) at 0°C. After stirring for 12 h at 0°C to room temperature, the reaction mixture was extracted by CH<sub>2</sub>Cl<sub>2</sub>, washed with brine, dried with Na<sub>2</sub>SO<sub>4</sub>, and evaporated. The crude product was purified by column chromatography (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>: MeOH= 20: 1). The resultant solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and washed with 1N HCl. Remaining water traces were removed by azeotropic distillation with benzene to give the product **8** (519.2 mg, 72% yield) as a white solid. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 8.40 (s, 10H), 8.13-7.94 (m, 12H), 7.54 (t, *J* = 7.8 Hz, 2H), 7.33 (t, *J* = 7.8 Hz, 2H), 7.15 (d, *J* = 8.6 Hz, 2H), <sup>13</sup>C NMR (101 MHz, CD<sub>3</sub>OD) δ 146.4, 145.1, 139.6, 139.0, 138.6, 138.4, 134.4, 133.3 (q, *J* = 32.8 Hz), 132.6, 131.3, 129.1, 128.8, 128.6, 128.0, 125.7, 125.1 (q, *J* = 274.9 Hz), 121.9, <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -64.1, HRMS(ESI-) calcd for C<sub>64</sub>H<sub>28</sub>F<sub>24</sub>O<sub>6</sub>S<sub>2</sub> [M-2H]<sup>2-</sup> 706.04775, found 706.04742.



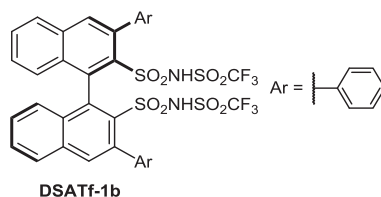
To a solution of (*R*)-3,3'-bis(3,3'',5,5''-tetrakis(trifluoromethyl)-[1,1':3',1''-terphenyl]-5'-yl)-[1,1'-binaphthalene]-2,2'-disulfonic acid **8** (471.3 mg, 0.33 mmol) in SOCl<sub>2</sub> (0.4 ml) was added several drops of dry DMF. After the mixture was refluxed for 2 h, SOCl<sub>2</sub> was removed under reduced pressure. The resulting crude product was purified by column chromatography (SiO<sub>2</sub>, hexane: EtOAc= 8: 1) to give the product **9** (426.2 mg, 89% yield) as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.43-8.35 (m, 10H), 8.20-8.10 (m, 6H), 8.03-7.97 (m, 6H), 7.79 (t, *J*= 7.7 Hz, 2H), 7.54 (t, *J*= 7.0 Hz, 2H), 7.30 (d, *J*= 8.0 Hz, 2H), <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -62.7, HRMS (ESI+) calcd for C<sub>64</sub>H<sub>28</sub>O<sub>4</sub>Cl<sub>2</sub>F<sub>24</sub>NaS<sub>2</sub> 1473.03150, found 1473.03296, [α]<sub>D</sub><sup>25</sup> (*c* 1.23, CHCl<sub>3</sub>) +3.1.



A mixture of (*R*)-3,3'-Bis(2,2'',5,5''-tetrakis(trifluoromethyl)-[1,1':3',1''-terphenyl]-5'-yl)-[1,1'-binaphthalene]-2,2'-disulfonyl dichloride **9** (392.0 mg, 0.27 mmol), trifluoromethanesulfonamide (148.5 mg, 0.10 mmol), K<sub>2</sub>CO<sub>3</sub> (161.4 mg, 1.17 mmol) was dissolved in MeCN (1.3 ml). The reaction mixture was stirred at 70°C for 12 h, quenched with 1N HCl, and extracted by EtOAc. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated to obtain the crude product which was purified by column chromatography (SiO<sub>2</sub>, hexane: EtOAc= 3: 1). The resultant solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and washed with 1N HCl. Remaining water traces were removed by azeotropic distillation with benzene to give **DSATf 1e** (270.1 mg, 60% yield) as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.48 (s, 4H), 8.40 (s, 4H), 8.17 (s, 2H), 8.07 (s, 2H), 8.01-7.89 (m, 10H), 7.54 (t, *J*= 8.0 Hz, 2H), 7.31 (d, *J*= 8.0 Hz, 2H), 7.26 (t, *J*= 8.0 Hz, 2H), <sup>13</sup>C NMR (CD<sub>3</sub>OD, 101 MHz) δ 145.5, 145.2, 139.3, 139.0, 138.7, 138.3, 134.7, 134.1, 133.5, 133.1 (q, *J*= 33.6 Hz), 131.9, 131.8, 130.1, 129.2, 128.4, 128.1, 125.9, 125.0 (q, *J*= 271.3 Hz), 121.3 (q, *J*= 319.1 Hz), 121.0 (q, *J*= 322.2 Hz), 120.9, <sup>19</sup>F NMR (376 MHz, CD<sub>3</sub>OD) δ -64.1, -79.4, HRMS (ESI-) *m/z* calcd for C<sub>66</sub>H<sub>28</sub>F<sub>30</sub>N<sub>2</sub>O<sub>8</sub>S<sub>4</sub> [M-2H]<sup>2-</sup> 837.01302, found 837.01172, [α]<sub>D</sub><sup>25</sup> (*c* 0.04, CHCl<sub>3</sub>) +75.8.

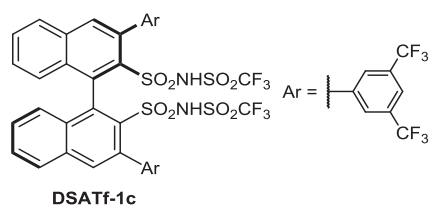


1,1'-Binaphthalene-2,2'-disulfonyl chloride (303.3 mg, 0.672 mmol), TfNH<sub>2</sub> (245.9 mg, 1.649 mmol) and K<sub>2</sub>CO<sub>3</sub> (241.9 mg, 1.75 mmol) were dissolved in MeCN 0.61 ml and stirred for 25 h at r.t.. The reaction mixture was quenched with water and extracted by CH<sub>2</sub>Cl<sub>2</sub>. The water layer was added 1 M HCl and extracted by EtOAc. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated to obtain N<sup>2</sup>, N<sup>2'</sup>-bis((trifluoromethyl)sulfonyl)-[1,1'-binaphthalene]-2,2'-disulfonamide **DSATf-1a** (424.6 mg, 0.628 mmol, 93%). *R*<sub>f</sub> = 0.15 (EtOAc: MeOH = 20:1), <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD) δ 9.0 (d, *J* = 9.0 Hz, 2 H), 8.18 (d, *J* = 7.6 Hz, 2 H), 7.98 (d, *J* = 8.4 Hz, 2 H), 7.61 (t, *J* = 7.6 Hz, 2 H), 7.35 (t, *J* = 7.6 Hz, 2 H), 6.99 (d, *J* = 8.6 Hz, 2 H) <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>OD) 140.5, 135.2, 134.7, 133.7, 129.1, 129.0, 128.3, 128.1, 127.4, 124.9, 120.8 (q, *J* = 320 Hz), 120.7 (q, *J* = 320 Hz), <sup>19</sup>F NMR (400 MHz, CD<sub>3</sub>OD), δ -79.65, IR (KBr): ν = 3391, 3250, 1638, 1321, 1192 cm<sup>-1</sup>. HRMS (APCI+) *m/z* calcd for C<sub>22</sub>H<sub>14</sub>O<sub>8</sub>N<sub>2</sub>F<sub>6</sub>NaS<sub>4</sub> [M+Na]<sup>+</sup> 698.94110 Found 698.94294, [α]<sub>D</sub><sup>23</sup> = -35.7° (c 1.25 in MeOH), m.p. 186~189°C.

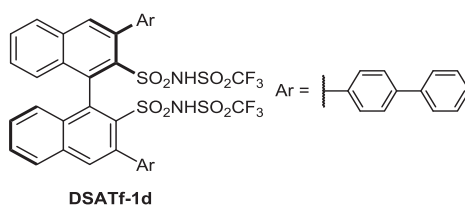


A mixture of 3,3'-diphenyl-[1,1'-binaphthalene]-2,2'-disulfonyl dichloride (101.9 mg, 0.17 mmol), trifluoromethanesulfonamide (47.7 mg, 0.320 mmol), K<sub>2</sub>CO<sub>3</sub> (67.0 mg, 0.491 mmol) was dissolved in MeCN 0.5 ml. The reaction mixture was stirred at 35°C for 48 h, quenched with 1 M HCl, extracted by EtOAc, washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, evaporated, and purified by column chromatography (SiO<sub>2</sub>, EtOAc) to get (*R*)-3,3'-diphenyl-N<sup>2</sup>, N<sup>2'</sup>-bis((trifluoromethyl)sulfonyl)-[1,1'-binaphthalene]-2,2'-disulfonamide **DSATf-1b** (969.7 mg, 1.58 mmol, 91%). *R*<sub>f</sub> = 0.10 (EtOAc), <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD) δ 7.85 (d, *J* = 8.2 Hz, 2 H), 7.77 (s, 2 H), 7.69 (d, *J* = 6.1 Hz, 4 H), 7.49 (dt, *J* = 7.5, 1.0 Hz, 2 H), 7.29-7.37 (m, 6 H), 7.23 (dt, *J* = 7.8, 1.0 Hz, 2 H), 7.12 (d, *J* = 8.7 Hz, 2 H), <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 142, 139, 138 x 2, 133 x 2, 130 x 2, 129, 126, 119 (q, *J* = 323 Hz), <sup>19</sup>F NMR (400 MHz, CD<sub>3</sub>OD) δ -77.8, IR (KBr): ν = 3434, 1620, 1308, 1190 cm<sup>-1</sup>, HRMS (ESI-) calcd for C<sub>34</sub>H<sub>20</sub>O<sub>8</sub>N<sub>2</sub>F<sub>6</sub>NaS<sub>4</sub> 848.99099, found 848.99127, [α]<sub>D</sub><sup>23</sup> = 55.9 (c 1.26 in CHCl<sub>3</sub>), m.p. 104-105°C.





A mixture of 3, 3'- bis(3,5-bis(trifluoromethyl)phenyl)-[1,1'-binaphthalene]-2, 2'-disulfonyl dichloride (814.9 mg, 0.93 mmol), trifluoromethanesulfonamide (386.3 mg, 2.59 mmol),  $K_2CO_3$  (479.6 mg, 3.47 mmol) was dissolved in MeCN 5.0 ml. The reaction mixture was stirred at r.t. for 48 h, quenched with 1 M HCl, and extracted by EtOAc, washed with brine, dried over  $Na_2SO_4$ , evaporated, and purified by column chromatography ( $SiO_2$ , EtOAc) to get (*R*)-3,3'-bis(3,5-bis(trifluoromethyl)phenyl)-N2,N2'-bis((trifluoromethyl)sulfonyl)-[1,1'-binaphthalene]-2,2'-disulfonamide **DSATf-1c** (1.08 g, 0.98 mmol, 98%).  $R_f = 0.3$  ( $CH_2Cl_2$ : MeOH = 6: 1),  $^1H$  NMR (400 MHz,  $CD_3OD$ )  $\delta$  8.19 (s, 4 H), 7.84 (d,  $J = 8.1$  Hz, 2 H), 7.78 (s, 2 H), 7.73 (s, 2 H), 7.46 (t,  $J = 7.8$  Hz, 2 H), 7.22 (t,  $J = 7.8$  Hz, 2 H), 7.11 (d,  $J = 8.1$  Hz, 2 H),  $^{13}C$  NMR (125 MHz,  $CDCl_3$ ) 145, 138, 137, 135, 133 x 2, 132 x 2, 130, 129, 128, 127 x 2, 125, 120, 119 (q,  $J = 322$  Hz),  $^{19}F$  NMR (400 MHz,  $CD_3OD$ )  $\delta$  -62.10, -78.29, IR (KBr):  $\nu = 3468$ , 1621, 1311, 1280, 1185  $cm^{-1}$ ,  $[\alpha]_D^{23} = 48.5$  ( $c$  1.27,  $CHCl_3$ ), m.p.141~143°C.

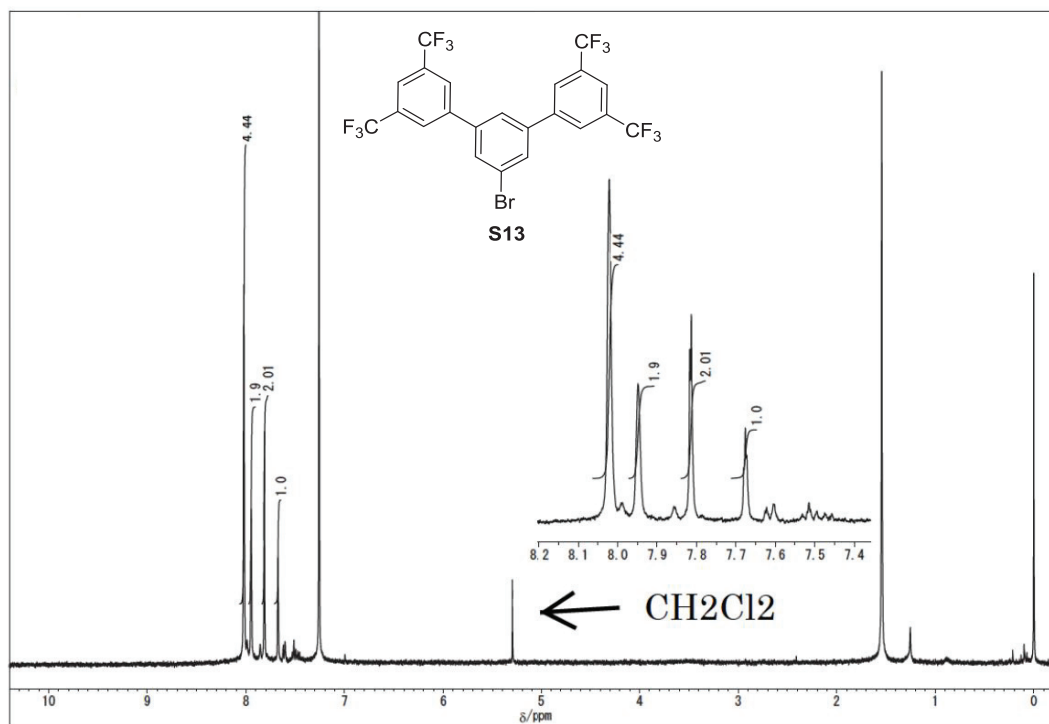


A mixture of 3, 3'- di([1,1'-biphenyl]-4-yl)-[1, 1'-binaphthalene]-2, 2'-disulfonyl dichloride (295.4 mg, 0.391 mmol), trifluoromethanesulfonamide (182.5 mg, 1.22 mmol),  $K_2CO_3$  (140.9 mg, 1.02 mmol) was dissolved in MeCN (3.0 ml). The reaction mixture was stirred at r.t. for 48 h, quenched with 1 M HCl, and extracted by EtOAc, washed with brine, dried over  $Na_2SO_4$ , evaporated, and purified by column chromatography ( $SiO_2$ , EtOAc) to get (*R*)-3,3'-di([1,1'-biphenyl]-4-yl)-N2,N2'-bis((trifluoromethyl)sulfonyl)-[1,1'-binaphthalene]-2,2'-disulfonamide **DSATf-1d** (192.5 mg, 0.20 mmol, 50%).  $R_f = 0.3$  ( $CH_2Cl_2$ /MeOH = 6/1),  $^1H$  NMR (400 MHz,  $CD_3OD$ )  $\delta$  7.86-7.91 (m, 2 H), 7.78 (d,  $J = 7.6$  Hz, 2 H), 7.70 (d,  $J = 7.7$  Hz, 2 H), 7.65 (d,  $J = 8.4$  Hz, 2 H), 7.53 (t,  $J = 7.3$  Hz, 2 H), 7.46 (t,  $J = 7.2$  Hz, 2 H), 7.33-7.36 (m, 2 H), 7.26 (t,  $J = 7.3$  Hz, 2 H), 7.10 (d,  $J = 8.2$  Hz, 2 H),  $^{13}C$  NMR (125 MHz,  $CDCl_3$ )  $\delta$  145, 143, 142 x 2, 141, 138, 137, 136, 135, 133, 132 x 2, 131 x 4, 130 x 3, 128, 123 (q,  $J = 322$  Hz),  $^{19}F$  NMR (400 MHz,  $CD_3OD$ )  $\delta$  -77.72, IR (KBr):  $\nu = 3435$ , 1629, 1307, 1190  $cm^{-1}$ ,  $[\alpha]_D^{23} = 71.9$  ( $c$  1.56 in  $CHCl_3$ ), m.p.190~192°C.

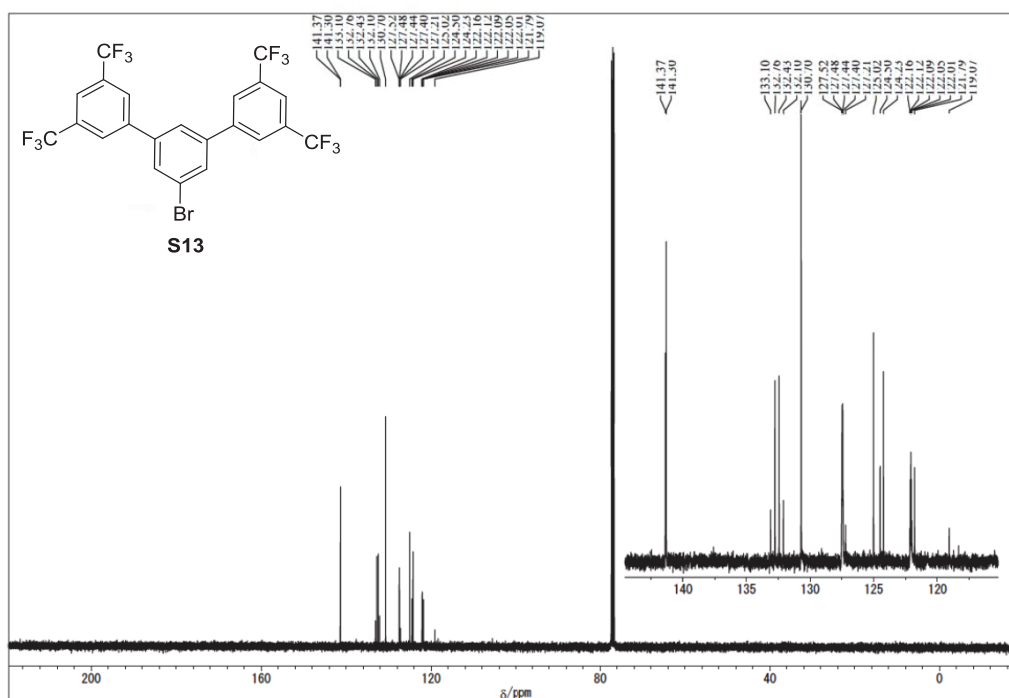


## 2-8-7. NMR spectra of new compounds

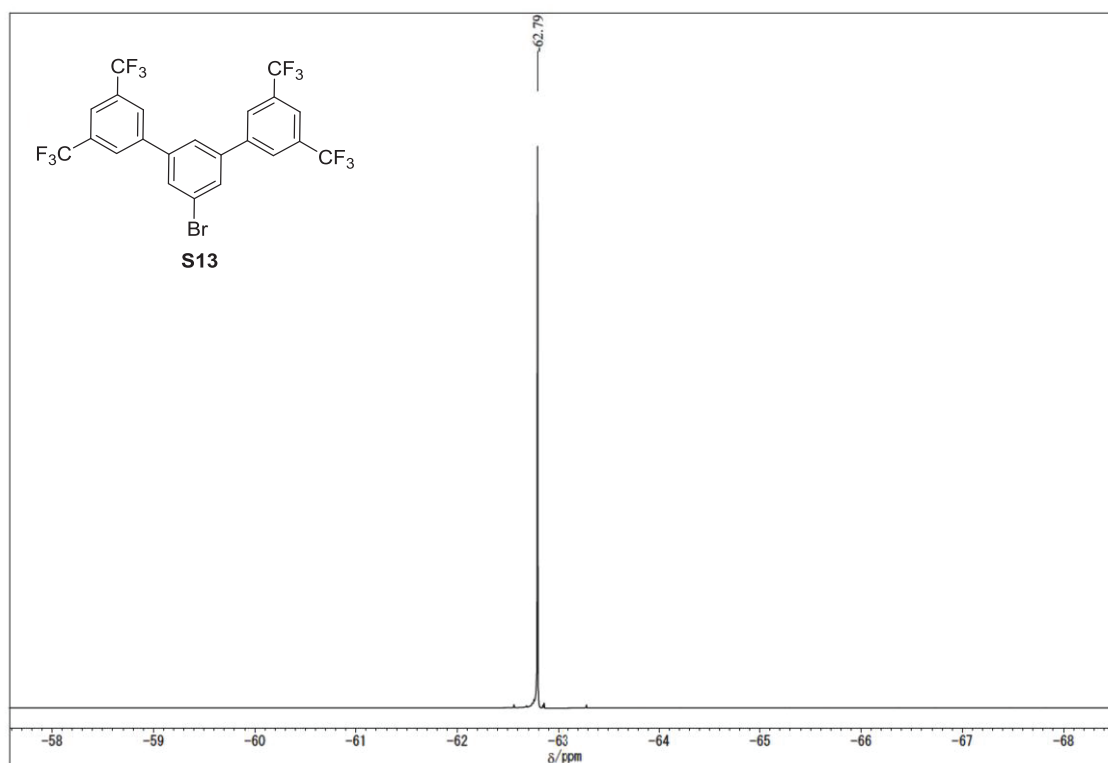
### $^1\text{H}$ NMR



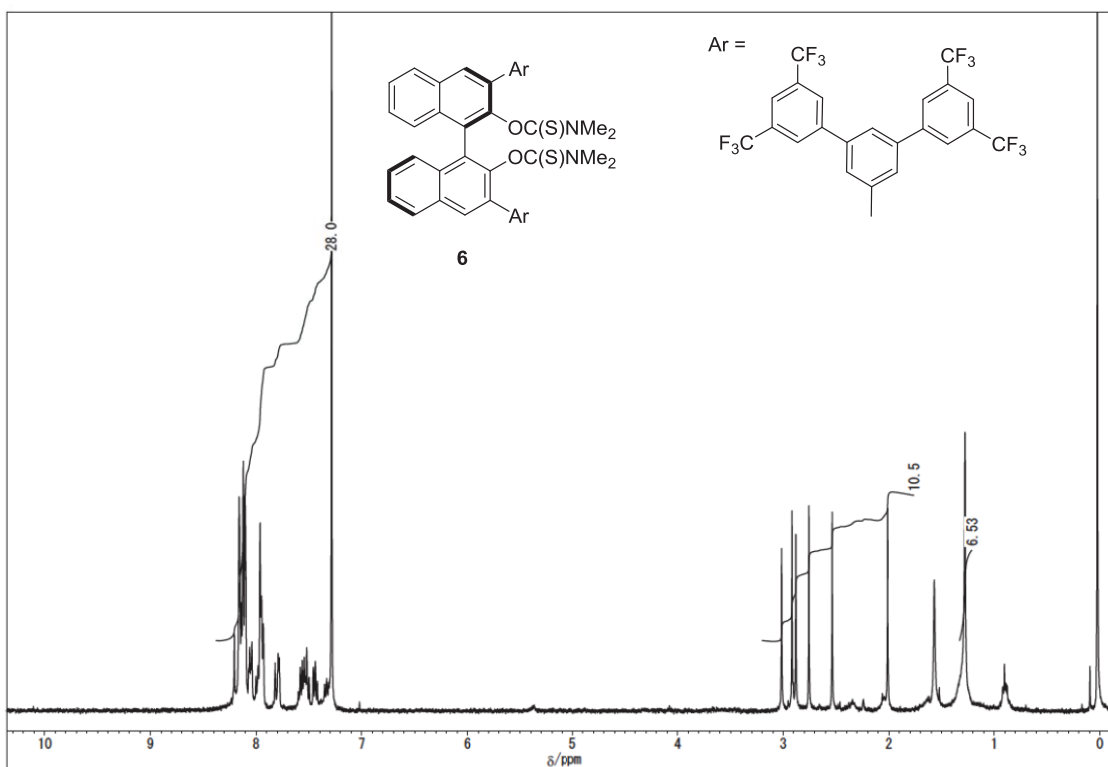
### $^{13}\text{C}$ NMR



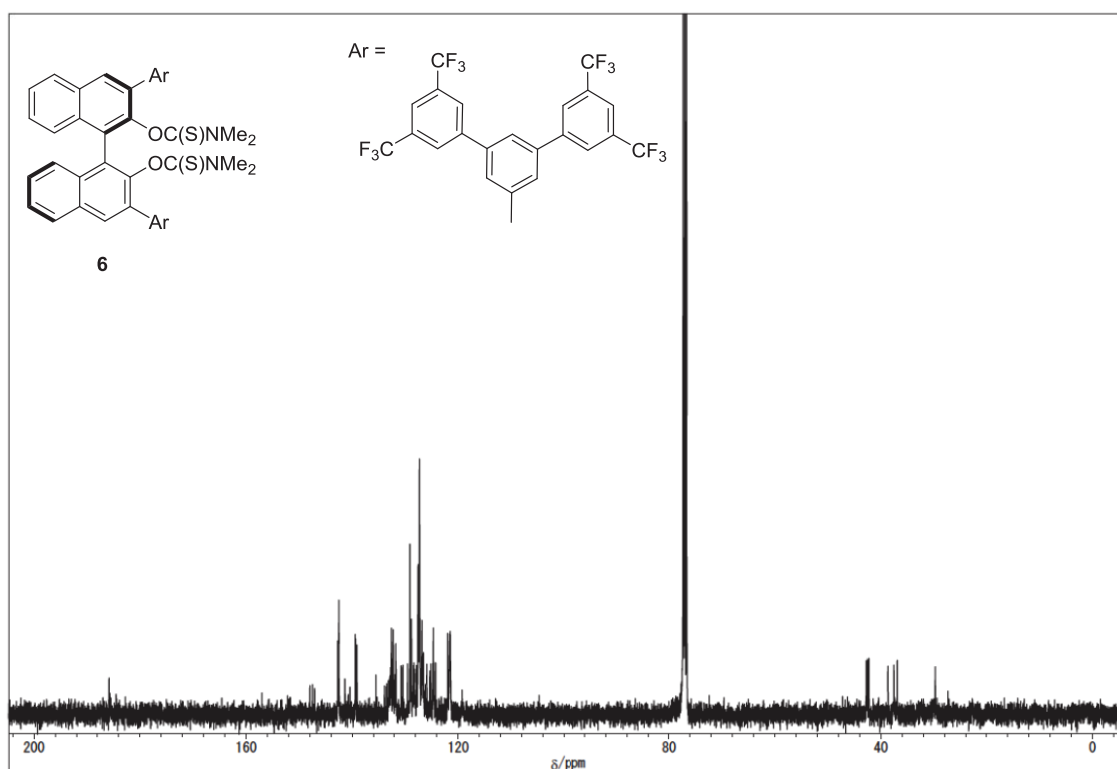
$^{19}\text{F}$  NMR



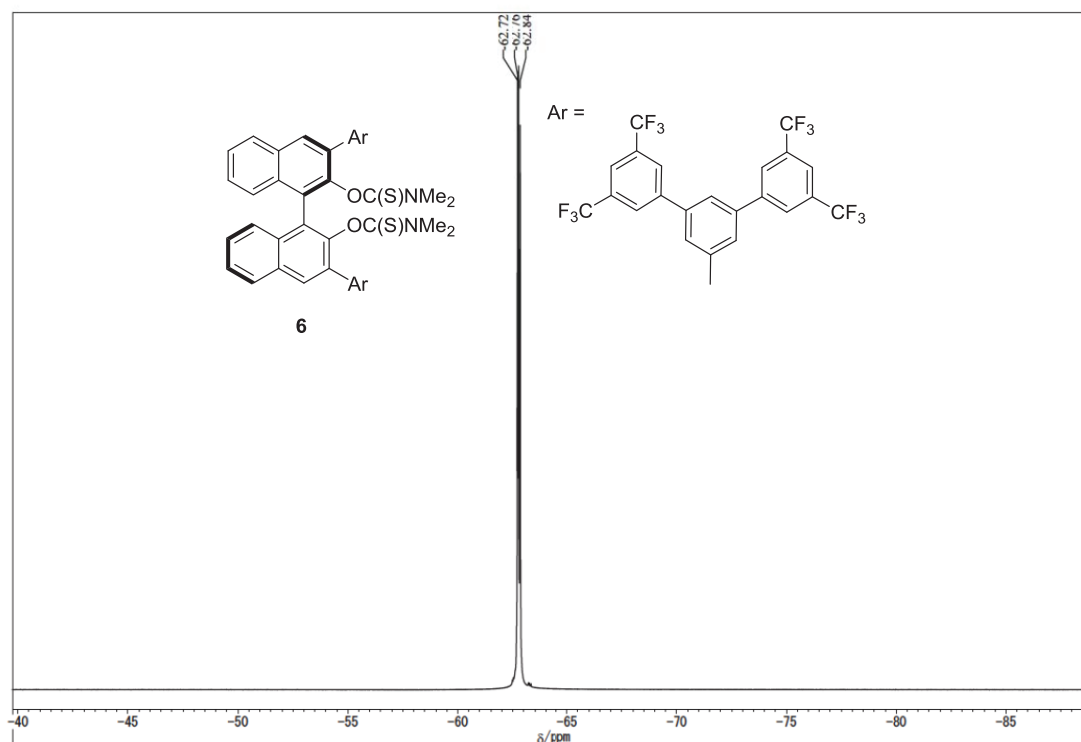
$^1\text{H}$  NMR



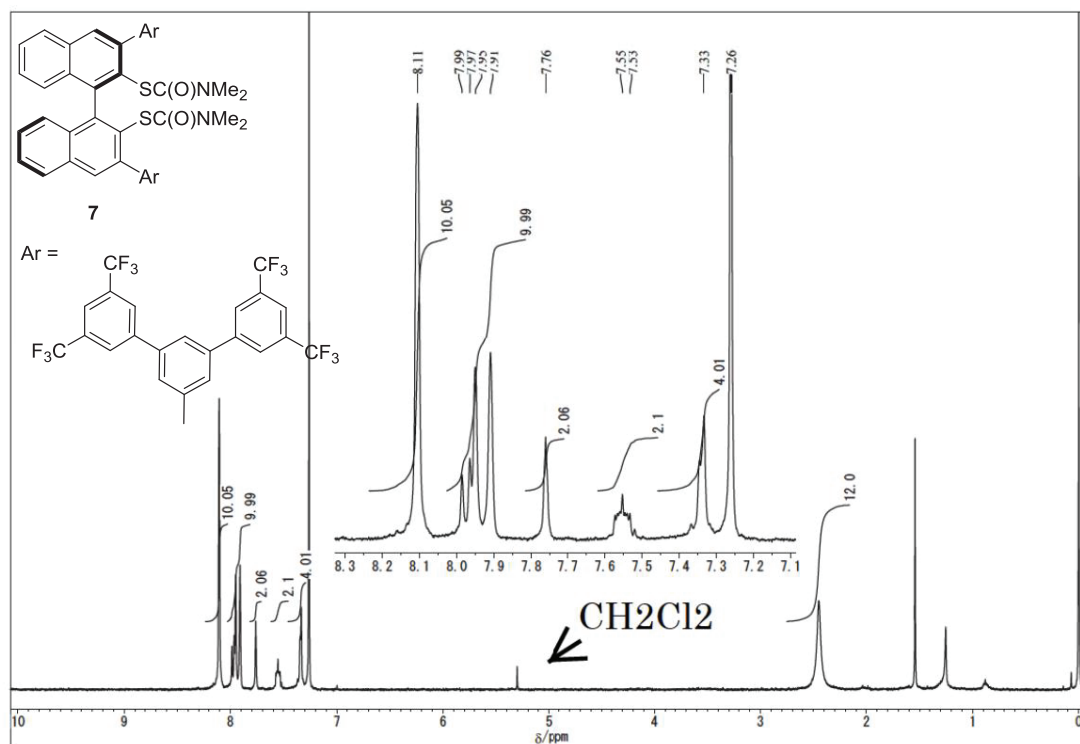
$^{13}\text{C}$  NMR



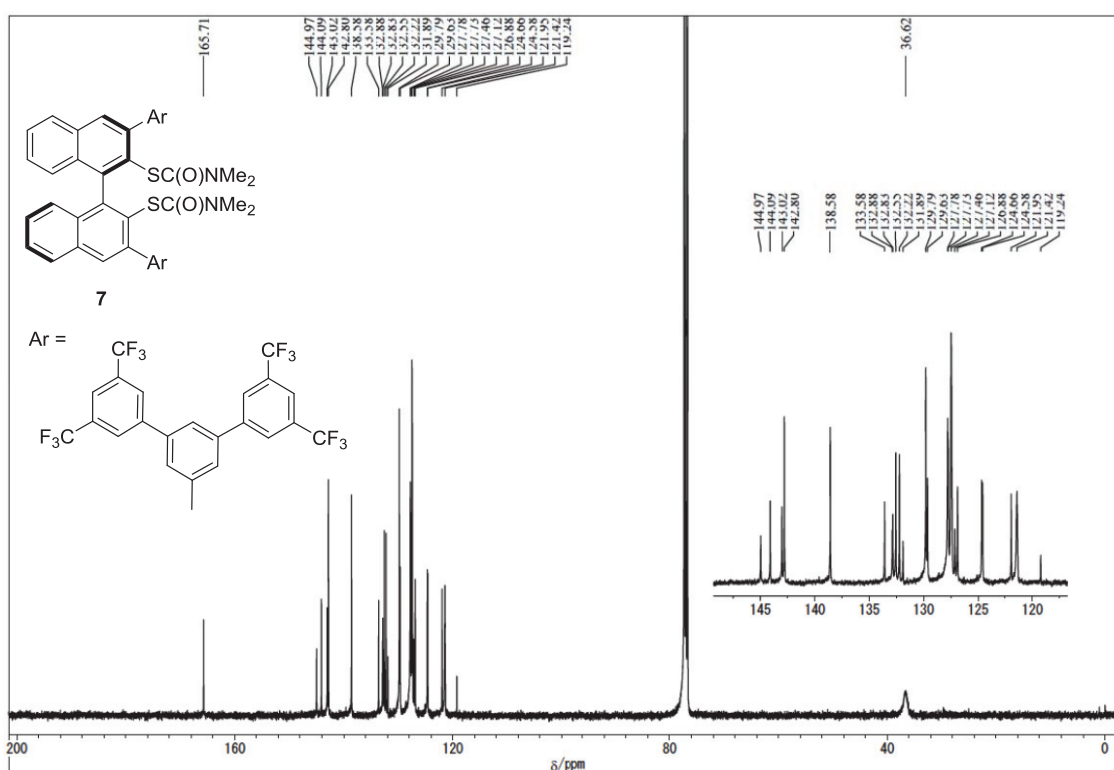
$^{19}\text{F}$  NMR



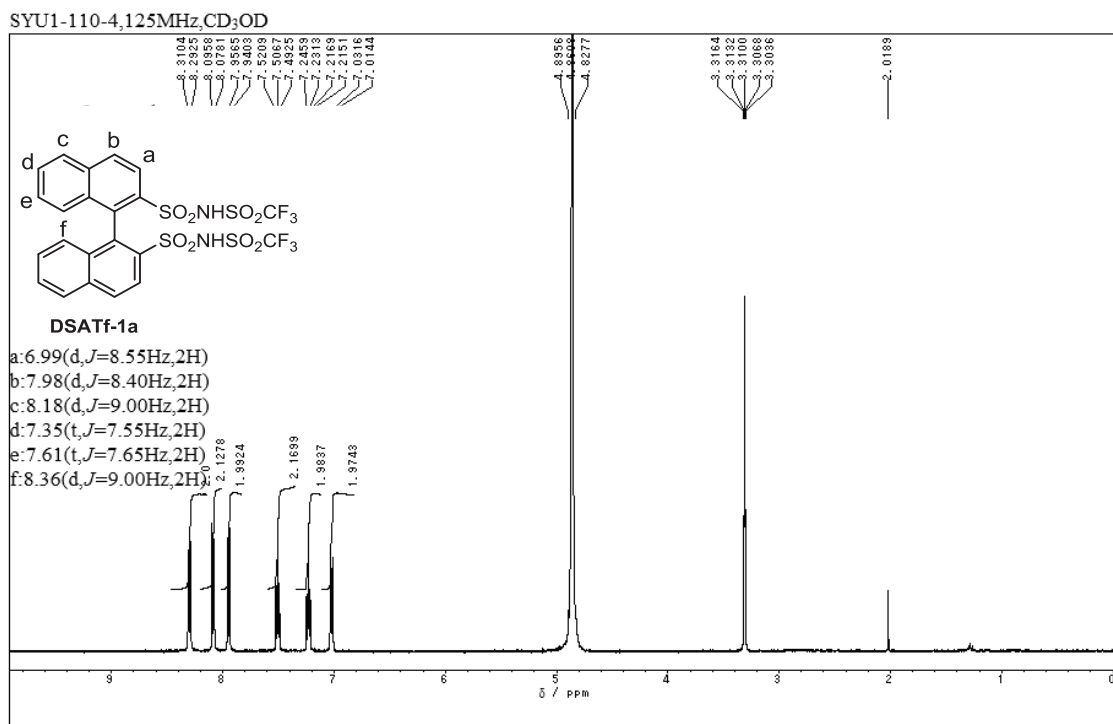
# $^1\text{H}$ NMR



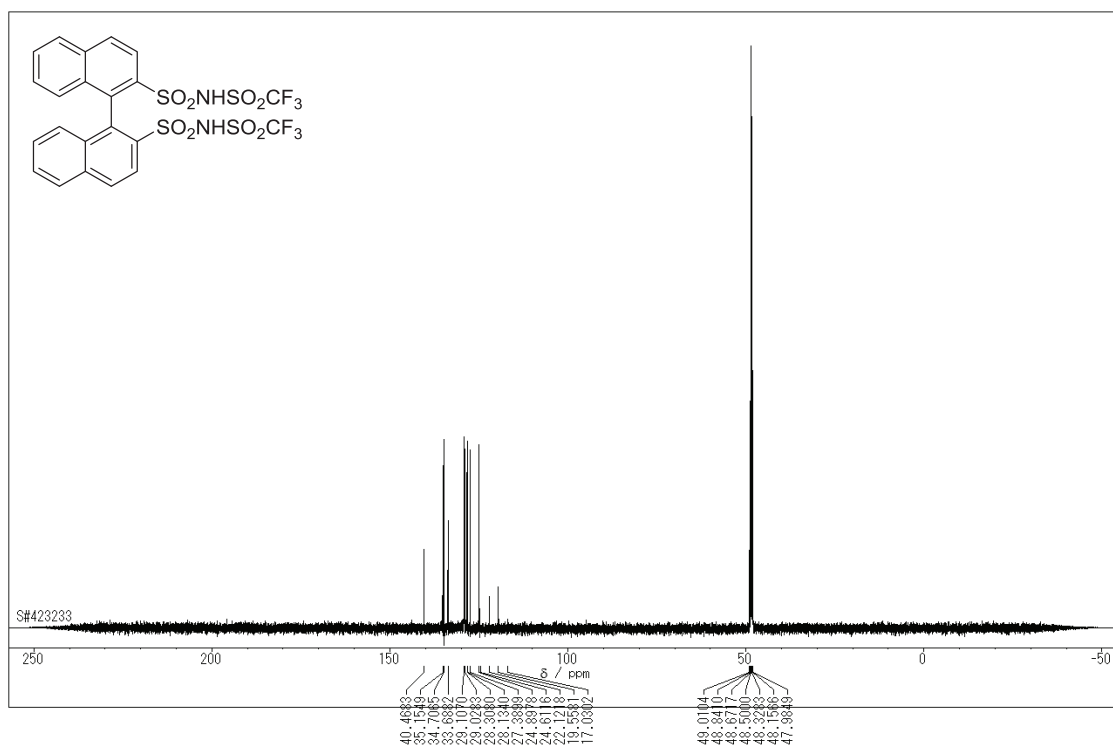
# $^{13}\text{C}$ NMR



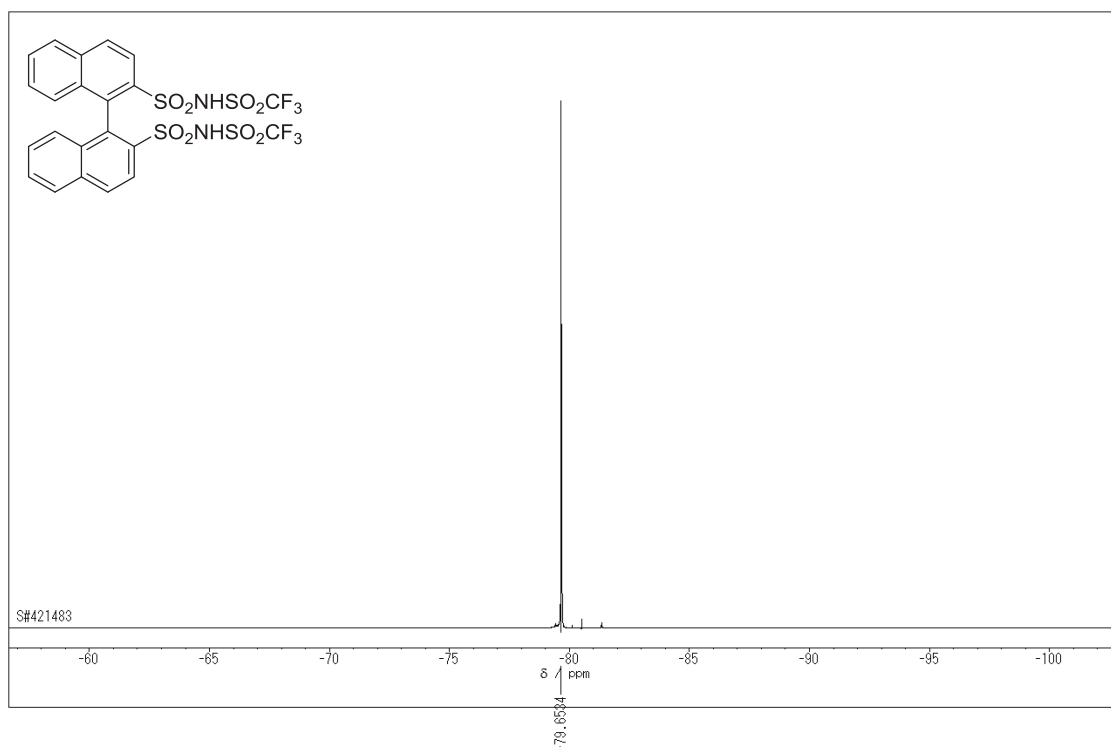
# <sup>1</sup>H NMR



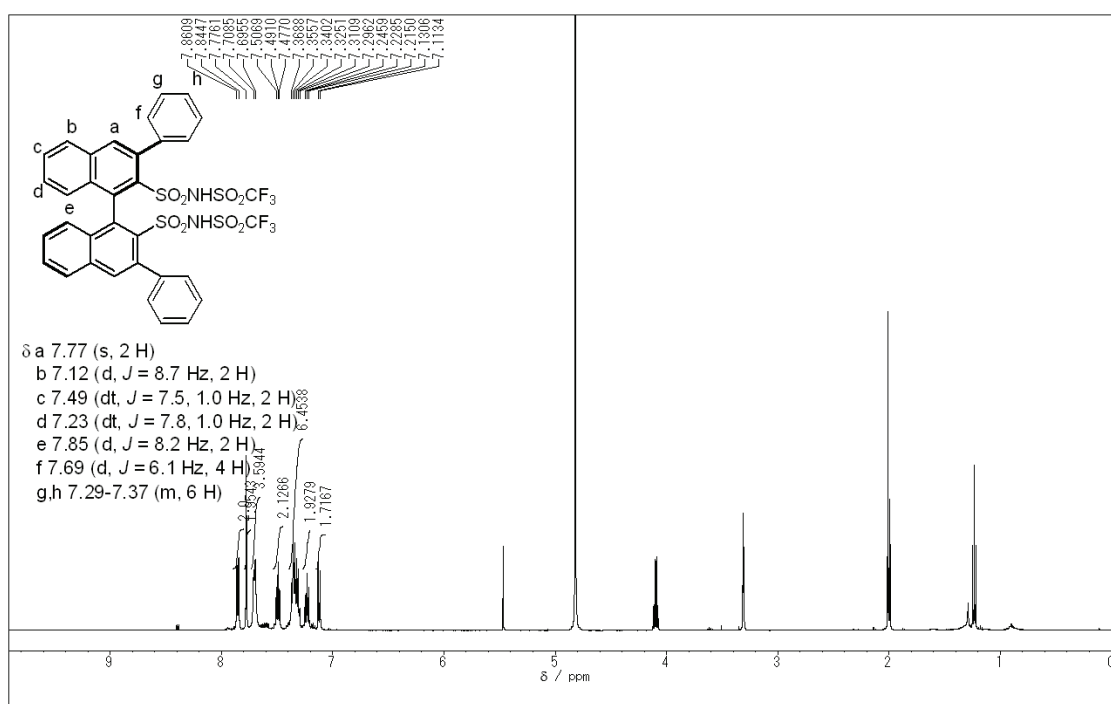
# <sup>13</sup>C NMR



# $^{19}\text{F}$ NMR

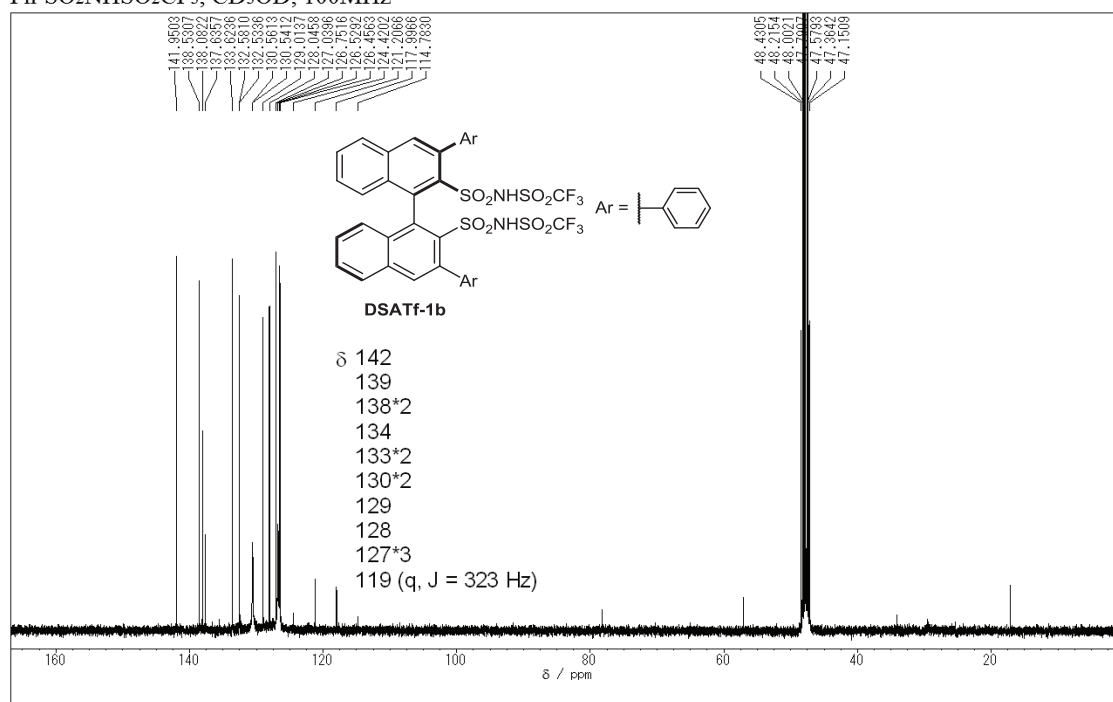


# $^1\text{H}$ NMR



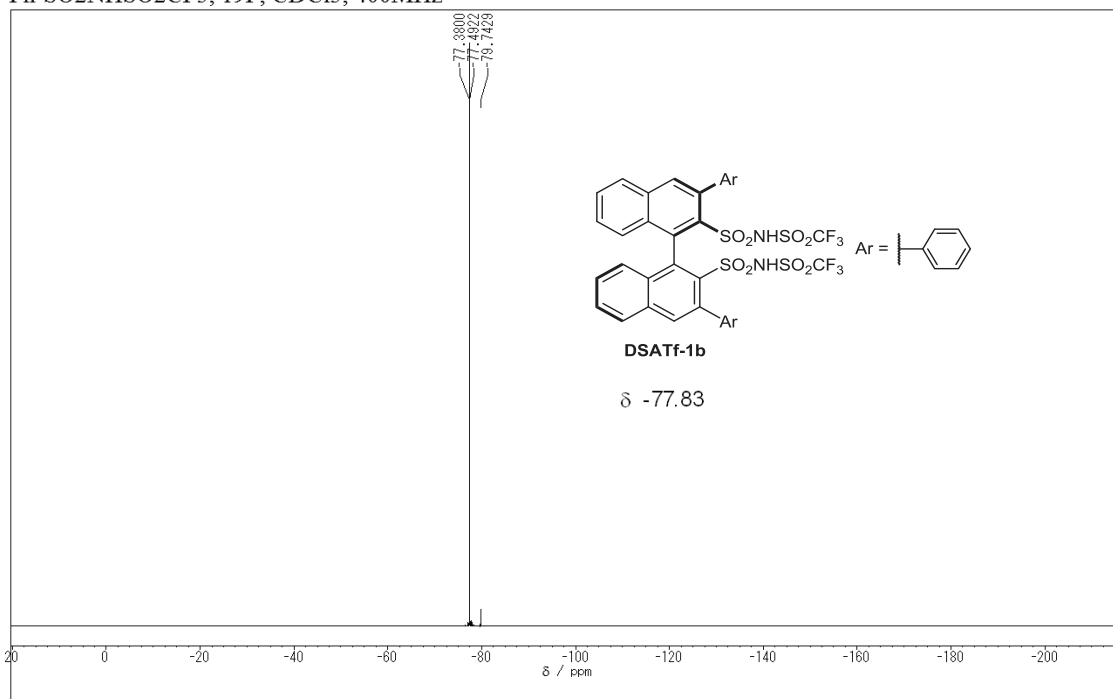
<sup>13</sup>C NMR

Ph-SO<sub>2</sub>NHSO<sub>2</sub>CF<sub>3</sub>, CD<sub>3</sub>OD, 100MHz



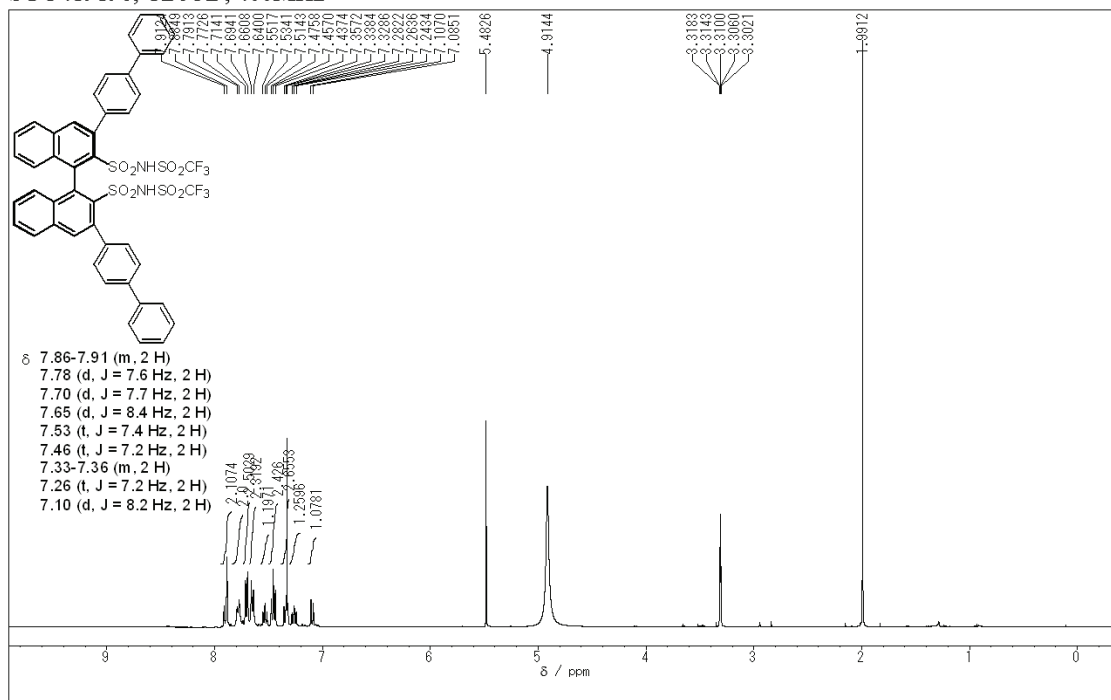
<sup>19</sup>F NMR

Ph-SO<sub>2</sub>NHSO<sub>2</sub>CF<sub>3</sub>, 19F, CDCl<sub>3</sub>, 400MHz



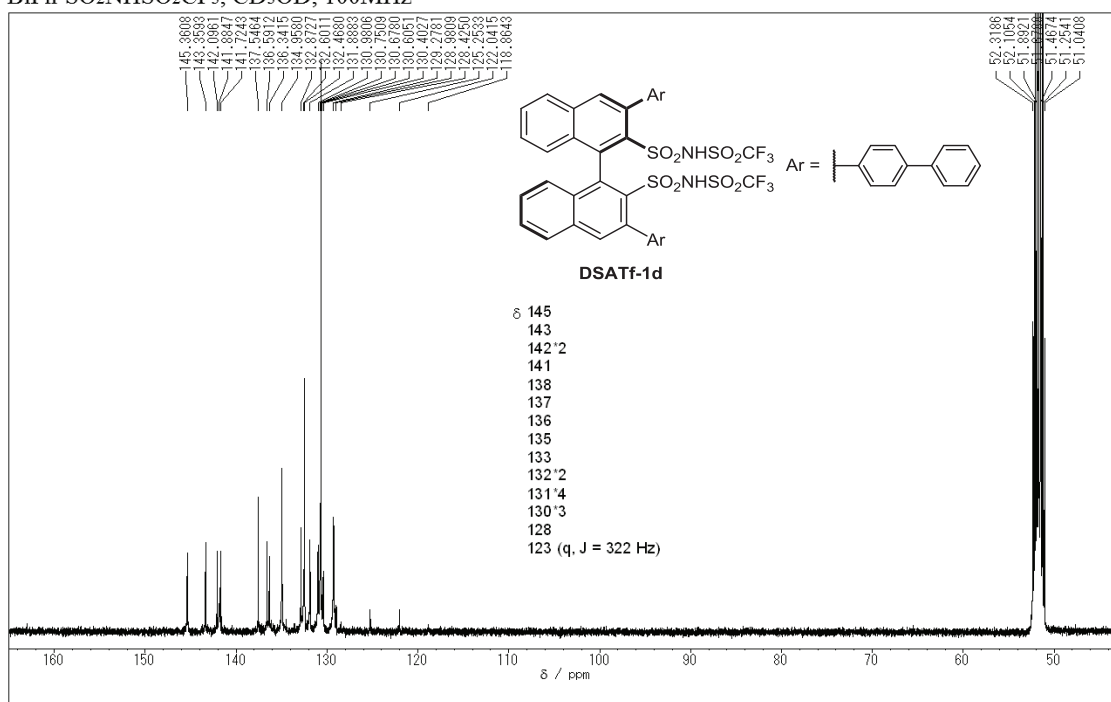
# <sup>1</sup>H NMR

SYU419196, CD<sub>3</sub>OD, 400MHz



# <sup>13</sup>C NMR

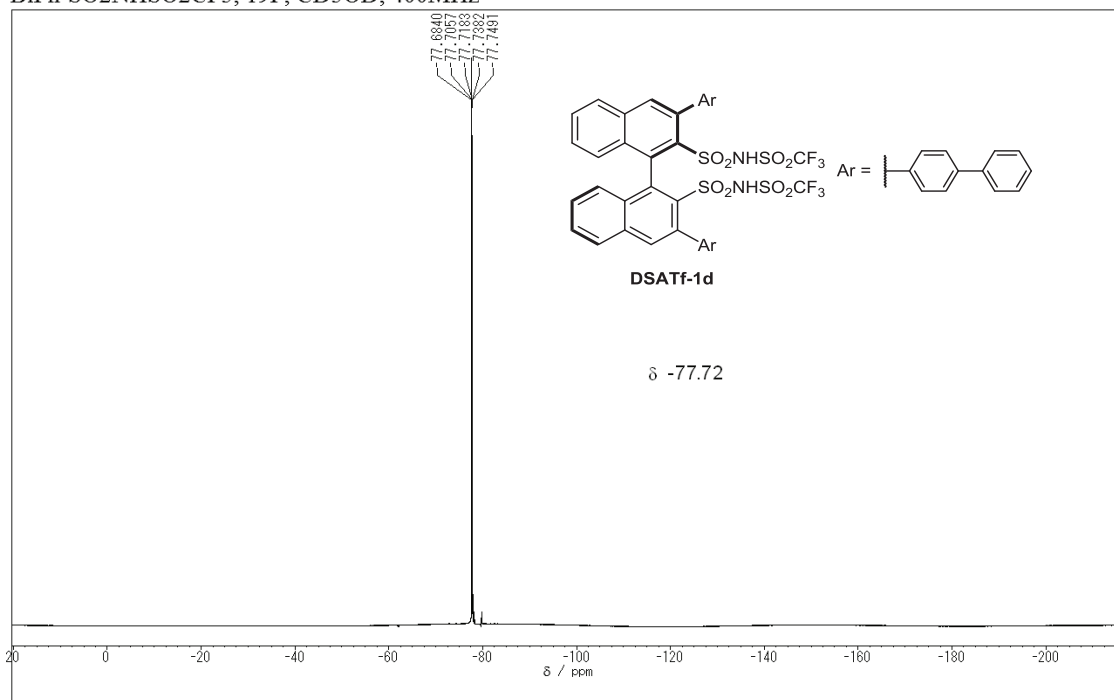
BiPh-SO<sub>2</sub>NHSO<sub>2</sub>CF<sub>3</sub>, CD<sub>3</sub>OD, 100MHz



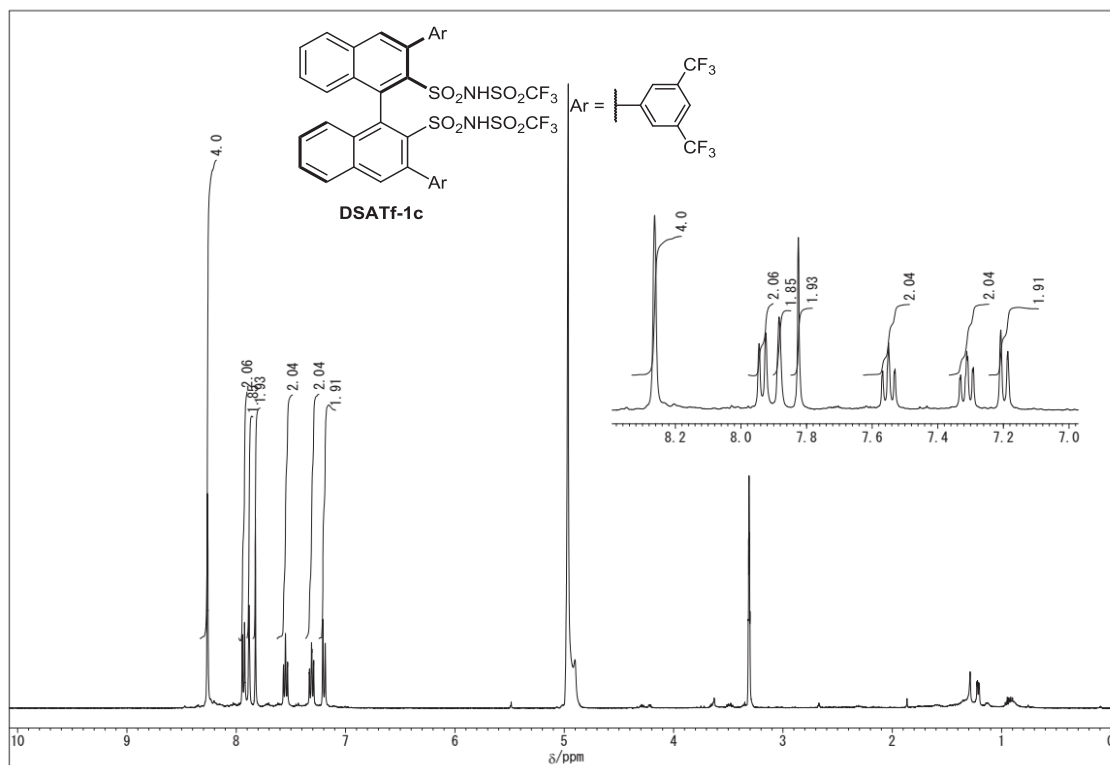


# $^{19}\text{F}$ NMR

BiPh-SO<sub>2</sub>NHSO<sub>2</sub>CF<sub>3</sub>, 19F, CD<sub>3</sub>OD, 400MHz



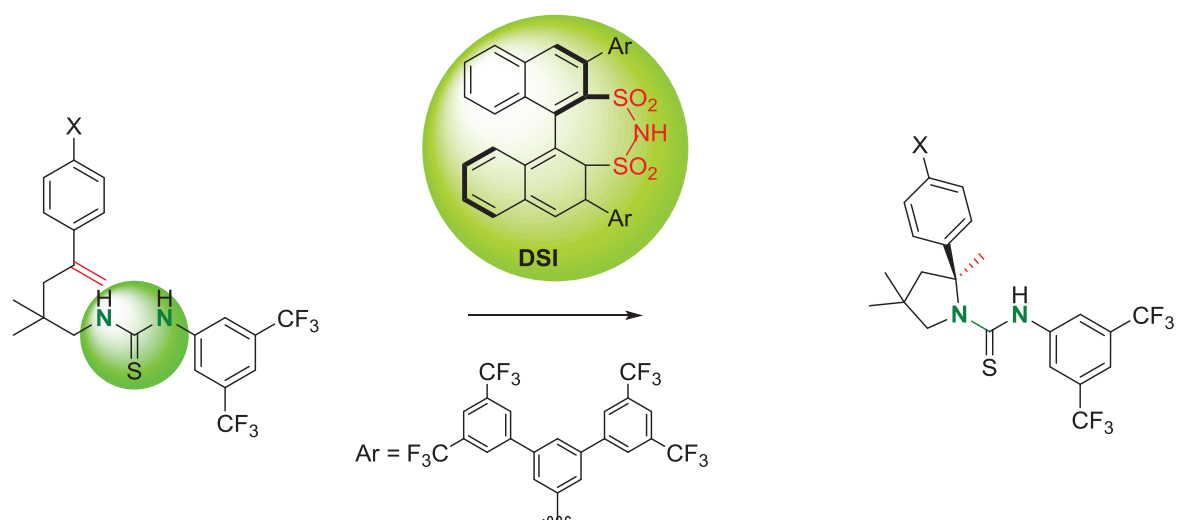
# $^1\text{H}$ NMR





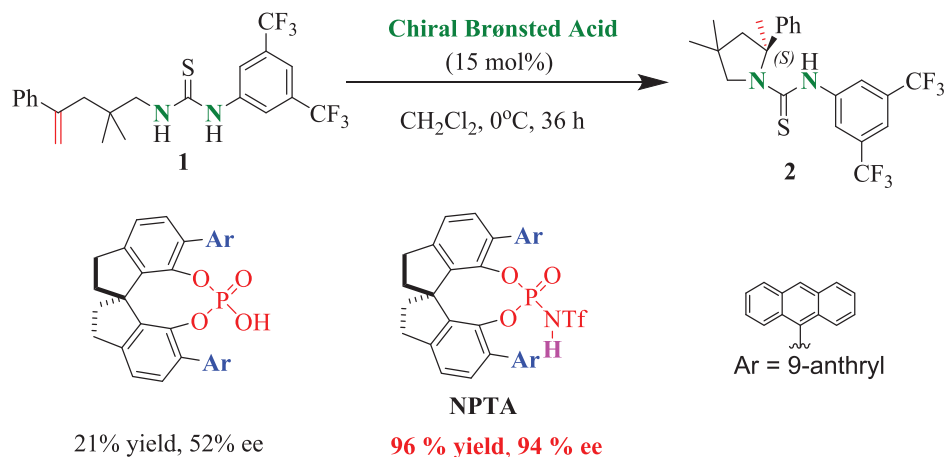
## Chapter 3.

### DSI Catalyzed Hydroamination Reaction



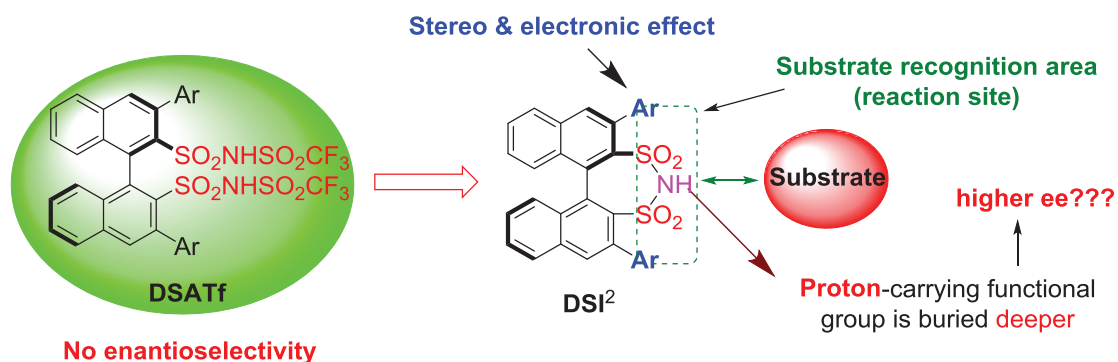
### 3-1. DSI Catalyzed Asymmetric Intramolecular Hydroamination Reaction

Liu and co-workers reported enantioselective hydroamination of alkenyl thioureas by **NPTA** (Scheme 1)<sup>24–26</sup>. The authors believe that the reaction proceeded by the activation of thiourea moiety by **NPTA**. In their research, the importance of acidity of Brønsted acid for good results was shown.



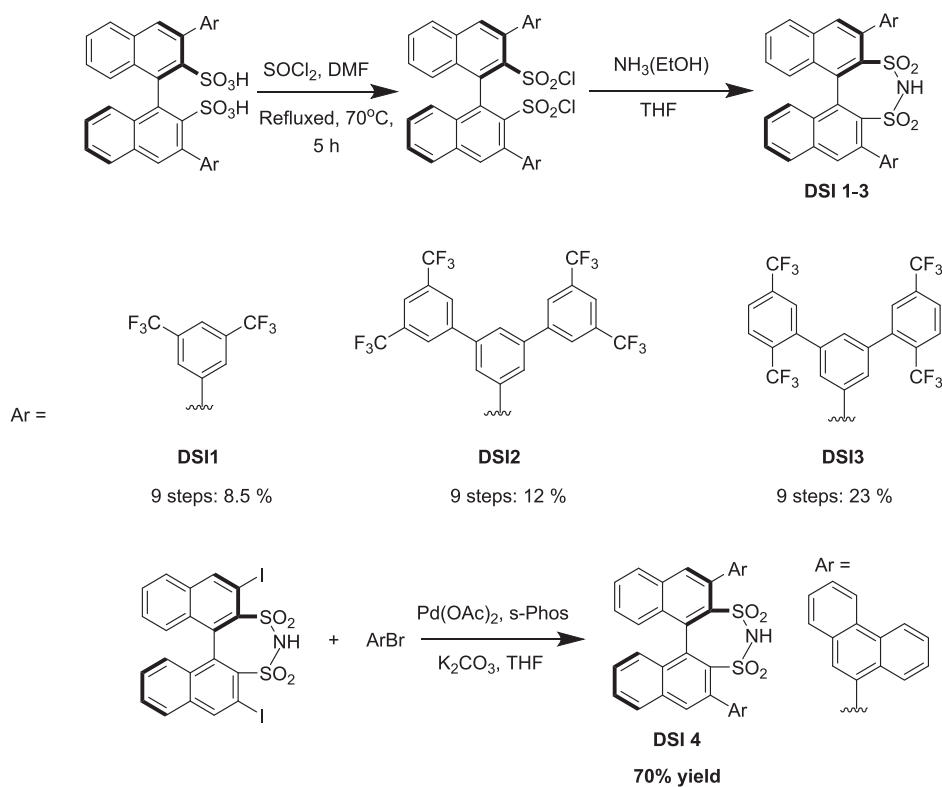
**Scheme 1.** NPTA-catalyzed hydroamination of alkenyl thiourea by Liu et al.

In chapter 2, **DSATf** was not effective for an asymmetric reaction in hydroamination of alkenyl amines. The flexibility of sulfonamide moiety is a reason for the no asymmetric induction. So a cyclic disulfonimide **DSI**<sup>28</sup> was focused as an asymmetric catalyst in order to achieve the enantioselective hydroamination. Because the proton-carrying functional group of **DSI** is buried deeper in the reaction site of the disulfonimide, leading to higher enantioselectivity by an enhanced stereochemical communication between catalyst and substrate (Scheme 2). In this study, **DSI**-catalyzed hydroamination of alkenyl thiourea was examined with an expectation of improving catalytic efficiency.



**Scheme 2.** Comparison of **DSATf** and **DSI**.

### 3-1.1. Synthesis of Brønsted Acid DSI



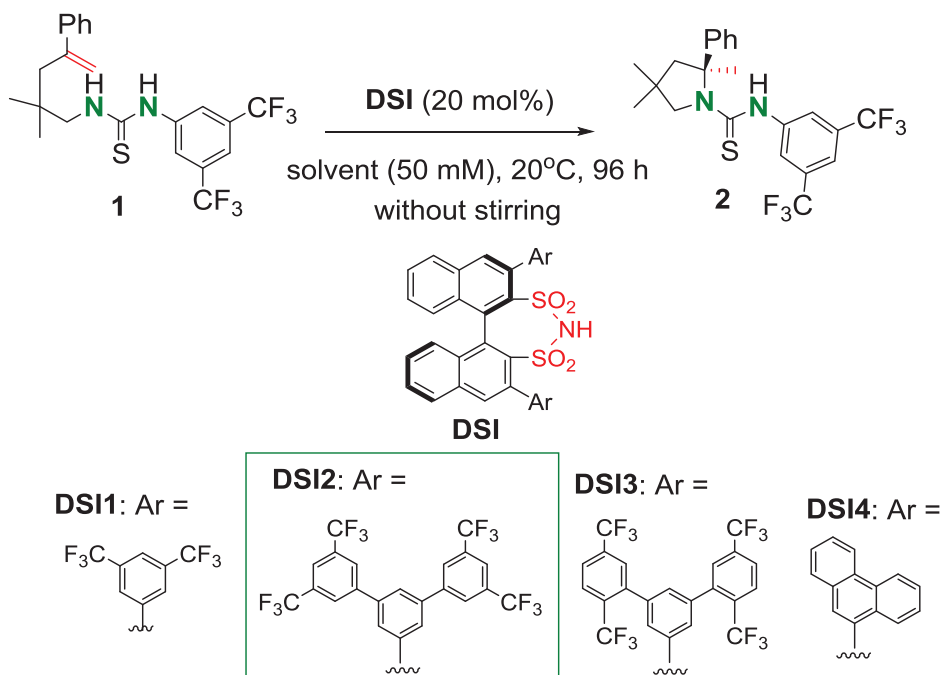
**Scheme 3.** Synthesis of DSI

DSI 1-3 were prepared according to the literature procedure from corresponding disulfonic acid (Scheme 3)<sup>28,88</sup>.

### 3-1.2. Optimization of 3,3'-substituents (Ar) of DSI and Solvent

At first, the aryl (Ar) group of **DSI** and solvents for the hydroamination reaction were screened (Table 1). The hydroamination reaction was performed using 20 mol% of **DSI** in 5-mm NMR tube without stirring at 20°C for 4 days. The screening of Ar group indicated that the size of Ar group affected the enantioselectivity, and the reaction using **DSI2** gave the hydroamination product with better enantioselectivity than **DSI1** (entries 1-2). The hydroamination using **DSI3** substituted with bulkier Ar group than **DSI2** gave hydroamination product **2** with lower chemical yield and enantioselectivity (entry 9). As other types of bulky Ar group, **DSI4** bearing 9-phenanthrenyl group was also examined (entry 10). On the hydroamination, the product was obtained with a good chemical yield (81%) but low enantioselectivity (18%ee).

The results of the solvent screening on the hydroamination catalyzed by **DSI2** showed that the use of benzene derived solvents led to better enantioselectivity than DCM (entries 2-8). When electron-donating benzene derivatives such as *o*-xylene and mesitylene were used as a solvent, the formation of a byproduct seemed to result from the reaction of the thiourea moiety on the alkenyl amine with the solvent was observed (entries 2 and 3). Toluene was selected as a suitable solvent on the basis of chemical yield and enantioselectivity (entry 4).

**Table 1.** Screening of aryl (Ar) group and solvent<sup>a</sup>.

entry	DSI	Solvent	yield (%) <sup>b</sup>	e.e.(%) <sup>c</sup>
1	<b>DSI1</b>	o-xylene	43	50
2	<b>DSI2</b>	o-xylene	31(45) <sup>d</sup>	62
3	<b>DSI2</b>	mesitylene	56(64) <sup>d</sup>	64
4	<b>DSI2</b>	toluene	68(68) <sup>d</sup>	67
5	<b>DSI2</b>	benzene	66(66) <sup>d</sup>	63
6	<b>DSI2</b>	C <sub>6</sub> H <sub>5</sub> F	75(75) <sup>d</sup>	50
7	<b>DSI2</b>	C <sub>6</sub> H <sub>5</sub> CF <sub>3</sub>	70(70) <sup>d</sup>	38
8	<b>DSI2</b>	DCM	56(56) <sup>d</sup>	25
9	<b>DSI3</b>	toluene	44	57
10	<b>DSI4</b>	toluene	81	18

<sup>a</sup>Reactions were performed with alkenyl thiourea (44 mmol), **DSI** (20 mol%) in solvent (0.88 mL) at 20°C for 96 h.

The reactions were performed in a 5-mm NMR tube under non-stirring.

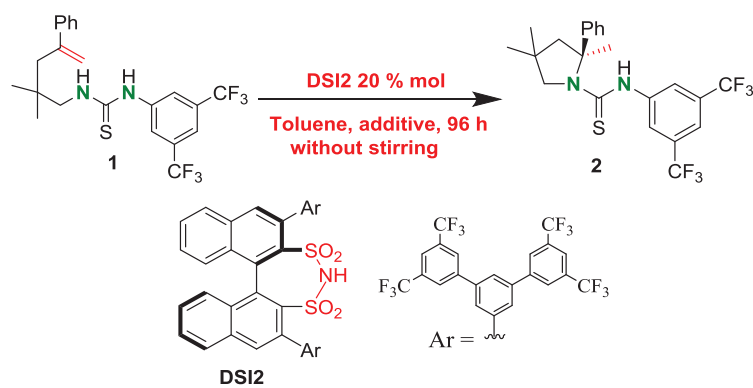
b. Isolated yield. c. Determined by chiral stationary phase HPLC analysis.

d. Conversion.

### 3-1.3. Effect of Additives

With the suitable Aryl group on **DSI** and solvent in hand, the effect of the additive was investigated (Table 2). Although the enantioselectivities were not affected by the use of the additives, the chemical yield increased in the presence of MS4Å as a drying agent (entry 2).

**Table 2.** Screening of Additives<sup>a</sup>.



Entry	Additive	yield (%) <sup>b</sup>	e.e.(%) <sup>c</sup>
-	-	68	67
1	Na <sub>2</sub> SO <sub>4</sub>	69	62
2	MS4Å	78	66
3	MS3Å	72	67

<sup>a</sup> Reactions were performed with alkenyl thiourea (44 mmol), **DSI2** (20 mol%), additive (50 mg) in toluene (0.88 mL) at 20°C for 96 h.

The reactions were performed in a 5-mm NMR tube under non-stirring.

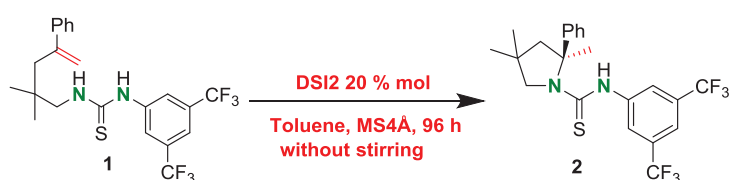
<sup>b</sup>. Isolated yield. <sup>c</sup>. Determined by chiral stationary phase HPLC.



### 3-1.4. Optimization of Reaction Conditions

The optimization of reaction conditions including concentration, temperature, and stirring of the reaction mixture on the **DSI** catalyzed hydroamination was performed (Table 3). When the reaction was conducted without stirring, the enantioselectivity of the product remained almost unchanged (70%ee) even when the reaction concentration decreased from 50 mM to 2.5 mM of concentration (entries 1, 4-7). The high concentration of 200 mM led to a decrease in enantioselectivity (entry 2). In entry 8, to improve the chemical yield, the amount of **DSI** was increased from 20 mol% to 30 mol%, the chemical yield increased and the product with the best chemical yield (86%) was obtained. At low temperature (0°C), the enantioselectivity could not be improved and the chemical yield also decreased (entries 3 and 9). When the reaction mixture was stirred, interestingly the chemical yield was dropped, but there was no change in enantioselectivity (entries 1, 3, and 8-9). These results showed that dilute (2.5 mM) concentration and non-stirring of the reaction mixture were important factors in controlling enantioselectivity.

**Table 3.** Optimization of Reaction Conditions<sup>a</sup>.



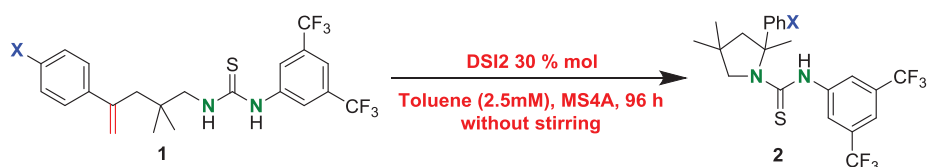
entry	Conc. (mM)	temp. (°C)	yield (%) <sup>b</sup>	e.e.(%) <sup>c</sup>
1	50	20	78(63) <sup>d</sup>	68(69) <sup>d</sup>
2	200	20	75	57
3	50	0	56(25) <sup>d</sup>	66(73) <sup>d</sup>
4 <sup>e</sup>	20	20	64	68
5 <sup>e</sup>	10	20	80	69
6 <sup>e</sup>	5	20	77	70
7 <sup>e</sup>	2.5	20	64	70
8 <sup>e,f</sup>	2.5	20	86 (27) <sup>d</sup>	70(70) <sup>d</sup>
9 <sup>e,f</sup>	2.5	0	24(0) <sup>d</sup>	75(---) <sup>d</sup>

<sup>a</sup> Unless otherwise noted, reactions were performed with alkenyl thiourea (44 mmol), **DSI2** (20 mol%), MS4A (50 mg) in toluene at 20°C for 96 h. The reactions were performed in a 5-mm NMR tube under non-stirring. <sup>b</sup> Isolated yield. <sup>c</sup> Determined by chiral stationary phase HPLC. <sup>d</sup> The reaction was performed in a test tube under stirring with a stirring bar. <sup>e</sup> The reaction was performed in a test tube under non-stirring. <sup>f</sup> 30 mol% of **DSI2** was used.

### 3-1.5. Scope of Substituent on Alkenyl Amine

Next, the substrate scope with a variety of X on the phenyl ring of alkenyl amines was examined (Table 4). The results showed that at 20°C the hydroamination of alkenyl amines with electron-donating groups (EDG) (MeO and Me) gave a product with excellent yields (98%, 84%) but low enantioselectivities (22%ee, 52%ee) (entries 1 and 3). In contrast, the hydroamination of alkenyl amines substituted with electron-withdrawing groups (EWG) (F, Cl, and Br) at room temperature gave a product with moderate enantioselectivity but low chemical yield (33%, 22%, 17%) (entries 6, 8, and 11). On the reaction of EWG-substituted alkenyl amines, increasing the reaction temperature to 50°C improved the chemical yield and interestingly, the enantioselectivity remained constant (entries 7, 9, and 12). However, when the reaction was performed at 70°C, a decline in both chemical yield and enantioselectivity was observed (entries 10 and 13). From these results, it was found that cation intermediate generated from protonation of the olefin moiety of alkenyl thiourea was affected by the substituent (X) on the phenyl ring of alkenyl thiourea.

**Table 4.** Scope of substituent on alkenyl amines<sup>a</sup>



entry	X	temp. (°C)	yield (%) <sup>b</sup>	e.e.(%) <sup>c</sup>
1	MeO	20	98	22
2	MeO	-20	28	12
3	Me	20	84	52
4	Me	0	45	49
5	H	20	86	70
6	F	20	33	70
7 <sup>e</sup>	F	50	81	70
8	Cl	20	22(0) <sup>d</sup>	73(---) <sup>d</sup>
9	Cl	50	66(58) <sup>e</sup>	64(---) <sup>e</sup>
10	Cl	70	74	61
11	Br	20	17(0) <sup>d</sup>	71(---) <sup>d</sup>
12	Br	50	43(45) <sup>e</sup>	71(67) <sup>e</sup>
13	Br	70	40	61

<sup>a</sup>Unless otherwise noted, reactions were performed with alkenyl thiourea (44 mmol), **DSI2** (30 mol%), MS4A (50 mg) in toluene (17.4 mL) at 20°C for 96 h. The reactions were performed in a test tube under non-stirring with a stirring bar.

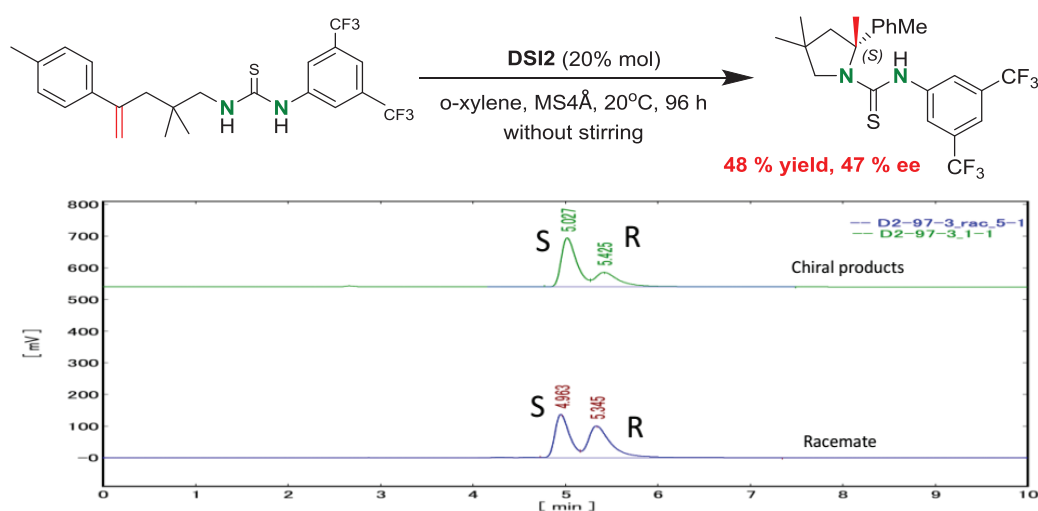
<sup>b</sup> Isolated yield. <sup>c</sup> Determined by chiral stationary phase HPLC.

<sup>d</sup> The reaction was conducted at 10 mM of the reaction mixture.

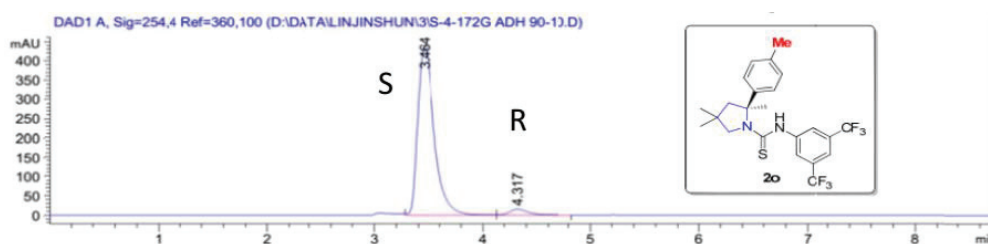
<sup>e</sup> The reaction was performed in a test tube under stirring with a stirring bar.

### 3-1.6. Determination of Absolute Configuration

To determine the absolute configuration of the hydroamination product, the retention time on the chiral stationary phase HPLC analysis of the product obtained from the hydroamination of alkenyl thiourea derivative (X= Me) was compared with the reported value<sup>26</sup> (Figures 1 and 2). The HPLC column using the same chiral selector with the literature was used as a stationary phase in this study. In the literature, the absolute stereochemistry of the product was determined by X-ray analysis. The comparison of the retention time on the HPLC analysis indicated that the **DSI**-catalyzed hydroamination afforded the hydroamination product having the same absolute stereochemistry as the literature.



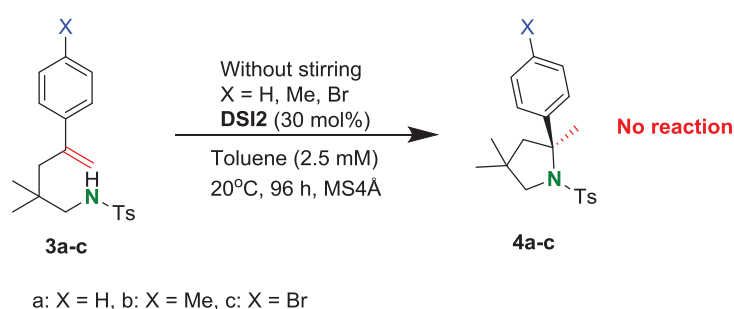
**Figure 1.** Chiral HPLC analysis using chiralpak IA column of the product on the hydroamination of alkenyl thiourea derivative (X = Me).



**Figure 2.** The reported chiral HPLC analysis using chiralpak AD-H column of the product on the hydroamination of alkenyl thiourea derivative (X = Me).

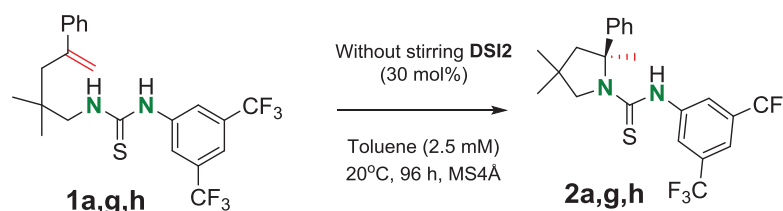
### 3-1.7. Hydroamination of *N*-Toluenesulfonyl (Ts) Alkenyl Amines and Effect of Thiourea Moiety

To understand the reaction mechanism, the hydroamination of alkenyl amines bearing several types of *N*-protective groups was examined. In the case of the hydroamination of *N*-toluenesulfonyl alkenyl amines **3a-c**, no reaction was observed (Scheme 4). It showed that the acidity of **DSI** was not sufficiently acidic to directly activate the alkene moiety of the alkenyl amines, even when the phenyl ring on the alkenyl amine was substituted with electron-donating groups (X). The results in Table 5 showed the importance of the 3,5-(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub> substituted thiourea moiety for the activation of the alkene moiety of the alkenyl thiourea and asymmetric induction in the hydroamination reaction. This suggested that the thiourea moiety acted as a hydrogen bond donor rather than as a hydrogen bond acceptor.



**Scheme 4.** Hydroamination of *N*-Toluenesulfonyl (Ts) alkenyl amines.

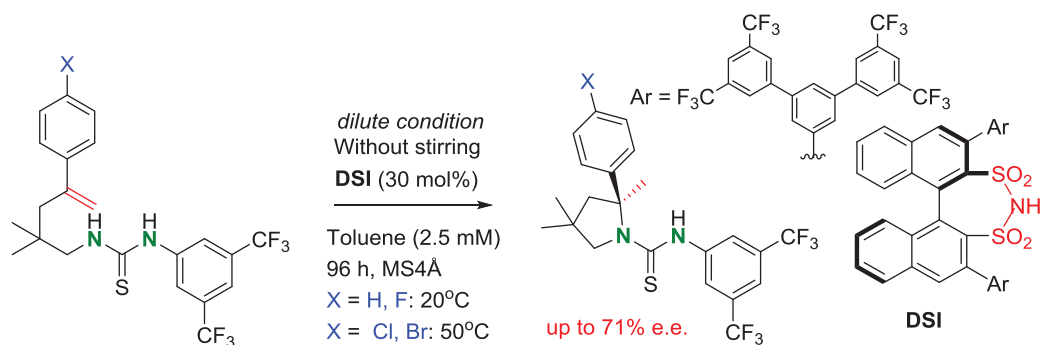
**Table 5.** Effect of Thiourea Moiety<sup>a</sup>.



entry	substituent	yield (%) <sup>b</sup>	e.e.(%) <sup>c</sup>
1	Y = S, R = CF <sub>3</sub> ( <b>a</b> )	86	70
2	Y = S, R = H ( <b>g</b> )	0	---
3	Y = O, R = CF <sub>3</sub> ( <b>h</b> )	10	58

<sup>a</sup>Reactions were performed with alkenyl amine (44 mmol), **DSI2** (30 mol%), MS4Å (50 mg) in toluene (17.4 mL) at 20°C for 96 h. The reactions were performed in a 5-mm NMR tube under non-stirring. <sup>b</sup> Isolated yield. <sup>c</sup> Determined by chiral stationary phase HPLC.

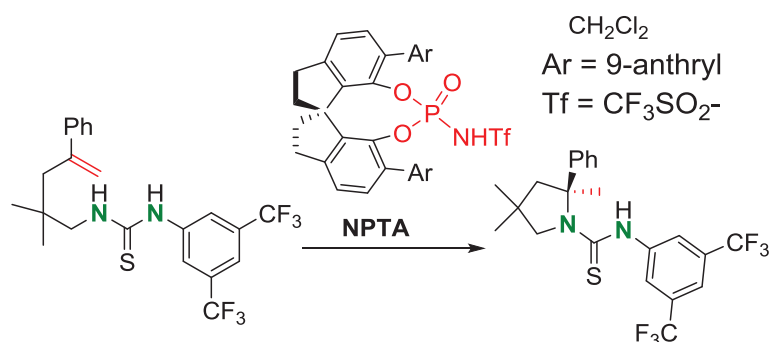
### 3-1.8. Conclusion



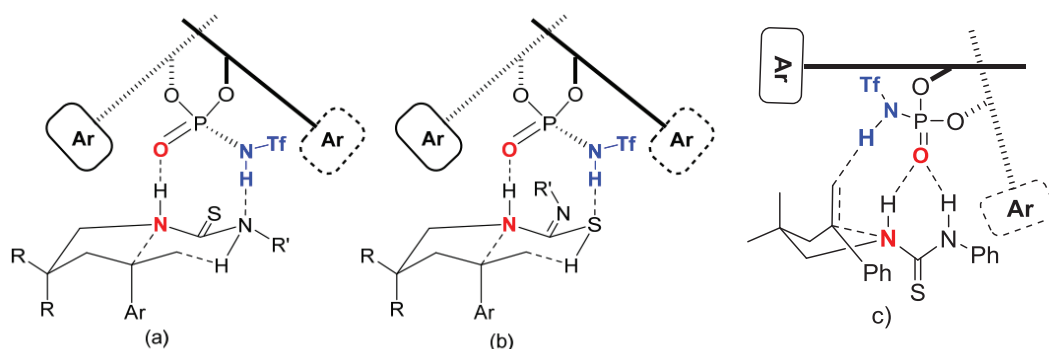
In conclusion, **DSI**-catalyzed hydroamination of alkenyl thiourea afforded the hydroamination product with good yield and moderate enantioselectivity under dilute concentration (2.5 mM) without stirring of the reaction mixture. The substituent on the phenyl ring of alkenyl thiourea significantly affected the reaction rate. When the electron-withdrawing group was introduced on the phenyl ring, the moderate enantioselectivity on the hydroamination product remained constant, although a higher reaction temperature was needed for the reaction progress.

## 3-2. Computational Mechanistic Study on DSI-catalyzed Asymmetric Intramolecular Hydroamination

### 3-2.1. Possible Mechanisms on DSI-Catalyzed Asymmetric Intramolecular Hydroamination



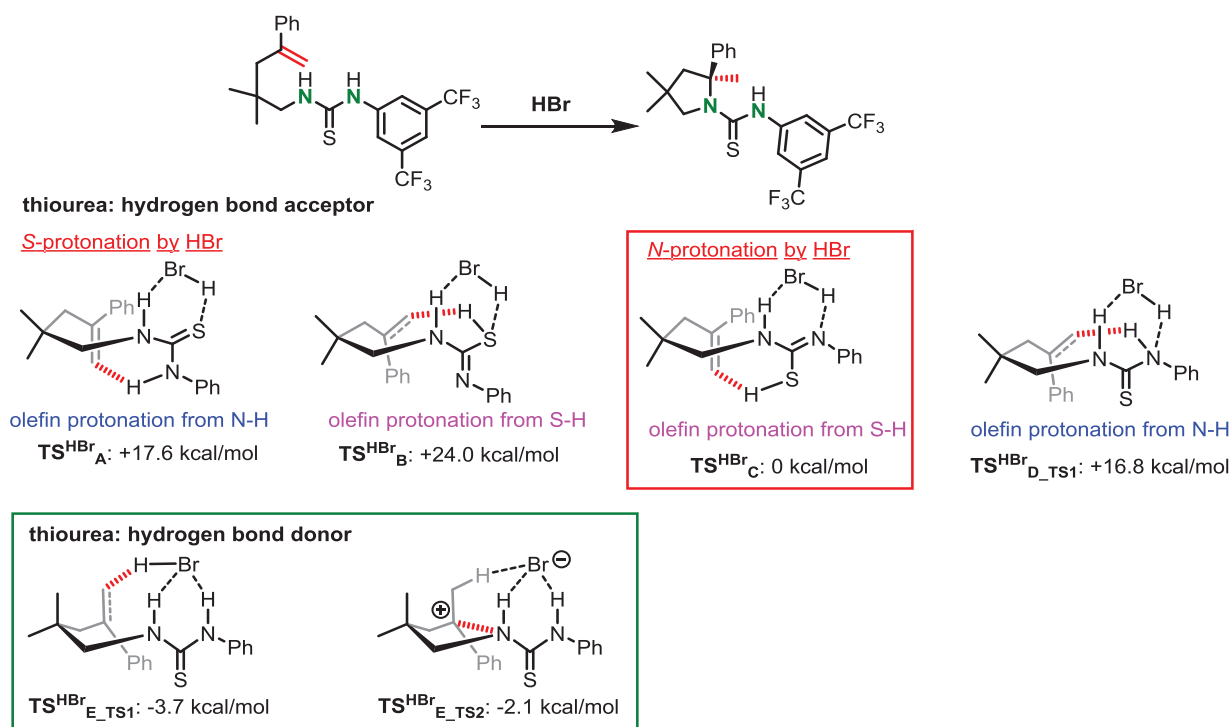
In previous work, Liu and coworkers proposed three possible mechanisms for hydroamination of alkenyl thiourea using NPTA (Figure 3)<sup>24–26</sup>. In modes **a** and **b**, the thiourea moiety of substrate acted as a hydrogen bond acceptor from NPTA. In mode **a**, the nitrogen atom of the thiourea moiety was activated by the hydrogen bonding from NPTA, then the acidity of the N-H bond of thiourea moiety was enhanced. The olefin of alkenyl thioureas was protonated by the activated N-H bond of the thiourea moiety. In mode **b**, the sulfur atom of the thiourea moiety made a hydrogen bond with NPTA, and the acidity of the S-H bond is enhanced. The olefin of alkenyl thioureas was protonated by the activated S-H bond. However, from considering the valency of the nitrogen atom in mode **a** and the nucleophilicity of the sulfur atom in mode **b**, it appeared that these structures were unsuitable. In mode **c**, thiourea acted as a dual hydrogen bond donor to activate NPTA. The olefin of alkenyl thioureas was activated by the activated NPTA. Liu and coworker only proposed three mechanisms and the detail of the mechanisms was not clear.



**Figure 3.** Possible mechanisms of NPTA catalyzed hydroamination of alkenyl thioureas: (a) and (b): Thiourea hydrogen bond acceptor, (c): thiourea hydrogen bond donor proposed by Liu et al.

### 3-2.2. DFT Calculation on HBr-Catalyzed Asymmetric Intramolecular Hydroamination

In order to determine hydrogen bond mode between alkenyl thiourea and Brønsted acid, density functional theory (DFT) calculations on the HBr-catalyzed intramolecular hydroamination of alkenyl thiourea as a simplified model were performed (Figure 4). Chiral phosphoric acids are generally known to create a chiral environment by dual hydrogen bonding with a substrate for the creation of high enantioselectivity<sup>48,89-91</sup>, and it was assumed that HBr forms dual hydrogen bonding with alkenyl thiourea during hydroamination. In this model study, five possible hydrogen bonding modes **A-E** based on differing complexation between HBr and the thiourea moiety were considered. Among these, structures **B**, **D**, and **E** were the same hydrogen bond mode mentioned by Liu et al. in their reports<sup>24,25</sup>. Additionally, modes **A** and **C** as the transition state structure were proposed as applicable structures. The exploration of transition state structures based on the possible hydrogen bond modes **A-E** was carried out at the B3LYP/6 - 31G(d,p) level of theory. The transition state structures ( $TS^{HBr}_{A-C}$  and  $TS^{HBr}_{D, E_{TS1}}$ ) having a single imaginary frequency related to the olefin protonation from the hydrogen bond donor moiety were optimized. The intrinsic reaction coordinate (IRC) calculation of the transition state structures and the estimation of the Gibbs energy profile for the reaction pathway via hydrogen bonding modes **D** and **E** indicated that the reaction pathways via hydrogen bond modes **A-D** proceeded through a concerted mechanism in which olefin protonation and C-N bond formation occur more synchronously. The reaction pathway via mode **E** involved the formation of cation intermediate. The relative Gibbs energies suggested that hydrogen bond modes **C** or **E** were favoured for the hydroamination reaction pathway.

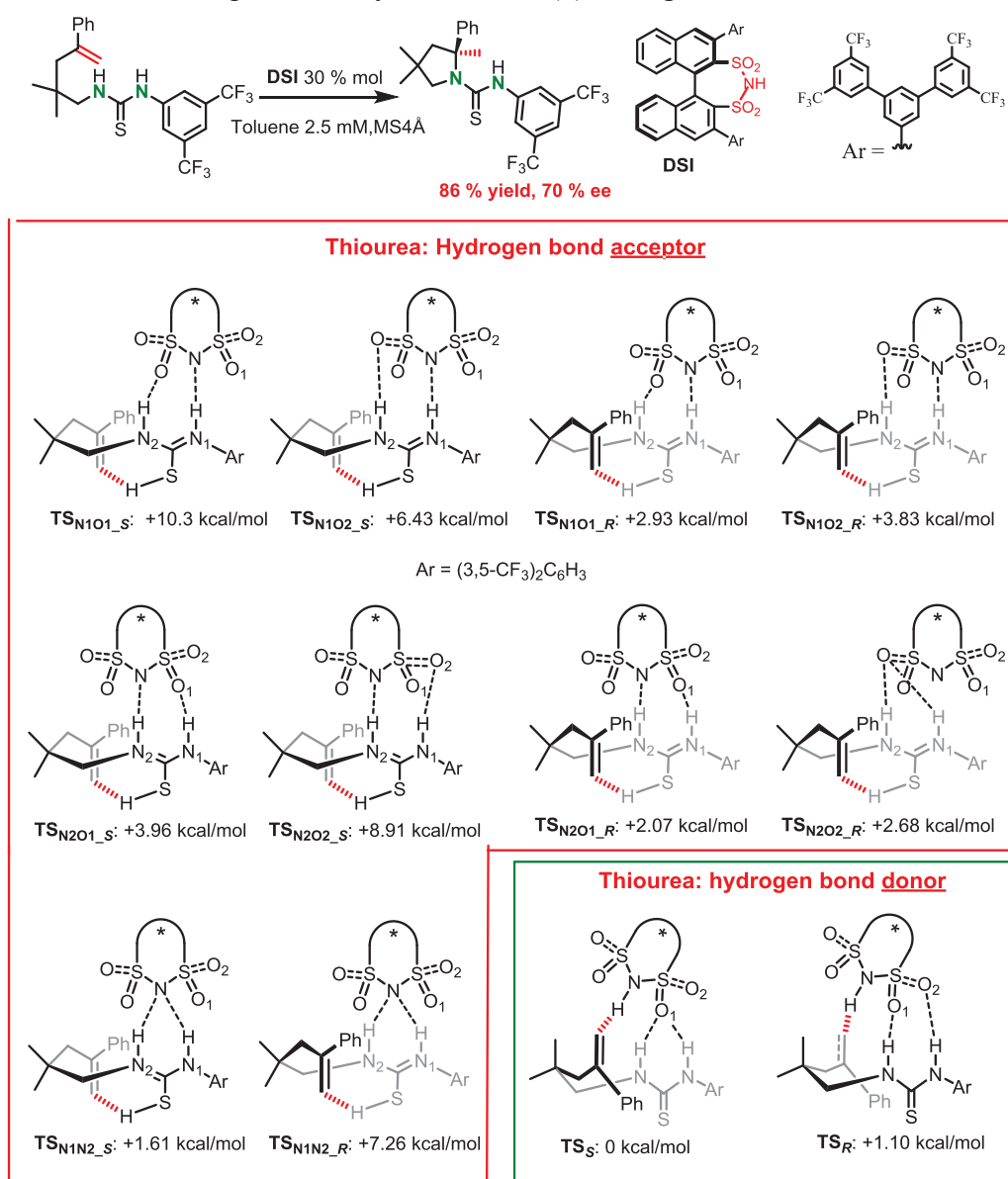


**Figure 4.** Schematic transition state structures based on possible hydrogen bonding modes in the HBr - catalyzed hydroamination of alkenyl thiourea and the relative Gibbs energies of transition states at 298.15 K at the B3LYP/6 - 31G(d,p) level of theory.



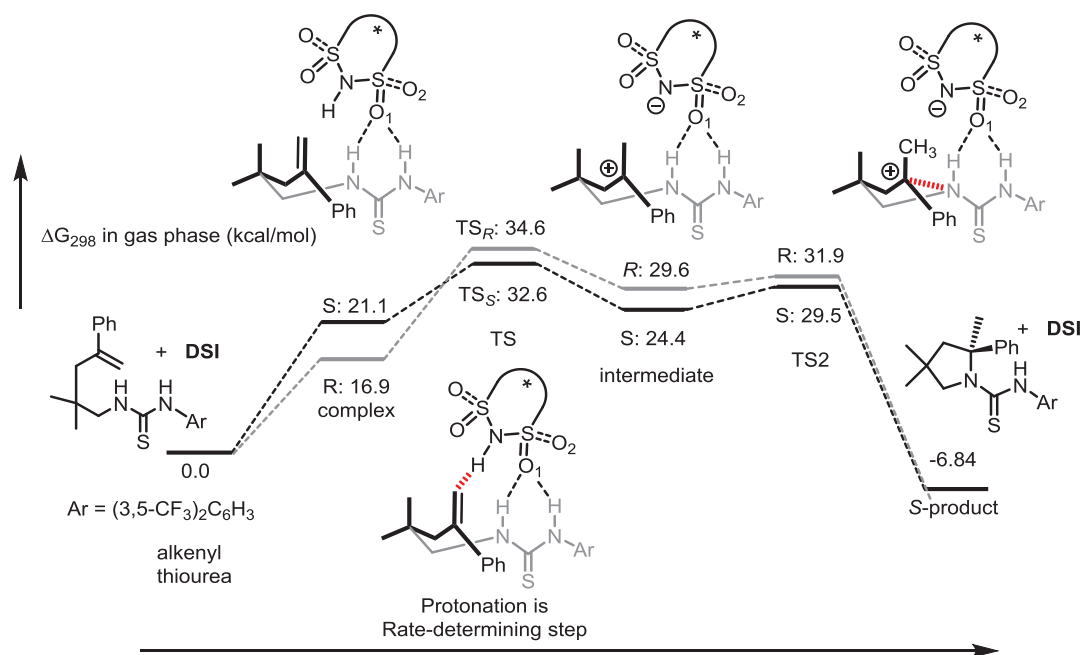
### 3-2.3. DFT Calculation on DSI-Catalyzed Asymmetric Intramolecular Hydroamination

Based on the above finding, computational studies on the enantioselective hydroamination of alkenyl thiourea catalyzed by **DSI** were conducted. There are 12 possible hydrogen bond modes for both enantiomers of the product (Figure 5). In the first 10 modes, thiourea acted as a hydrogen bonding acceptor from the catalyst and olefin was protonated by the activated S-H bond. In the last 2 cases, thiourea acted as a dual hydrogen bond donor and the olefin of alkenyl amine was protonated by the activated **DSI**. From the relative Gibbs energies of these structures, the transition state structure TSs was energetically most favoured for the hydroamination reaction pathway. The structure TSs gave the hydroamination product in agreement with the experimentally determined (*S*)-configuration.



**Figure 5.** Schematic structures and relative energies of possible transition states (298.15 K in toluene) of the **DSI**-catalyzed hydroamination at the M06-2X/6-311+G(d,p)/SMD(toluene)//ONIOM(B3LYP/6-31G(d): HF/3-21G\*) level of theory.

The Gibbs energy profile in toluene via  $TS_S$  and  $TS_R$  was examined to gain insights into the lowest energy path on the **DSI**-catalyzed hydroamination (Figure 6). The results showed that the reaction pathways proceeded through a stepwise mechanism in which the protonation of the olefin was a rate-determining step. **DSI** was activated by the dual hydrogen bond from thiourea moiety and then the olefin of alkenyl thiourea was protonated by the activated **DSI** to form the carbocation intermediate. In the second transition state, the carbocation was then attracted by the nucleophilic nitrogen to give the cyclized product.



**Figure 6.** Calculated Gibbs energy profile for the **DSI**-catalysed hydroamination of alkenyl thiourea at 298.15 K in toluene at the M06-2X/6 311+G(d, p)/SMD (toluene)//ONIOM(M06-2X/6-31G(d): HF/3-21G\*) level of theory.

### 3-2.4. Distortion/interaction analysis for TS<sub>R</sub> and TS<sub>S</sub>

Distortion/interaction analysis for TS<sub>R</sub> and TS<sub>S</sub> was conducted to understand the origin of enantioselectivity (Figure 7). The analysis indicated that the total distortion energy of substrate ( $\Delta\Delta G_{\text{dist\_sub}}^\ddagger$ ) and catalyst ( $\Delta\Delta G_{\text{dist\_cat}}^\ddagger$ ) was 0.2 kcal/mol and noncovalent interactions ( $\Delta\Delta G_{\text{int}}^\ddagger$ ) between substrate and catalyst predominantly contributed to the enantioselectivity. The visualization of noncovalent interactions of TS<sub>R</sub> and TS<sub>S</sub> using IMGPlot showed the presence of the dual hydrogen bond and other noncovalent interactions between DSI and alkenyl thiourea (Figure 8).

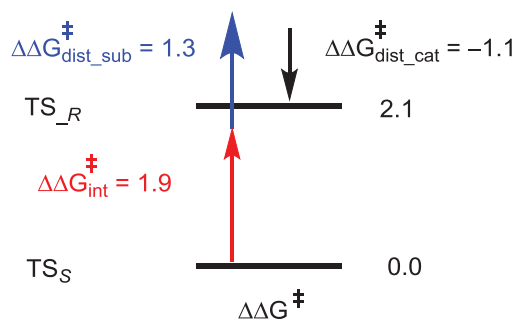


Figure 7. Distortion/interaction analysis for TS<sub>R</sub> and TS<sub>S</sub>.

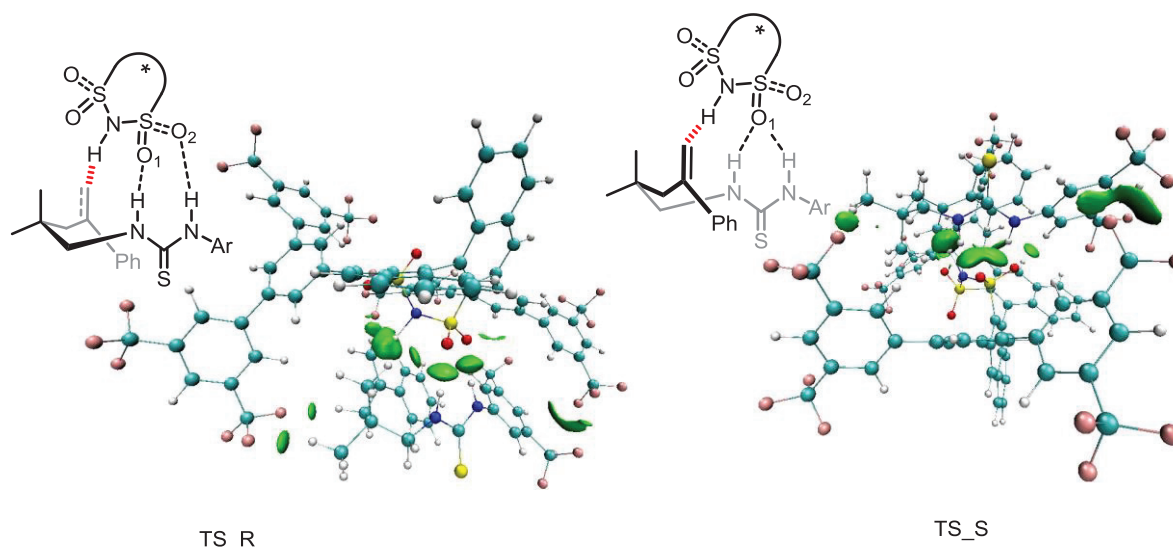
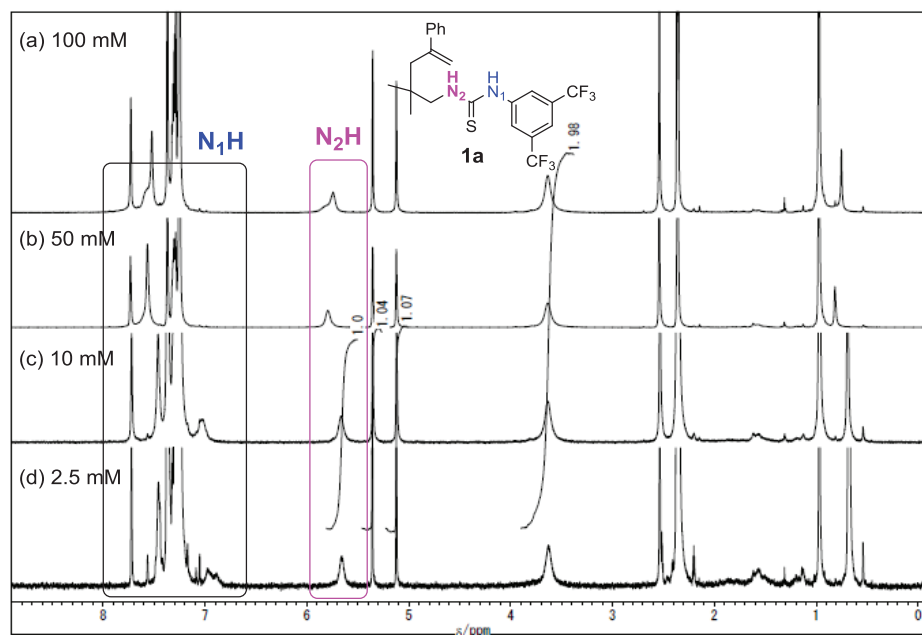


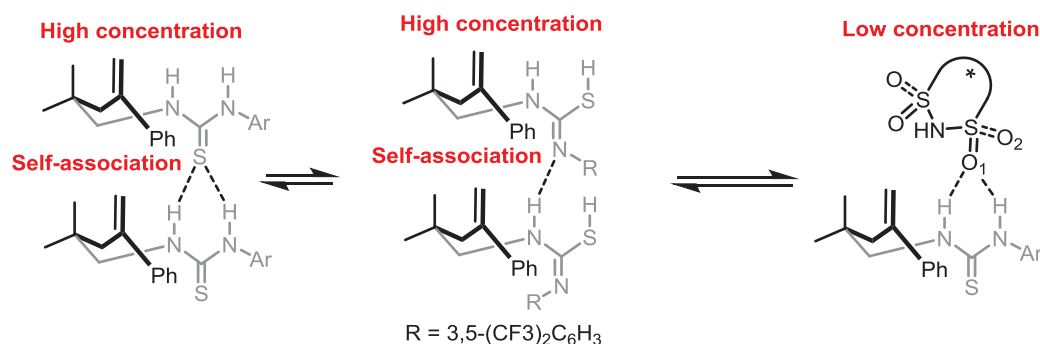
Figure 8. Nonbonding interactions visualized by IGMPLOT.

### 3-2.5. Concentration Effect on Alkenyl Thiourea

To support the calculational result, the  $^1\text{H-NMR}$  measurement of alkenyl amines from high concentration 100 mM to low concentration 2.5 mM was carried out (Figure 9). The spectra showed that the peak of the  $\text{N}_1\text{-H}$  moiety appeared at a higher magnetic field in a high concentration. The phenomena suggested that the self-association of thiourea was possible at high concentration<sup>92,93</sup>. In the diluted condition of the **DSI**-catalyzed hydroamination, the self-association of alkenyl thiourea can be avoided and **DSI** can be activated by a dual hydrogen bond from the thiourea moiety.

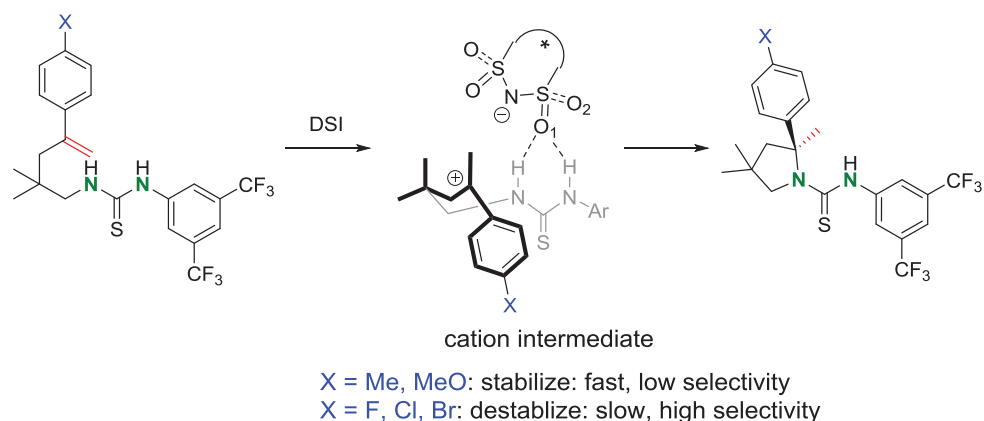


**Figure 9.**  $^1\text{H}$  NMR spectra of alkenyl thiourea **1a** in toluene- $d_8$  at  $25^\circ\text{C}$ , (a) 100 mM, (b) 50 mM, (c) 10 mM, and (d) 2.5 mM.



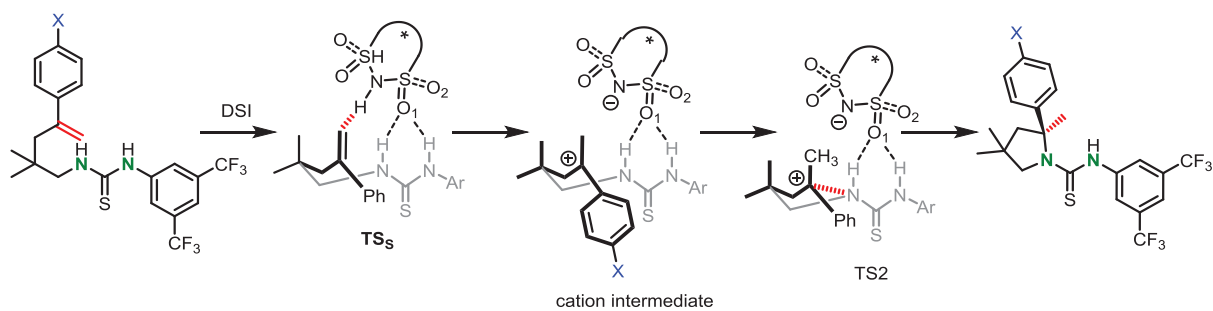
**Scheme 5.** Concentration effect on the **DSI**-catalyzed hydroamination.

### 3-2.6. Substituent Effect



The substituent effect of the phenyl ring of alkenyl thiourea on the **DSI**-catalyzed hydroamination is able to be explained by the stepwise mechanism. The electron-donating groups (MeO, Me) on the phenyl ring stabilized the cation intermediate and the reaction rate was accelerated. As a result, the asymmetric induction was not generated. In the case of the electron-withdrawing group (F, Cl, Br), although the intermediate was destabilized by the substituents, the hydroamination product was obtained with moderate enantioselectivity.

### 3-2.7. Conclusion



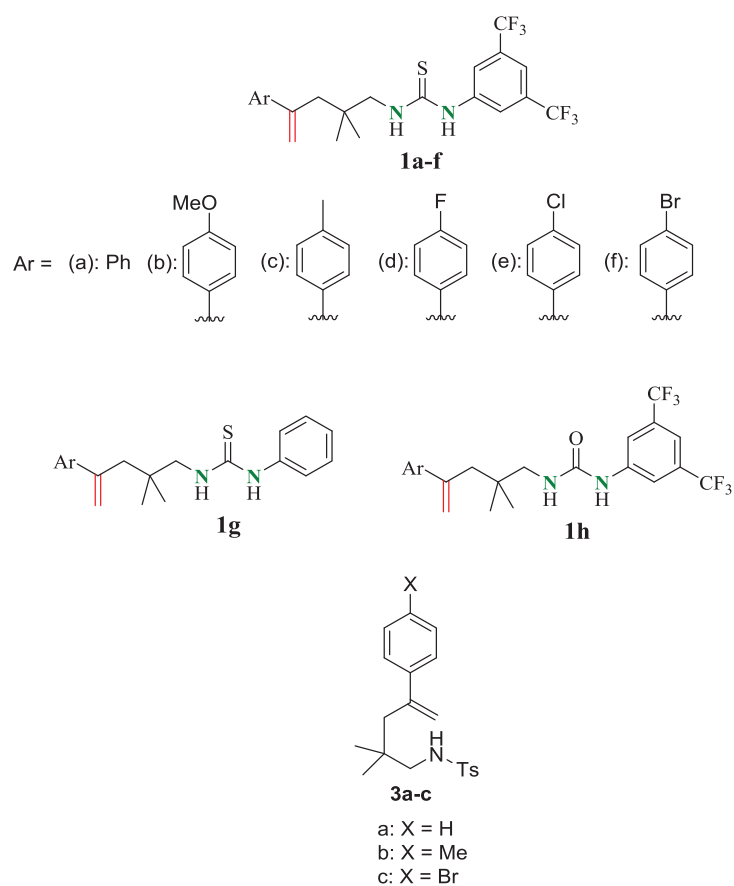
The mechanism for the intramolecular hydroamination of alkenyl thioureas catalyzed by **DSI** was investigated. The DFT calculations on the reaction supported the stepwise mechanism, in which thiourea moiety of alkenyl amines acted as a dual hydrogen bond donor to **DSI**.

### 3-2.8. Experimental Section

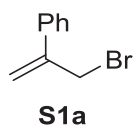
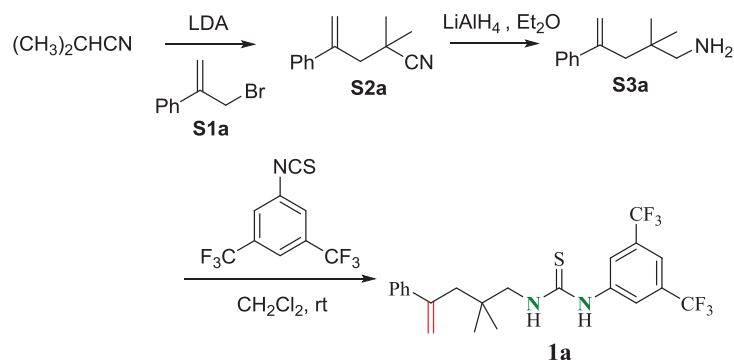
#### General experiment

Internal references for  $^1\text{H}$  NMR spectra were 0.0 ppm ( $\text{Me}_4\text{Si}$ ) for  $\text{CDCl}_3$ ,  $(\text{CD}_3)_2\text{CO}$  (2.05 ppm) and  $\text{MeOD}$  (3.31 ppm and 4.78 ppm). Chemical shifts for  $^{13}\text{C}$  NMR spectra were referenced to  $\text{CDCl}_3$  (77.0 ppm). High resolution mass spectral (HRMS) data were recorded with an LTQ Orbitrap trap mass spectrometer using the electrospray ionization (ESI) method. The enantiomeric excess (ee) of the products was determined by high performance liquid chromatography (HPLC) measured on a digital polarimeter with a 0.1 dm cell at room temperature. All reactions were monitored by analytical thin-layer chromatography (TLC), which was visualized by ultraviolet light (254 nm).

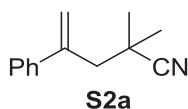
#### 3-2.8.1. Preparation of alkenyl amine



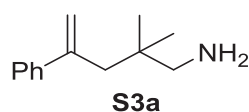
## Preparation of 1a



Alpha-methylstyrene (7 ml, 53.9 mmol) and TsOH.H<sub>2</sub>O (0.15 g, 0.77 mmol) were mixed in THF (140 ml). The mixture was refluxed at 100°C for 4 h. The reaction mixture was extracted with EtOAc, washed with 1N HCl and water, and dried over Na<sub>2</sub>SO<sub>4</sub>. The crude product was purified by hexane to give the desired product **S1a** (6.53 g, 62% yield). *R*<sub>f</sub> = 0.8 (hexane), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.53 – 7.29 (m, 5H), 5.56 (s, 1H), 5.49 (s, 1H), 4.39 (s, 2H).



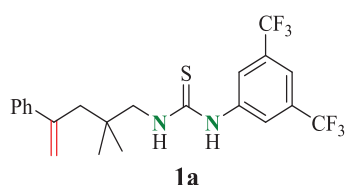
Isobutyronitrile (3.8 ml, 42 mmol) was added to a solution of LDA [generated in situ from *n*-BuLi (31 ml, 81 mmol) (in hexane, 2.6 M) and diisopropylamine (8.4 ml, 60 mmol) in THF (115 ml)] at -78°C and stirred for 45 min. To the resulting solution was added (3-bromoprop-1-en-2-yl)benzene **S1a** (6.53 g, 33 mmol) at -78°C. The solution was warmed up to room temperature overnight with stirring. The mixture was extracted with DCM, washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The crude product was purified by silica gel column chromatography (Hexane: EtOAc = 20:1) to give the desired product **S2a** as a liquid (6 g, 98 % yield). *R*<sub>f</sub> = 0.3 (hexane: EtOAc = 5: 1), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.42 – 7.26 (m, 5H), 5.44 (d, *J* = 1.2 Hz, 1H), 5.29 (d, *J* = 1.0 Hz, 1H), 2.77 (d, *J* = 0.8 Hz, 2H), 1.26 (s, 6H).



LiAlH<sub>4</sub> (2.5 g, 66 mmol) was added slowly to Et<sub>2</sub>O (75 ml) at 0°C and then added 2,2-dimethyl-

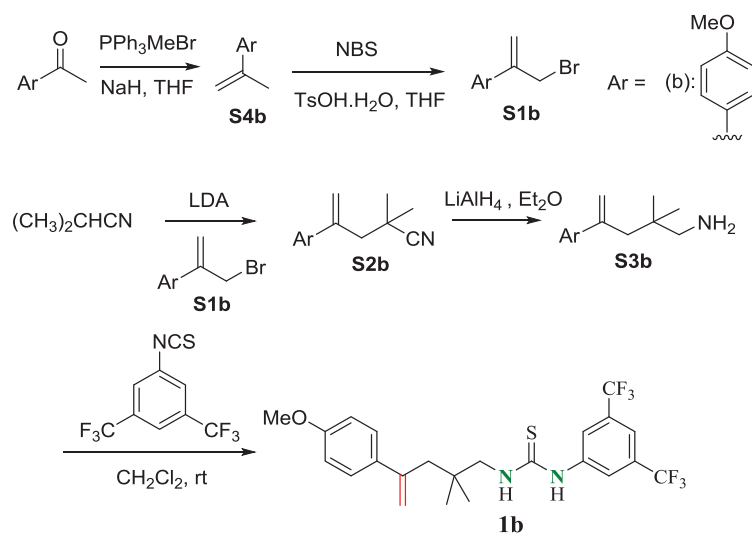


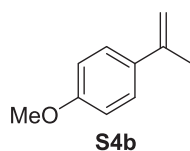
4-phenylpent-4-enitrile **S2a** (6 g, 32 mmol) slowly to this mixture at 0°C. This mixture was stirred at room temperature overnight. The reaction mixture was quenched by adding dropwise KOH until the grey solid changed completely to the white solid, and then was filtrated. The filtrate was evaporated and 2,2-dimethyl-4-phenylpent-4-en-1-amine **S3a** was used for the next step (6.1 g, 100 %).  $R_f = 0.3$  (hexane: EtOAc = 5: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.42 – 7.24 (m, 5H), 5.24 (d,  $J = 2.0$  Hz, 1H), 5.10 – 4.98 (m, 1H), 2.48 (d,  $J = 0.5$  Hz, 2H), 2.34 (d,  $J = 9.2$  Hz, 2H), 0.74 (s, 6H).



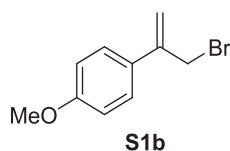
To a stirred solution 2,2-dimethyl-4-phenylpent-4-en-1-amine **S3a** (0.94 g, 5 mmol) in  $\text{CH}_2\text{Cl}_2$  (37 ml) was added aryl isothiocyanates (1 ml, 5 mmol) at room temperature. The reaction mixture was stirred overnight. The reaction mixture was concentrated to obtain the solid. The solid was washed with hexane and filtrated. The filtrate was concentrated to give the desired product **1a**<sup>7</sup> (2 g, 88 % yield).  $R_f = 0.58$  (hexane: EtOAc = 5: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.73 (d,  $J = 19.2$  Hz, 3H), 7.21 (s, 5H), 5.89 (s, 1H), 5.04 (s, 1H), 3.37 (s, 2H), 2.51 (s, 2H), 0.92 (s, 6H).

### Preparation of 1b

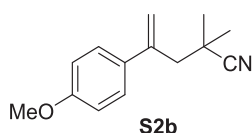




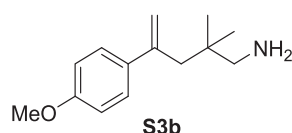
**S4b** was prepared according to the preparation of **S4a**. 74% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.46 – 7.31 (m, 2H), 6.93 – 6.77 (m, 2H), 5.28 (dd,  $J = 1.5, 0.7$  Hz, 1H), 5.05 – 4.89 (m, 1H), 3.82 (s, 3H), 2.13 (dd,  $J = 1.4, 0.8$  Hz, 3H).



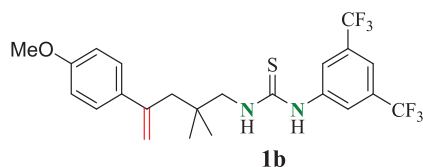
**S1b** was prepared according to the preparation of **S1a**. 50% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.47-7.41 (m, 2H), 6.93-6.88 (m, 2H), 5.48 (s, 1H), 5.40 (m, 1H), 4.37 (m, 2H), 3.83 (m, 3H).



**S2b** was prepared according to the preparation of **S2a**. 76% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 – 7.30 (m, 2H), 6.89 – 6.84 (m, 2H), 5.37 (d,  $J = 1.3$  Hz, 1H), 5.19 (d,  $J = 1.0$  Hz, 1H), 3.81 (s, 3H), 2.74 (d,  $J = 0.6$  Hz, 2H), 1.26 (s, 6H).



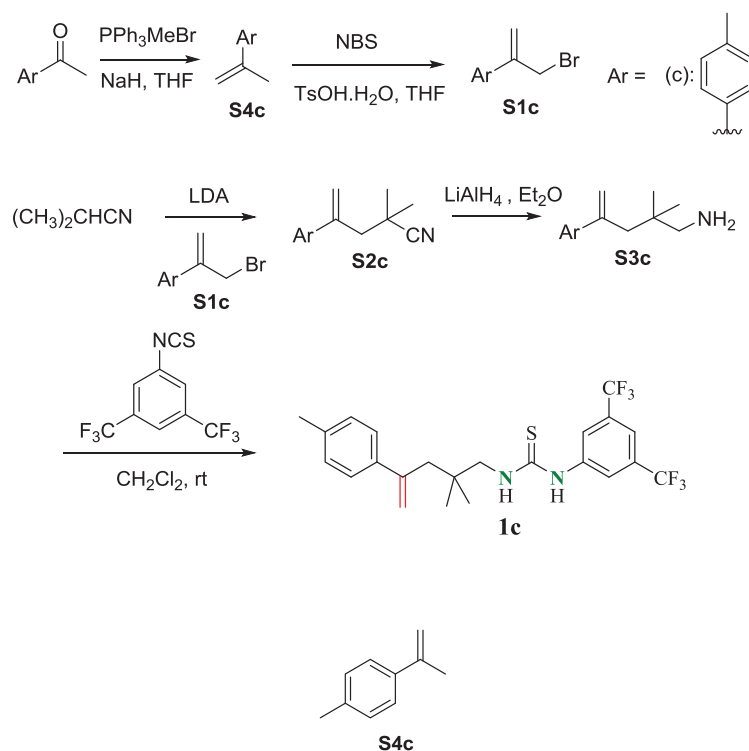
**S3b** was prepared according to the preparation of **S3a**. 100% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.30 (d,  $J = 8.8$  Hz, 2H), 6.83 (d,  $J = 8.8$  Hz, 2H), 5.17 (d,  $J = 2.0$  Hz, 1H), 5.00 – 4.90 (m, 1H), 3.78 (s, 3H), 2.43 (s, 2H), 2.33 (s, 2H), 0.74 (s, 6H).



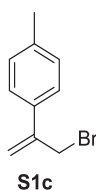
**1b** was prepared according to the preparation of **3a**. 50% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.74 (d,  $J = 9.3$  Hz, 3H), 7.16 (d,  $J = 7.7$  Hz, 2H), 6.74 (d,  $J = 8.4$  Hz, 2H), 5.95 (s, 1H), 5.15 (s, 1H), 4.95 (s, 1H), 3.75 (s, 3H), 3.38 (s, 2H), 2.47 (s, 2H), 0.92 (s, 6H).  $^{13}\text{C NMR}$  (101 MHz,

CDCl<sub>3</sub>)  $\delta$  180.79, 159.13, 145.29, 138.98, 135.07, 133.25 (q,  $J = 34.9$  Hz) (x2), 127.33 (x2), 123.86, 122.79 (q,  $J = 273.0$  Hz) (x2), 119.26, 116.18, 113.80 (x2), 55.16, 54.89, 45.54, 35.76, 26.16 (x2). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>)  $\delta$  -63.01 (s). HRMS (ESI+) calcd for C<sub>23</sub>H<sub>24</sub>ON<sub>2</sub>F<sub>6</sub>NaS [M+Na]<sup>+</sup> 513.14057, found 513.14112.

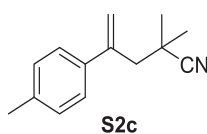
### Preparation of 1c



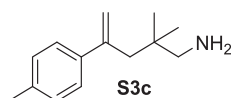
NaH (2 g, 0.083 mol) was added to the mixture of Ph<sub>3</sub>PCH<sub>3</sub>Br in THF at room temperature and stirred for 1 hour. Then the solution of 4'-methylacetophenone (5 ml, 0.037 mol) in THF was added dropwise to the reaction mixture at 0°C and refluxed overnight at 66°C. The reaction mixture was diluted with hexane and filtrated with silica gel. The filtrate was concentrated and purified by column chromatography (hexane) to give the product **S4c** (2.8 g, 64% yield). R<sub>f</sub> = 0.55 (hexane), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 (d,  $J = 8.2$  Hz, 2H), 7.14 (d,  $J = 7.9$  Hz, 2H), 5.33 (s, 1H), 5.07 – 4.98 (m, 1H), 2.34 (s, 3H), 2.14 (s, 3H).



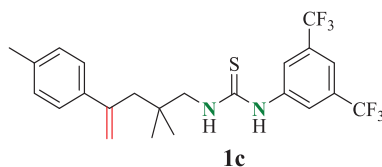
**S1c** was prepared according to the preparation of **S1a**. 59% yield. R<sub>f</sub> = 0.55 (hexane), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.46 – 7.32 (m, 2H), 7.18 (d,  $J = 7.9$  Hz, 2H), 5.52 (s, 1H), 5.44 (s, 1H), 4.38 (s, 2H), 2.36 (s, 3H).



**S2c** was prepared according to the preparation of **S2a**. 62% yield.  $R_f = 0.59$  (hexane: EtOAc= 5: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.28 (d,  $J=8.2\text{Hz}$ , 2H), 7.14 (d,  $J= 8.3\text{ Hz}$ , 2H), 5.41 (d,  $J= 1.1\text{ Hz}$ , 1H), 2.75 (s, 2H), 1.26 (s, 6H).

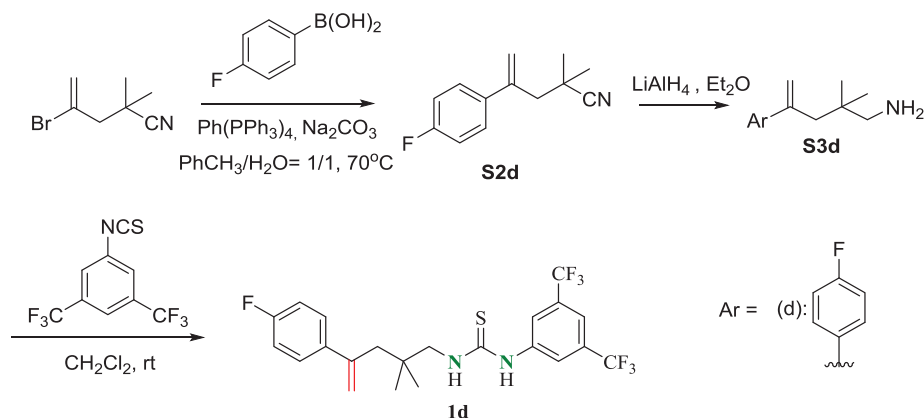


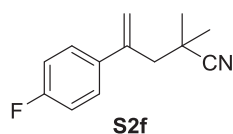
**S3c** was prepared according to the preparation of **S3a**. 100% yield.  $R_f = 0.28$  ( $\text{CHCl}_3$ : MeOH= 5: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.27 (d,  $J= 1.6\text{ Hz}$ , 2H), 7.11 (d,  $J= 8.0\text{ Hz}$ , 2H), 5.21 (d,  $J= 2.0\text{ Hz}$ , 1H), 4.99 (d,  $J= 1.9\text{ Hz}$ , 1H), 2.45 (s, 2H), 2.34 (d,  $J= 4.6\text{ Hz}$ , 4H), 0.75 (s, 6H).



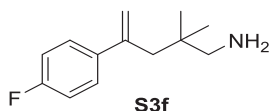
**1c**<sup>26</sup> was prepared according to the preparation of **1a**. 95% yield.  $R_f = 0.14$  (hexane: EtOAc= 10: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.74 (d,  $J= 17.9\text{ Hz}$ , 3H), 7.10 (d,  $J= 7.5\text{ Hz}$ , 2H), 6.98 (d,  $J= 7.6\text{ Hz}$ , 2H), 5.88 (s, 1H), 5.19 (s, 1H), 4.99 (s, 1H), 3.36 (s, 2H), 2.48 (s, 2H), 2.26 (s, 3H), 0.92 (s, 6H).

### Preparation of 1d

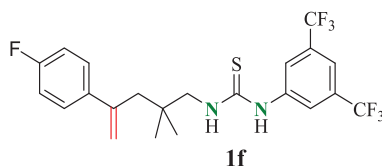




4-fluorophenylboronic acid (0.77 g, 5.5 mmol), sodium carbonate (1.78 g, 16.8 mmol) and tetrakis(triphenylphosphine)palladium(0) (0.24 g, 0.21 mmol) were added to 4-bromo-2,2-dimethylpent-4-enitrile (0.79 g, 4.2 mmol) in toluene/water (1/1) (12 ml) and then this mixture was degassed and refluxed at 70°C for 18 h under nitrogen atmosphere. The reaction mixture was diluted by ethyl acetate and washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The crude product was purified by column chromatography (Hexane: EtOAc= 40: 1) to give the desired product **S2f** as a colorless liquid (0.51 g, 60 % yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.40 – 7.31 (m, 2H), 7.08 – 6.97 (m, 1H), 5.39 (d, *J* = 0.9 Hz, 1H), 5.26 (s, 2H), 2.73 (d, *J* = 0.5 Hz, 2H), 1.26 (s, 6H).

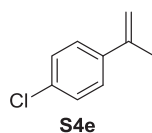
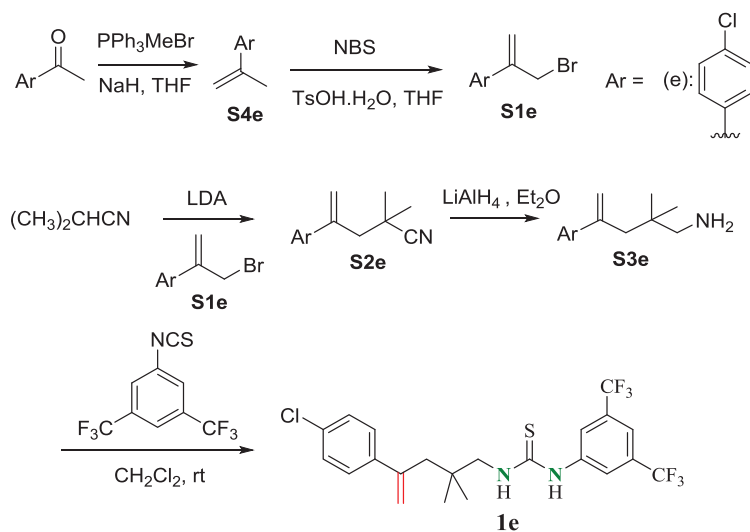


**S3f** was prepared according to the preparation of **S3a**. 99% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.37 – 7.30 (m, 2H), 7.08 – 6.90 (m, 2H), 5.20 (d, *J* = 1.8 Hz, 1H), 5.03 (d, *J* = 0.8 Hz, 1H), 2.45 (s, 2H), 2.35 (s, 2H), 0.73 (s, 6H).

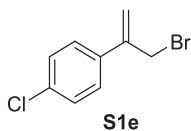


**1b**<sup>26</sup> was prepared according to the preparation of **1a**. 76% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.74 (d, *J* = 15.5 Hz, 3H), 7.25 – 7.18 (m, 2H), 6.92 (t, *J* = 8.5 Hz, 2H), 5.99 (s, 1H), 5.19 (s, 1H), 5.01 (s, 1H), 3.42 (s, 2H), 2.48 (s, 2H), 0.87 (s, 6H).

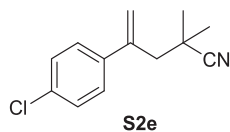
## Preparation of 1e



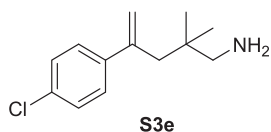
**S4e** was prepared according to the preparation of **S4a**. 55% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.45-7.30 (m, 4H),  $\delta$  5.35 (s, 1H), 5.09 (m, 1H), 2.12 (m, 3H).



**S1e** was prepared according to the preparation of **S1a**. 65% yield.  $^1\text{H NMR}$  (400MHz,  $\text{CDCl}_3$ )  $\delta$  7.44-7.30 (m, 4H), 5.53-5.50 (m, 2H), 4.34 (m, 2H).

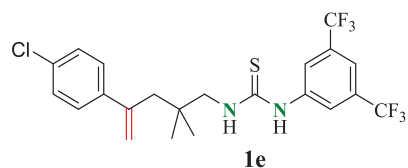


**S2e** was prepared according to the preparation of **S2a**. 46% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.34-7.29 (m, 4H), 5.43 (d, 1H), 5.30 (d, 1H), 2.73 (s, 2H), 1.26 (s, 6H).



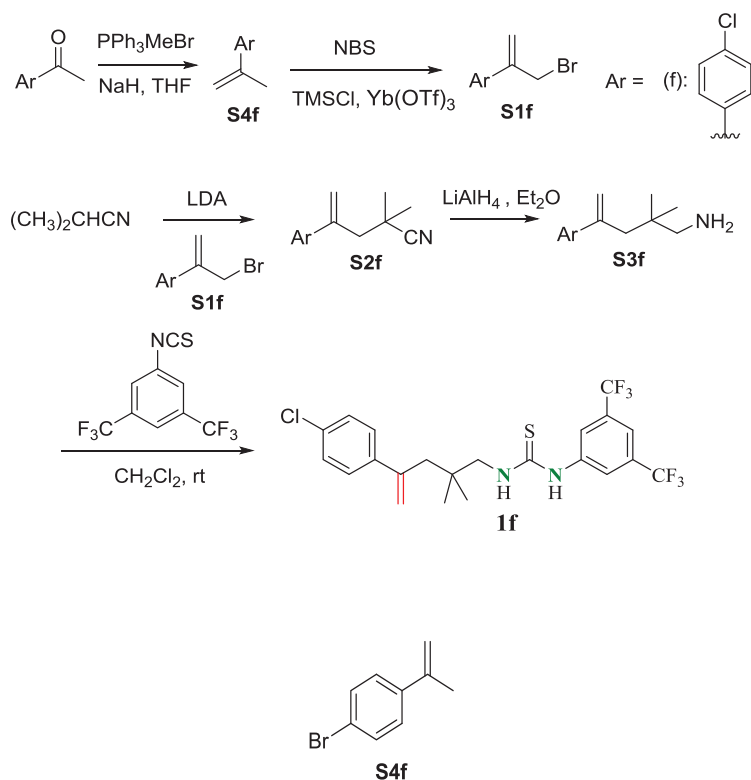
**S3e** was prepared according to the preparation of **S3a**. 97% yield.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.34 – 7.27 (m, 4H), 5.23 (d,  $J = 1.8$  Hz, 1H), 5.06 (dd,  $J = 1.9, 1.0$ Hz, 1H), 2.45 (d,  $J = 0.9$

Hz, 2H), 2.35 (s, 2H), 0.73 (s, 6H).

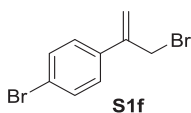


**1e** was prepared according to the preparation of **1a**. 75% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.74 (d,  $J = 14.7$  Hz, 3H), 7.20 (s, 4H), 5.96 (s, 1H), 5.23 (s, 1H), 5.05 (s, 1H), 3.41 (s, 2H), 2.49 (s, 2H), 0.87 (s, 6H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  180.83, 144.79, 141.24, 138.85, 133.34, 133.13(q,  $J = 30.7$  Hz) (x2), 128.57(x2), 127.57(x2), 123.85(x2), 122.74 (q,  $J = 273.1$  Hz) (x2), 119.48 – 118.33, 118.10, 55.09, 45.34, 35.81, 25.92(x2).  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.02 (s). HRMS (ESI+)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{22}\text{N}_2\text{ClF}_6\text{S}$   $[\text{M}+\text{H}]^+$  495.10909 found 495.10904.

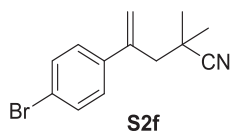
### Preparation of **1f**



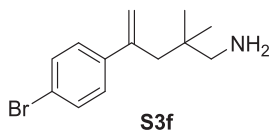
**S4f** was prepared according to the preparation of **S4a**. 84% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.47 – 7.40 (m, 2H), 7.36 – 7.28 (m, 2H), 5.39 – 5.33 (m, 1H), 5.12 – 5.07 (m, 1H), 2.13 (dd,  $J = 1.4, 0.8$  Hz, 3H).



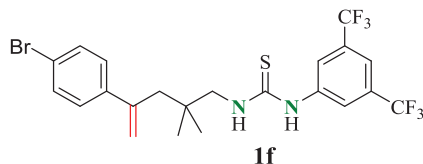
To a mixture of the 1-bromo-4-(prop-1-en-2-yl)benzene (0.98 g, 5 mmol) **S4f** and TMSCl (6.7 mg, 0.06 mmol) in CH<sub>2</sub>Cl<sub>2</sub>/THF (4:1) (15 ml) under an nitrogen atmosphere were added NBS (1 g, 5.8 mmol) and Yb(OTf)<sub>3</sub> (39 mg, 0.06 mmol) in one portion. After stirring for 1.5 h, the mixture was concentrated under reduced pressure. The resulting residue was filtered three times with hexane, and the combined filtrates were concentrated under reduced pressure. The crude product mixture was then purified by silica gel chromatography (hexane) to give the product **S1f** (0.54 g, 39%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.49 (d, *J* = 8.6 Hz, 2H), 7.35 (d, *J* = 8.6 Hz, 2H), 5.54 (s, 1H), 5.50 (s, 1H), 4.33 (s, 2H).



**S2f** was prepared according to the preparation of **S2a**. 77% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.46 (d, *J* = 8.6 Hz, 2H), 7.26 (d, *J* = 8.6 Hz, 2H), 5.44 (d, *J* = 1.0 Hz, 1H), 5.30 (d, *J* = 0.8 Hz, 1H), 2.73 (d, *J* = 0.6 Hz, 2H), 1.26 (s, 6H).



**S3f** was prepared according to the preparation of **S3a**. 91% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.45 – 7.40 (m, 2H), 7.27 – 7.22 (m, 2H), 5.24 (d, *J* = 1.7 Hz, 1H), 5.07 – 5.02 (m, 1H), 2.44 (s, 2H), 2.34 (s, 2H), 0.73 (d, *J* = 5.3 Hz, 6H).

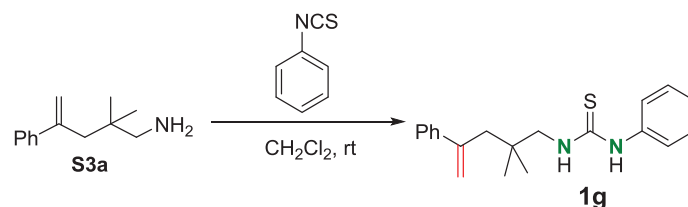


**1f** was prepared according to the preparation of **1a**. 50% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.43 (s, 1H), 7.75 (s, 3H), 7.35 (d, *J* = 8.4 Hz, 2H), 7.13 (d, *J* = 8.3 Hz, 2H), 6.05 (s, 1H), 5.23 (s, 1H), 5.04 (s, 1H), 3.42 (s, 2H), 2.47 (s, 2H), 0.85 (s, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 180.99, 144.88, 141.74, 138.69, 133.32 (q, *J* = 67.0, 33.4 Hz) (x2), 131.59 (x2), 127.92 (x2), 123.93 (x2), 122.51 (q, *J* = 128.4 Hz) (x2), 121.51, 119.65 – 119.51, 118.22, 55.12, 45.40,



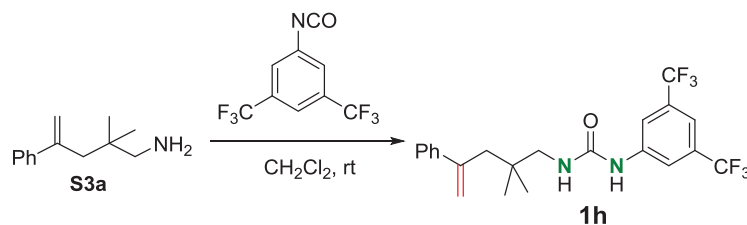
35.87, 26.03(x2).  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.00 (s). HRMS (ESI+),  $m/z$  calcd for  $\text{C}_{22}\text{H}_{22}\text{N}_2\text{BrF}_6\text{S}$   $[\text{M}+\text{H}]^+$  539.05858, found 539.05908.

### Preparation of **1g**



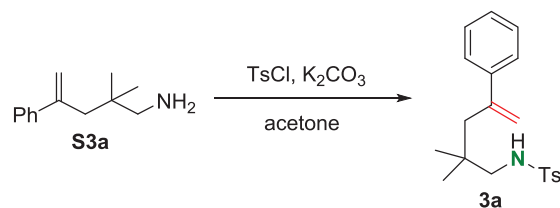
To a stirred solution amine (0.23 g, 0.0012 mol) in  $\text{CH}_2\text{Cl}_2$  (9.4 ml) was added aryl isothiocyanates (0.2 g, 0.0002 mol) at room temperature. The reaction mixture was stirred overnight. The reaction mixture was concentrated and purified by column chromatography (Hexane: EtOAc= 5: 1) give the product **1g** (0.32 g, 80%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.76 (s, 1H), 7.73 (bs, 2H), 7.20 (bs, 4H), 5.96 (s, 1H), 5.23 (s, 1H), 5.05 (s, 1H), 3.41 (bs, 2H), 2.49 (s, 2H), 0.87 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  180.8, 144.8, 141.2, 138.9, 133.3, 133.1 (q,  $J = 29.3$  Hz, x2), 128.6 (x2), 127.6 (x2), 123.8 (x2), 122.7 (q,  $J = 273.3$  Hz, x2), 119.3, 118.1, 55.1, 45.3, 35.8, 26.0 (x2).  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.0. HRMS (ESI+)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{22}\text{N}_2\text{ClF}_6\text{S}$   $[\text{M}+\text{H}]^+$  495.10909, found 495.10904.

### Preparation of **1h**



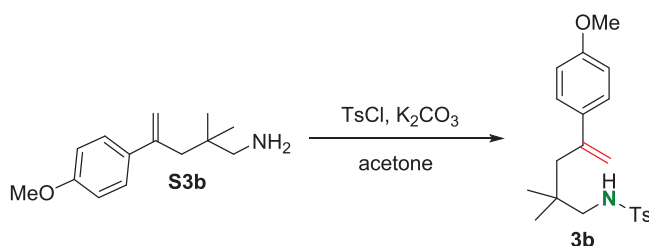
**1h**<sup>94</sup> was prepared according to the preparation of **1g**. 95% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.89 – 7.29 (m, 7H), 6.41 (d,  $J = 75.3$  Hz, 1H), 5.28 (s, 1H), 5.08 (s, 1H), 4.50 (d,  $J = 41.4$  Hz, 1H), 2.99 (d,  $J = 6.3$  Hz, 2H), 2.51 (s, 2H), 0.84 (d,  $J = 3.2$  Hz, 5H).

## Preparation of 3a



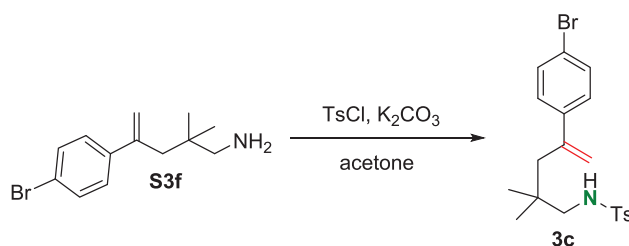
2,2-Dimethyl-4-phenylpent-4-en-1-amine (100.6 mg, 0.56 mmol) **S2a**,  $K_2CO_3$  (144 mg, 1 mmol), and 4-methylbenzenesulfonyl chloride (111 mg, 0.58 mmol) in acetone (0.53 ml) were refluxed at 60°C for 12 h, quenched with saturated  $NH_4Cl$ , extracted with EtOAc and brine, dried over  $Na_2SO_4$ , and then concentrated in vacuo. The crude products were isolated by column chromatography on silica gel using EtOAc: hexane= 5: 1 to give the product **3a**<sup>26</sup> (84 mg, 42%).  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.53 – 7.47 (m, 2H), 7.31 – 7.22 (m, 7H), 5.22 (d,  $J$  = 1.8 Hz, 1H), 5.04 – 4.99 (m, 1H), 4.13 (dd,  $J$  = 8.7, 5.6 Hz, 1H), 2.46 (dd,  $J$  = 22.9, 10.8 Hz, 7H), 0.79 (s, 6H).

## Preparation of 3b



**3b** was prepared according to the preparation of **3a**. 85% yield.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.54 – 7.37 (m, 2H), 7.26 (s, 4H), 7.11 (d,  $J$  = 7.9 Hz, 2H), 5.18 (d,  $J$  = 1.8 Hz, 1H), 5.03 – 4.87 (m, 1H), 4.08 (t,  $J$  = 7.2 Hz, 1H), 2.79 – 1.89 (m, 10H), 0.80 (s, 6H).  $^{13}C$  NMR (101 MHz,  $CDCl_3$ )  $\delta$  145.89, 143.06, 140.40, 137.02, 129.53(x2), 129.19(x2), 126.95(x2), 126.28(x2), 116.83, 52.97, 44.82, 35.18(x2), 25.65(x2), 21.52, 21.12. HRMS (ESI+)  $m/z$  calcd for  $C_{21}H_{27}O_2NNaS$   $[M+Na]^+$  380.16547, found 380.16580.

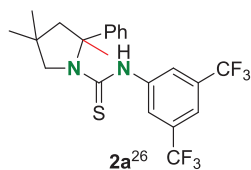
## Preparation of 3c



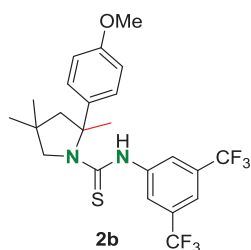
**3c** was prepared according to the preparation of **3a**. 60% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.54 (d,  $J = 8.3$  Hz, 2H), 7.44 – 7.37 (m, 2H), 7.34 – 7.24 (m, 2H), 7.21 – 7.12 (m, 2H), 5.22 (d,  $J = 1.6$  Hz, 1H), 5.05 (s, 1H), 4.14 (t,  $J = 7.2$  Hz, 1H), 2.47 (dd,  $J = 27.4, 4.9$  Hz, 6H), 0.79 (d,  $J = 5.3$  Hz, 6H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  144.96, 143.30, 142.18, 136.79(x2), 131.59(x2), 129.67(x2), 128.03(x2), 126.90(x2), 121.36, 118.22, 53.05, 44.85, 35.24, 25.52(x2), 21.54. HRMS (ESI+),  $m/z$  calcd for  $\text{C}_{20}\text{H}_{24}\text{O}_2\text{NBrNaS}$   $[\text{M}+\text{Na}]^+$  444.06033, found 444.06042.

### 3-2.8.2. Typical procedure for asymmetric hydroamination of alkenyl thioureas

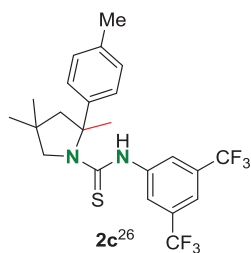
The substrate (20.2 mg, 44  $\mu\text{mol}$ ), **DSI** (12.3 mg, 8.8  $\mu\text{mol}$ ), and molecular sieves MS4Å (55.8 mg) and toluene (17.4 mL) were added to the test tube and was kept for 1 h at 20°C and then alkenyl thiourea (20.2 mg, 44  $\mu\text{mol}$ ) was added and kept for 4 days. The reaction mixture was added  $\text{NaHCO}_3$  and kept for 15 min and filtered. The filtrate was concentrated and purified by column chromatography (Hexane: EtOAc= 40: 1) to give the chiral product (17.4 mg, 86%). The ee value was determined by the chiral stationary phase HPLC of the isolated product.



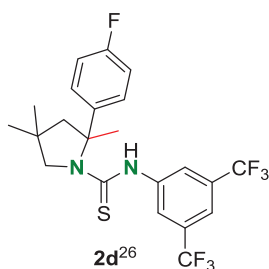
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.89 – 7.27 (m, 8H), 6.84 (s, 1H), 4.37 (d,  $J = 12.6$  Hz, 1H), 4.04 (d,  $J = 12.6$  Hz, 1H), 2.36 (dd,  $J = 33.1, 13.7$  Hz, 2H), 2.04 (s, 3H), 1.33 (s, 3H), 1.17 (s, 3H).



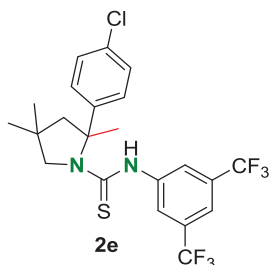
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.53 (s, 1H), 7.45 – 7.29 (m, 4H), 7.06 – 6.99 (m, 2H), 6.98 (s, 1H), 4.34 (d,  $J = 12.6$  Hz, 1H), 4.02 (d,  $J = 12.7$  Hz, 1H), 3.86 (s, 3H), 2.33 (dd,  $J = 31.1, 13.5$  Hz, 2H), 2.00 (s, 3H), 1.32 (s, 3H), 1.17 (s, 3H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  178.28, 159.73, 140.76, 136.73, 131.44 (q,  $J = 33.5$  Hz) (x2), 126.85(x2), 123.07 (q,  $J = 272.8$  Hz) (x2), 124.12-124.09(x2), 118.23-118.09, 115.11(x2), 69.00, 68.58, 61.44, 55.44, 34.99, 29.07, 28.92, 26.0.  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.04 (s). HRMS (ESI+)  $m/z$  calcd for  $\text{C}_{23}\text{H}_{25}\text{ON}_2\text{F}_6\text{S}$   $[\text{M}+\text{H}]^+$  491.15863, found 491.15918.



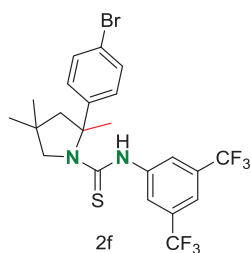
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.02 (s, 1H), 7.91 (d, *J* = 8.2 Hz, 3H), 7.79 (dd, *J* = 13.8, 5.0 Hz, 4H), 7.71 (s, 2H), 7.43 (ddd, *J* = 8.1, 4.8, 3.2 Hz, 1H), 7.32 (s, 1H), 7.31 – 7.27 (m, 2H), 4.43 (dd, *J* = 15.7, 5.7 Hz, 2H), 2.41 (s, 3H).



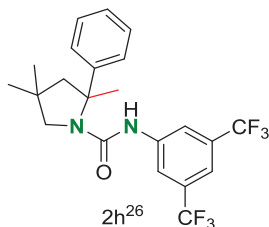
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.61 – 7.35 (m, 4H), 7.19 (t, *J* = 8.5 Hz, 3H), N-H: 6.81 (s, 1H), 4.35 (d, *J* = 12.4 Hz, 1H), 4.02 (d, *J* = 12.6 Hz, 1H), 2.32 (s, 2H), 2.04 (s, 3H), 1.31 (s, 3H), 1.14 (s, 3H).



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.57 (s, 1H), 7.47 (d, *J* = 8.6 Hz, 3H), 7.35 (d, *J* = 8.7 Hz, 2H), 6.77 (s, 1H), 4.33 (d, *J* = 12.8 Hz, 1H), 4.02 (d, *J* = 12.6 Hz, 1H), 2.32 (s, 2H), 2.04 (s, 3H), 1.30 (s, 3H), 1.13 (s, 3H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 178.68, 143.75, 140.52, 134.59, 131.66 (q, *J* = 33.6 Hz) (x2), 129.95 (x2), 126.85 (x2), 124.59-124.56 (x2), 122.99 (q, *J* = 272.8 Hz) (x2), 121.63, 118.78-118.63, 68.79 (d, *J* = 20.2 Hz), 61.57, 35.16, 28.88, 28.57, 26.73. <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -63.01 (s). HRMS (ESI+) calcd for C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>ClF<sub>6</sub>S [M+H]<sup>+</sup> 495.10909, found 495.10909.



$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 – 7.56 (m, 2H), 7.48 (s, 2H), 7.33 – 7.23 (m, 3H), 6.76 (s, 1H), 4.32 (d,  $J = 12.6$  Hz, 1H), 4.02 (d,  $J = 12.6$  Hz, 1H), 2.32 (s, 2H), 2.03 (s, 3H), 1.30 (s, 3H), 1.12 (s, 3H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  178.71(x2), 144.30, 140.50, 132.93(x2), 131.67 (q,  $J = 33.6$  Hz) (x2), 127.14(x2), 123.25 (q,  $J = 272.7$  Hz) (x2), 124.63-124.60(x2), 122.58, 118.82-118.67(x2), 68.75 (d,  $J = 24.9$  Hz), 61.53, 35.17, 28.88, 28.56, 26.71.  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -62.99. HRMS (ESI+)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{22}\text{N}_2\text{BrF}_6\text{S}$   $[\text{M}+\text{H}]^+$  539.05858, found 539.05865.



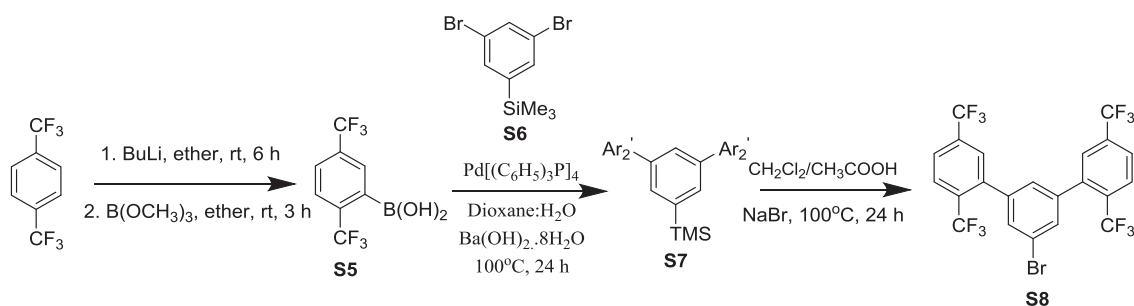
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.58 – 7.29 (m, 8H), 6.21 (s, 1H), 3.74 (d,  $J = 10.6$  Hz, 1H), 3.64 (d,  $J = 10.6$  Hz, 1H), 2.35 – 2.07 (m, 2H), 1.94 (s, 3H), 1.27 (d,  $J = 6.0$  Hz, 3H), 1.12 (s, 3H).

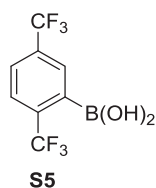
### 3-2.8.3. Preparation of DSI

DSIs were prepared according to the literature procedure<sup>28,88</sup> and the preparation of DSATf. DSI was dried in vacuo at 50°C for 3 h before using.

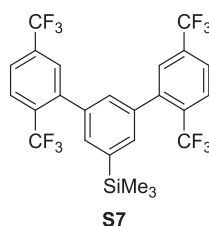
### Preparation of DSI via the final sulfonimide formation

#### Preparation of Ar substituent for DSI3

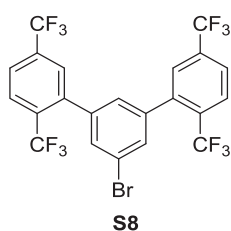




n-BuLi (2.7 M in hexane, 37 mL, 0.1 mol) in hexane was mixed with bis(trifluoromethyl)benzene (14 ml, 0.09 mol) was added dropwise to this solution at room temperature. After refluxing for 6 h, this solution was added dropwise, over a 1 h period, from a dropping funnel to a flask in an ice bath containing trimethyl borate (25 ml, 0.23 mol) mixed with ether (180 ml). The mixture was further stirred for 2 h at room temperature. Then, the mixture was poured into 1N HCl. The ether layer was separated and washed with water. The product was dried over MgSO<sub>4</sub>. The solvent was evaporated and the crude product was washed with hexane to give the product (11.6 g, 50%). *R*<sub>f</sub> = 0.27 (hexane: EtOAc = 4: 1) <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.04 (s, 1H), 7.81 (q, *J* = 8.4 Hz, 2H), 4.81 (s, 2H).

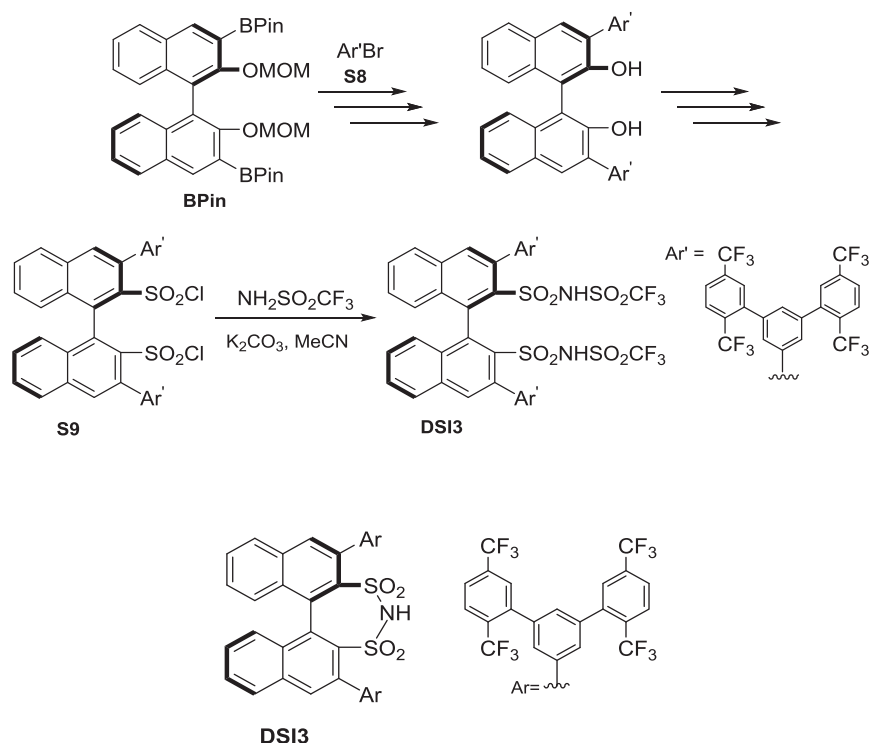


3,5-dibromo trisilyl benzene **S6** (0.85 g, 2.75 mmol), barium hydroxide octahydrate (3 g, 9.6 mmol), ArB(OH)<sub>2</sub> **S5** (2.86 g, 11 mmol), and Pd[(C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>P]<sub>4</sub> (0.26 g, 0.2 mmol) were mixed in dioxane: H<sub>2</sub>O (5:1) (48 ml). The mixture was stirred at 100°C for 24 h under nitrogen atmosphere. The reaction mixture was quenched with 1M HCl for 1 h, extracted with CH<sub>2</sub>Cl<sub>2</sub>, washed with brine, and dried with Na<sub>2</sub>SO<sub>4</sub>. The crude product was purified by column chromatography (hexane) to give the desired product **S7** (1.4 g, 89%). *R*<sub>f</sub> = 0.45 (hexane), <sup>1</sup>H NMR (400 MHz, Acetone) δ 7.94 (d, *J* = 8.3 Hz, 2H), 7.83 (d, *J* = 8.3 Hz, 2H), 7.65 (s, 2H), 7.50 (d, *J* = 1.4 Hz, 2H), 7.29 (s, 1H), 0.15 (s, 9H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 140.02 (s), 139.73 (s), 134.29 (s), 133.96 (s), 133.67 (s), 133.47 (d, *J* = 33.3 Hz), 132.45 (s), 132.14 (s), 131.83 (s), 131.64 (d, *J* = 22.4 Hz), 128.71 (q, *J* = 3.6 Hz), 128.19 (s), 127.76 – 126.71 (m), 125.15 (q, *J* = 3.7 Hz), 124.47 (d, *J* = 4.3 Hz), 121.75 (s), 119.03 (s). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -57.54 (s), -63.19 (s). HRMS (ESI+) *m/z* calcd for C<sub>25</sub>H<sub>18</sub>F<sub>12</sub>Si<sub>1</sub>[M]<sup>+</sup> 574.09861, found 574.09898, m.p 98-108°C.



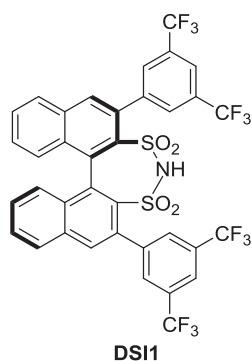
Trimethyl(2,2'',5,5''-tetrakis(trifluoromethyl)-[1,1':3',1''-terphenyl]-5'-yl)silane **S7** (1 g, 1.76 mmol) in CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>COOH (8 ml) was added NaBr (0.8 g, 7.83 mmol), NBS (1 g, 5.85 mmol) at r.t.. The mixture was stirred at 100°C for 1 day, quenched with aqueous NaHSO<sub>3</sub>, extracted with Et<sub>2</sub>O, washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The crude product was purified by washing with hexane to give the product **S8** (0.99 g, 98%). R<sub>f</sub> = 0.33 (hexane), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.94 (d, *J* = 8.2 Hz, 2H), 7.81 (d, *J* = 8.8 Hz, 2H), 7.66 (s, 2H), 7.59 (d, *J* = 1.5 Hz, 2H), 7.27 (s, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 140.02 (s), 139.73 (s), 134.29 (s), 133.79 (d, *J* = 33.3 Hz), 131.95 (d, *J* = 38.1 Hz), 131.57 – 131.08 (m), 128.74 (dd, *J* = 7.3, 3.7 Hz), 128.21 (s), 127.14 (d, *J* = 5.4 Hz), 125.19 (d, *J* = 3.7 Hz), 124.49 (d, *J* = 3.6 Hz), 121.78 (s), 119.05 (s). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -57.52 (s), -63.15 (s). HRMS (ESI+) *m/z* calcd for C<sub>22</sub>H<sub>10</sub>BrF<sub>12</sub>[M]<sup>2+</sup> 580.09861, found 580.96674, m.p 125-132°C.

### Preparation of DSI3

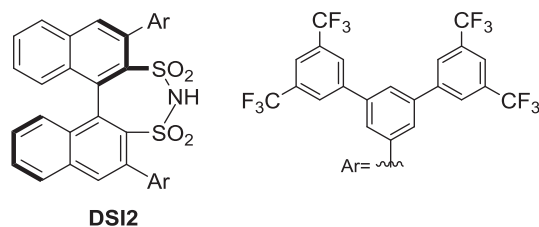


A mixture of **S8** (0.13 g, 0.09 mmol) in THF (14 ml) was kept at -20°C for 20 minutes and then NH<sub>3</sub> in EtOH (1.5 ml, 3 mmol) was added to this solution and was continued to stir overnight

at  $-20^{\circ}\text{C}$ . The crude product was purified by column chromatography (Hexane: AcOEt= 3: 1) and the obtained product was extracted with 1N HCl to give **DSI3** (0.07 g, 60%).  $R_f= 0.6$  ( $\text{CHCl}_3$ : MeOH= 5: 1),  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.13 (s, 1H), 8.04 (d,  $J = 8.3$  Hz, 1H), 7.90 (dd,  $J = 24.0, 8.2$  Hz, 2H), 7.82 – 7.65 (m, 5H), 7.61 (s, 1H), 7.49 (s, 1H), 7.39 (dd,  $J = 16.7, 9.2$  Hz, 2H), 7.16 (d,  $J = 8.6$  Hz, 1H).  $^{13}\text{C NMR}$  (101 MHz,  $\text{CDCl}_3$ )  $\delta$  140.26 (s), 137.70 (d,  $J = 43.4$  Hz), 136.72 (s), 136.11 (s), 134.52 (s), 133.32 (s), 132.67 (d,  $J = 33.0$  Hz), 132.44 – 131.78 (m), 131.01 (t,  $J = 15.3$  Hz), 129.99 (s), 128.93 (s), 127.75 (d,  $J = 42.6$  Hz), 127.06 (s), 126.81 – 124.92 (m), 123.67 (d,  $J = 17.9$  Hz), 121.40 – 120.43 (m), 118.25 – 117.28 (m).



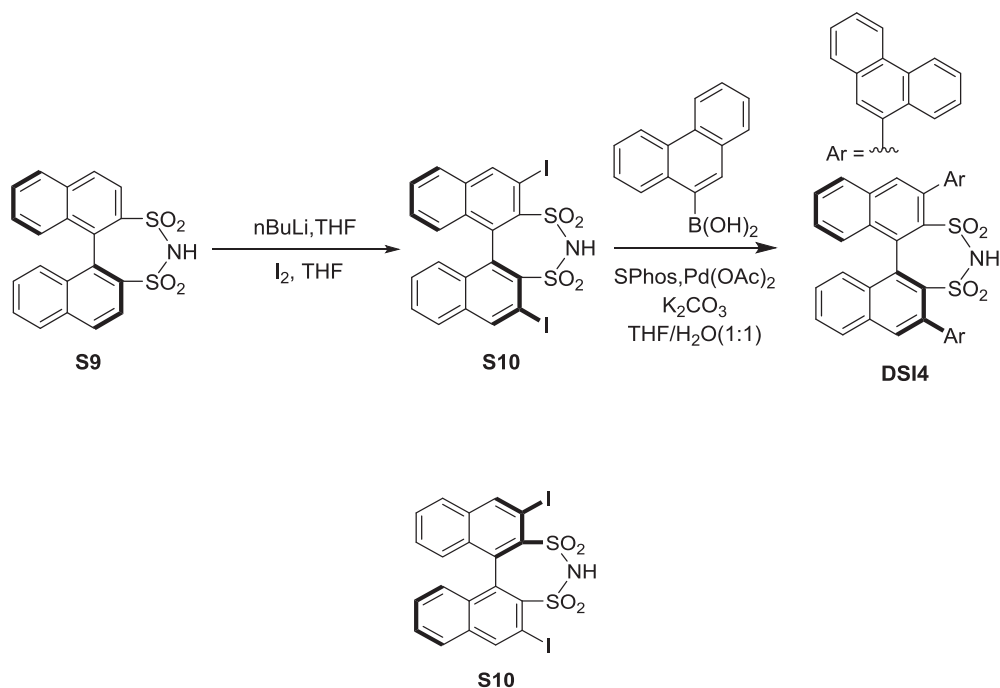
$^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.10 (d,  $J = 6.1$  Hz, 4H), 8.05 – 7.89 (m, 6H), 7.84 – 7.76 (m, 2H), 7.54 (ddd,  $J = 8.4, 6.9, 1.2$  Hz, 2H), 7.25 (d,  $J = 7.9$  Hz, 2H).



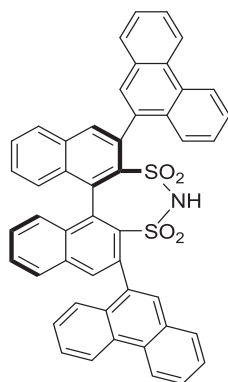
$^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.21 (s, 2H), 8.15 (s, 4H), 8.09 (d,  $J = 12$  Hz, 6H), 7.82 (ddd,  $J = 20.2, 8.3$  Hz, 12H), 7.51 (t,  $J = 7.8$  Hz, 2H), 7.28 (s, 2H).



## Preparation of DSI via the direct introduction of Ar-group to DSI skeleton (Preparation of DSI4)



To a dry-ice-cooled solution of **S9** (149 mg, 0.38 mmol) in dry THF at  $-40^\circ\text{C}$  under  $\text{N}_2$  was dropwise added  $n\text{-BuLi}$  (10.4 ml, 3.45 mmol). After this dark green solution was kept stirring at  $-40^\circ\text{C}$  for 2.5 h. After the mixture was cooled at  $-78^\circ\text{C}$ , a solution of  $\text{I}_2$  (1 g, 3.94 mmol) in THF (4.5 ml) was slowly added dropwise. After the addition, the reddish-brown solution was warmed to r.t and stirred for 12 h. The resulting solution was quenched by saturated  $\text{NH}_4\text{Cl}$ . The mixture was then acidified with 4 N  $\text{HCl}$  and extracted with  $\text{AcOEt}$  three times. The combined organic layers were washed with saturated  $\text{Na}_2\text{S}_2\text{O}_3$ , brine, and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After the removal of the solvent under reduced pressure, the crude product was purified by a flash column chromatography (hexane:  $\text{AcOEt}$  = 1:6) to give the desired product **S10** (87.3 mg, 35%).  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.88 (s, 2H), 7.96 (d,  $J = 8.3$  Hz, 2H), 7.60 – 7.51 (m, 2H), 7.31 – 7.21 (m, 2H), 6.78 (d,  $J = 8.6$  Hz, 2H).

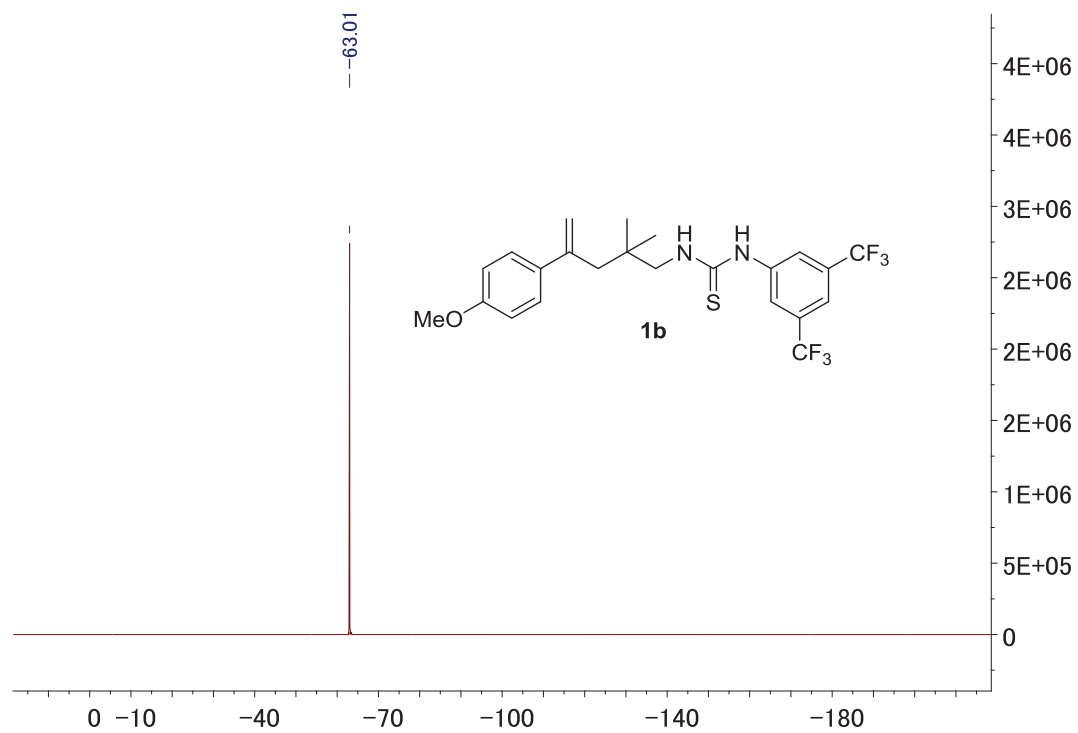


**DSi4**

(*S*)-3,3'-diiododisulfonamide (87.3 mg, 0.13 mmol) **S10**, 9-phenanthracenyl boronic acid (124 mg, 0.56 mmol), K<sub>2</sub>CO<sub>3</sub> (77 mg, 0.56 mmol), palladium(II)-acetate (3.3 mg, 0.015 mmol) were suspended in THF/water (3 ml) under an atmosphere of nitrogen was heated for 1 day at 70°C. The reaction was poured into HCl 1M and extracted with DCM, and the combined organic layers were washed with water and brine. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, and the product was purified by flash column chromatography (hexane: AcOEt= 1: 6). The purified product was dissolved in DCM, acidified with HCl and the solvent was removed in vacuo to obtain **DSi4** (68.2 mg, 70%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.76 (dd, *J* = 18.0, 8.2 Hz, 2H), 8.27 – 8.11 (m, 1H), 8.03 (dd, *J* = 12.5, 7.9 Hz, 1H), 7.96 – 7.81 (m, 2H), 7.73 – 7.49 (m, 6H), 7.46 – 7.33 (m, 2H).

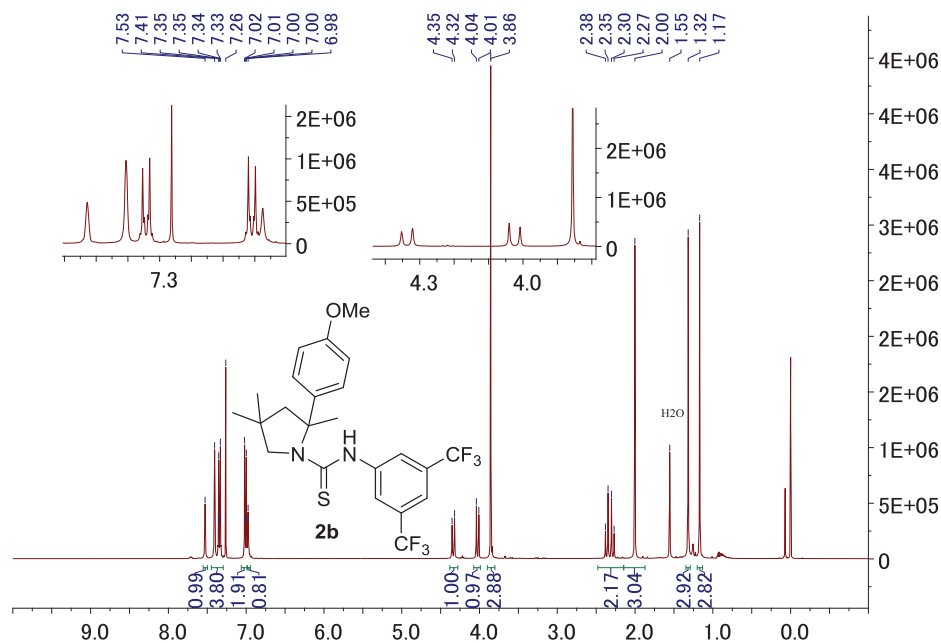


$^{19}\text{F}$  NMR



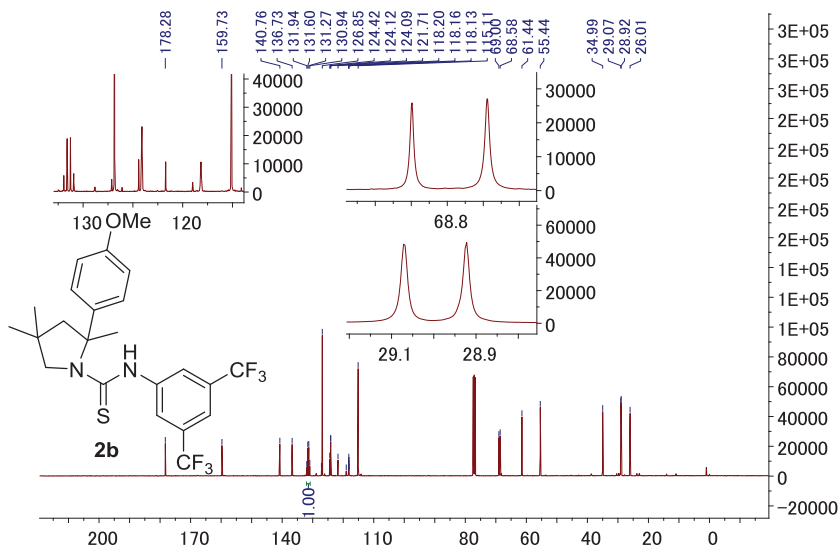
$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.01 (s).

## <sup>1</sup>H NMR



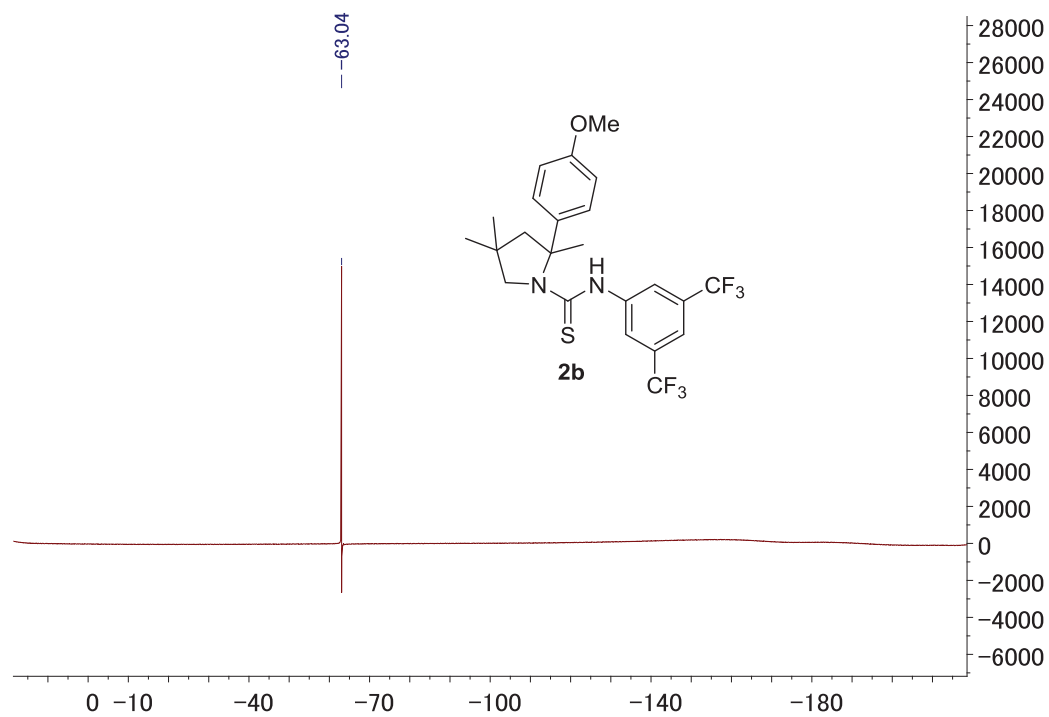
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.53 (s, 1H), 7.45 – 7.29 (m, 4H), 7.06 – 6.99 (m, 2H), 6.98 (s, 1H), 4.34 (d, *J* = 12.6 Hz, 1H), 4.02 (d, *J* = 12.7 Hz, 1H), 3.86 (s, 3H), 2.33 (dd, *J* = 31.1, 13.5 Hz, 2H), 2.00 (s, 3H), 1.32 (s, 3H), 1.17 (s, 3H).

## <sup>13</sup>C NMR



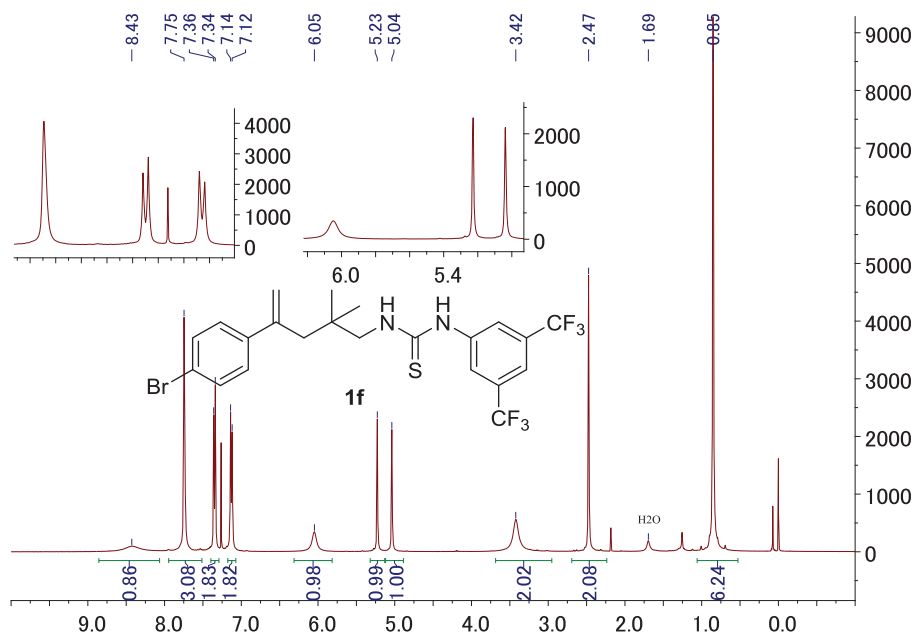
<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 178.28, 159.73, 140.76, 136.73, 131.44 (q, *J* = 33.5 Hz) (x2), 126.85(x2), 123.07 (q, *J* = 272.8 Hz) (x2), 124.12-124.09(x2), 118.23-118.09, 115.11(x2), 69.00, 68.58, 61.44, 55.44, 34.99, 29.07, 28.92, 26.0.

$^{19}\text{F}$  NMR



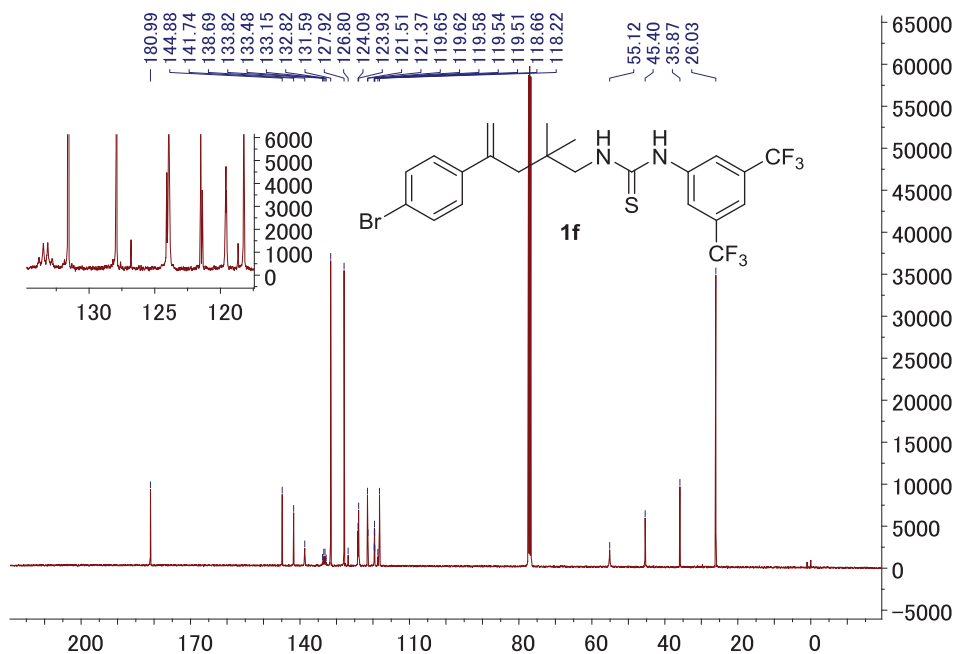
$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta -63.04$  (s).

## <sup>1</sup>H NMR



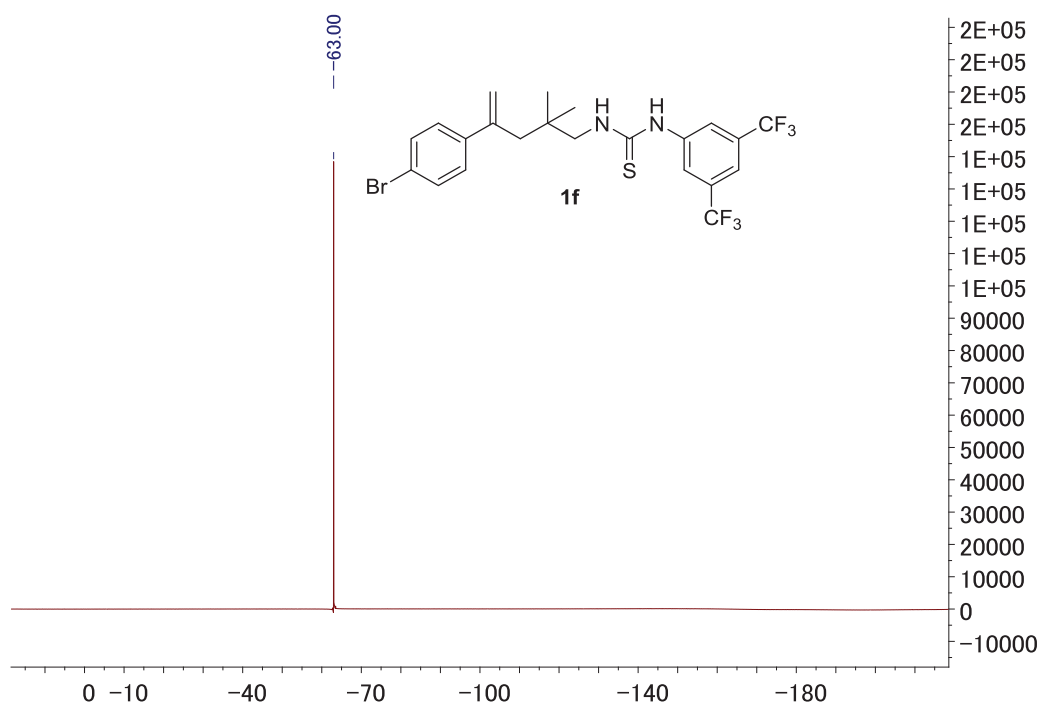
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.43 (s, 1H), 7.75 (s, 3H), 7.35 (d, *J* = 8.4 Hz, 2H), 7.13 (d, *J* = 8.3 Hz, 2H), 6.05 (s, 1H), 5.23 (s, 1H), 5.04 (s, 1H), 3.42 (s, 2H), 2.47 (s, 2H), 0.85 (s, 6H).

## <sup>13</sup>C NMR



<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 180.99, 144.88, 141.74, 138.69, 133.32 (q, *J* = 67.0, 33.4 Hz) (x2), 131.59 (x2), 127.92 (x2), 123.93 (x2), 122.51 (q, *J* = 128.4 Hz) (x2), 121.51, 119.65 – 119.51, 118.22, 55.12, 45.40, 35.87, 26.03(x2).

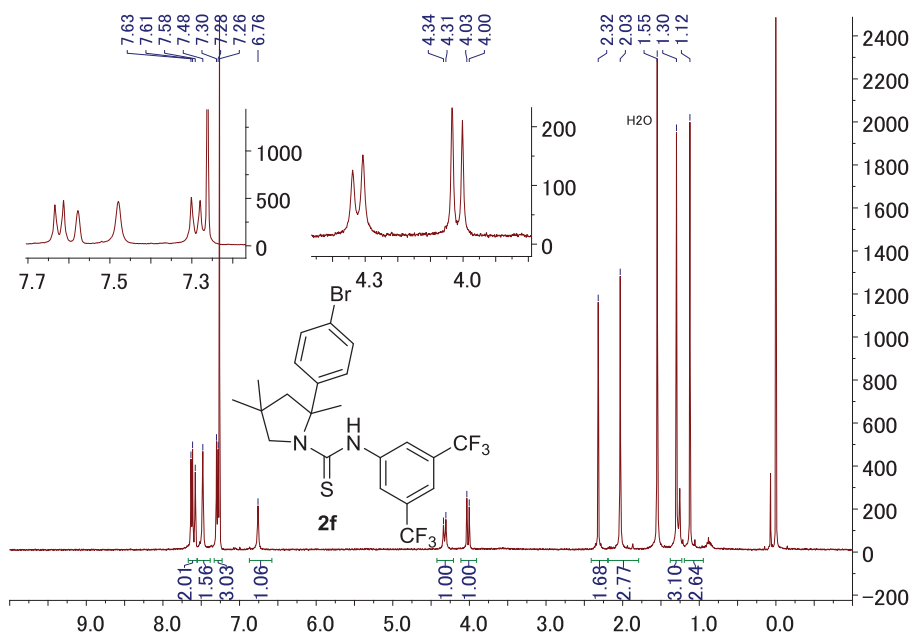
$^{19}\text{F}$  NMR



$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.00 (s).

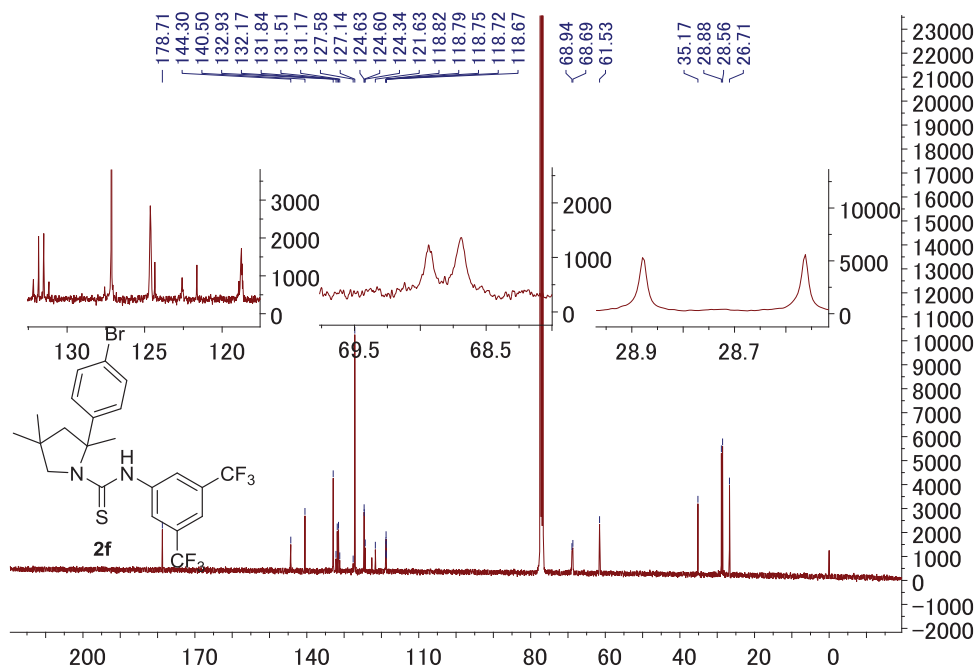


# <sup>1</sup>H NMR



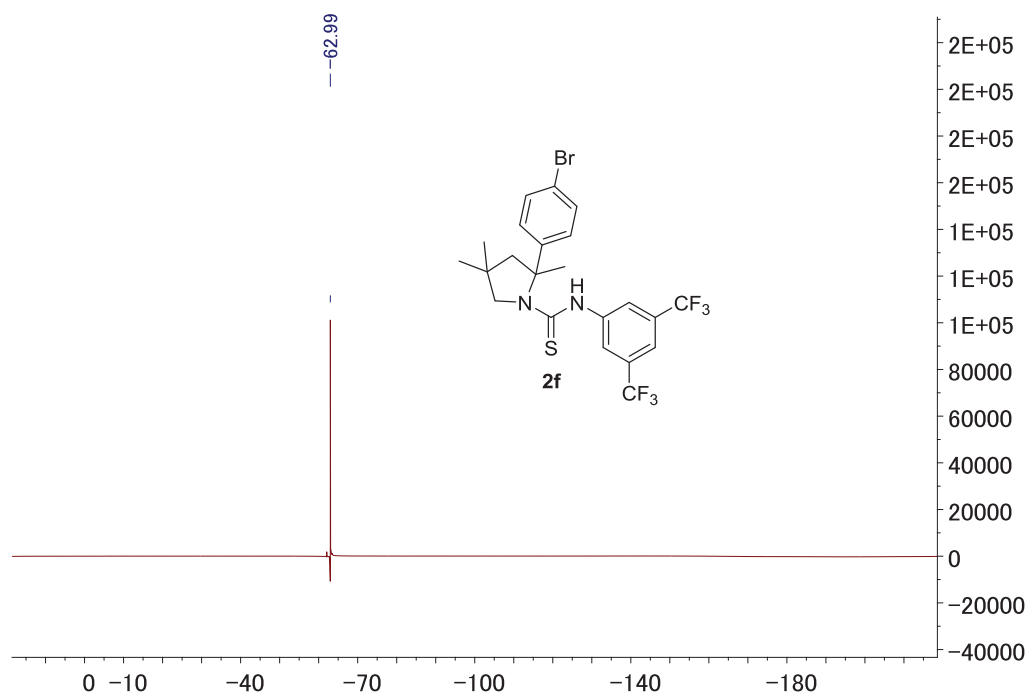
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.67 – 7.56 (m, 2H), 7.48 (s, 2H), 7.33 – 7.23 (m, 3H), 6.76 (s, 1H), 4.32 (d, *J* = 12.6 Hz, 1H), 4.02 (d, *J* = 12.6 Hz, 1H), 2.32 (s, 2H), 2.03 (s, 3H), 1.30 (s, 3H), 1.12 (s, 3H).

# <sup>13</sup>C NMR



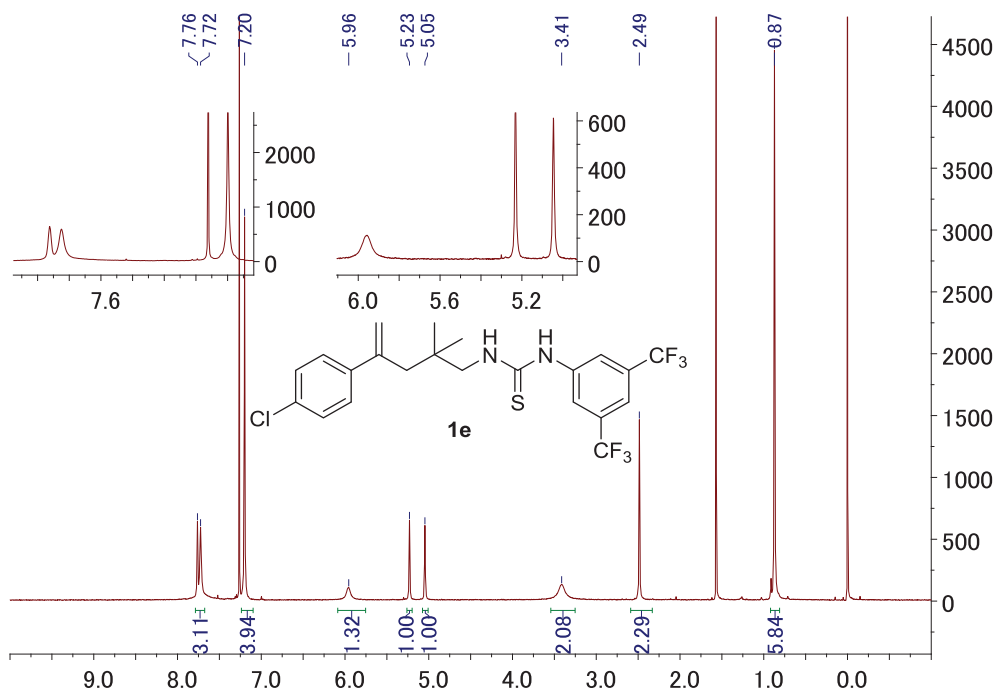
<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 178.71(x2), 144.30, 140.50, 132.93(x2), 131.67 (q, *J* = 33.6 Hz) (x2), 127.14(x2), 123.25 (q, *J* = 272.7 Hz) (x2), 124.63-124.60 (x2), 122.58, 118.82-118.67 (x2), 68.75 (d, *J* = 24.9 Hz), 61.53, 35.17, 28.88, 28.56, 26.71.

$^{19}\text{F}$  NMR



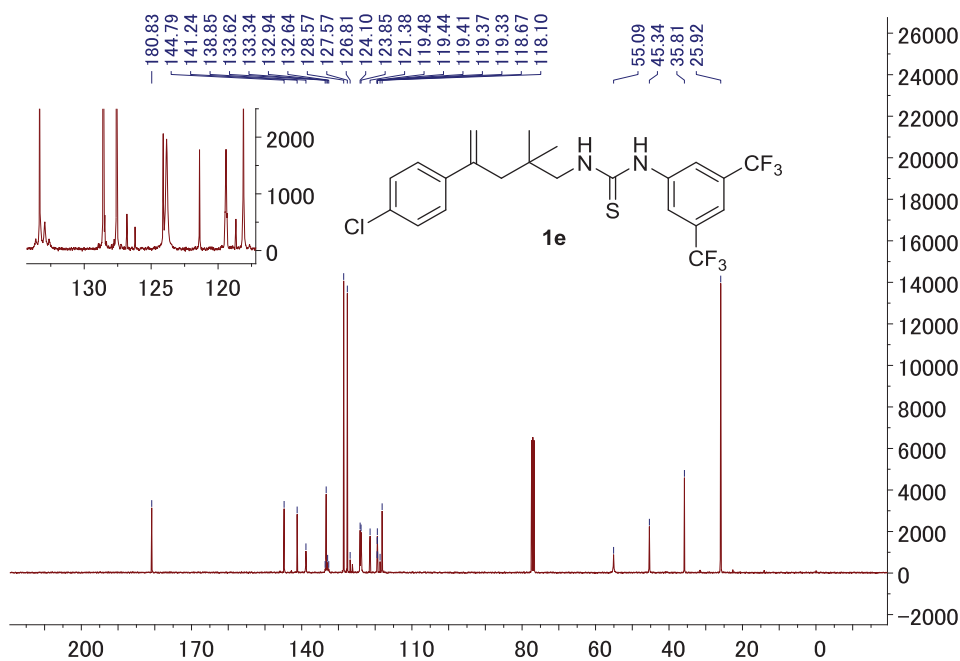
$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -62.99 (s)

# <sup>1</sup>H NMR



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.74 (d,  $J = 14.7$  Hz, 3H), 7.20 (s, 4H), 5.96 (s, 1H), 5.23 (s, 1H), 5.05 (s, 1H), 3.41 (s, 2H), 2.49 (s, 2H), 0.87 (s, 6H).

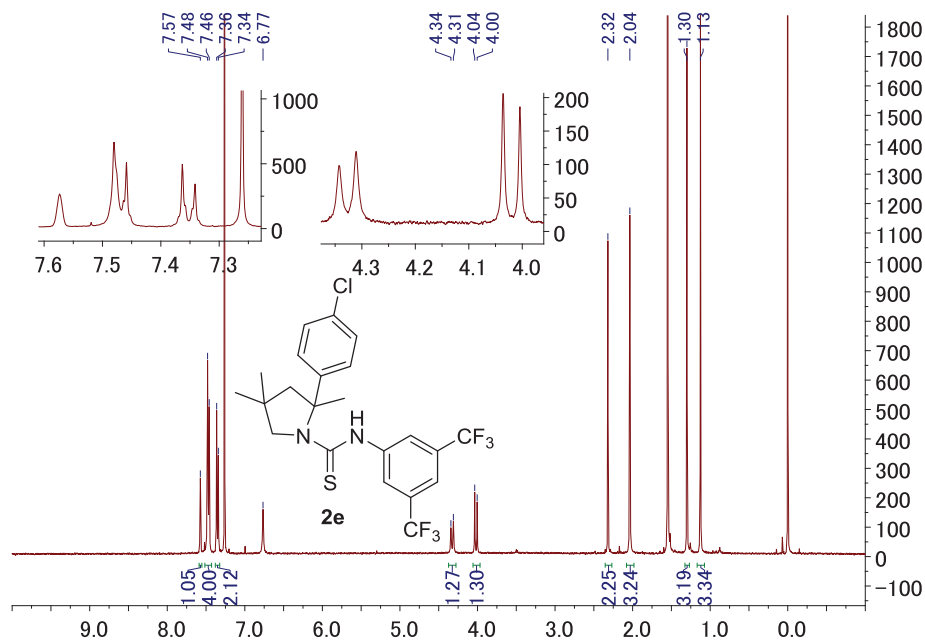
# <sup>13</sup>C NMR



<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 180.83, 144.79, 141.24, 138.85, 133.34, 133.13 (q,  $J = 30.7$  Hz) (x2), 128.57(x2), 127.57(x2), 123.85(x2), 122.74 (q,  $J = 273.1$  Hz) (x2), 119.48 – 118.33, 118.10, 55.09, 45.34, 35.81, 25.92(x2).

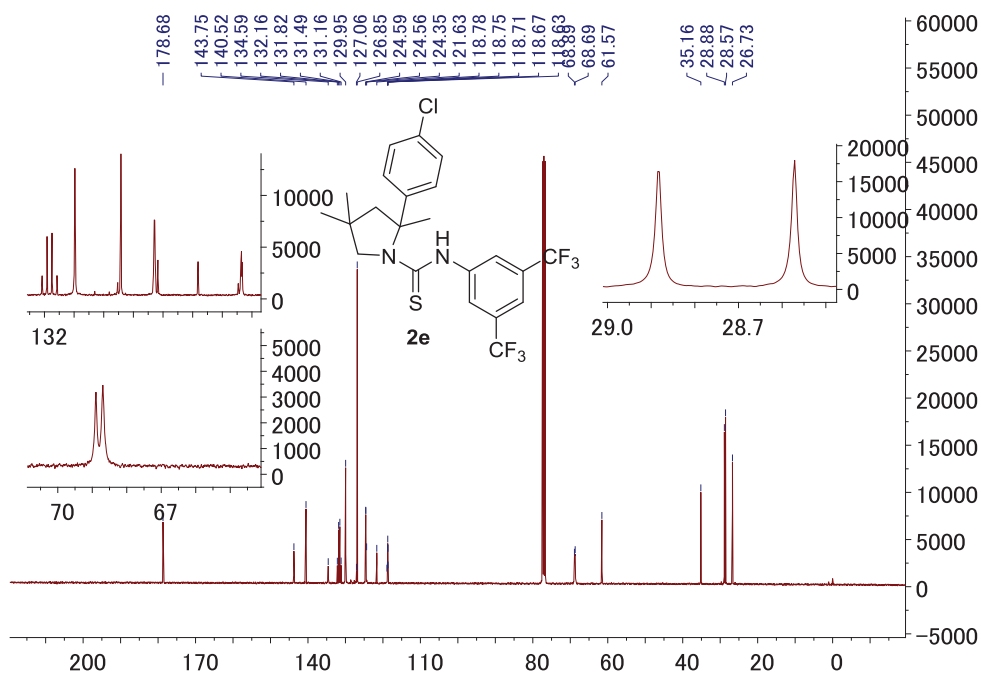


# <sup>1</sup>H NMR



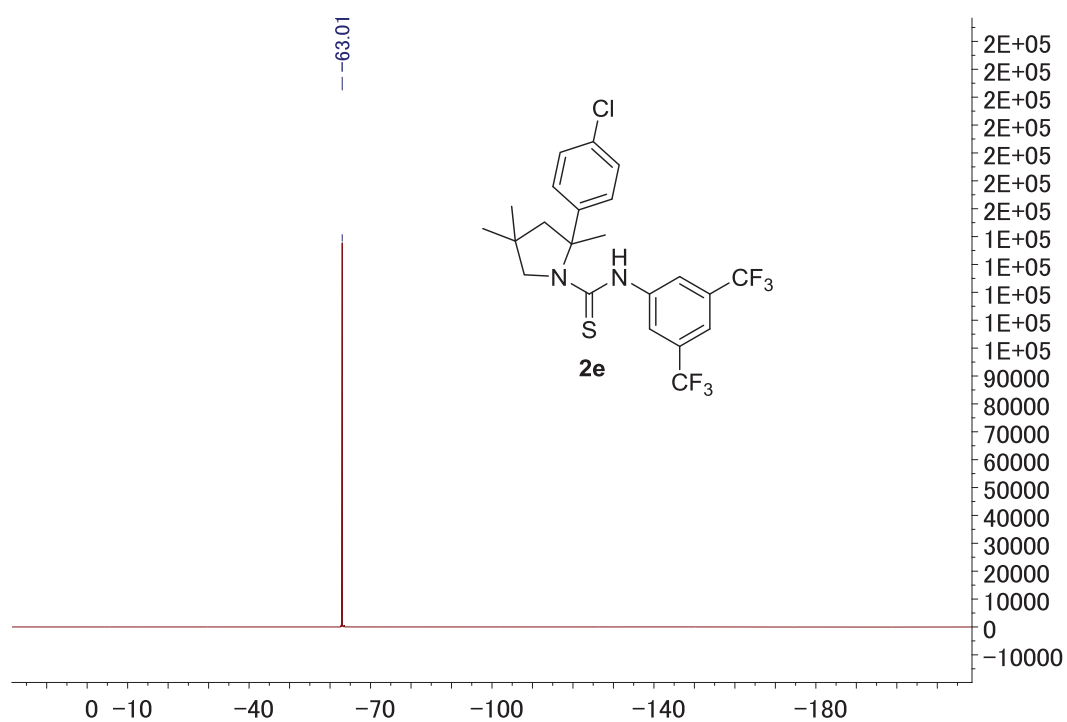
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.57 (s, 1H), 7.47 (d, *J* = 8.6 Hz, 4H), 7.35 (d, *J* = 8.7 Hz, 2H), 4.33 (d, *J* = 12.8 Hz, 1H), 4.02 (d, *J* = 12.6 Hz, 1H), 2.32 (s, 2H), 2.04 (s, 3H), 1.30 (s, 3H), 1.13 (s, 3H).

# <sup>13</sup>C NMR



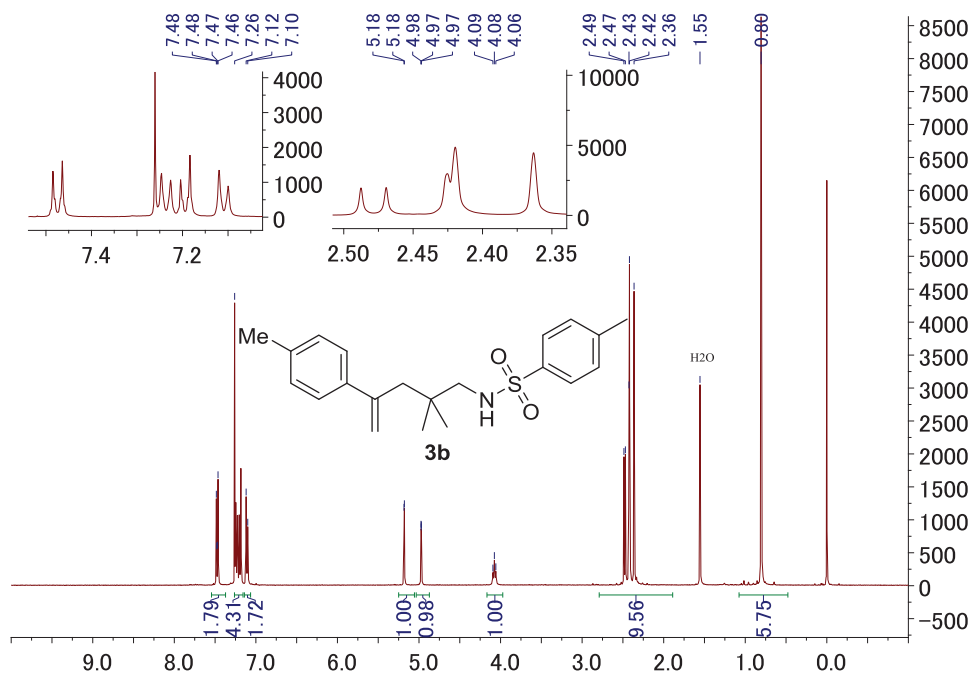
<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 178.68, 143.75, 140.52, 134.59, 131.66 (q, *J* = 33.6 Hz) (x2), 129.95 (x2), 126.85 (x2), 124.59-124.56 (x2), 122.99 (q, *J* = 272.8 Hz) (x2), 121.63, 118.78-118.63, 68.79 (d, *J* = 20.2 Hz), 61.57, 35.16, 28.88, 28.57, 26.73.

$^{19}\text{F}$  NMR



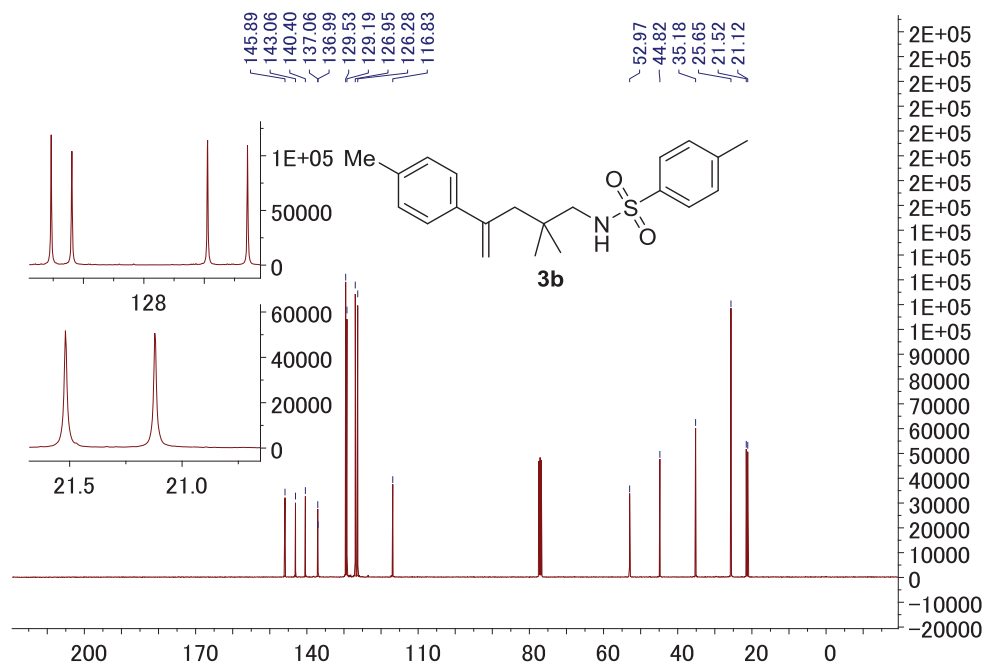
$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -63.01 (s).

# <sup>1</sup>H NMR



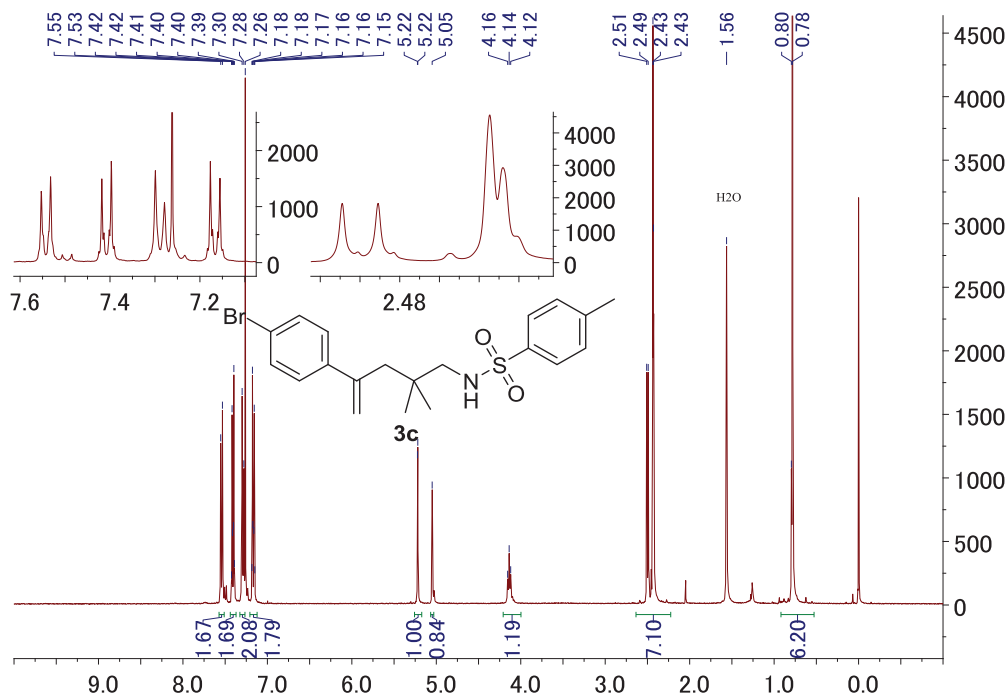
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.54 – 7.37 (m, 2H), 7.26 (s, 4H), 7.11 (d, *J* = 7.9 Hz, 2H), 5.18 (d, *J* = 1.8 Hz, 1H), 5.03 – 4.87 (m, 1H), 4.08 (t, *J* = 7.2 Hz, 1H), 2.79 – 1.89 (m, 10H), 0.80 (s, 6H).

# <sup>13</sup>C NMR



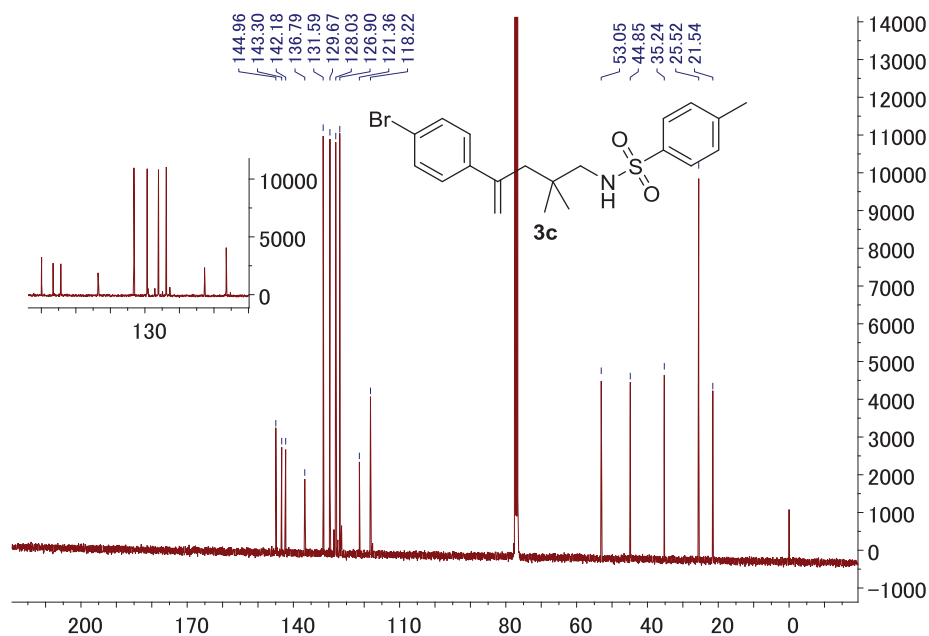
<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 145.89, 143.06, 140.40, 137.02, 129.53(x2), 129.19(x2), 126.95(x2), 126.28(x2), 116.83, 52.97, 44.82, 35.18(x2), 25.65(x2), 21.52, 21.12.

### $^1\text{H}$ NMR



$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.54 (d,  $J = 8.3$  Hz, 2H), 7.44 – 7.37 (m, 2H), 7.34 – 7.24 (m, 2H), 7.21 – 7.12 (m, 2H), 5.22 (d,  $J = 1.6$  Hz, 1H), 5.05 (s, 1H), 4.14 (t,  $J = 7.2$  Hz, 1H), 2.47 (dd,  $J = 27.4, 4.9$  Hz, 6H), 0.79 (d,  $J = 5.3$  Hz, 6H).

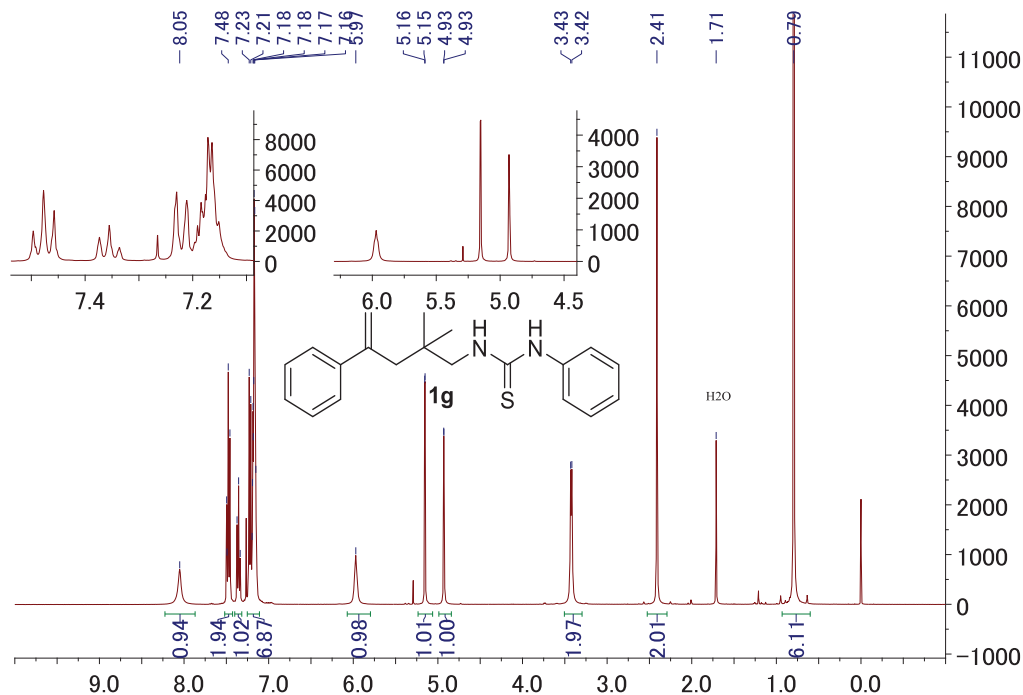
### $^{13}\text{C}$ NMR



$^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  144.96, 143.30, 142.18, 136.79(x2), 131.59(x2), 129.67(x2), 128.03(x2), 126.90(x2), 121.36, 118.22, 53.05, 44.85, 35.24, 25.52(x2), 21.54.

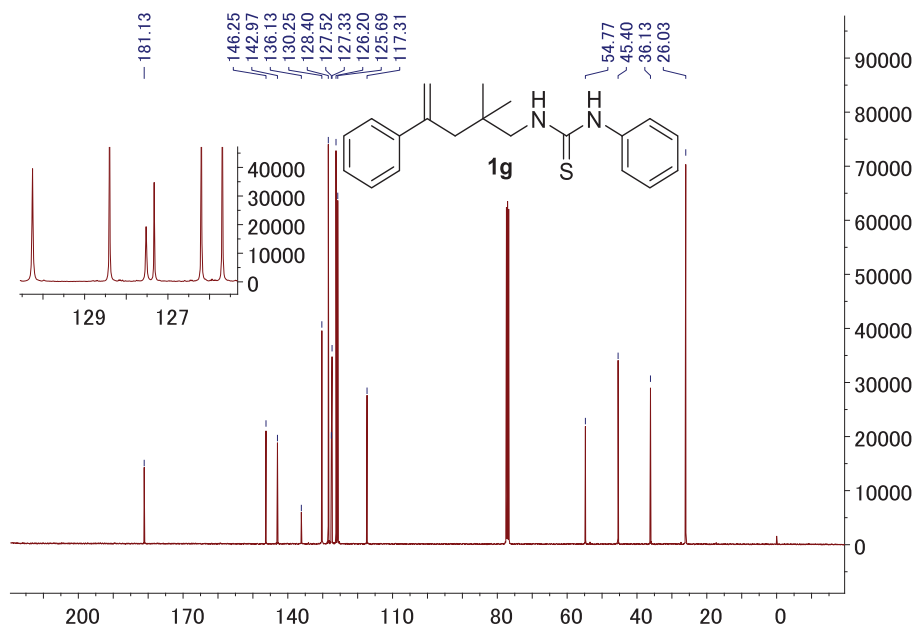


# <sup>1</sup>H NMR



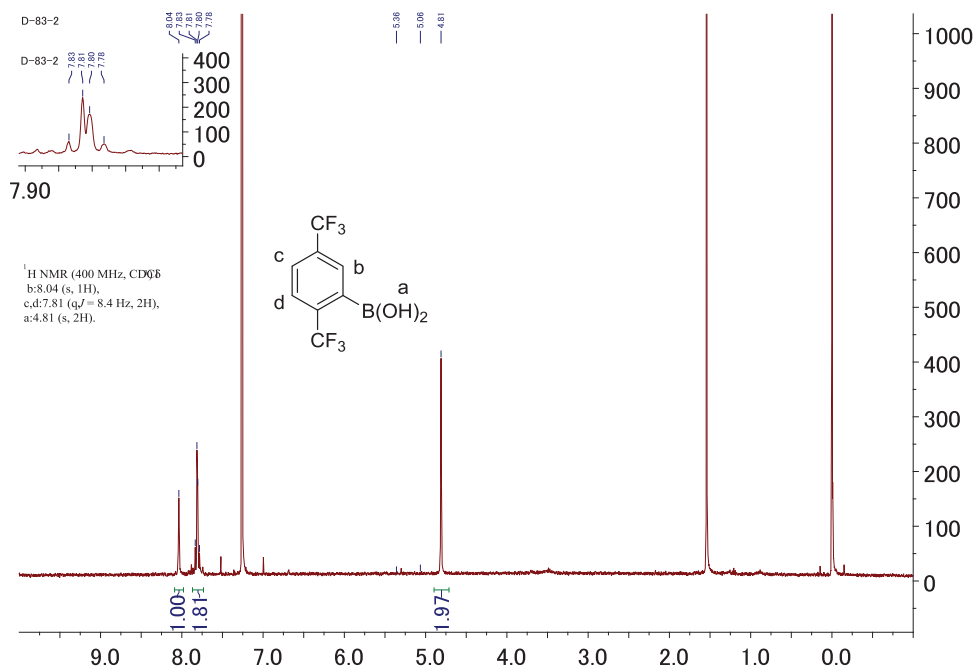
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.05 (s, 1H), 7.48 (dd, *J* = 10.6, 4.9 Hz, 2H), 7.35 (t, *J* = 7.4 Hz, 1H), 7.25 – 7.11 (m, 7H), 5.97 (s, 1H), 5.15 (d, *J* = 1.7 Hz, 1H), 4.93 (d, *J* = 0.7 Hz, 1H), 3.42 (d, *J* = 5.7 Hz, 2H), 2.41 (s, 2H), 0.79 (s, 6H).

# <sup>13</sup>C NMR

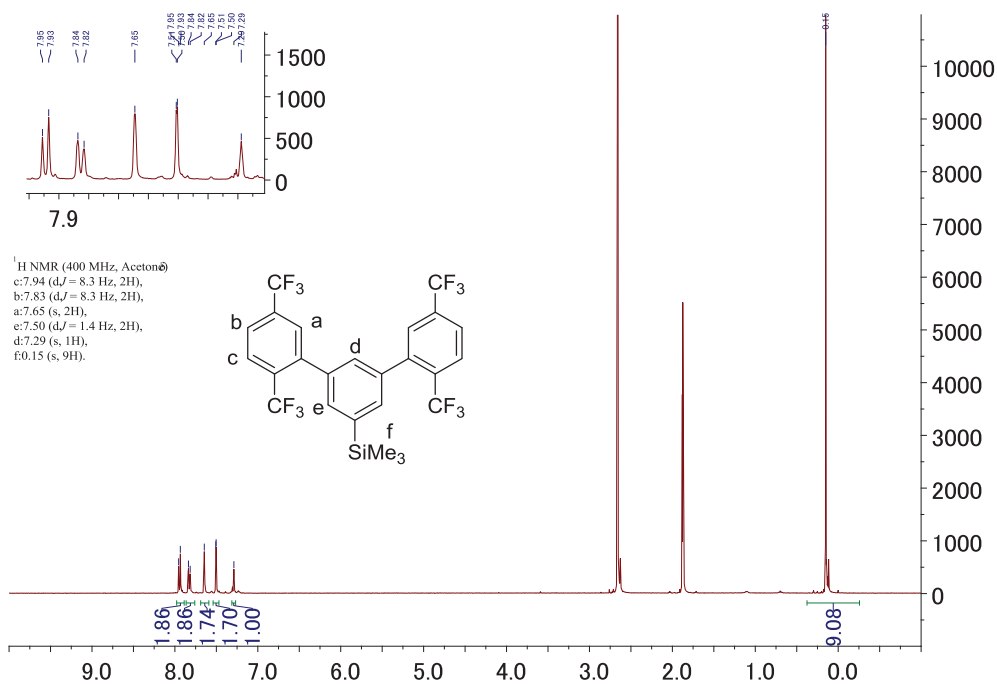


<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 181.13, 146.25, 142.9, 136.13, 130.25 (x2), 128.40 (x2), 127.52, 127.33, 126.20(x2)

# <sup>1</sup>H NMR

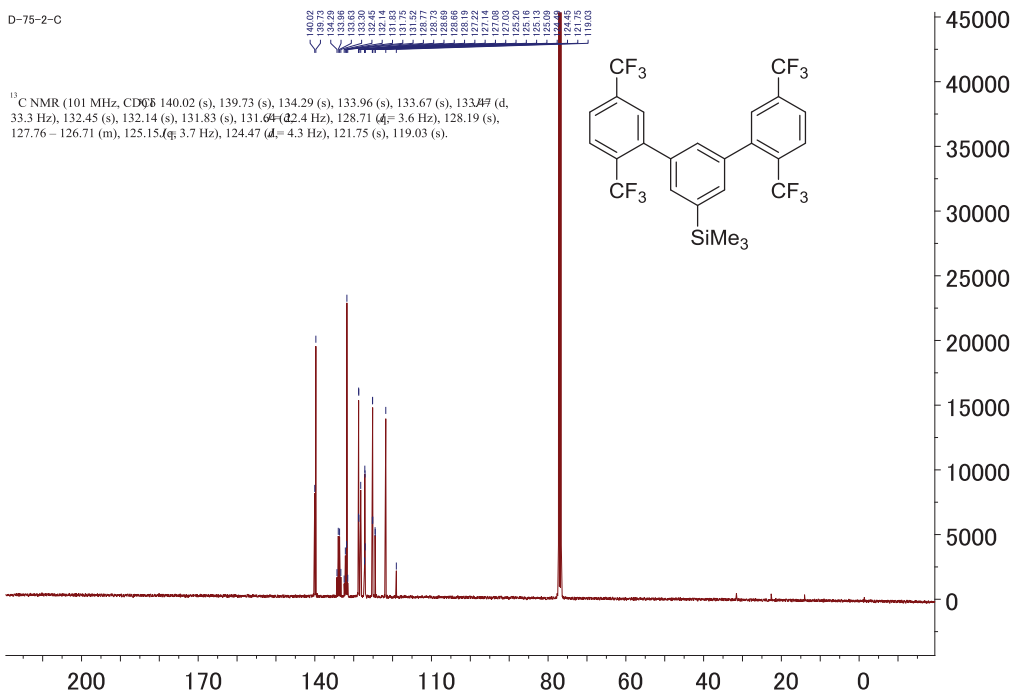


# <sup>1</sup>H NMR



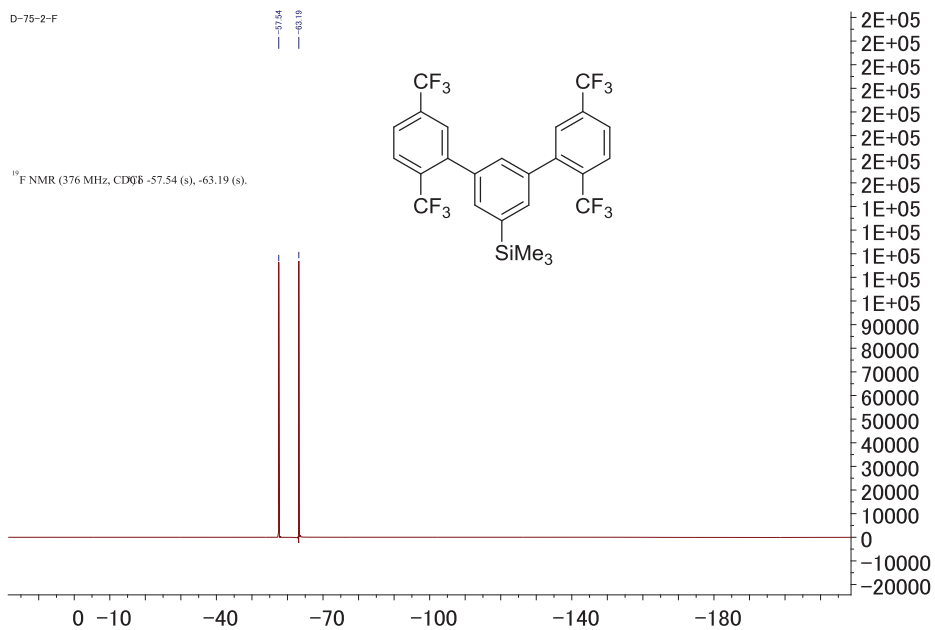
# <sup>13</sup>C NMR

D-75-2-C

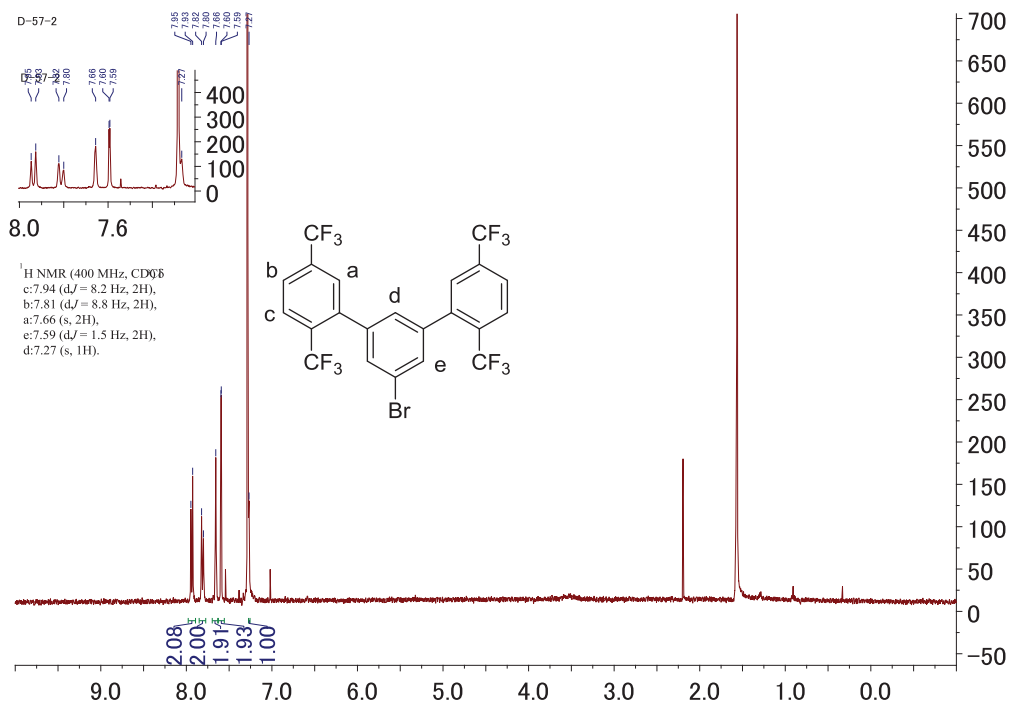


# <sup>19</sup>F NMR

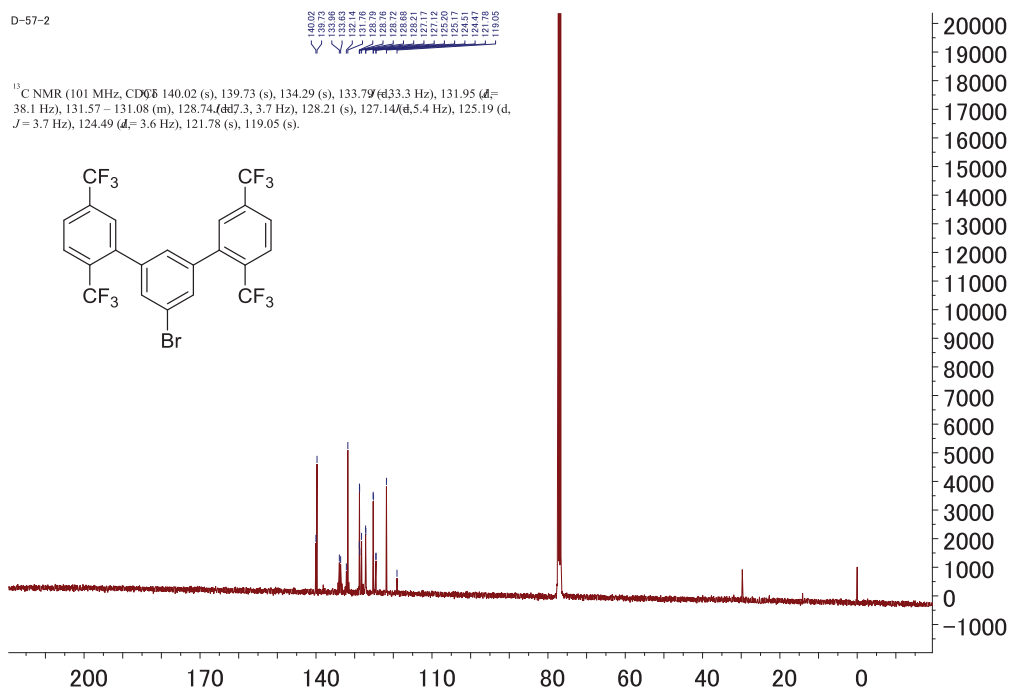
D-75-2-F



# <sup>1</sup>H NMR

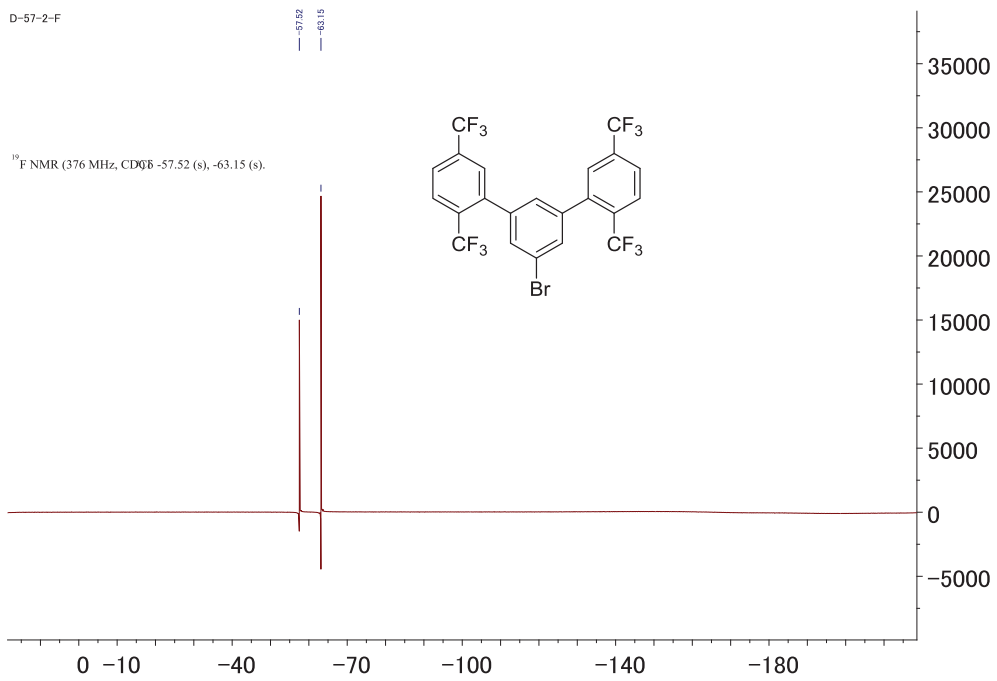


# <sup>13</sup>C NMR

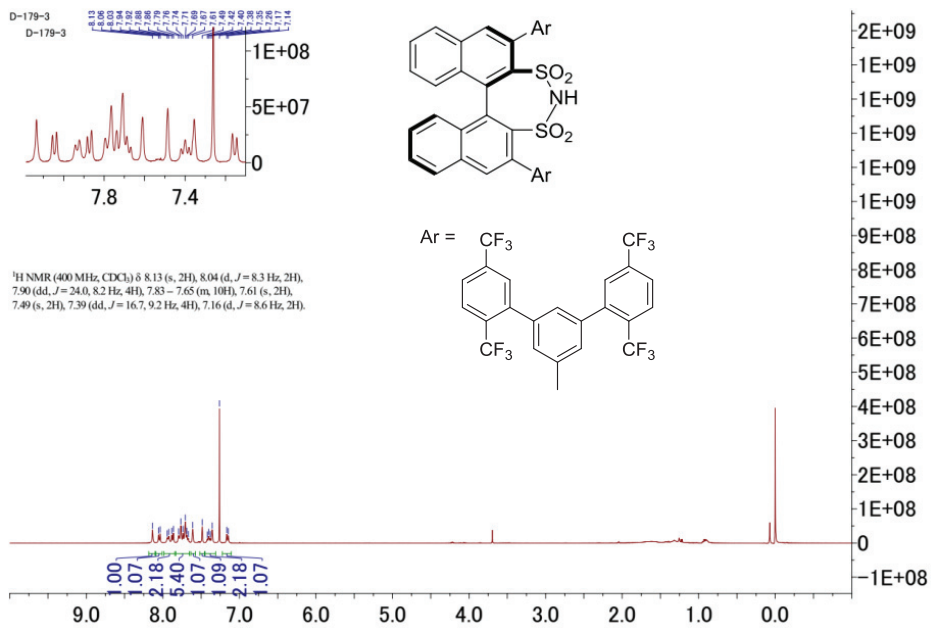


# $^{19}\text{F}$ NMR

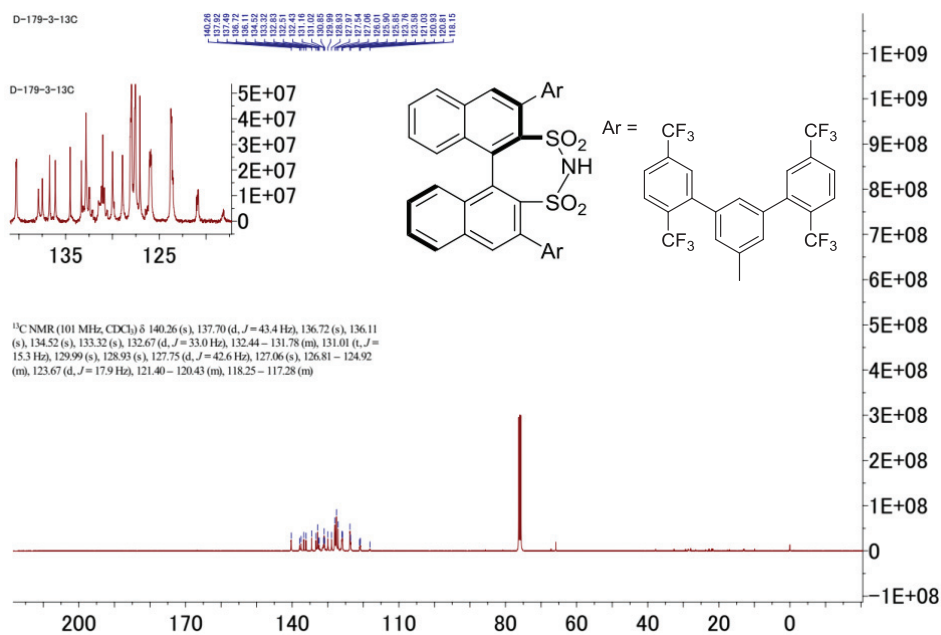
D-57-2-F



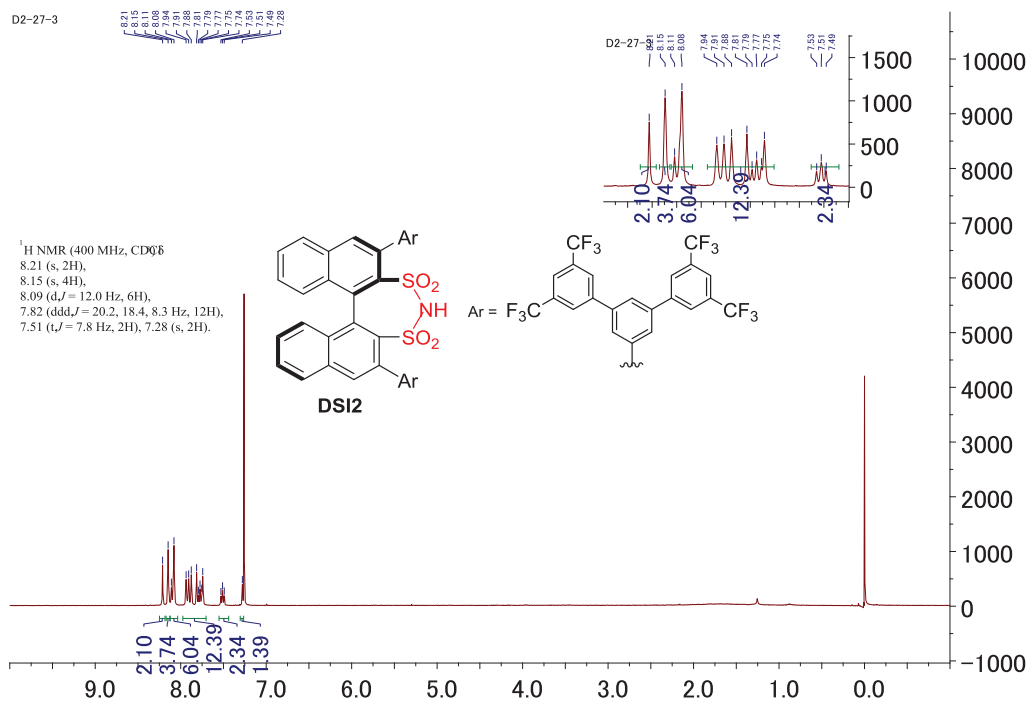
# <sup>1</sup>H NMR



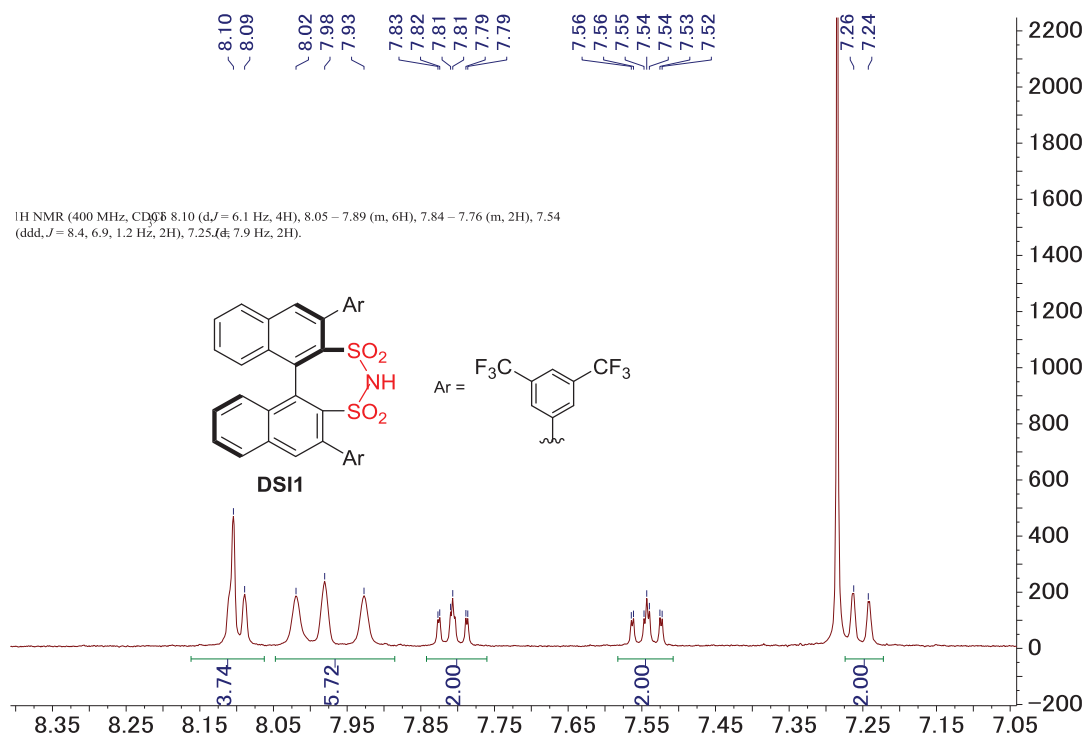
# <sup>13</sup>C NMR



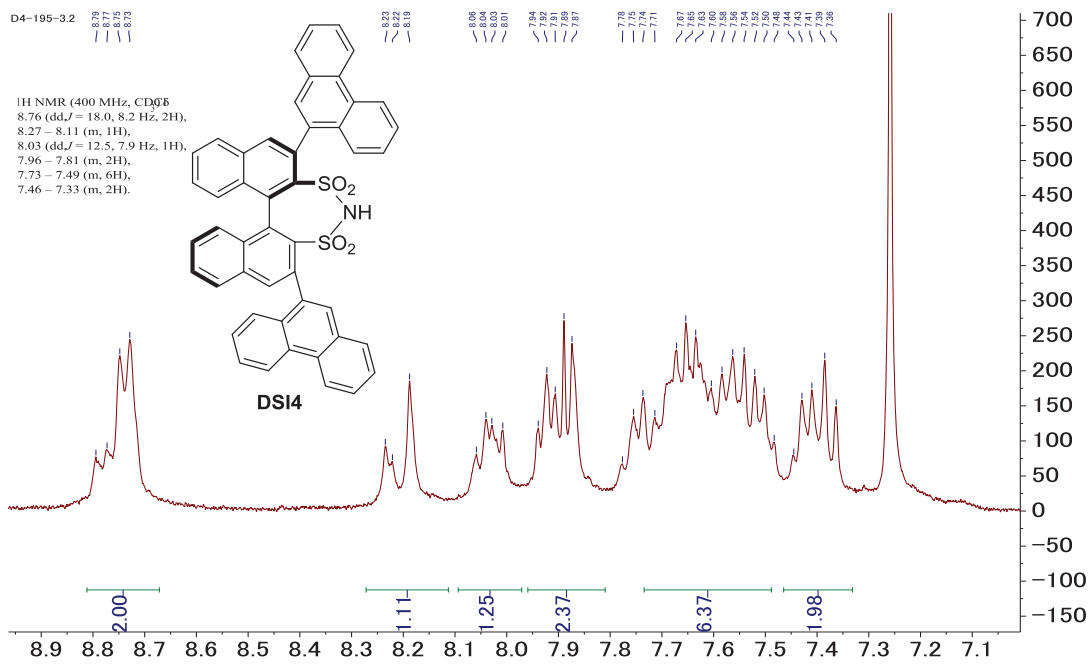
# <sup>1</sup>H NMR



# <sup>1</sup>H NMR



# <sup>1</sup>H NMR

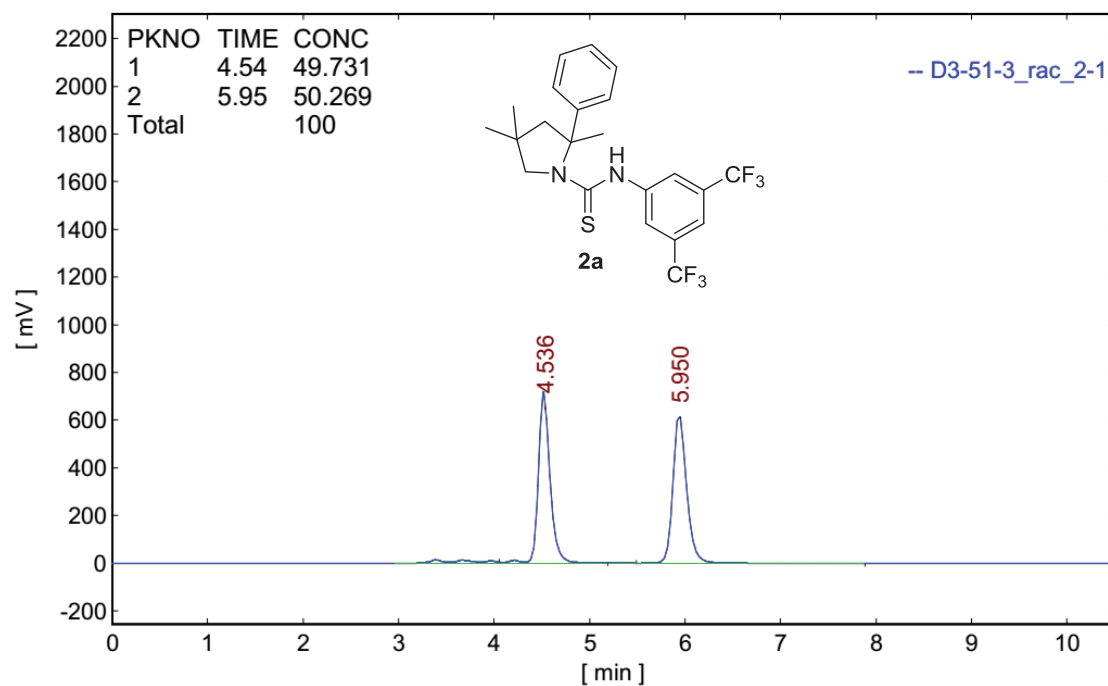




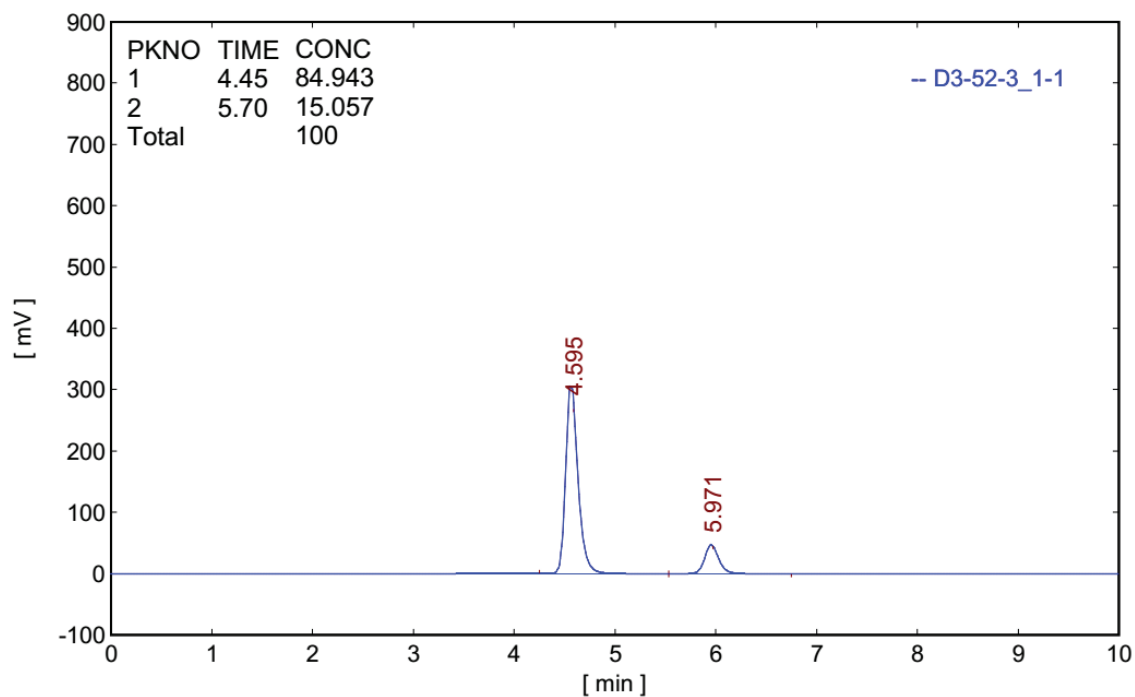
## HPLC spectra

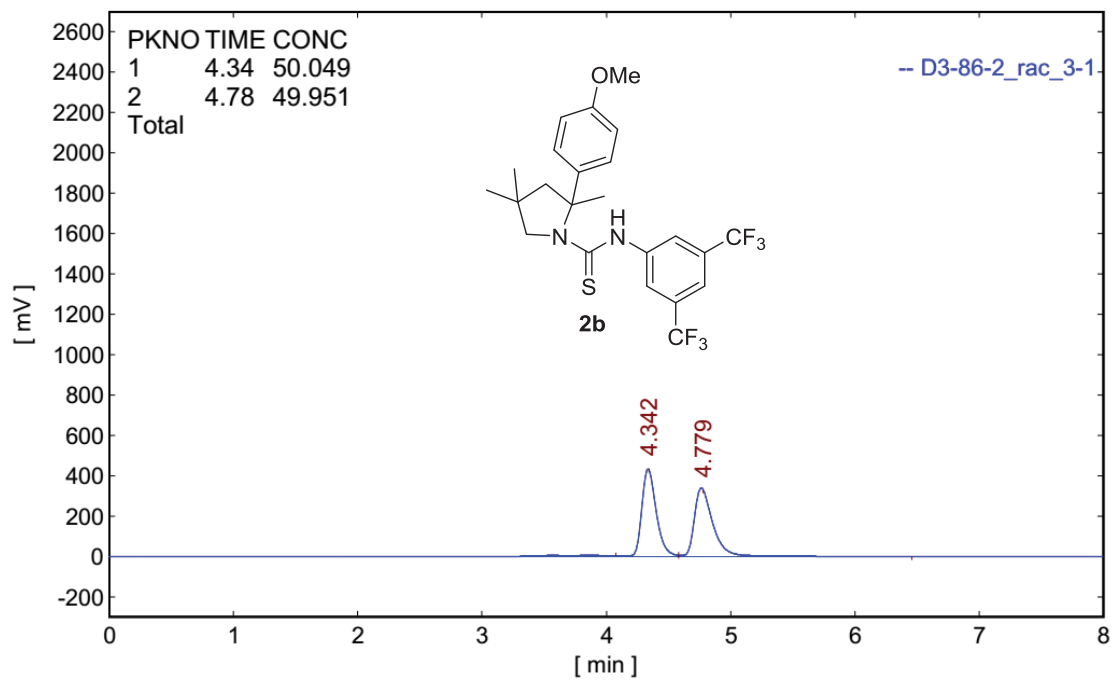
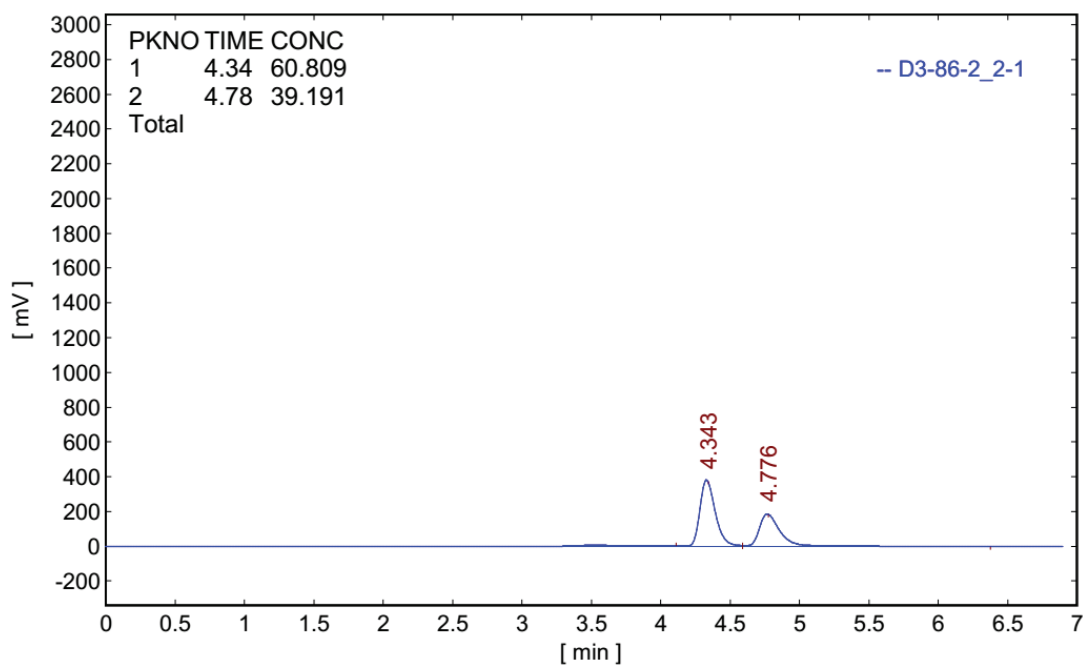
**Table 3**

Entry 8: *Rac-2a* Chiralpak IB, hexane: EtOH= 10: 1, 1.0 mL/ min 254 nm

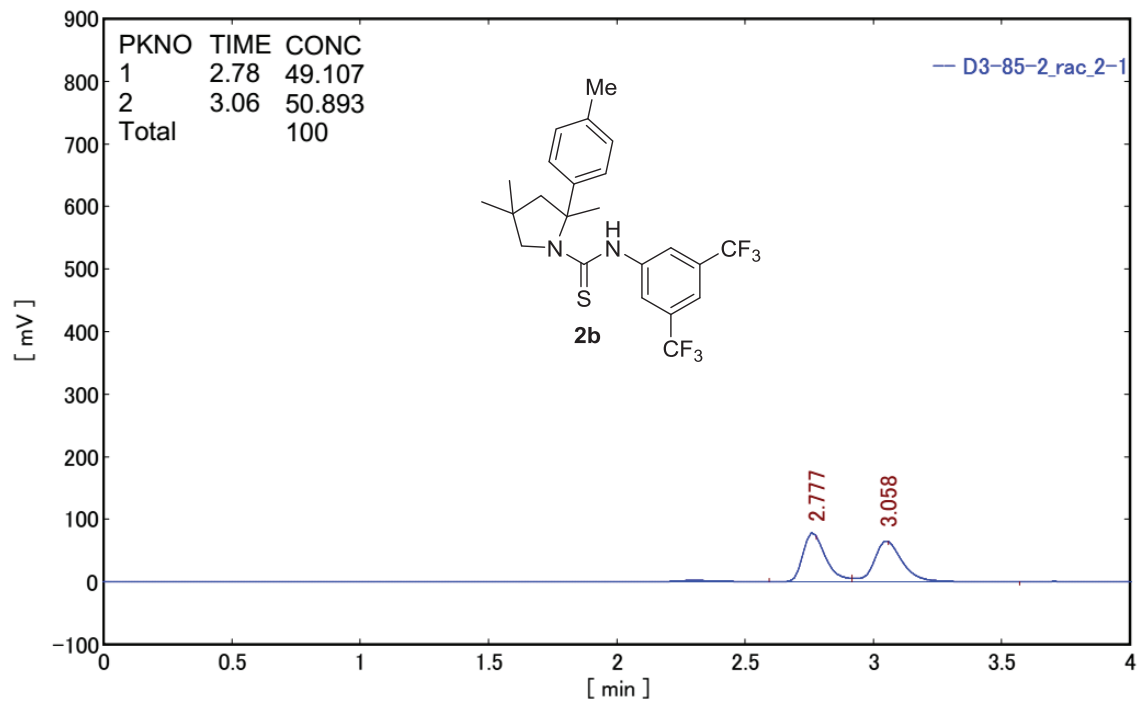


**2a**: 70% ee

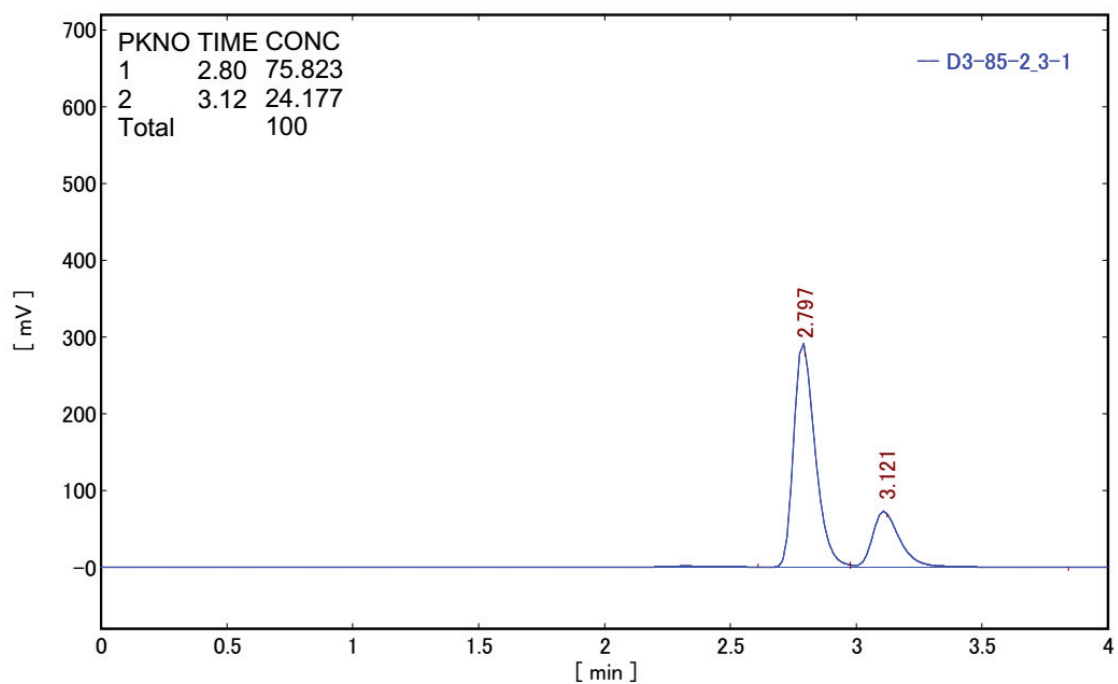


**Table 4***Rac-2b* Chiralpak ID, hexane: EtOH= 20: 1, 1.0 mL/ min 254 nm**2b**: 22% ee (Table 4, entry 1)

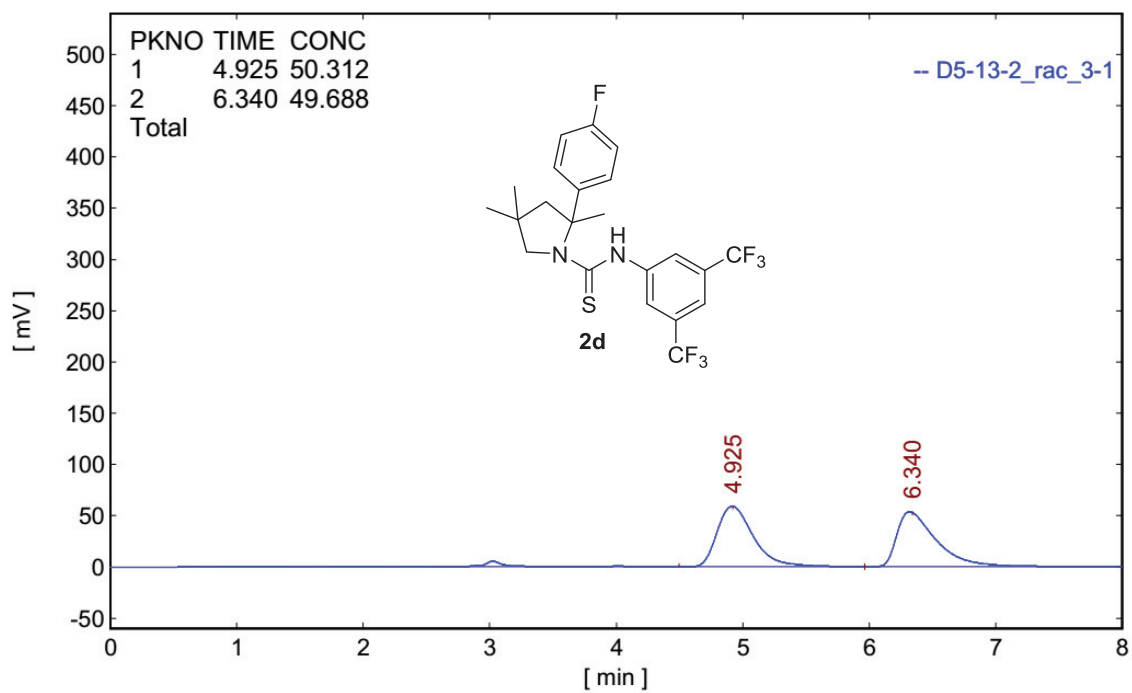
*Rac-2c* Chiralpak ID, hexane: EtOH= 40: 1, 1.5 mL/ min 254 nm



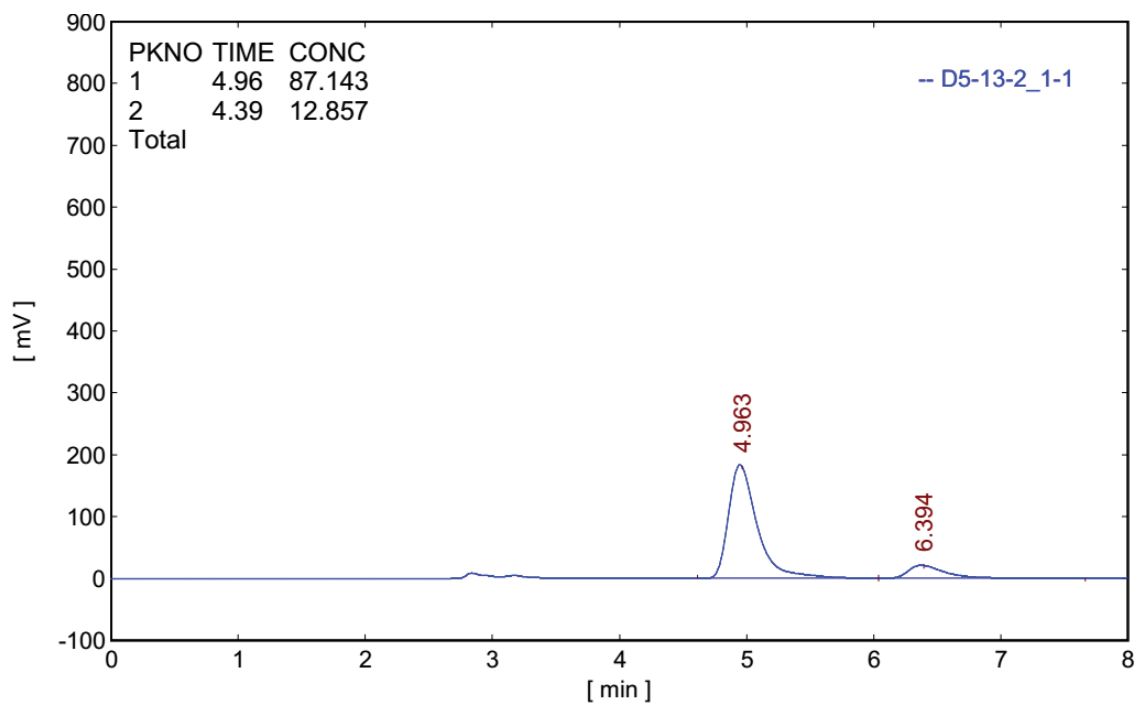
**2c**: 52% ee (Table 4, entry 3)



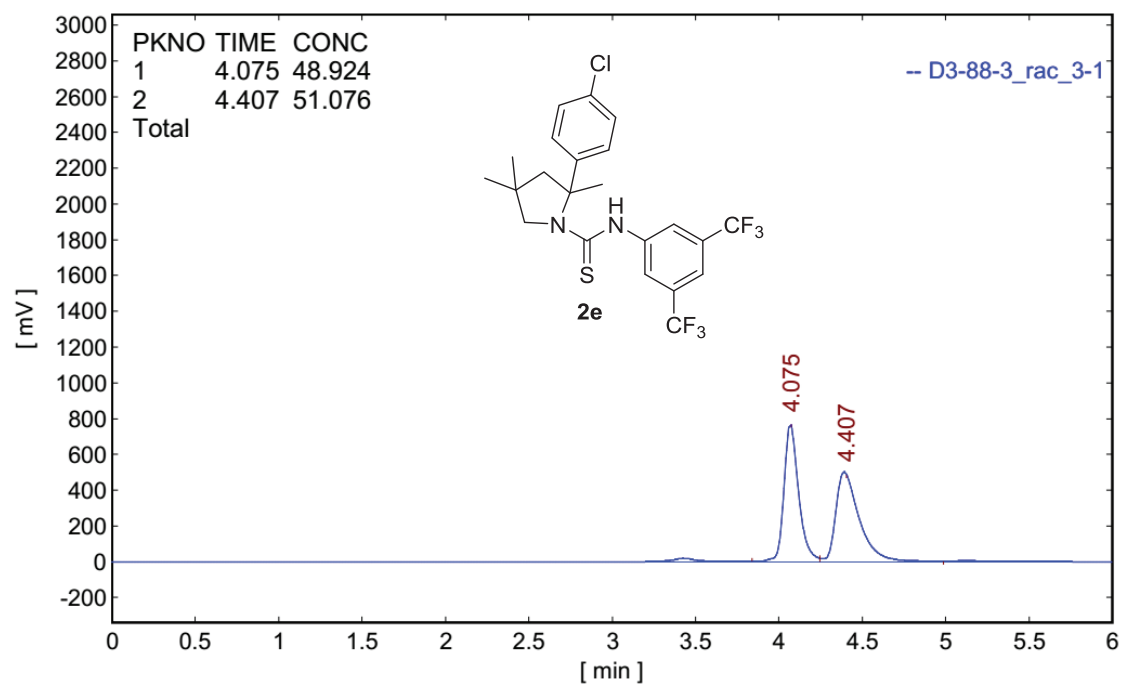
*Rac-2d* Chiralpak IG-3, hexane: EtOH= 80: 1, 1.1 mL/ min 254 nm



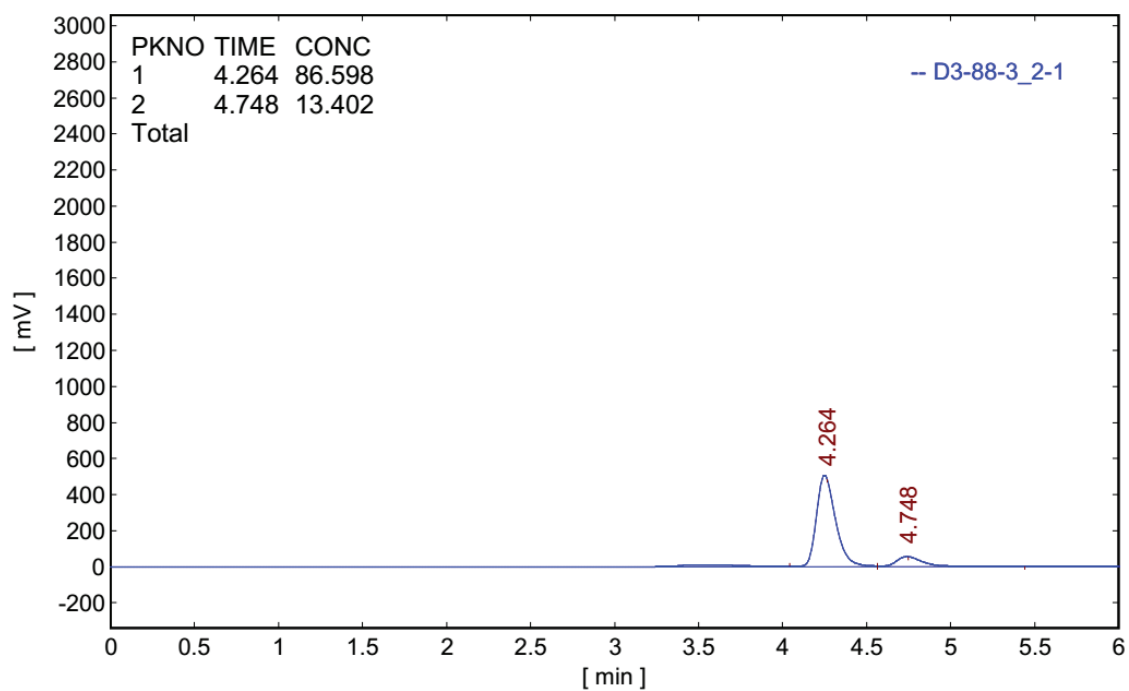
**2d**: 74% ee (Table 4, entry 6)



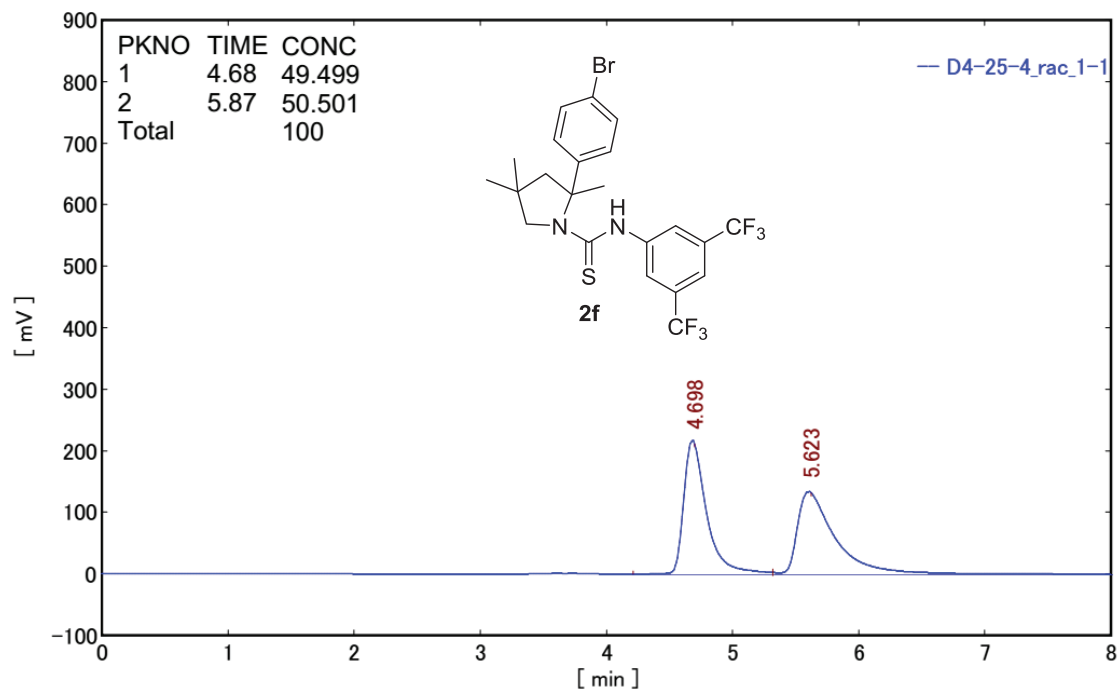
*Rac-2e* Chiralpak ID, hexane: EtOH= 20: 1, 1.0 mL/ min 254 nm



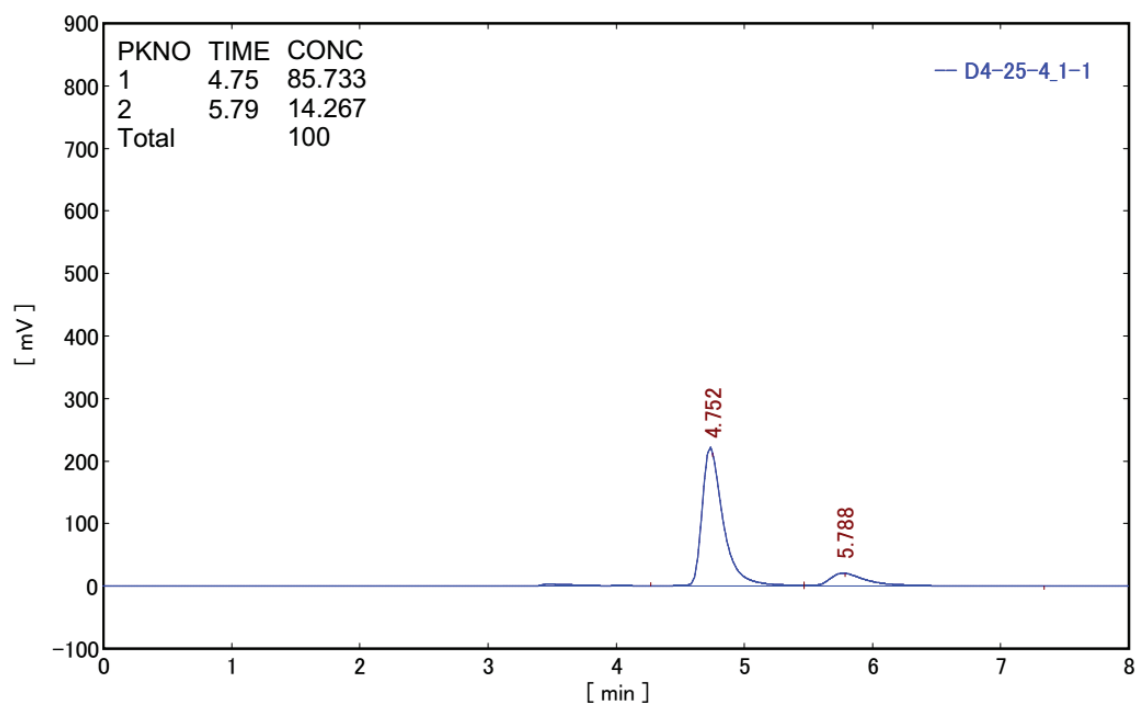
**2e**: 73% ee (Table 4, entry 8)

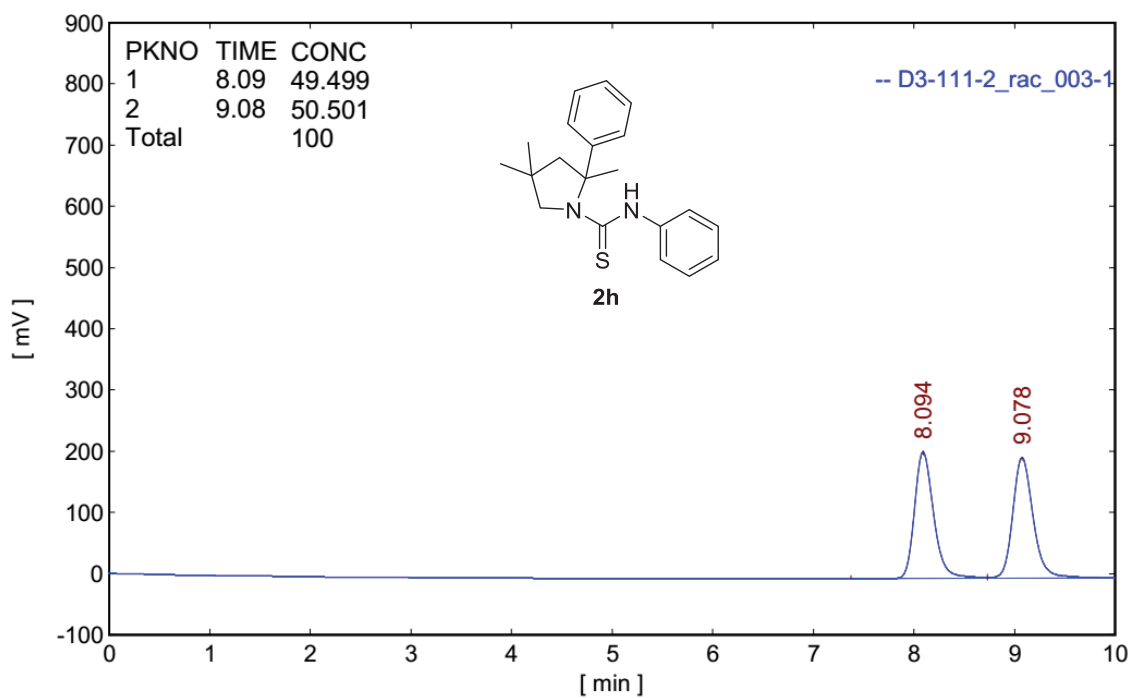
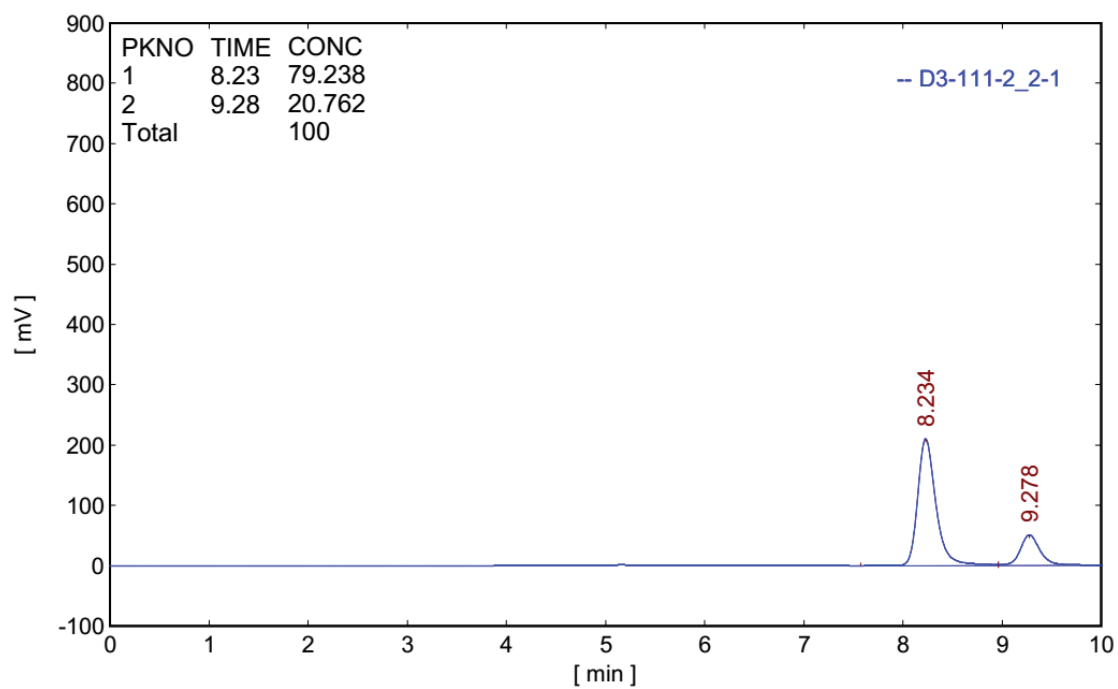


Entry 11: *Rac*-**2f** Chiralpak ID, hexane: EtOH= 40: 1, 1.0 mL/ min 254 nm



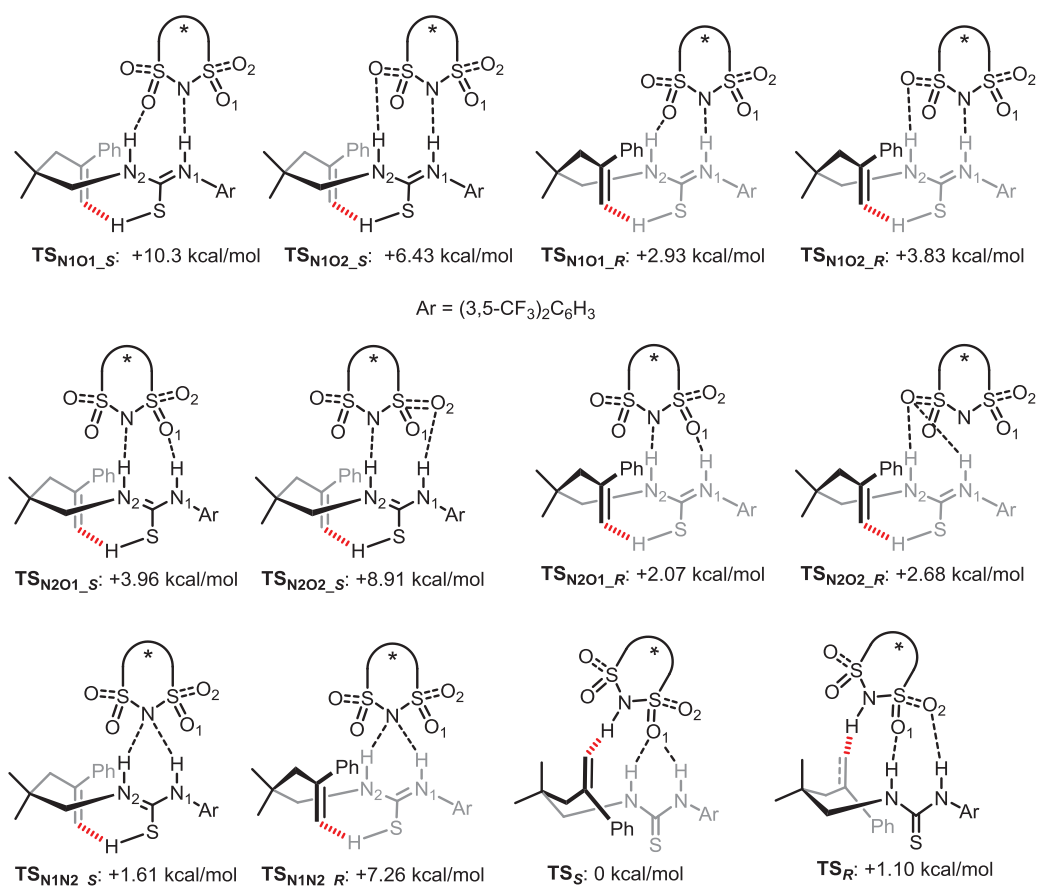
**2f**: 71% ee



**Table 5**Entry 3: *Rac-2h* Chiralpak IB, hexane: EtOH= 50: 1, 0.75 mL/min 254 nm**2h**: 58% ee

## Computational Study

### Summary of the calculated energies

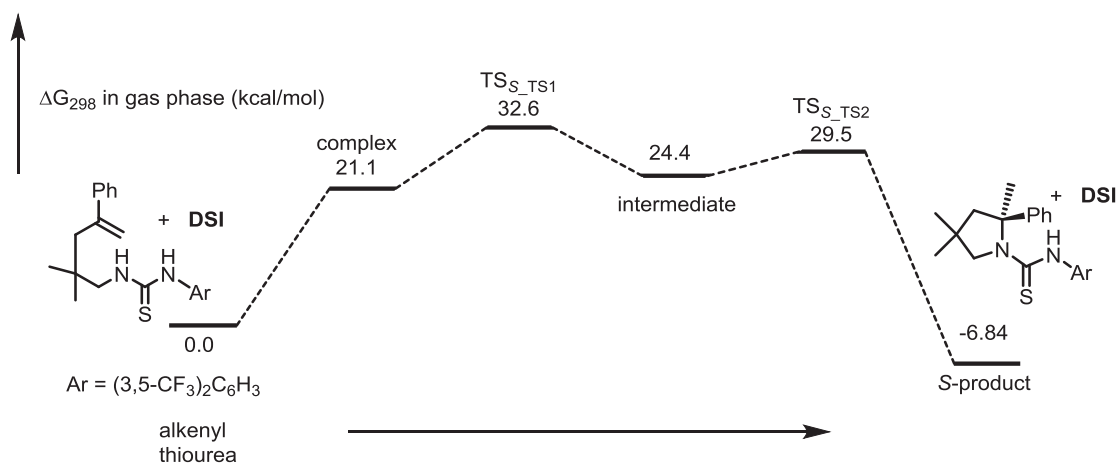


**Figure S1.** Schematic structures and relative energies of possible transition states (298.15 K in toluene) of the **DSI**-catalyzed hydroamination at the M06-2X/6-311+G(d,p)/SMD(toluene)//ONIOM(B3LYP/6-31G(d):HF/3-21G\*) level of theory.



**Table S1.** Electronic energies of possible transition states at 298.15 K in toluene at the M06-2X/6-311+G(d,p)/SMD (toluene)//ONIOM(B3LYP/6-31G(d): HF/3-21G\*) level of theory.

	$E_{\text{sp}}^{\text{toluene}}$ (a.u.)	$\Delta E$ (kcal/mol)		$E_{\text{sp}}^{\text{toluene}}$ (a.u.)	$\Delta E$ (kcal/mol)
<b>TS<sub>N102_S</sub></b>	-7962.461166	+10.312	<b>TS<sub>N101_R</sub></b>	-7962.472932	+2.939
<b>TS<sub>N102_S</sub></b>	-7962.467356	+6.427	<b>TS<sub>N102_R</sub></b>	-7962.471501	+3.826
<b>TS<sub>N201_S</sub></b>	-7962.471295	+3.956	<b>TS<sub>N201_R</sub></b>	-7962.474300	+2.070
<b>TS<sub>N202_S</sub></b>	-7962.463404	+8.907	<b>TS<sub>N202_R</sub></b>	-7962.473332	+2.678
<b>TS<sub>N1N2_S</sub></b>	-7962.475041	+1.605	<b>TS<sub>N1N2_R</sub></b>	-7962.466025	+7.263
<b>TS<sub>S</sub></b>	-7962.477599	0	<b>TS<sub>R</sub></b>	-7962.475851	+1.097



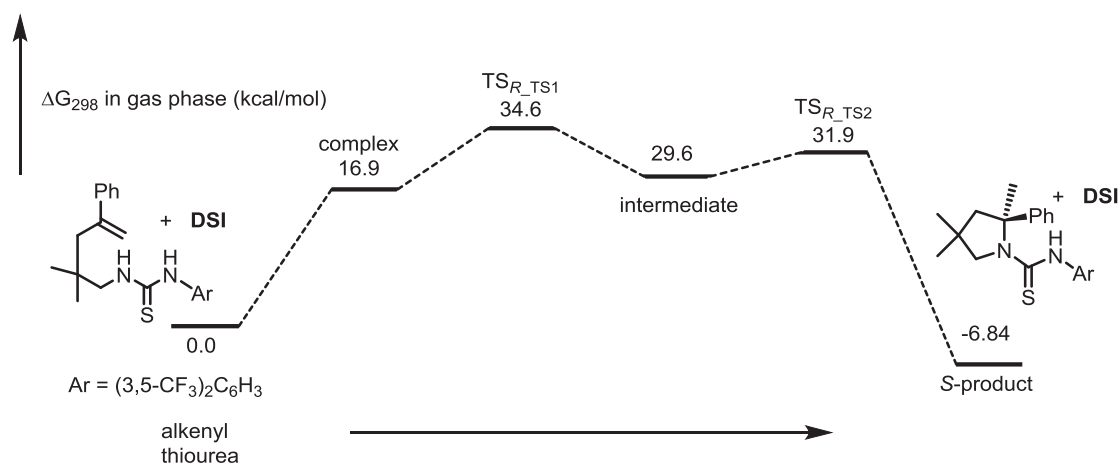
**Figure. S2** Calculated Gibbs energy profile for the DSI-catalyzed hydroamination of alkenyl thiourea via transition state structure TS<sub>S</sub> at 298.15 K in toluene at the M06-2X/6-311+G(d,p)/SMD(toluene)// ONIOM(M06-2X/6-31G(d): HF/3-21G\*) level of theory.

**Table S3.** Calculated Gibbs energies for the **DSI**-catalyzed hydroamination via transition state  $TS_S$  at 298.15 K in toluene at the M06-2X/6-311+G(d,p)/SMD (toluene)//ONIOM(M06-2X/6-31G(d):HF/3-21G\*) level of theory.

	$G_{298}$ (a.u.)	$\Delta G_{298}$ (kcal/mol)
thiourea + <b>DSI</b>	-7961.397859	0
complex	-7961.364227	21.104080
$TS_{S\_TS1}$	-7961.345959	32.567187
intermediate	-7961.358931	24.427320
$TS_{S\_TS2}$	-7961.350799	29.530150
product + <b>DSI</b>	-7961.408841	-6.891205

**Table S4.** Calculated Gibbs energies for the **DSI**-catalyzed hydroamination via transition state  $TS_R$  at 298.15 K in toluene at the M06-2X/6-311+G(d,p)/SMD (toluene)//ONIOM(M06-2X/6-31G(d):HF/3-21G\*) level of theory.

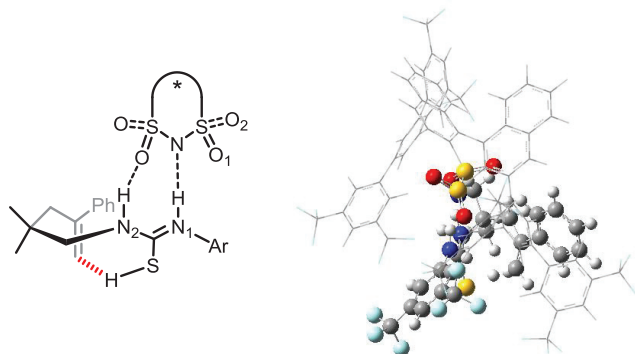
	$G_{298}$ (a.u.)	$\Delta G_{298}$ (kcal/mol)
thiourea + <b>DSI</b>	-7961.397859	0
complex	-7961.370872	16.934343
$TS_{S\_TS1}$	-7961.342649	34.644275
intermediate	-7961.350616	29.6449825
$TS_{S\_TS2}$	-7961.347031	31.89457
product + <b>DSI</b>	-7961.408841	-6.891205



**Figure. S3** Calculated Gibbs energy profile for the **DSI**-catalyzed hydroamination of alkenyl thiourea via transition state structure  $TS_R$  at 298.15 K in toluene at the M06-2X/6-311+G(d,p)/SMD(toluene)//ONIOM(M06-2X/6-31G(d):HF/3-21G\*) level of theory.

## Cartesian coordinates of the structures

TS<sub>N102\_S</sub>



E (ONIOM(B3LYP/6-31G(d):HF/3-21G\*)) = -7911.096155 a.u.

Imaginary Freq = 1: -832.46 cm<sup>-1</sup>

Zero-point Energy Correction = 1.285824

Thermal Correction to Energy = 1.38963

Thermal Correction to Enthalpy = 1.390574

Thermal Correction to Free Energy = 1.120923

EE + Zero-point Energy = -7909.810331

EE + Thermal Energy Correction = -7909.706525

EE + Thermal Enthalpy Correction = -7909.705581

EE + Thermal Free Energy Correction = -7909.975232

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.461166

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.116469	-2.360290	4.167778
2	6	0	2.453948	-1.749989	3.813709
3	6	0	3.496447	-2.470015	2.908859
4	7	0	1.734926	-2.500346	1.042901
5	1	0	2.943414	-1.539375	4.779416
6	1	0	2.273409	-0.760416	3.387729
7	6	0	4.470905	-1.374203	2.416424
8	1	0	4.896800	-0.829084	3.267328

9	1	0	3.984251	-0.650390	1.755945
10	1	0	5.305111	-1.831026	1.870670
11	6	0	4.337357	-3.523205	3.664825
12	1	0	4.945243	-3.044770	4.440955
13	1	0	5.017526	-4.027938	2.968541
14	1	0	3.732845	-4.297631	4.146810
15	6	0	0.037712	-1.436091	4.549535
16	6	0	-2.001987	0.326329	5.377773
17	6	0	-0.926700	-1.830786	5.508309
18	6	0	-0.056514	-0.131433	4.014603
19	6	0	-1.065799	0.735489	4.427201
20	6	0	-1.931514	-0.963125	5.914016
21	1	0	-0.856228	-2.808081	5.974214
22	1	0	0.620360	0.220552	3.244190
23	1	0	-1.108823	1.730631	3.992990
24	1	0	-2.663559	-1.269247	6.652368
25	1	0	-2.787393	0.991204	5.718704
26	6	0	0.890906	-3.738392	4.159106
27	1	0	0.531471	-4.060547	2.883714
28	1	0	1.745871	-4.405007	4.237252
29	6	0	0.593214	-3.090356	0.655781
30	7	0	-0.185045	-2.454550	-0.244374
31	6	0	-1.262617	-3.084465	-0.935368
32	6	0	-3.327344	-4.312927	-2.370913
33	6	0	-1.020664	-4.265226	-1.647798
34	6	0	-2.529070	-2.500291	-0.957807
35	6	0	-3.550908	-3.122304	-1.682894
36	6	0	-2.053567	-4.881203	-2.348655
37	1	0	-0.024705	-4.691845	-1.668575
38	1	0	-2.700346	-1.557128	-0.452307
39	1	0	-4.125616	-4.780444	-2.934218
40	16	0	0.070925	-4.607272	1.352710
41	1	0	-0.011673	-4.113510	4.639012
42	6	0	2.852993	-3.194349	1.697083
43	1	0	2.497203	-4.181127	1.984681
44	1	0	3.633255	-3.365584	0.947902
45	1	0	1.967057	-1.594290	0.607580
46	1	0	-0.015005	-1.446156	-0.434980

47	6	0	-1.204044	2.826374	-0.129441
48	6	0	-0.124194	3.661315	-0.082545
49	6	0	-2.301096	4.088993	1.581992
50	6	0	-0.066618	4.719260	0.873486
51	6	0	-2.337375	3.054762	0.700695
52	6	0	-1.172069	4.925589	1.715037
53	6	0	1.070729	5.556591	1.027441
54	1	0	-1.992661	6.123418	3.300306
55	1	0	-3.160756	4.285066	2.191884
56	6	0	1.081681	6.542617	1.958596
57	1	0	1.922438	5.396443	0.403293
58	1	0	1.946909	7.164471	2.070163
59	6	0	-0.040182	6.759211	2.790738
60	1	0	-0.014231	7.546878	3.516701
61	6	0	-1.136917	5.971731	2.672163
62	6	0	3.144055	2.505634	-1.697018
63	1	0	4.200932	3.495065	-3.217785
64	6	0	3.305692	3.481360	-2.628451
65	6	0	1.043860	3.521624	-1.026178
66	6	0	2.336222	4.482969	-2.850823
67	6	0	1.955031	2.509564	-0.908881
68	6	0	1.195466	4.519383	-2.034159
69	6	0	2.499501	5.456857	-3.867727
70	1	0	-0.658481	5.548272	-1.661968
71	6	0	1.555831	6.410464	-4.062970
72	1	0	3.377362	5.418315	-4.482056
73	1	0	1.677038	7.141998	-4.836379
74	6	0	0.398564	6.440933	-3.252441
75	1	0	-0.346327	7.190653	-3.427402
76	6	0	0.222096	5.526638	-2.266131
77	16	0	1.487330	1.141887	0.173980
78	8	0	1.296677	1.663564	1.541854
79	8	0	2.515466	0.071774	0.065489
80	16	0	-1.060278	1.366280	-1.186302
81	8	0	-0.719623	1.821926	-2.538220
82	8	0	-2.267875	0.544490	-1.034934
83	7	0	0.155304	0.489636	-0.486756
84	6	0	-3.621732	2.285773	0.624776

85	6	0	-6.076301	0.996429	0.566633
86	6	0	-3.971400	1.411907	1.636725
87	6	0	-4.512101	2.528544	-0.408457
88	6	0	-5.736428	1.885014	-0.444052
89	6	0	-5.199590	0.763824	1.613665
90	1	0	-3.279368	1.209678	2.429004
91	1	0	-4.244435	3.205151	-1.195092
92	1	0	-7.032119	0.511328	0.549931
93	6	0	4.314556	1.586004	-1.498879
94	6	0	6.582237	0.025911	-1.169973
95	6	0	4.356850	0.305138	-2.025470
96	6	0	5.411889	2.080962	-0.817620
97	6	0	6.554945	1.307630	-0.650929
98	6	0	5.481255	-0.482272	-1.850567
99	1	0	3.494397	-0.091687	-2.518583
100	1	0	5.369507	3.064872	-0.394955
101	1	0	7.465980	-0.573037	-1.067860
102	6	0	5.488719	-1.897485	-2.310524
103	6	0	5.407890	-4.583601	-3.061977
104	6	0	5.195808	-2.255645	-3.617839
105	6	0	5.748922	-2.901295	-1.392100
106	6	0	5.701946	-4.227325	-1.765432
107	6	0	5.167449	-3.586299	-3.982651
108	1	0	4.984202	-1.501197	-4.347444
109	1	0	5.951581	-2.643486	-0.373842
110	1	0	5.345213	-5.612586	-3.345724
111	6	0	7.729724	1.851637	0.083737
112	6	0	9.907124	2.884336	1.488045
113	6	0	8.288224	3.070621	-0.271609
114	6	0	8.278694	1.156910	1.151023
115	6	0	9.358606	1.671157	1.838023
116	6	0	9.366126	3.574093	0.426008
117	1	0	7.880763	3.627338	-1.090150
118	1	0	7.843517	0.231086	1.466039
119	1	0	10.724231	3.293369	2.043375
120	6	0	-6.694015	2.159097	-1.555414
121	6	0	-8.463203	2.680321	-3.638952
122	6	0	-6.946621	1.195102	-2.519475

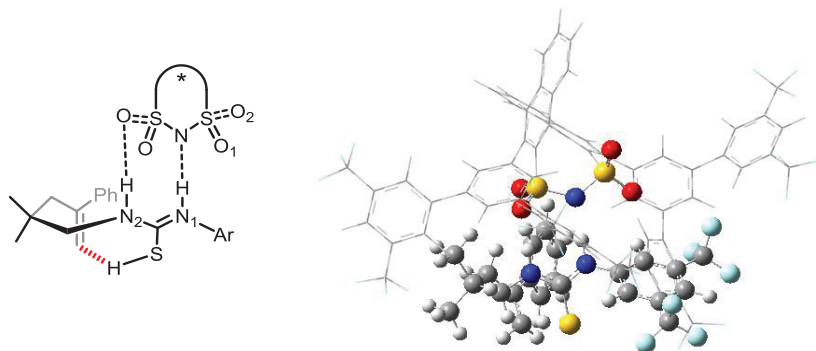
123	6	0	-7.331660	3.384518	-1.646755
124	6	0	-8.210450	3.635887	-2.681313
125	6	0	-7.829864	1.460804	-3.545160
126	1	0	-6.426784	0.262215	-2.480700
127	1	0	-7.137193	4.144674	-0.918214
128	1	0	-9.122618	2.888962	-4.454400
129	6	0	-5.570363	-0.150930	2.730296
130	6	0	-6.226333	-1.840425	4.847551
131	6	0	-5.633481	0.331710	4.027121
132	6	0	-5.843838	-1.492198	2.501894
133	6	0	-6.177083	-2.317979	3.556443
134	6	0	-5.947587	-0.511972	5.072857
135	1	0	-5.451630	1.369828	4.217260
136	1	0	-5.775166	-1.890888	1.511355
137	1	0	-6.475929	-2.490637	5.658337
138	6	0	4.913066	-3.948926	-5.402137
139	6	0	5.981799	-5.266525	-0.742699
140	6	0	9.979771	4.857187	-0.004780
141	6	0	9.952390	0.889099	2.953216
142	6	0	-5.932533	0.018997	6.456875
143	6	0	-6.520435	-3.740127	3.292324
144	6	0	-8.922027	4.937848	-2.734370
145	6	0	-8.126672	0.404180	-4.546464
146	6	0	-1.801428	-6.185993	-3.059507
147	6	0	-4.925548	-2.511869	-1.684104
148	9	0	4.350478	-5.171396	-5.502353
149	9	0	4.099323	-3.053094	-5.999550
150	9	0	6.050291	-3.983917	-6.133381
151	9	0	7.302682	-5.455413	-0.532971
152	9	0	5.462211	-4.910189	0.463967
153	9	0	5.459276	-6.459507	-1.084630
154	9	0	9.012412	0.164970	3.600412
155	9	0	10.884997	0.007571	2.525917
156	9	0	10.560797	1.692614	3.850554
157	9	0	10.577082	5.487775	1.028494
158	9	0	9.056369	5.689624	-0.533804
159	9	0	10.927497	4.684055	-0.953649
160	9	0	-8.501856	0.938703	-5.729335



161	9	0	-9.137012	-0.406776	-4.154225
162	9	0	-7.058063	-0.390979	-4.759977
163	9	0	-8.159134	5.937493	-2.238282
164	9	0	-9.271111	5.260332	-3.998146
165	9	0	-10.063668	4.935898	-2.007414
166	9	0	-5.781256	-4.250450	2.286843
167	9	0	-7.816087	-3.901336	2.940372
168	9	0	-6.321416	-4.502102	4.391668
169	9	0	-6.652666	-0.745421	7.300416
170	9	0	-4.670190	0.073167	6.972888
171	9	0	-6.401651	1.282941	6.515009
172	9	0	-5.569845	-2.727500	-0.500339
173	9	0	-5.717188	-3.013535	-2.648069
174	9	0	-4.880604	-1.170139	-1.849463
175	9	0	-2.589074	-6.320313	-4.147139
176	9	0	-2.051724	-7.243303	-2.255518
177	9	0	-0.517394	-6.292525	-3.465501

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**TS<sub>N102\_S</sub>**



E (ONIOM(B3LYP/6-31G(d):HF/3-21G\*)) = -7911.104351 a.u.

Imaginary Freq = 1: -820.27 cm<sup>-1</sup>

Electronic Energy (EE) = -7911.1044

Zero-point Energy Correction = 1.286352

Thermal Correction to Energy = 1.389699

Thermal Correction to Enthalpy = 1.390643

Thermal Correction to Free Energy = 1.127158

EE + Zero-point Energy = -7909.817999

EE + Thermal Energy Correction = -7909.714652  
 EE + Thermal Enthalpy Correction = -7909.713708  
 EE + Thermal Free Energy Correction = -7909.977193

$E_{sp}$  (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.467356

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.892911	-3.533619	-3.588126
2	6	0	2.130445	-3.701446	-2.738272
3	6	0	3.463357	-2.953514	-3.009685
4	7	0	2.228198	-0.776128	-2.460976
5	1	0	2.351734	-4.781263	-2.749325
6	1	0	1.854418	-3.503280	-1.698602
7	6	0	4.328824	-3.158751	-1.742886
8	1	0	4.392258	-4.224249	-1.491130
9	1	0	3.921383	-2.633185	-0.873597
10	1	0	5.346744	-2.806512	-1.926345
11	6	0	4.262472	-3.537769	-4.196076
12	1	0	4.448096	-4.606172	-4.036484
13	1	0	5.239741	-3.047913	-4.270524
14	1	0	3.761262	-3.429065	-5.162916
15	6	0	-0.364088	-4.049615	-3.024155
16	6	0	-2.718055	-5.215070	-1.995160
17	6	0	-0.581391	-4.138523	-1.629370
18	6	0	-1.372182	-4.549058	-3.884030
19	6	0	-2.533308	-5.117046	-3.377098
20	6	0	-1.742226	-4.722030	-1.127486
21	1	0	0.126070	-3.727799	-0.917019
22	1	0	-1.220197	-4.538008	-4.957986
23	1	0	-3.299382	-5.500247	-4.041502
24	1	0	-1.873947	-4.783293	-0.050865
25	1	0	-3.622367	-5.680967	-1.620854
26	6	0	0.919817	-2.940119	-4.852664
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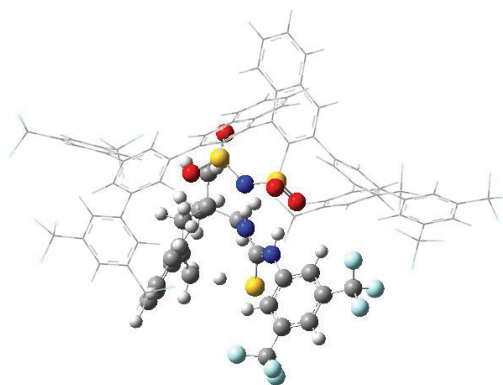
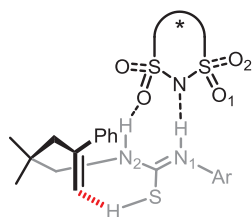
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174	9	0	11.094723	-3.678603	1.732207
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TS<sub>N101\_R</sub>



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.129054 a.u.

Imaginary Freq = 1: -842.02 cm<sup>-1</sup>

Zero-point Energy Correction = 1.28867

Thermal Correction to Energy = 1.391795

Thermal Correction to Enthalpy = 1.39274

Thermal Correction to Free Energy = 1.128899

EE + Zero-point Energy = -7909.840385

EE + Thermal Energy Correction = -7909.737259

EE + Thermal Enthalpy Correction = -7909.736315

EE + Thermal Free Energy Correction = -7910.000155

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.472932

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
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2	6	0	-1.988307	2.958284	-2.547188
3	6	0	-0.718230	3.817402	-2.262122
4	7	0	0.558315	1.599211	-2.097809
5	1	0	-2.840391	3.650295	-2.628113
6	1	0	-2.206746	2.378660	-1.647457
7	6	0	-0.809121	4.284579	-0.791775
8	1	0	-1.768677	4.761423	-0.582117
9	1	0	-0.691423	3.465568	-0.077594
10	1	0	-0.021126	5.018990	-0.585599

11	6	0	-0.626567	5.093582	-3.128436
12	1	0	-1.456960	5.769174	-2.895255
13	1	0	0.310024	5.617927	-2.913129
14	1	0	-0.646849	4.899255	-4.204539
15	6	0	-2.835741	0.783683	-3.608787
16	6	0	-4.173436	-1.697484	-3.434017
17	6	0	-3.525099	0.245410	-4.719962
18	6	0	-2.859736	0.058635	-2.398433
19	6	0	-3.506908	-1.171156	-2.326004
20	6	0	-4.190717	-0.971487	-4.629445
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23	1	0	-3.478628	-1.711625	-1.386068
24	1	0	-4.724307	-1.358391	-5.492728
25	1	0	-4.673133	-2.659294	-3.372696
26	6	0	-1.396795	2.278883	-4.934047
27	1	0	-1.034541	3.278868	-5.150242
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39	1	0	2.744086	-5.402794	-3.789538
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53	1	0	-3.083810	-3.335183	6.203653
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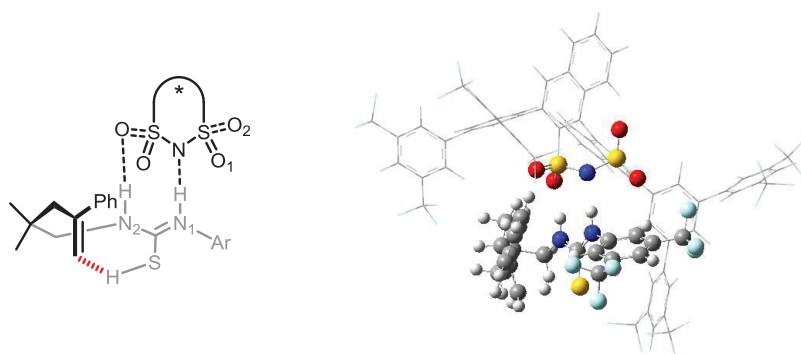
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97	6	0	5.303816	1.849473	0.288074
98	1	0	3.579498	2.832205	1.081107
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141	6	0	6.582957	-5.215167	1.313109
142	6	0	8.495216	5.126467	-1.683032
143	6	0	3.776822	5.034036	-3.192260
144	6	0	-5.792488	-5.389645	-1.903312
145	6	0	-10.097966	-4.141388	0.207214
146	6	0	-9.530151	4.368726	0.987007
147	6	0	-4.907429	5.418364	-0.434191
148	9	0	5.390776	-5.358521	1.925278
149	9	0	7.439181	-6.112210	1.851882
150	9	0	6.403517	-5.581698	0.022591
151	9	0	10.939283	-3.135838	2.691228
152	9	0	10.586264	-0.975320	2.717686
153	9	0	11.061366	-1.981929	0.860993
154	9	0	8.763410	5.651149	-2.897444
155	9	0	9.450158	4.218229	-1.390017
156	9	0	8.632921	6.131760	-0.788596
157	9	0	3.244319	4.094780	-4.008274
158	9	0	4.176542	6.064988	-3.959136
159	9	0	2.740358	5.472078	-2.426067
160	9	0	-4.560465	-5.640211	-1.414583
161	9	0	-5.600175	-4.782843	-3.107324
162	9	0	-6.402920	-6.562988	-2.155665

163	9	0	-10.662092	-4.782983	-0.837948
164	9	0	-10.315678	-4.910932	1.297687
165	9	0	-10.770366	-2.987935	0.411492
166	9	0	-5.415251	6.657704	-0.583668
167	9	0	-4.498320	5.010239	-1.665774
168	9	0	-3.779090	5.518741	0.306109
169	9	0	-9.666821	5.703091	1.140214
170	9	0	-10.113441	3.757813	2.041084
171	9	0	-10.258277	4.028912	-0.101992
172	9	0	3.997703	-4.319690	-0.309750
173	9	0	4.270101	-5.765850	-1.926753
174	9	0	5.376723	-3.886524	-1.932018
175	9	0	0.754234	-5.713216	-5.253450
176	9	0	-0.471520	-3.935223	-5.553441
177	9	0	1.472586	-4.094193	-6.510732

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**TS<sub>N102\_R</sub>**



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.100988 a.u.

Imaginary Freq = 1: -671.87 cm<sup>-1</sup>

Electronic Energy (EE) = -7911.101

Zero-point Energy Correction = 1.28567

Thermal Correction to Energy = 1.389474

Thermal Correction to Enthalpy = 1.390418

Thermal Correction to Free Energy = 1.119879

EE + Zero-point Energy = -7909.815318

EE + Thermal Energy Correction = -7909.711514

EE + Thermal Enthalpy Correction = -7909.710570

EE + Thermal Free Energy Correction = -7909.981109

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.471501

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.194573	4.371739	0.697289
2	6	0	2.625199	3.927167	-0.678685
3	6	0	1.780721	4.218255	-1.959287
4	7	0	-0.026130	2.733937	-0.932067
5	1	0	3.624415	4.360325	-0.840477
6	1	0	2.810275	2.850085	-0.642182
7	6	0	2.366523	3.319909	-3.074389
8	1	0	3.429202	3.528514	-3.225035
9	1	0	2.243013	2.255945	-2.854262
10	1	0	1.857259	3.521282	-4.024437
11	6	0	1.903236	5.676080	-2.455570
12	1	0	2.952764	5.926038	-2.646535
13	1	0	1.361602	5.796444	-3.401149
14	1	0	1.500966	6.418818	-1.761197
15	6	0	2.607678	3.573422	1.859471
16	6	0	3.220142	2.059032	4.160813
17	6	0	2.862525	4.201144	3.102666
18	6	0	2.698505	2.164412	1.796376
19	6	0	2.979994	1.425096	2.940850
20	6	0	3.176452	3.454718	4.232544
21	1	0	2.851939	5.284349	3.169770
22	1	0	2.495156	1.622431	0.879627
23	1	0	2.993242	0.343709	2.885985
24	1	0	3.380052	3.960485	5.171730
25	1	0	3.433355	1.457298	5.037915
26	6	0	1.347280	5.466315	0.907071
27	1	0	1.255735	6.218759	0.129777
28	1	0	1.260957	5.856958	1.920534
29	6	0	-0.782978	2.710694	0.178818
30	7	0	-1.050301	1.510819	0.729337
31	6	0	-1.682622	1.310409	1.997742

32	6	0	-2.904270	0.847876	4.470277
33	6	0	-1.082761	1.776072	3.171837
34	6	0	-2.871614	0.586932	2.055112
35	6	0	-3.478773	0.366985	3.296566
36	6	0	-1.698263	1.548701	4.400020
37	1	0	-0.138985	2.304651	3.123310
38	1	0	-3.299864	0.187726	1.143549
39	1	0	-3.379100	0.670326	5.427322
40	16	0	-1.423711	4.168578	0.896366
41	1	0	0.145967	4.897284	0.853870
42	6	0	0.263547	3.902334	-1.766736
43	1	0	-0.264845	4.751876	-1.335750
44	1	0	-0.171511	3.723872	-2.758040
45	1	0	0.287752	1.836054	-1.329649
46	6	0	1.789581	-2.158077	-2.199725
47	6	0	0.842918	-2.684332	-3.034921
48	6	0	3.172965	-4.092196	-2.478857
49	6	0	1.004492	-3.998241	-3.566382
50	6	0	2.999803	-2.861416	-1.933248
51	6	0	2.182367	-4.705491	-3.275809
52	6	0	0.003926	-4.632894	-4.349568
53	1	0	3.262735	-6.540754	-3.564849
54	1	0	4.089436	-4.618700	-2.300523
55	6	0	0.190911	-5.891260	-4.819343
56	1	0	-0.905053	-4.110665	-4.554009
57	1	0	-0.574270	-6.365611	-5.399860
58	6	0	1.386695	-6.591921	-4.542245
59	1	0	1.516325	-7.584688	-4.923914
60	6	0	2.356027	-6.015013	-3.790537
61	6	0	-2.562761	-0.939370	-3.051308
62	1	0	-3.509874	-0.059458	-4.707555
63	6	0	-2.628363	-0.556569	-4.353986
64	6	0	-0.376195	-1.904569	-3.458360
65	6	0	-1.570189	-0.781218	-5.258710
66	6	0	-1.401282	-1.626918	-2.597816
67	6	0	-0.421286	-1.454217	-4.811989
68	6	0	-1.640889	-0.331339	-6.602370
69	1	0	1.540881	-2.130922	-5.387997

70	6	0	-0.602273	-0.529523	-7.449828
71	1	0	-2.529410	0.171694	-6.929906
72	1	0	-0.656203	-0.186624	-8.463553
73	6	0	0.565300	-1.186402	-6.998137
74	1	0	1.384885	-1.325811	-7.673968
75	6	0	0.655184	-1.636569	-5.722097
76	16	0	-1.126836	-2.026564	-0.850641
77	8	0	-0.716871	-3.431431	-0.776676
78	8	0	-2.300196	-1.614642	-0.071313
79	16	0	1.366854	-0.645886	-1.311587
80	8	0	1.045341	0.406389	-2.320241
81	8	0	2.465056	-0.276070	-0.399609
82	7	0	0.063371	-0.973430	-0.401363
83	6	0	4.149121	-2.299383	-1.153854
84	6	0	6.360925	-1.348634	0.218538
85	6	0	4.420875	-2.755067	0.122952
86	6	0	4.987656	-1.368430	-1.743798
87	6	0	6.090858	-0.886413	-1.060507
88	6	0	5.522547	-2.275522	0.817123
89	1	0	3.763700	-3.465778	0.583016
90	1	0	4.773984	-1.011681	-2.731766
91	1	0	7.190876	-0.947847	0.766283
92	6	0	-3.750848	-0.612000	-2.198857
93	6	0	-5.988543	0.051405	-0.703134
94	6	0	-3.981723	0.701917	-1.836358
95	6	0	-4.656217	-1.593651	-1.830827
96	6	0	-5.771596	-1.267430	-1.080791
97	6	0	-5.101795	1.045405	-1.086893
98	1	0	-3.273607	1.457628	-2.113123
99	1	0	-4.475764	-2.614531	-2.100243
100	1	0	-6.859359	0.301155	-0.131308
101	1	0	-0.714068	0.644593	0.260677
102	6	0	6.938871	0.184601	-1.657258
103	6	0	8.501046	2.230678	-2.723290
104	6	0	6.404514	1.446213	-1.861840
105	6	0	8.265561	-0.041475	-1.989895
106	6	0	9.033429	0.977251	-2.514312
107	6	0	7.183901	2.455106	-2.392712

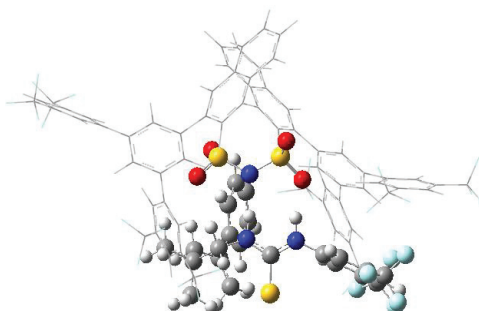
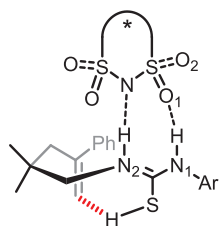
108	1	0	5.385693	1.635004	-1.588810
109	1	0	8.692971	-1.013657	-1.853675
110	1	0	9.100523	3.013560	-3.135674
111	6	0	5.748723	-2.655751	2.239340
112	6	0	6.168904	-3.195897	4.940536
113	6	0	6.915609	-3.273381	2.659974
114	6	0	4.792533	-2.323453	3.184762
115	6	0	5.007965	-2.586402	4.520871
116	6	0	7.115174	-3.538786	3.999587
117	1	0	7.660976	-3.556742	1.945387
118	1	0	3.891968	-1.838944	2.869360
119	1	0	6.334482	-3.397453	5.977235
120	6	0	-5.337759	2.466847	-0.710269
121	6	0	-5.776207	5.133003	-0.024650
122	6	0	-5.271197	3.465497	-1.669192
123	6	0	-5.628206	2.820427	0.601218
124	6	0	-5.840816	4.142179	0.930018
125	6	0	-5.484096	4.783749	-1.322628
126	1	0	-5.081668	3.212976	-2.692124
127	1	0	-5.685687	2.067513	1.358142
128	1	0	-5.968930	6.152150	0.235136
129	6	0	-6.740291	-2.326731	-0.675708
130	6	0	-8.555730	-4.292070	0.086817
131	6	0	-7.524819	-2.963813	-1.621483
132	6	0	-6.865129	-2.687424	0.657450
133	6	0	-7.765307	-3.665493	1.025011
134	6	0	-8.428208	-3.934004	-1.235767
135	1	0	-7.433779	-2.703900	-2.656200
136	1	0	-6.250759	-2.220247	1.396143
137	1	0	-9.255383	-5.045181	0.381291
138	6	0	-5.359367	5.837512	-2.360681
139	6	0	-6.089170	4.515318	2.346589
140	6	0	-9.310982	-4.554924	-2.255067
141	6	0	-7.847429	-4.086237	2.447760
142	6	0	-4.771103	-0.402015	3.342973
143	6	0	-1.082463	2.106681	5.656643
144	6	0	3.965465	-2.200303	5.506327
145	6	0	8.397388	-4.148228	4.438268



146	6	0	10.468478	0.732451	-2.812983
147	6	0	6.578783	3.790318	-2.622643
148	9	0	-6.086890	6.930994	-2.049675
149	9	0	-5.763025	5.390252	-3.570899
150	9	0	-4.081748	6.257516	-2.522301
151	9	0	-6.864677	5.619351	2.432573
152	9	0	-6.697633	3.517524	3.021845
153	9	0	-4.948973	4.796561	3.016660
154	9	0	-8.689204	-4.652982	-3.451359
155	9	0	-10.438978	-3.838924	-2.471779
156	9	0	-9.705930	-5.790901	-1.881618
157	9	0	-9.078700	-4.558305	2.746895
158	9	0	-6.975048	-5.073571	2.750451
159	9	0	-7.574991	-3.060321	3.281486
160	9	0	-5.287916	-0.486310	4.578555
161	9	0	-4.626587	-1.661720	2.866142
162	9	0	-5.722736	0.174203	2.545990
163	9	0	-1.392237	1.361660	6.738576
164	9	0	-1.510680	3.364509	5.905887
165	9	0	0.266760	2.162450	5.570636
166	9	0	2.928868	-3.062673	5.542687
167	9	0	4.467532	-2.126544	6.755476
168	9	0	3.434849	-0.987347	5.198164
169	9	0	8.249045	-4.833317	5.591386
170	9	0	8.887092	-4.989768	3.502289
171	9	0	9.358448	-3.221473	4.659021
172	9	0	11.261712	0.945071	-1.738317
173	9	0	10.678941	-0.544488	-3.198102
174	9	0	10.913393	1.550836	-3.790091
175	9	0	7.508839	4.733573	-2.863030
176	9	0	5.842334	4.195101	-1.555717
177	9	0	5.722673	3.799935	-3.675455

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TS<sub>N201\_S</sub>



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.099991 a.u.

Imaginary Freq = 1: -931.48 cm<sup>-1</sup>

Zero-point Energy Correction = 1.287466

Thermal Correction to Energy = 1.390238

Thermal Correction to Enthalpy = 1.391182

Thermal Correction to Free Energy = 1.131316

EE + Zero-point Energy = -7909.812525

EE + Thermal Energy Correction = -7909.709753

EE + Thermal Enthalpy Correction = -7909.708809

EE + Thermal Free Energy Correction = -7909.968675

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.471295

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.507610	-1.271484	-4.325260
2	6	0	0.985223	-1.030564	-4.229379
3	6	0	1.573180	0.407607	-4.150494
4	7	0	-0.156681	0.837848	-2.181509
5	1	0	1.417923	-1.501460	-5.126483
6	1	0	1.389204	-1.601946	-3.391117
7	6	0	3.107994	0.232950	-4.019145
8	1	0	3.517409	-0.263501	-4.906830
9	1	0	3.374555	-0.354420	-3.134213
10	1	0	3.588154	1.213491	-3.944748
11	6	0	1.350046	1.264676	-5.418001
12	1	0	1.629592	0.707521	-6.320641

13	1	0	1.990849	2.149460	-5.363195
14	1	0	0.328191	1.625800	-5.531970
15	6	0	-1.079709	-2.399227	-3.565089
16	6	0	-2.077775	-4.693971	-2.260891
17	6	0	-2.163269	-3.132908	-4.102624
18	6	0	-0.523690	-2.843763	-2.346809
19	6	0	-1.018196	-3.975786	-1.702676
20	6	0	-2.654422	-4.262469	-3.459164
21	1	0	-2.584812	-2.848313	-5.060949
22	1	0	0.289470	-2.311542	-1.868482
23	1	0	-0.559214	-4.284929	-0.767980
24	1	0	-3.474306	-4.829604	-3.884831
25	1	0	-2.458483	-5.595527	-1.790828
26	6	0	-1.293412	-0.516655	-5.185276
27	1	0	-0.816784	-0.033565	-6.034432
28	1	0	-2.330322	-0.804147	-5.345101
29	6	0	-1.310156	1.516131	-2.296251
30	7	0	-2.086844	1.639354	-1.208152
31	6	0	-3.373456	2.267283	-1.181447
32	6	0	-5.871891	3.508139	-1.078963
33	6	0	-4.496095	1.603263	-1.674244
34	6	0	-3.495965	3.536038	-0.609873
35	6	0	-4.746634	4.154779	-0.570951
36	6	0	-5.742525	2.228023	-1.616454
37	1	0	-4.389871	0.612845	-2.100815
38	1	0	-2.626451	4.035702	-0.200966
39	1	0	-6.839007	3.998008	-1.059489
40	16	0	-1.825461	2.163868	-3.826327
41	1	0	-1.498267	0.741749	-4.512871
42	6	0	1.115996	1.179234	-2.864447
43	1	0	1.120861	2.250303	-3.070241
44	1	0	1.855888	0.988704	-2.084477
45	1	0	0.015306	0.422846	-1.248843
46	6	0	2.735103	-0.992511	2.147328
47	6	0	1.937340	-1.500733	3.134293
48	6	0	4.335731	-0.268065	3.773063
49	6	0	2.281560	-1.326055	4.507805
50	6	0	3.985719	-0.387173	2.465547

51	6	0	3.496061	-0.697186	4.821992
52	6	0	1.432316	-1.735729	5.570292
53	1	0	4.789003	-0.029102	6.403550
54	1	0	5.279149	0.176364	4.020255
55	6	0	1.796053	-1.533116	6.860955
56	1	0	0.496196	-2.195973	5.341401
57	1	0	1.143779	-1.836218	7.654870
58	6	0	3.028536	-0.915601	7.173240
59	1	0	3.298381	-0.766156	8.199329
60	6	0	3.856025	-0.507840	6.180331
61	6	0	-1.657572	-2.476804	2.232035
62	1	0	-2.520470	-4.373862	2.500263
63	6	0	-1.617236	-3.799707	2.552804
64	6	0	0.686545	-2.292097	2.841361
65	6	0	-0.434883	-4.447659	2.963149
66	6	0	-0.460002	-1.710615	2.371590
67	6	0	0.738880	-3.692686	3.107280
68	6	0	-0.410906	-5.839952	3.234117
69	1	0	2.836436	-3.796567	3.583370
70	6	0	0.742067	-6.446721	3.607049
71	1	0	-1.319258	-6.400580	3.132065
72	1	0	0.760999	-7.499411	3.805842
73	6	0	1.932476	-5.694045	3.729624
74	1	0	2.840416	-6.188488	4.010225
75	6	0	1.933311	-4.358881	3.491781
76	16	0	2.071028	-1.026707	0.465729
77	8	0	2.993509	-0.324105	-0.444707
78	8	0	1.755742	-2.424587	0.119793
79	16	0	-0.265325	0.008402	1.819064
80	8	0	-1.556254	0.512399	1.272626
81	8	0	0.259162	0.783272	2.947913
82	7	0	0.718234	-0.083694	0.523550
83	6	0	5.013625	0.068003	1.471318
84	6	0	7.064460	0.938609	-0.178487
85	6	0	6.126457	-0.725691	1.262836
86	6	0	4.931807	1.305895	0.857178
87	6	0	5.947565	1.740182	0.024445
88	6	0	7.160576	-0.296532	0.438520

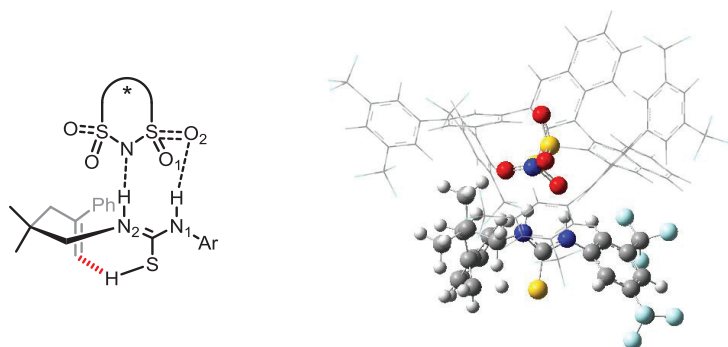
89	1	0	6.196659	-1.675730	1.753899
90	1	0	4.069808	1.919508	1.018520
91	1	0	7.833644	1.260051	-0.852940
92	6	0	-2.989395	-1.938109	1.817726
93	6	0	-5.523087	-1.051890	1.091014
94	6	0	-3.688297	-2.561639	0.799065
95	6	0	-3.576526	-0.879253	2.483329
96	6	0	-4.813304	-0.397909	2.091136
97	6	0	-4.976243	-2.159461	0.455644
98	1	0	-3.208686	-3.339574	0.243337
99	1	0	-3.056092	-0.401613	3.285648
100	1	0	-6.477863	-0.669394	0.790997
101	1	0	-1.736641	1.279087	-0.302131
102	6	0	8.345844	-1.166625	0.207574
103	6	0	10.564143	-2.799928	-0.227600
104	6	0	8.186475	-2.486886	-0.188771
105	6	0	9.631504	-0.676752	0.382154
106	6	0	10.723996	-1.491896	0.170552
107	6	0	9.288596	-3.286786	-0.407109
108	1	0	7.203645	-2.885775	-0.335412
109	1	0	9.780592	0.337493	0.690642
110	1	0	11.414563	-3.426259	-0.394380
111	6	0	5.810879	3.021853	-0.718811
112	6	0	5.557567	5.359073	-2.216090
113	6	0	6.789033	4.003997	-0.666046
114	6	0	4.700490	3.229024	-1.521043
115	6	0	4.587604	4.384398	-2.268586
116	6	0	6.651559	5.162001	-1.401610
117	1	0	7.655503	3.865297	-0.052664
118	1	0	3.944096	2.471284	-1.568214
119	1	0	5.468771	6.248652	-2.801691
120	6	0	-5.303522	0.869379	2.695160
121	6	0	-6.132318	3.304442	3.765299
122	6	0	-6.607699	1.037431	3.132613
123	6	0	-4.418115	1.934691	2.801011
124	6	0	-4.835975	3.134477	3.336111
125	6	0	-7.008993	2.245862	3.664102
126	1	0	-7.309353	0.231851	3.060773

127	1	0	-3.414856	1.818323	2.438126
128	1	0	-6.456360	4.246010	4.153080
129	6	0	-5.712192	-2.909689	-0.602570
130	6	0	-7.011818	-4.352188	-2.620995
131	6	0	-5.511989	-4.275023	-0.752296
132	6	0	-6.599109	-2.285524	-1.470955
133	6	0	-7.235161	-3.002433	-2.461791
134	6	0	-6.140777	-4.976169	-1.758816
135	1	0	-4.878767	-4.802689	-0.070925
136	1	0	-6.807753	-1.245877	-1.376654
137	1	0	-7.519307	-4.904871	-3.382467
138	6	0	-6.970163	1.487222	-2.068220
139	6	0	-4.915878	5.521159	0.046712
140	6	0	-3.853217	4.242532	3.453845
141	6	0	-8.393307	2.391056	4.182399
142	6	0	-5.813067	-6.410931	-1.939187
143	6	0	-8.131884	-2.296837	-3.416386
144	6	0	9.096686	-4.675508	-0.899598
145	6	0	12.088331	-0.966840	0.435589
146	6	0	7.676130	6.231459	-1.281210
147	6	0	3.431202	4.535146	-3.185376
148	9	0	-6.763579	-7.065193	-2.633189
149	9	0	-5.634176	-7.038508	-0.757293
150	9	0	-4.647133	-6.579112	-2.625552
151	9	0	-8.818594	3.670117	4.103356
152	9	0	-8.501473	2.028941	5.481719
153	9	0	-9.260286	1.613517	3.496878
154	9	0	-7.461431	-1.797725	-4.479300
155	9	0	-8.766404	-1.259467	-2.833116
156	9	0	-9.066539	-3.136816	-3.914949
157	9	0	-2.902211	3.976920	4.377053
158	9	0	-3.186335	4.442837	2.288268
159	9	0	-4.437766	5.402291	3.806797
160	9	0	-7.509333	0.786626	-1.019232
161	9	0	-6.685581	0.570469	-3.018853
162	9	0	-7.933588	2.295350	-2.531832
163	9	0	-5.463837	5.437289	1.282060
164	9	0	-3.752814	6.183811	0.154887

165	9	0	-5.755623	6.280952	-0.693254
166	9	0	7.746714	6.972408	-2.407764
167	9	0	8.899288	5.713581	-1.035868
168	9	0	7.412531	7.089260	-0.269764
169	9	0	3.501764	3.669824	-4.233993
170	9	0	2.256226	4.257802	-2.570185
171	9	0	3.348284	5.774815	-3.701714
172	9	0	12.165300	0.355830	0.172072
173	9	0	13.014800	-1.603600	-0.311625
174	9	0	12.459699	-1.121794	1.727171
175	9	0	7.957829	-5.215977	-0.414677
176	9	0	9.007701	-4.739213	-2.247575
177	9	0	10.128377	-5.470314	-0.542616

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**TS<sub>N2O2\_S</sub>**



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.101681 a.u.

Imaginary Freq = 1: -807.61 cm<sup>-1</sup>

Zero-point Energy Correction = 1.286605

Thermal Correction to Energy = 1.39004

Thermal Correction to Enthalpy = 1.390985

Thermal Correction to Free Energy = 1.124268

EE + Zero-point Energy = -7909.815076

EE + Thermal Energy Correction = -7909.711641

EE + Thermal Enthalpy Correction = -7909.710697

EE + Thermal Free Energy Correction = -7909.977413

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.463404

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.696034	0.946904	-4.310333
2	6	0	2.288686	1.883055	-3.199931
3	6	0	0.931696	2.645206	-3.209464
4	7	0	-0.276859	0.438727	-2.791748
5	1	0	3.083276	2.644160	-3.142888
6	1	0	2.383571	1.338217	-2.253051
7	6	0	0.739467	3.277577	-1.810342
8	1	0	1.502418	4.034627	-1.622268
9	1	0	0.772268	2.535841	-1.007617
10	1	0	-0.229858	3.784148	-1.758018
11	6	0	0.922080	3.810632	-4.224949
12	1	0	1.711403	4.523910	-3.965174
13	1	0	-0.037973	4.330591	-4.186309
14	1	0	1.083797	3.498466	-5.261374
15	6	0	3.807548	0.014220	-4.051840
16	6	0	5.920020	-1.802157	-3.650108
17	6	0	3.912157	-0.706309	-2.843948
18	6	0	4.783507	-0.197144	-5.050685
19	6	0	5.842516	-1.074197	-4.840019
20	6	0	4.940313	-1.628903	-2.670277
21	1	0	3.178346	-0.581954	-2.055068
22	1	0	4.733773	0.371570	-5.974417
23	1	0	6.605117	-1.192621	-5.604358
24	1	0	4.978257	-2.236013	-1.772955
25	1	0	6.743794	-2.482945	-3.463444
26	6	0	2.045169	0.859613	-5.540322
27	1	0	1.018944	0.004006	-5.255156
28	1	0	1.453710	1.697328	-5.897316
29	6	0	-0.371590	-0.781794	-3.334839
30	7	0	-0.489356	-1.819098	-2.468826
31	6	0	-0.662731	-3.194185	-2.762054
32	6	0	-1.016961	-5.940512	-3.189349
33	6	0	-1.576419	-3.641334	-3.719520
34	6	0	0.049805	-4.122658	-1.987597



35	6	0	-0.135316	-5.484682	-2.206265
36	6	0	-1.735005	-5.011612	-3.937063
37	1	0	-2.177920	-2.936706	-4.279028
38	1	0	0.728340	-3.773461	-1.217844
39	1	0	-1.153780	-7.002853	-3.356366
40	16	0	-0.303150	-1.048321	-5.060181
41	1	0	2.520821	0.269430	-6.323326
42	6	0	-0.279660	1.718730	-3.512425
43	1	0	-0.366385	1.496507	-4.573717
44	1	0	-1.188823	2.258994	-3.238681
45	1	0	-0.302670	0.507221	-1.759956
46	1	0	-0.184674	-1.642119	-1.499785
47	6	0	-1.886821	-0.874631	1.510711
48	6	0	-0.975461	-1.721780	2.078015
49	6	0	-3.324209	-2.725489	1.004265
50	6	0	-1.149695	-3.135278	1.996436
51	6	0	-3.138828	-1.379189	1.035240
52	6	0	-2.325687	-3.636154	1.416538
53	6	0	-0.150215	-4.050822	2.415977
54	1	0	-3.392891	-5.405129	0.824545
55	1	0	-4.260907	-3.113690	0.656303
56	6	0	-0.323505	-5.385283	2.242153
57	1	0	0.752547	-3.673703	2.845679
58	1	0	0.448224	-6.068510	2.532793
59	6	0	-1.510519	-5.887883	1.662350
60	1	0	-1.618110	-6.942392	1.516508
61	6	0	-2.490006	-5.035934	1.269618
62	6	0	2.397597	-0.206826	3.116002
63	1	0	3.141906	-0.051259	5.073308
64	6	0	2.317479	-0.361284	4.462665
65	6	0	0.195042	-1.207130	2.882138
66	6	0	1.177092	-0.905328	5.090438
67	6	0	1.299852	-0.634935	2.318498
68	6	0	0.097276	-1.334248	4.300155
69	6	0	1.099763	-1.019284	6.501766
70	1	0	-1.887742	-2.168669	4.358604
71	6	0	-0.010466	-1.525868	7.090996
72	1	0	1.932863	-0.689646	7.090541

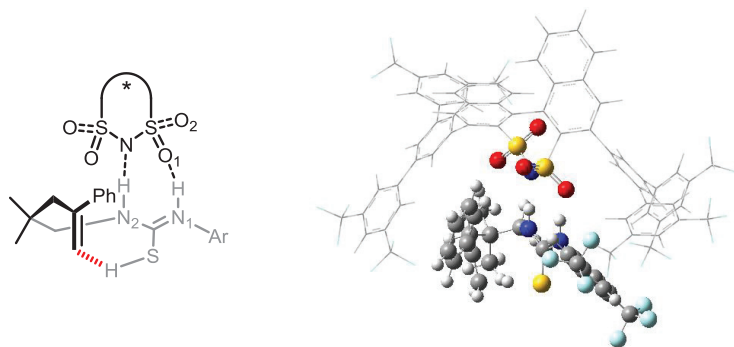
73	1	0	-0.069203	-1.606122	8.157784
74	6	0	-1.105007	-1.944536	6.300764
75	1	0	-1.981650	-2.331347	6.779777
76	6	0	-1.055475	-1.854176	4.948709
77	16	0	1.229722	-0.377281	0.542708
78	8	0	0.990225	-1.716584	-0.070428
79	8	0	2.454469	0.290009	0.063484
80	16	0	-1.318772	0.836312	1.245718
81	8	0	-0.972559	1.437091	2.535721
82	8	0	-2.281615	1.549023	0.394252
83	7	0	-0.028806	0.603691	0.242718
84	6	0	-4.337081	-0.524417	0.738878
85	6	0	-6.605497	1.023934	0.330383
86	6	0	-4.455709	0.250584	-0.398196
87	6	0	-5.380356	-0.550058	1.651995
88	6	0	-6.527379	0.205322	1.445903
89	6	0	-5.564214	1.059200	-0.585651
90	1	0	-5.298390	-1.162948	2.527333
91	1	0	-7.447941	1.675616	0.206141
92	6	0	3.679143	0.357158	2.577893
93	6	0	6.114614	1.365044	1.720937
94	6	0	3.804972	1.704377	2.278144
95	6	0	4.772057	-0.478744	2.446464
96	6	0	5.993547	0.018441	2.011090
97	6	0	5.018328	2.208560	1.844136
98	1	0	2.953106	2.348777	2.361313
99	1	0	4.672315	-1.521657	2.673390
100	1	0	7.042308	1.750588	1.346500
101	1	0	-3.659283	0.276982	-1.106299
102	6	0	-5.552075	2.061859	-1.683061
103	6	0	-5.493525	4.049239	-3.630878
104	6	0	-4.438866	2.887704	-1.811915
105	6	0	-6.624177	2.238796	-2.539196
106	6	0	-6.586987	3.225305	-3.506131
107	6	0	-4.420630	3.864396	-2.783543
108	1	0	-3.615473	2.769851	-1.133101
109	1	0	-7.490758	1.615556	-2.453338
110	1	0	-5.484703	4.832250	-4.359581

111	6	0	-7.637008	0.171879	2.437352
112	6	0	-9.717169	0.102146	4.294075
113	6	0	-8.941786	-0.065309	2.031287
114	6	0	-7.388841	0.375861	3.787293
115	6	0	-8.423358	0.343834	4.698879
116	6	0	-9.964920	-0.103502	2.955666
117	1	0	-9.158369	-0.228047	0.995549
118	1	0	-6.391029	0.569082	4.124197
119	1	0	-10.514479	0.074934	5.006016
120	6	0	7.125686	-0.912696	1.746959
121	6	0	9.204435	-2.661744	1.127519
122	6	0	8.340231	-0.818405	2.403041
123	6	0	6.965296	-1.890199	0.776974
124	6	0	7.998384	-2.744804	0.464668
125	6	0	9.362410	-1.697733	2.097192
126	1	0	8.492146	-0.066477	3.150047
127	1	0	6.032112	-1.961239	0.262192
128	1	0	10.005206	-3.330956	0.891865
129	6	0	5.143787	3.625766	1.397932
130	6	0	5.445096	6.225259	0.450916
131	6	0	4.505303	4.028844	0.235342
132	6	0	5.929256	4.536894	2.083823
133	6	0	6.075197	5.823824	1.607206
134	6	0	4.660238	5.319644	-0.228674
135	1	0	3.906084	3.322233	-0.302749
136	1	0	6.417195	4.247457	2.991914
137	1	0	5.564067	7.223373	0.086944
138	6	0	6.970263	6.770359	2.321570
139	6	0	3.952163	5.747994	-1.461349
140	6	0	10.633242	-1.631456	2.864965
141	6	0	7.827137	-3.725263	-0.637060
142	6	0	-3.216908	4.713204	-2.952942
143	6	0	-7.725347	3.366729	-4.449703
144	6	0	-11.342228	-0.432889	2.506509
145	6	0	-8.144749	0.631186	6.129879
146	6	0	0.619603	-6.508234	-1.397979
147	6	0	-2.672852	-5.481230	-5.019754
148	9	0	8.257915	6.663657	1.919219

149	9	0	6.966247	6.541423	3.652341
150	9	0	6.603936	8.051699	2.108236
151	9	0	10.584401	-2.354744	4.006659
152	9	0	10.920785	-0.362699	3.227777
153	9	0	11.668505	-2.113094	2.145020
154	9	0	3.926935	4.758871	-2.393935
155	9	0	2.654876	6.065672	-1.236607
156	9	0	4.528151	6.828988	-2.022811
157	9	0	8.289623	-3.258079	-1.827225
158	9	0	6.522094	-4.020698	-0.837830
159	9	0	8.493677	-4.871573	-0.397948
160	9	0	-8.181135	1.953125	6.410812
161	9	0	-6.916496	0.196833	6.487712
162	9	0	-9.053419	0.039811	6.935279
163	9	0	-12.269627	0.099862	3.330112
164	9	0	-11.571045	-1.766048	2.477525
165	9	0	-11.573145	0.021431	1.255572
166	9	0	1.382104	-7.292656	-2.193734
167	9	0	-0.227420	-7.340336	-0.742056
168	9	0	1.431344	-5.952735	-0.478900
169	9	0	-3.156278	-6.715488	-4.760467
170	9	0	-3.729271	-4.650223	-5.158255
171	9	0	-2.055924	-5.540514	-6.220151
172	9	0	-2.364271	4.195802	-3.885434
173	9	0	-2.497945	4.818712	-1.816364
174	9	0	-3.525648	5.952779	-3.383134
175	9	0	-7.603949	2.562851	-5.531223
176	9	0	-8.896268	3.039103	-3.860982
177	9	0	-7.824111	4.629548	-4.917307

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TS<sub>N2O1\_R</sub>



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.100950 a.u.

Imaginary Freq = 1: -889.01 cm<sup>-1</sup>

Zero-point Energy Correction = 1.286709

Thermal Correction to Energy = 1.390098

Thermal Correction to Enthalpy = 1.391043

Thermal Correction to Free Energy = 1.124413

EE + Zero-point Energy = -7909.814241

EE + Thermal Energy Correction = -7909.710852

EE + Thermal Enthalpy Correction = -7909.709908

EE + Thermal Free Energy Correction = -7909.976537

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.474300

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.070995	-3.955131	0.426521
2	6	0	3.001698	-2.964976	1.089089
3	6	0	2.731663	-2.414731	2.523300
4	7	0	0.374209	-1.742853	1.573774
5	1	0	3.965482	-3.488514	1.143320
6	1	0	3.157619	-2.112971	0.423608
7	6	0	3.883605	-1.424002	2.821312
8	1	0	4.840404	-1.955678	2.889153
9	1	0	3.954436	-0.647623	2.053260
10	1	0	3.721369	-0.931628	3.787408
11	6	0	2.792013	-3.502755	3.620818

12	1	0	3.662258	-4.152873	3.474566
13	1	0	2.897294	-3.025597	4.602406
14	1	0	1.891438	-4.117691	3.669248
15	6	0	1.775870	-3.802535	-1.008288
16	6	0	1.219285	-3.571448	-3.760669
17	6	0	1.639227	-4.949420	-1.822044
18	6	0	1.640665	-2.536467	-1.616730
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20	6	0	1.375800	-4.833705	-3.183014
21	1	0	1.787979	-5.934640	-1.392216
22	1	0	1.711894	-1.630193	-1.026047
23	1	0	1.183178	-1.438816	-3.398019
24	1	0	1.293810	-5.728933	-3.792926
25	1	0	0.996410	-3.481470	-4.820236
26	6	0	1.543794	-5.033283	1.129254
27	1	0	2.057474	-5.377860	2.022569
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32	6	0	-5.552740	-3.701559	0.018952
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37	1	0	-2.747385	-2.422826	-1.433280
38	1	0	-3.671523	-2.877930	2.740876
39	1	0	-6.530203	-4.087419	-0.242991
40	16	0	-0.871531	-3.900145	2.592242
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44	1	0	1.639331	-0.535966	2.608000
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46	6	0	2.127691	1.919026	-1.967385
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72	1	0	-0.156798	8.533222	-2.385791
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79	16	0	1.672598	1.596515	-0.240546
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82	7	0	0.364228	0.599594	-0.348576
83	6	0	4.512948	0.978830	-1.701764
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86	6	0	5.187361	1.896185	-0.912533
87	6	0	6.307573	1.519012	-0.192743

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93	6	0	-5.574517	1.548050	0.816700
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96	6	0	-5.633242	1.826816	-0.538229
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106	6	0	6.907029	-4.167649	0.170105
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109	1	0	6.452473	-2.294154	1.062534
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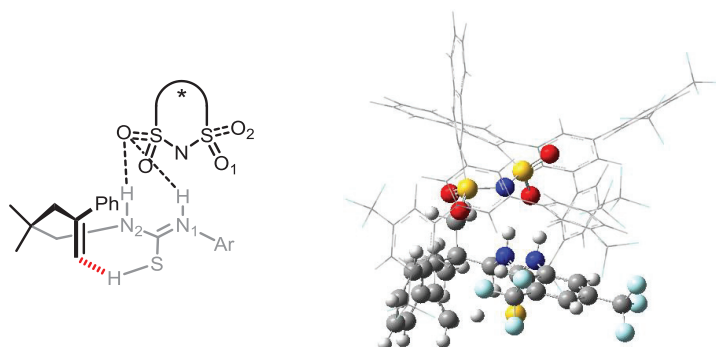


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133	6	0	-8.308169	-0.158151	-2.283215
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136	1	0	-6.421116	-0.586062	-1.410947
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140	6	0	-8.567742	-1.572640	-2.662671
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142	6	0	6.165091	4.899472	3.446278
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144	6	0	7.356445	-4.907275	-3.457684
145	6	0	6.871992	-4.867634	1.477136
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147	6	0	-6.092905	-4.045500	2.450489
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171	9	0	8.666426	-4.844000	-3.786453
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175	9	0	6.939914	5.277972	4.486266
176	9	0	5.644476	6.031467	2.920706
177	9	0	5.127088	4.185474	3.929319

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**TS<sub>N2O2\_R</sub>**



E (ONIOM(B3LYP/6-31G(d): HF/3-21G\*)) = -7911.104119 a.u.

Imaginary Freq = 1: -622.35 cm<sup>-1</sup>

Zero-point Energy Correction = 1.286952

Thermal Correction to Energy = 1.389983

Thermal Correction to Enthalpy = 1.390927

Thermal Correction to Free Energy = 1.132228

EE + Zero-point Energy = -7909.817166

EE + Thermal Energy Correction = -7909.714136

EE + Thermal Enthalpy Correction = -7909.713191

EE + Thermal Free Energy Correction = -7909.971891

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.473332

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
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2	6	0	0.287999	-5.224860	-0.651193
3	6	0	-1.256824	-5.111779	-0.836586
4	7	0	-1.055140	-2.718853	-1.686574
5	1	0	0.478851	-6.143907	-0.071614
6	1	0	0.604145	-4.424838	0.024812
7	6	0	-1.827325	-4.750659	0.554977
8	1	0	-1.550977	-5.510148	1.295142
9	1	0	-1.452863	-3.791852	0.925710
10	1	0	-2.922677	-4.709966	0.517548
11	6	0	-1.911251	-6.441260	-1.274678
12	1	0	-1.781548	-7.202607	-0.497251
13	1	0	-2.985789	-6.300987	-1.434155
14	1	0	-1.497817	-6.846839	-2.202836
15	6	0	2.663757	-4.953287	-1.564103
16	6	0	5.361529	-4.175548	-1.276280
17	6	0	3.045496	-4.000695	-0.593687
18	6	0	3.681347	-5.537200	-2.357876
19	6	0	5.008640	-5.159999	-2.207140
20	6	0	4.374835	-3.605542	-0.469627
21	1	0	2.312456	-3.513111	0.035710
22	1	0	3.429626	-6.315917	-3.070740
23	1	0	4.618959	-2.834594	0.254017
24	1	0	6.393539	-3.849884	-1.198602
25	6	0	0.843760	-5.512102	-3.126366
26	1	0	0.532418	-4.277437	-3.505073
27	1	0	-0.097520	-6.022304	-3.306777
28	6	0	-0.375717	-2.035572	-2.620762
29	7	0	-0.117234	-0.740967	-2.350575
30	6	0	0.806545	0.078163	-3.077155
31	6	0	2.603196	1.766047	-4.386980
32	6	0	2.173688	-0.036763	-2.821210
33	6	0	0.331162	1.024165	-3.984651

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37	1	0	-0.733156	1.117012	-4.161649
38	1	0	3.292557	2.464446	-4.848941
39	16	0	0.150956	-2.704797	-4.135863
40	1	0	1.609453	-5.710186	-3.875248
41	6	0	-1.692304	-4.022895	-1.857838
42	1	0	-1.522570	-4.340351	-2.885139
43	1	0	-2.775320	-3.897400	-1.759734
44	1	0	-1.297477	-2.185396	-0.842489
45	1	0	-0.564464	-0.321147	-1.531805
46	6	0	1.504455	-0.596777	3.418924
47	6	0	0.415378	-0.734295	4.223441
48	6	0	2.643051	0.733230	5.054591
49	6	0	0.407130	-0.110086	5.520904
50	6	0	2.604809	0.230030	3.799761
51	6	0	1.565557	0.559840	5.955062
52	6	0	-0.723486	-0.115803	6.381334
53	1	0	2.493167	1.646002	7.557617
54	1	0	3.462632	1.359661	5.346523
55	6	0	-0.670687	0.469996	7.604819
56	1	0	-1.625361	-0.584364	6.058133
57	1	0	-1.535241	0.460264	8.237492
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59	1	0	0.534769	1.550271	9.023986
60	6	0	1.596787	1.148588	7.243430
61	6	0	-3.059132	-1.459233	2.807942
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63	6	0	-3.455125	-2.443796	3.650583
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67	6	0	-1.239815	-2.574783	4.625868
68	6	0	-2.925710	-4.192781	5.314612
69	1	0	0.675205	-2.851362	5.564548
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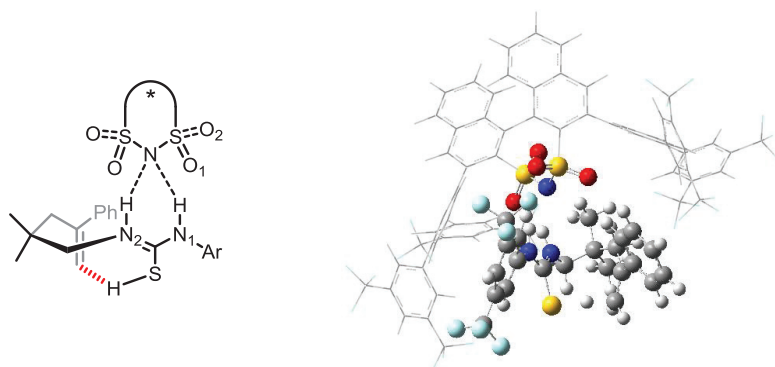
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79	16	0	1.362286	-1.264980	1.746397
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81	8	0	2.652053	-1.214457	1.050624
82	7	0	0.422090	-0.108140	1.111564
83	6	0	3.532529	0.762157	2.755664
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85	6	0	2.969383	1.705360	1.917473
86	6	0	4.822025	0.329243	2.535880
87	6	0	5.521121	0.793130	1.427120
88	6	0	3.623434	2.129675	0.775756
89	1	0	1.979515	2.051016	2.126122
90	1	0	5.258168	-0.415153	3.171862
91	1	0	5.414459	1.933634	-0.369651
92	6	0	-4.087802	-0.790202	1.957406
93	6	0	-5.987580	0.460465	0.368273
94	6	0	-4.873408	-1.532384	1.095185
95	6	0	-4.284540	0.576691	2.050443
96	6	0	-5.214618	1.212446	1.241674
97	6	0	-5.827355	-0.916746	0.299628
98	1	0	-4.728428	-2.591973	1.032341
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101	1	0	5.774068	-5.627053	-2.820066
102	6	0	6.889990	0.270673	1.159029
103	6	0	9.440651	-0.752522	0.685917
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105	6	0	7.191560	-0.340423	-0.050517
106	6	0	8.454575	-0.850055	-0.271113
107	6	0	9.141295	-0.145213	1.883619
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109	1	0	6.444241	-0.414726	-0.812632

110	1	0	10.424830	-1.124286	0.494756
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116	6	0	2.766692	4.582738	-1.972280
117	1	0	4.544074	4.089607	-0.927913
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119	1	0	0.865011	4.896379	-2.872419
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143	9	0	10.028563	1.039880	3.702224
144	9	0	10.109785	-1.120779	3.801655
145	9	0	11.420261	-0.072898	2.429861
146	6	0	-0.657429	3.185257	-1.459570
147	6	0	3.470826	5.455146	-2.942757

148	9	0	-1.442719	3.817344	-0.564490
149	9	0	-0.927671	1.858706	-1.326342
150	9	0	-1.092348	3.548832	-2.689711
151	9	0	4.601317	5.977580	-2.421297
152	9	0	3.849809	4.768683	-4.056020
153	9	0	2.691937	6.468038	-3.371571
154	6	0	0.743565	2.922876	-5.593492
155	6	0	4.550433	0.694335	-3.251367
156	9	0	-0.559752	2.773329	-5.893503
157	9	0	1.439987	2.895138	-6.749110
158	9	0	0.899244	4.170575	-5.080701
159	9	0	4.858164	-0.323196	-2.408174
160	9	0	5.063267	1.825507	-2.696543
161	9	0	5.224396	0.490949	-4.397755
162	6	0	-10.170824	-2.321370	-1.714905
163	6	0	-8.028041	5.285428	1.654195
164	6	0	-3.134316	5.744579	1.183905
165	9	0	-8.073531	6.536983	1.150027
166	9	0	-8.373133	5.383392	2.959572
167	9	0	-8.998530	4.554477	1.060725
168	9	0	-3.343108	6.920569	1.825501
169	9	0	-2.043142	5.168825	1.721612
170	9	0	-2.829358	6.065134	-0.092924
171	9	0	-10.752347	-3.482829	-2.083821
172	9	0	-10.714165	-1.928109	-0.542801
173	6	0	-5.941775	-4.473472	-3.137932
174	9	0	-4.870366	-4.981867	-2.469421
175	9	0	-5.446179	-3.931497	-4.270751
176	9	0	-6.726702	-5.505870	-3.504496
177	9	0	-10.530945	-1.398355	-2.634704

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TS<sub>N1N2\_S</sub>



Imaginary Freq = 1: -747.33 cm-1

Electronic Energy (EE) at ONIOM(B3LYP/6-31G(d): HF/3-21G\*) =  
-7911.108826 Hartree

Zero-point Energy Correction = 1.287518 Hartree

Thermal Correction to Energy = 1.390557 Hartree

Thermal Correction to Enthalpy = 1.391501 Hartree

Thermal Correction to Free Energy = 1.129979 Hartree

EE + Zero-point Energy = -7909.821308 Hartree

EE + Thermal Energy Correction = -7909.718269 Hartree

EE + Thermal Enthalpy Correction = -7909.717325 Hartree

EE + Thermal Free Energy Correction = -7909.978847 Hartree

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.475041

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.327792	1.267907	-4.020686
2	6	0	1.356707	2.284298	-2.910488
3	6	0	0.167202	3.239885	-2.582053
4	7	0	-1.297510	1.284878	-1.879674
5	1	0	2.223791	2.926777	-3.127442
6	1	0	1.631901	1.759201	-1.991282
7	6	0	0.454038	3.787536	-1.163862
8	1	0	1.380954	4.367840	-1.162209
9	1	0	0.556665	2.986992	-0.423589



10	1	0	-0.341421	4.455344	-0.828008
11	6	0	0.096943	4.454822	-3.533574
12	1	0	-0.143020	4.183366	-4.567461
13	1	0	1.056755	4.980314	-3.539526
14	1	0	-0.675183	5.150095	-3.186903
15	6	0	2.211887	0.100482	-3.905249
16	6	0	3.835084	-2.197186	-3.728999
17	6	0	2.822618	-0.447632	-5.057163
18	6	0	2.443554	-0.526950	-2.663132
19	6	0	3.230374	-1.673117	-2.584190
20	6	0	3.635981	-1.570454	-4.964743
21	1	0	2.692627	0.041055	-6.017960
22	1	0	1.975090	-0.174970	-1.751399
23	1	0	3.350698	-2.149648	-1.616487
24	1	0	4.115995	-1.962557	-5.856742
25	1	0	4.460547	-3.081305	-3.658157
26	6	0	0.443094	1.341434	-5.097138
27	1	0	-0.012971	2.295911	-5.345995
28	1	0	0.613136	0.681365	-5.947358
29	6	0	-1.645075	0.078403	-2.346212
30	7	0	-1.721348	-0.897134	-1.403338
31	6	0	-2.046623	-2.271717	-1.529570
32	6	0	-2.584169	-5.024516	-1.643803
33	6	0	-1.410814	-3.154109	-0.641761
34	6	0	-2.992858	-2.763150	-2.437018
35	6	0	-3.230607	-4.139326	-2.502087
36	6	0	-1.689286	-4.517329	-0.701241
37	1	0	-0.699722	-2.784686	0.092881
38	1	0	-3.556096	-2.103704	-3.081328
39	1	0	-2.770727	-6.088847	-1.707469
40	16	0	-1.961655	-0.225063	-4.031115
41	1	0	-0.611093	0.683054	-4.571469
42	6	0	-1.226444	2.549673	-2.607352
43	1	0	-1.571156	2.373615	-3.625406
44	1	0	-1.937303	3.243945	-2.146295
45	1	0	-1.251000	1.371782	-0.857961
46	6	0	1.833066	-0.041949	2.994438
47	6	0	0.947250	0.154583	4.018747

48	6	0	3.540868	1.125851	4.198021
49	6	0	1.328968	0.908354	5.169744
50	6	0	3.162486	0.463901	3.073017
51	6	0	2.649504	1.376467	5.260741
52	6	0	0.423594	1.227119	6.218294
53	1	0	4.070198	2.457202	6.458017
54	1	0	4.545458	1.491547	4.274731
55	6	0	0.834953	1.945693	7.292076
56	1	0	-0.589857	0.899038	6.149368
57	1	0	0.141524	2.185124	8.072783
58	6	0	2.173004	2.390350	7.391959
59	1	0	2.479447	2.952939	8.250819
60	6	0	3.055682	2.114957	6.401635
61	6	0	-2.787272	-0.278959	3.269877
62	1	0	-4.126640	-1.508427	4.323088
63	6	0	-3.103045	-1.212406	4.208261
64	6	0	-0.468136	-0.373131	3.993930
65	6	0	-2.131320	-1.818669	5.029318
66	6	0	-1.426596	0.141384	3.159395
67	6	0	-0.793607	-1.408214	4.919104
68	6	0	-2.477190	-2.838369	5.952926
69	1	0	1.209761	-1.770043	5.620651
70	6	0	-1.522456	-3.432373	6.708892
71	1	0	-3.504233	-3.136029	6.030434
72	1	0	-1.783351	-4.208892	7.399442
73	6	0	-0.170049	-3.039806	6.581878
74	1	0	0.577459	-3.531303	7.171134
75	6	0	0.185682	-2.057272	5.717961
76	16	0	1.195337	-0.831460	1.499240
77	8	0	2.267589	-0.922552	0.496187
78	8	0	0.564951	-2.101624	1.898716
79	16	0	-0.797998	1.218347	1.834627
80	8	0	-1.893632	1.635110	0.916933
81	8	0	-0.065063	2.324740	2.463001
82	7	0	0.085438	0.225675	0.891790
83	6	0	4.202521	0.354420	1.997292
84	6	0	6.226063	0.271787	0.105906
85	6	0	4.853298	-0.838604	1.728075

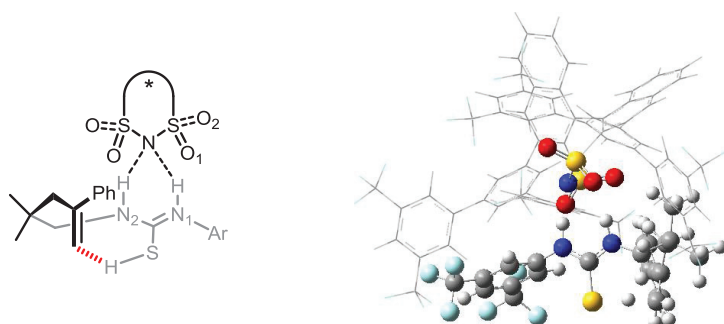
86	6	0	4.573777	1.502978	1.322985
87	6	0	5.584291	1.469429	0.368828
88	6	0	5.862518	-0.887174	0.781491
89	1	0	4.541852	-1.735106	2.224041
90	1	0	4.075626	2.425244	1.546612
91	1	0	7.001330	0.232584	-0.633701
92	6	0	-3.903367	0.239476	2.421982
93	6	0	-5.929914	1.148816	0.758320
94	6	0	-4.664463	-0.657410	1.689697
95	6	0	-4.183159	1.592550	2.336813
96	6	0	-5.163264	2.054390	1.478227
97	6	0	-5.690819	-0.210289	0.868097
98	1	0	-4.422581	-1.701287	1.714598
99	1	0	-3.593973	2.294407	2.888854
100	1	0	-6.681716	1.505679	0.081876
101	1	0	-1.297588	-0.640693	-0.501511
102	6	0	6.518137	-2.179882	0.443635
103	6	0	7.698066	-4.620752	-0.218645
104	6	0	6.629312	-2.578174	-0.879471
105	6	0	7.012524	-3.019421	1.431939
106	6	0	7.600514	-4.220349	1.096296
107	6	0	7.207591	-3.789600	-1.199220
108	1	0	6.246747	-1.949568	-1.657571
109	1	0	6.941472	-2.735314	2.461631
110	1	0	8.143100	-5.559360	-0.470870
111	6	0	5.955121	2.701457	-0.378769
112	6	0	6.633375	4.987270	-1.832857
113	6	0	4.973622	3.503594	-0.938943
114	6	0	7.284371	3.067026	-0.550164
115	6	0	7.610845	4.198555	-1.265221
116	6	0	5.316321	4.629964	-1.660865
117	1	0	3.942926	3.236500	-0.821226
118	1	0	8.062983	2.470260	-0.121695
119	1	0	6.894445	5.855871	-2.397972
120	6	0	-6.435377	-1.181969	0.021578
121	6	0	-7.725573	-3.042867	-1.604668
122	6	0	-7.136930	-2.239962	0.579117
123	6	0	-6.399456	-1.063026	-1.358092

124	6	0	-7.037169	-1.988410	-2.156590
125	6	0	-7.772574	-3.156361	-0.232349
126	1	0	-7.198607	-2.339959	1.643233
127	1	0	-5.852327	-0.257902	-1.806811
128	1	0	-8.209573	-3.760700	-2.230210
129	6	0	-5.252669	3.507819	1.169094
130	6	0	-5.338019	6.165366	0.344698
131	6	0	-4.113282	4.129028	0.675621
132	6	0	-6.429023	4.232434	1.260358
133	6	0	-6.458879	5.554545	0.863135
134	6	0	-4.170140	5.441150	0.254343
135	1	0	-3.202728	3.564054	0.612101
136	1	0	-7.319803	3.769209	1.632891
137	1	0	-5.377620	7.179913	0.010904
138	6	0	-6.939973	-1.834046	-3.629809
139	6	0	-8.471146	-4.313489	0.384038
140	6	0	-7.708614	6.339694	1.030078
141	6	0	-2.961809	6.042280	-0.361252
142	6	0	4.237594	5.441794	-2.277054
143	6	0	9.035792	4.605444	-1.380060
144	6	0	8.187811	-5.069844	2.164848
145	6	0	7.245926	-4.209521	-2.622295
146	6	0	-1.043703	-5.451442	0.292631
147	6	0	-4.223848	-4.660570	-3.511098
148	9	0	3.596014	4.771441	-3.275927
149	9	0	3.271165	5.756203	-1.382739
150	9	0	4.702913	6.586094	-2.809125
151	9	0	9.854293	3.531418	-1.381427
152	9	0	9.432762	5.390387	-0.353143
153	9	0	9.254605	5.308821	-2.511462
154	9	0	-2.672359	5.475454	-1.568839
155	9	0	-1.855334	5.849868	0.391635
156	9	0	-3.100529	7.364544	-0.574454
157	9	0	-7.786018	7.340275	0.126236
158	9	0	-7.800982	6.914845	2.250963
159	9	0	-8.802380	5.558713	0.892275
160	9	0	-8.994814	-3.987152	1.586603
161	9	0	-9.469621	-4.764515	-0.404248

162	9	0	-7.644761	-5.362262	0.600745
163	9	0	-7.460551	-0.653916	-4.040836
164	9	0	-7.580182	-2.814617	-4.288990
165	9	0	-5.647744	-1.821378	-4.049411
166	9	0	-4.039978	-4.112083	-4.726970
167	9	0	-5.503615	-4.386768	-3.144360
168	9	0	-4.141708	-6.000700	-3.651913
169	9	0	-1.764146	-5.521706	1.435585
170	9	0	-0.958854	-6.711046	-0.195575
171	9	0	0.198015	-5.057551	0.624951
172	9	0	8.122473	-5.210786	-2.829034
173	9	0	6.035511	-4.652162	-3.060566
174	9	0	7.572144	-3.181046	-3.435076
175	9	0	9.474341	-4.746445	2.430808
176	9	0	7.512067	-4.936621	3.326375
177	9	0	8.182164	-6.373514	1.815496

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**TS<sub>N1N2\_R</sub>**



Imaginary Freq = 1: -603.16 cm<sup>-1</sup>

Electronic Energy (EE) at ONIOM(B3LYP/6-31G(d): HF/3-21G\*)=

-7911.1038 Hartree

Zero-point Energy Correction = 1.286535 Hartree

Thermal Correction to Energy = 1.390098 Hartree

Thermal Correction to Enthalpy = 1.391042 Hartree

Thermal Correction to Free Energy = 1.12498 Hartree

EE + Zero-point Energy = -7909.8173259 Hartree

EE + Thermal Energy Correction = -7909.713696 Hartree

EE + Thermal Enthalpy Correction = -7909.712751 Hartree

EE + Thermal Free Energy Correction = -7909.978813 Hartree

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.466025

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.088599	-5.173450	-2.223077
2	6	0	-0.977374	-4.832913	-1.208022
3	6	0	-2.438907	-4.465781	-1.583133
4	7	0	-1.713592	-2.114312	-2.226412
5	1	0	-1.021694	-5.683456	-0.511266
6	1	0	-0.584771	-4.012748	-0.593458
7	6	0	-3.117003	-4.079550	-0.247936
8	1	0	-3.066627	-4.909772	0.465875
9	1	0	-2.648008	-3.210314	0.223106
10	1	0	-4.175458	-3.852151	-0.413395
11	6	0	-3.225740	-5.656875	-2.171145
12	1	0	-3.215136	-6.498005	-1.468123
13	1	0	-4.271309	-5.389747	-2.344913
14	1	0	-2.823122	-6.021349	-3.120909
15	6	0	1.439256	-5.424939	-1.712888
16	6	0	4.039622	-5.964191	-0.753557
17	6	0	1.855562	-4.965240	-0.439994
18	6	0	2.363604	-6.178030	-2.477333
19	6	0	3.643841	-6.437794	-2.007280
20	6	0	3.136681	-5.237129	0.025730
21	1	0	1.202573	-4.361834	0.180430
22	1	0	2.063051	-6.597640	-3.430945
23	1	0	4.331927	-7.020369	-2.612825
24	1	0	3.440444	-4.889594	1.005975
25	1	0	5.042398	-6.143376	-0.384100
26	6	0	-0.160201	-5.153522	-3.601382
27	1	0	-0.154982	-3.868129	-3.915844
28	1	0	-1.166852	-5.372900	-3.944533
29	6	0	-0.737082	-1.579440	-2.977030
30	7	0	-0.155840	-0.462398	-2.507393

31	6	0	0.958395	0.203666	-3.109574
32	6	0	3.131197	1.594558	-4.183009
33	6	0	2.231745	-0.359300	-3.059478
34	6	0	0.763542	1.461804	-3.681646
35	6	0	1.852460	2.154824	-4.208207
36	6	0	3.309578	0.334920	-3.612432
37	1	0	2.376002	-1.319585	-2.579266
38	1	0	-0.229672	1.895171	-3.701969
39	1	0	3.977161	2.145782	-4.573208
40	16	0	-0.225525	-2.242688	-4.510776
41	1	0	0.617035	-5.512918	-4.272851
42	6	0	-2.547370	-3.264660	-2.571276
43	1	0	-2.302048	-3.556320	-3.591135
44	1	0	-3.593072	-2.939649	-2.597688
45	1	0	-2.019549	-1.559195	-1.414897
46	1	0	-0.365886	-0.184051	-1.528036
47	6	0	0.986240	-0.018663	2.566251
48	6	0	-0.066272	0.063100	3.431391
49	6	0	2.194834	1.622073	3.815261
50	6	0	-0.045321	1.002741	4.510253
51	6	0	2.156257	0.765823	2.760163
52	6	0	1.106519	1.785106	4.695546
53	6	0	-1.149141	1.207854	5.383408
54	1	0	2.042780	3.313189	5.881687
55	1	0	3.079137	2.204744	3.980254
56	6	0	-1.081838	2.126756	6.378230
57	1	0	-2.039413	0.635511	5.245545
58	1	0	-1.923937	2.278184	7.022969
59	6	0	0.087773	2.896058	6.573457
60	1	0	0.120669	3.615108	7.366938
61	6	0	1.152825	2.729288	5.753265
62	6	0	-3.611367	-1.052172	2.611506
63	1	0	-4.843235	-2.252219	3.819062
64	6	0	-3.843675	-1.914085	3.629119
65	6	0	-1.311675	-0.781100	3.331891
66	6	0	-2.800041	-2.343571	4.486668
67	6	0	-2.280156	-0.564481	2.400084
68	6	0	-1.534734	-1.745529	4.369096

69	6	0	-3.020846	-3.310746	5.496320
70	1	0	0.459792	-1.695095	5.168321
71	6	0	-2.016213	-3.674065	6.334214
72	1	0	-3.994039	-3.752541	5.582974
73	1	0	-2.185231	-4.411346	7.093218
74	6	0	-0.740876	-3.083138	6.210991
75	1	0	0.045871	-3.384129	6.872876
76	6	0	-0.507230	-2.140341	5.261577
77	16	0	-1.938081	0.284890	0.828161
78	8	0	-2.110852	1.739410	0.993550
79	8	0	-2.832308	-0.366492	-0.167114
80	16	0	0.702689	-0.973015	1.066622
81	8	0	0.234039	-2.323344	1.448355
82	8	0	1.896671	-0.910304	0.216282
83	7	0	-0.420294	-0.058745	0.330995
84	6	0	3.405620	0.705667	1.930409
85	6	0	5.869913	0.777430	0.665161
86	6	0	3.711016	1.738730	1.060510
87	6	0	4.337583	-0.288796	2.166046
88	6	0	5.571187	-0.261527	1.532976
89	6	0	4.945219	1.783074	0.424944
90	1	0	2.980967	2.499692	0.870153
91	1	0	4.114031	-1.067586	2.867580
92	1	0	6.824787	0.806312	0.179365
93	6	0	-4.758919	-0.465612	1.859452
94	6	0	-6.831612	0.726479	0.452867
95	6	0	-5.644601	-1.243782	1.139928
96	6	0	-4.938360	0.906573	1.904790
97	6	0	-5.956341	1.513545	1.187712
98	6	0	-6.679683	-0.652010	0.430544
99	1	0	-5.509567	-2.305624	1.101818
100	1	0	-4.257044	1.510258	2.465999
101	1	0	-7.588513	1.187942	-0.150207
102	6	0	-7.534574	-1.496575	-0.446550
103	6	0	-9.054453	-3.095388	-2.153180
104	6	0	-8.917893	-1.495132	-0.366507
105	6	0	-6.925264	-2.302360	-1.395286
106	6	0	-7.679450	-3.097556	-2.229837

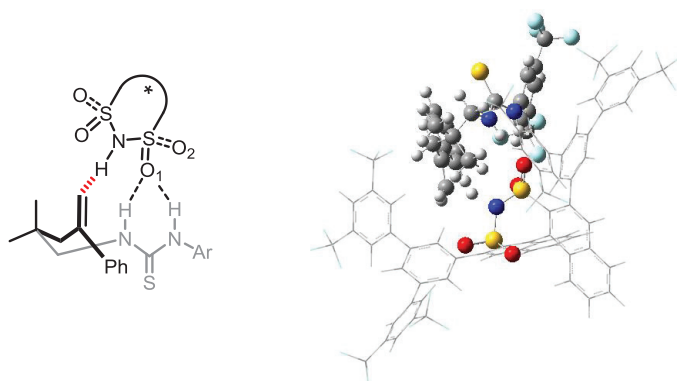


107	6	0	-9.662533	-2.291832	-1.213788
108	1	0	-9.414212	-0.874581	0.351153
109	1	0	-5.860410	-2.277576	-1.492642
110	1	0	-9.640423	-3.689227	-2.822244
111	6	0	-6.037310	2.998221	1.150813
112	6	0	-6.158625	5.775674	1.026560
113	6	0	-4.884261	3.742650	0.921932
114	6	0	-7.242735	3.658994	1.319422
115	6	0	-7.295637	5.036422	1.259698
116	6	0	-4.957986	5.118962	0.866048
117	1	0	-3.944740	3.243121	0.774437
118	1	0	-8.140732	3.103716	1.497680
119	1	0	-6.208653	6.840879	0.952209
120	6	0	6.577715	-1.318664	1.822625
121	6	0	8.441366	-3.305306	2.425451
122	6	0	7.873628	-0.987644	2.190636
123	6	0	6.230209	-2.659567	1.752339
124	6	0	7.157510	-3.636406	2.052202
125	6	0	8.789138	-1.975199	2.489518
126	1	0	8.165623	0.040388	2.254190
127	1	0	5.238735	-2.930450	1.453795
128	1	0	9.155027	-4.065693	2.662197
129	6	0	5.295792	2.913746	-0.479503
130	6	0	6.001635	5.042780	-2.137321
131	6	0	5.871746	2.682481	-1.721204
132	6	0	5.076097	4.224884	-0.083444
133	6	0	5.430600	5.272681	-0.906450
134	6	0	6.216661	3.742056	-2.535367
135	1	0	6.025742	1.679695	-2.053742
136	1	0	4.635794	4.430557	0.870268
137	1	0	6.272897	5.859009	-2.772106
138	6	0	-6.985324	-3.985052	-3.194698
139	6	0	-11.142167	-2.320523	-1.075317
140	6	0	-3.716449	5.909872	0.656594
141	6	0	-8.587233	5.728093	1.498720
142	6	0	6.781025	-5.066632	1.937514
143	6	0	10.148708	-1.595287	2.953897
144	6	0	6.796213	3.479180	-3.878241

145	6	0	5.257017	6.667913	-0.426644
146	6	0	1.662827	3.504483	-4.852115
147	6	0	4.672665	-0.300031	-3.561371
148	9	0	7.576651	2.381498	-3.878674
149	9	0	5.845289	3.276003	-4.830637
150	9	0	7.533742	4.524905	-4.308148
151	9	0	4.188544	6.777970	0.391821
152	9	0	6.325534	7.104424	0.281027
153	9	0	5.098646	7.527718	-1.454189
154	9	0	5.467423	-5.259331	2.210627
155	9	0	6.970375	-5.565514	0.686277
156	9	0	7.501151	-5.845497	2.767998
157	9	0	11.051037	-2.558763	2.670741
158	9	0	10.559960	-0.444190	2.380946
159	9	0	10.200000	-1.398341	4.291263
160	9	0	5.120406	-0.416768	-2.279184
161	9	0	5.605337	0.431109	-4.212568
162	9	0	4.678485	-1.541193	-4.084784
163	9	0	0.463131	4.043514	-4.552865
164	9	0	1.743205	3.424788	-6.199186
165	9	0	2.614773	4.376778	-4.453236
166	9	0	-9.631567	4.984532	1.069385
167	9	0	-8.810739	5.973063	2.811124
168	9	0	-8.633010	6.922889	0.871793
169	9	0	-3.996794	7.133476	0.152995
170	9	0	-2.867564	5.291965	-0.187839
171	9	0	-3.036891	6.115030	1.808405
172	9	0	-11.620974	-1.126804	-0.665032
173	9	0	-11.738898	-2.640932	-2.242892
174	9	0	-11.555673	-3.232981	-0.166093
175	9	0	-7.756698	-4.286140	-4.255816
176	9	0	-6.616984	-5.167053	-2.632264
177	9	0	-5.834859	-3.428215	-3.644754

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TS<sub>s</sub>



Imaginary Freq = 1: -1336.18 cm<sup>-1</sup>

Electronic Energy (EE) at ONIOM(B3LYP/6-31G(d): HF/3-21G\*)=  
-7911.0943 Hartree

Zero-point Energy Correction = 1.286001 Hartree

Thermal Correction to Energy = 1.389864 Hartree

Thermal Correction to Enthalpy = 1.390808 Hartree

Thermal Correction to Free Energy = 1.122679 Hartree

EE + Zero-point Energy = -7909.808269 Hartree

EE + Thermal Energy Correction = -7909.704404 Hartree

EE + Thermal Enthalpy Correction = -7909.703460 Hartree

EE + Thermal Free Energy Correction = -7909.971589 Hartree

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.477599

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.068870	-0.576082	-2.729139
2	6	0	-0.549708	0.016156	-4.023209
3	6	0	0.675990	1.004126	-4.047376
4	7	0	2.176932	-0.429626	-2.605612
5	1	0	-1.392468	0.577143	-4.445455
6	1	0	-0.315619	-0.775277	-4.736396
7	6	0	0.702744	1.613386	-5.467903
8	1	0	-0.186516	2.228957	-5.645026
9	1	0	0.734251	0.833894	-6.238561
10	1	0	1.584883	2.245820	-5.594378

11	6	0	0.547084	2.153313	-3.031910
12	1	0	-0.382813	2.713249	-3.183436
13	1	0	1.375557	2.856804	-3.170375
14	1	0	0.565609	1.820837	-1.990571
15	6	0	-0.909733	-1.997754	-2.418166
16	6	0	-0.729163	-4.747194	-1.790402
17	6	0	-1.218948	-2.478734	-1.117733
18	6	0	-0.484846	-2.942616	-3.378993
19	6	0	-0.410119	-4.298571	-3.069730
20	6	0	-1.120018	-3.826682	-0.808876
21	1	0	-1.482840	-1.785466	-0.328385
22	1	0	-0.223605	-2.634516	-4.382595
23	1	0	-0.086947	-5.000277	-3.831612
24	1	0	-1.315671	-4.166562	0.203116
25	1	0	-0.658697	-5.801907	-1.544040
26	6	0	-1.805017	0.256115	-1.867677
27	1	0	-2.086428	1.233464	-2.253377
28	1	0	-2.597838	-0.204392	-1.279572
29	6	0	2.726523	-1.668183	-2.451013
30	7	0	2.875507	-2.006692	-1.119125
31	16	0	3.158269	-2.668495	-3.735682
32	6	0	3.124777	-3.300258	-0.595935
33	6	0	3.570972	-5.824991	0.544465
34	6	0	2.275701	-3.780445	0.405393
35	6	0	4.224681	-4.067376	-1.000675
36	6	0	4.437943	-5.318389	-0.426251
37	6	0	2.497046	-5.043032	0.960986
38	1	0	1.440393	-3.174592	0.742906
39	1	0	4.914818	-3.681297	-1.735258
40	1	0	3.738862	-6.804983	0.974002
41	6	0	2.018996	0.256450	-3.876802
42	1	0	2.827703	0.988662	-4.014468
43	1	0	2.137867	-0.495795	-4.659218
44	1	0	2.033259	0.119221	-1.761990
45	1	0	2.421830	-1.372936	-0.462679
46	1	0	-1.103013	0.615077	-0.779872
47	6	0	-2.207996	0.504394	2.646653
48	6	0	-1.200232	0.352626	3.563554

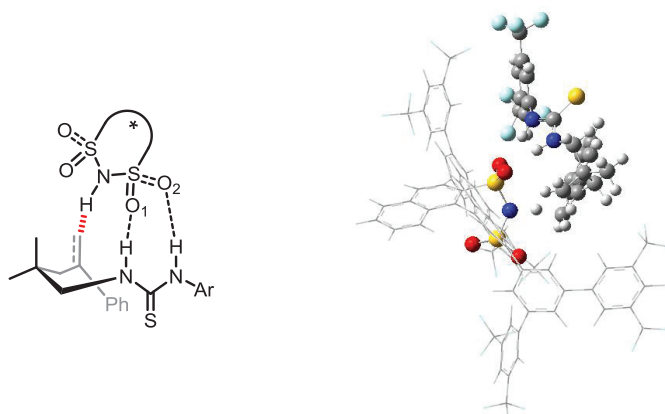
49	6	0	-3.371618	-1.369108	3.581070
50	6	0	-1.224500	-0.720849	4.501260
51	6	0	-3.325727	-0.384952	2.642094
52	6	0	-2.336909	-1.577172	4.513929
53	6	0	-0.159541	-0.976623	5.407099
54	1	0	-3.250372	-3.284161	5.448442
55	1	0	-4.223775	-2.018197	3.609831
56	6	0	-0.229771	-2.012846	6.278454
57	1	0	0.702833	-0.347259	5.390197
58	1	0	0.582276	-2.202774	6.950587
59	6	0	-1.364222	-2.856230	6.305787
60	1	0	-1.400643	-3.667880	7.003992
61	6	0	-2.389910	-2.644768	5.446239
62	6	0	2.085547	2.127258	2.720068
63	1	0	3.082055	3.549893	3.898264
64	6	0	2.216546	2.923842	3.812054
65	6	0	-0.011219	1.278943	3.612521
66	6	0	1.238703	2.976753	4.827721
67	6	0	0.941454	1.285435	2.629718
68	6	0	0.101445	2.158528	4.727086
69	6	0	1.372333	3.854318	5.933515
70	1	0	-1.781619	1.658207	5.645945
71	6	0	0.403220	3.920341	6.878556
72	1	0	2.248323	4.468579	6.000282
73	1	0	0.502794	4.588272	7.710275
74	6	0	-0.753998	3.115491	6.768194
75	1	0	-1.519747	3.190856	7.513374
76	6	0	-0.902725	2.259674	5.727288
77	16	0	0.598635	0.303217	1.162131
78	8	0	0.280783	-1.076717	1.561499
79	8	0	1.709109	0.426352	0.193211
80	16	0	-1.934892	1.765529	1.369082
81	8	0	-1.529947	3.006983	2.025852
82	8	0	-3.056832	1.834866	0.425562
83	7	0	-0.682671	1.096331	0.458816
84	6	0	-4.474372	-0.320407	1.685411
85	6	0	-6.648693	-0.275470	-0.039176
86	6	0	-4.784885	-1.434832	0.920675

87	6	0	-5.274884	0.803655	1.598613
88	6	0	-6.337680	0.847681	0.715144
89	6	0	-5.883647	-1.426632	0.068926
90	1	0	-4.177386	-2.315003	0.995004
91	1	0	-5.050867	1.663373	2.193419
92	1	0	-7.454924	-0.236066	-0.744895
93	6	0	3.180552	2.208650	1.697874
94	6	0	5.265441	2.485207	-0.101240
95	6	0	3.119370	3.176211	0.710136
96	6	0	4.288625	1.384964	1.789599
97	6	0	5.336638	1.519207	0.889870
98	6	0	4.156849	3.313989	-0.199326
99	1	0	2.248773	3.796699	0.632111
100	1	0	4.342244	0.645883	2.564090
101	1	0	6.073640	2.589777	-0.798026
102	6	0	4.061419	4.307193	-1.303257
103	6	0	3.830953	6.125264	-3.405537
104	6	0	3.788498	5.644466	-1.054504
105	6	0	4.221646	3.895504	-2.616589
106	6	0	4.105236	4.799466	-3.652440
107	6	0	3.673880	6.536632	-2.099826
108	1	0	3.681818	5.992089	-0.047609
109	1	0	4.414276	2.863672	-2.828341
110	1	0	3.740661	6.822719	-4.210768
111	6	0	6.535048	0.637501	0.980151
112	6	0	8.780213	-1.002967	1.142244
113	6	0	6.416756	-0.734273	0.830468
114	6	0	7.791413	1.180219	1.209108
115	6	0	8.898228	0.362911	1.284520
116	6	0	7.535706	-1.540656	0.912113
117	1	0	5.456164	-1.169425	0.638200
118	1	0	7.900354	2.236070	1.349348
119	1	0	9.637854	-1.635455	1.223041
120	6	0	-7.057025	2.133020	0.512323
121	6	0	-8.330375	4.569370	0.090541
122	6	0	-8.436609	2.223915	0.577705
123	6	0	-6.315733	3.274509	0.230099
124	6	0	-6.956470	4.478688	0.030737

125	6	0	-9.061894	3.435430	0.360938
126	1	0	-9.023355	1.360526	0.816478
127	1	0	-5.247739	3.201719	0.154538
128	1	0	-8.820092	5.505888	-0.071416
129	6	0	-6.224703	-2.634248	-0.733563
130	6	0	-6.885783	-4.899664	-2.220687
131	6	0	-5.276440	-3.259707	-1.529130
132	6	0	-7.509032	-3.159512	-0.694230
133	6	0	-7.830682	-4.274823	-1.437194
134	6	0	-5.610395	-4.384384	-2.257021
135	1	0	-4.282750	-2.865431	-1.594272
136	1	0	-8.251023	-2.710254	-0.066797
137	1	0	-7.138071	-5.768884	-2.789245
138	6	0	3.327519	7.954336	-1.816421
139	6	0	4.300909	4.324601	-5.045671
140	6	0	10.240519	0.967387	1.485384
141	6	0	7.377696	-3.007078	0.719389
142	6	0	-4.563677	-5.059308	-3.068406
143	6	0	-9.226651	-4.783995	-1.423700
144	6	0	-10.545500	3.503205	0.376318
145	6	0	-6.152297	5.704177	-0.214842
146	6	0	1.506758	-5.569129	1.962481
147	6	0	5.671682	-6.105983	-0.784619
148	9	0	3.811727	8.777324	-2.770514
149	9	0	3.819733	8.349614	-0.622720
150	9	0	1.992016	8.158741	-1.769458
151	9	0	5.602627	4.197176	-5.378964
152	9	0	3.748125	3.094316	-5.220135
153	9	0	3.740703	5.161780	-5.940515
154	9	0	10.166135	2.092285	2.230326
155	9	0	10.827974	1.319857	0.319080
156	9	0	11.081603	0.107817	2.097982
157	9	0	8.492287	-3.675784	1.067920
158	9	0	6.345845	-3.480491	1.448884
159	9	0	7.106899	-3.311158	-0.573416
160	9	0	6.671306	-5.900207	0.109551
161	9	0	6.150451	-5.788570	-2.000673
162	9	0	5.423892	-7.437841	-0.777405

163	9	0	1.951948	-6.671686	2.599010
164	9	0	1.202097	-4.649463	2.902530
165	9	0	0.329735	-5.909907	1.365399
166	9	0	-3.751481	-5.839785	-2.318416
167	9	0	-3.765798	-4.161621	-3.685825
168	9	0	-5.108539	-5.856731	-4.011492
169	9	0	-9.270132	-6.108020	-1.682099
170	9	0	-10.003872	-4.181859	-2.352527
171	9	0	-9.817826	-4.563962	-0.229228
172	9	0	-11.068897	2.615430	1.250438
173	9	0	-10.980998	4.735934	0.711831
174	9	0	-11.088702	3.215744	-0.829264
175	9	0	-6.865642	6.630124	-0.893236
176	9	0	-5.747358	6.294127	0.932767
177	9	0	-5.036532	5.429474	-0.921684

### TS<sub>R</sub>



Imaginary Freq = 1: -1404.23 cm<sup>-1</sup>

Electronic Energy (EE) at ONIOM(B3LYP/6-31G(d): HF/3-21G\*) =  
-7911.0993 Hartree

Zero-point Energy Correction = 1.28726 Hartree

Thermal Correction to Energy = 1.390473 Hartree

Thermal Correction to Enthalpy = 1.391417 Hartree

Thermal Correction to Free Energy = 1.129337 Hartree

EE + Zero-point Energy = -7909.812020 Hartree

EE + Thermal Energy Correction = -7909.708807 Hartree



EE + Thermal Enthalpy Correction = -7909.707863 Hartree

EE + Thermal Free Energy Correction = -7909.969943 Hartree

E<sub>sp</sub> (RM06-2X/6-311+G(d,p)/SMD (toluene)) = -7962.475851

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.846994	-2.690366	-1.913894
2	6	0	-1.334194	-4.126236	-1.772590
3	6	0	-1.390876	-4.904667	-0.407788
4	7	0	1.031922	-4.384957	0.175225
5	1	0	-2.367028	-4.095207	-2.143763
6	1	0	-0.797573	-4.764395	-2.476102
7	6	0	-2.213804	-6.184878	-0.685873
8	1	0	-3.242586	-5.957728	-0.975550
9	1	0	-1.765555	-6.777538	-1.492185
10	1	0	-2.249122	-6.818174	0.208319
11	6	0	-2.088119	-4.118824	0.715339
12	1	0	-3.101309	-3.827241	0.414937
13	1	0	-2.183522	-4.752922	1.604775
14	1	0	-1.547482	-3.219890	1.016987
15	6	0	0.304812	-2.400514	-2.780551
16	6	0	2.565904	-1.828504	-4.381569
17	6	0	0.542103	-1.093964	-3.275815
18	6	0	1.225743	-3.410360	-3.141243
19	6	0	2.343111	-3.126212	-3.919926
20	6	0	1.653726	-0.817408	-4.061108
21	1	0	-0.152357	-0.289426	-3.076996
22	1	0	1.114125	-4.422708	-2.775982
23	1	0	3.046197	-3.921777	-4.145016
24	1	0	1.808714	0.192584	-4.429417
25	1	0	3.446795	-1.601754	-4.974203
26	6	0	-1.580580	-1.652671	-1.321170
27	1	0	-2.543449	-1.913543	-0.889038
28	1	0	-1.592601	-0.680609	-1.812760
29	6	0	2.321224	-4.551034	-0.248143

30	7	0	3.102921	-3.450636	0.007963
31	16	0	2.888693	-5.928405	-1.043882
32	6	0	4.413643	-3.228105	-0.478289
33	6	0	7.002247	-2.650996	-1.390677
34	6	0	4.650464	-2.082485	-1.240863
35	6	0	5.475078	-4.085520	-0.161688
36	6	0	6.757361	-3.787107	-0.612927
37	6	0	5.941268	-1.804454	-1.701705
38	1	0	3.828858	-1.414771	-1.476865
39	1	0	5.294916	-4.963134	0.443929
40	1	0	8.010266	-2.403349	-1.694954
41	6	0	0.004509	-5.405593	0.045080
42	1	0	-0.120034	-5.932349	1.002879
43	1	0	0.385422	-6.137962	-0.669200
44	1	0	0.858990	-3.657654	0.868841
45	1	0	2.618832	-2.594440	0.280838
46	1	0	-0.939367	-1.070585	-0.257610
47	6	0	-2.271951	-0.259476	2.798325
48	6	0	-1.365106	-0.566459	3.784614
49	6	0	-3.875357	-1.755236	3.768230
50	6	0	-1.650803	-1.599103	4.722522
51	6	0	-3.580127	-0.835554	2.809708
52	6	0	-2.921608	-2.196222	4.708671
53	6	0	-0.683564	-2.067402	5.651348
54	1	0	-4.199064	-3.659087	5.632131
55	1	0	-4.866148	-2.161368	3.812118
56	6	0	-0.990086	-3.061522	6.520624
57	1	0	0.294118	-1.636025	5.644567
58	1	0	-0.251315	-3.418643	7.209034
59	6	0	-2.280489	-3.641175	6.524356
60	1	0	-2.505061	-4.420390	7.224549
61	6	0	-3.220848	-3.220557	5.643595
62	6	0	1.963032	1.209776	3.115747
63	1	0	2.879517	2.658607	4.326303
64	6	0	2.086442	1.941920	4.249744
65	6	0	-0.101134	0.237978	3.962578
66	6	0	1.129134	1.876798	5.286459
67	6	0	0.895404	0.272122	3.030499

68	6	0	-0.018063	1.079042	5.116858
69	6	0	1.254077	2.685869	6.443655
70	1	0	-1.936690	0.584710	5.965752
71	6	0	0.268417	2.715674	7.373465
72	1	0	2.137446	3.281807	6.559965
73	1	0	0.364381	3.330820	8.245316
74	6	0	-0.902914	1.948262	7.188262
75	1	0	-1.686228	1.998915	7.917156
76	6	0	-1.045666	1.156925	6.096039
77	16	0	0.748348	-0.850705	1.637912
78	8	0	0.362475	-2.173729	2.174336
79	8	0	1.963135	-0.854265	0.807541
80	16	0	-1.693149	0.746558	1.391985
81	8	0	-1.157089	2.022632	1.879055
82	8	0	-2.691629	0.795597	0.321362
83	7	0	-0.474556	-0.201730	0.705938
84	6	0	-4.667137	-0.444919	1.861714
85	6	0	-6.642679	0.251636	0.039219
86	6	0	-5.311452	-1.419668	1.121585
87	6	0	-5.041709	0.879427	1.707630
88	6	0	-6.000097	1.235874	0.777921
89	6	0	-6.305569	-1.080545	0.212043
90	1	0	-5.026794	-2.446050	1.237691
91	1	0	-4.552110	1.644551	2.275218
92	1	0	-7.354498	0.531774	-0.711955
93	6	0	2.773717	1.600179	1.923100
94	6	0	4.094696	2.481786	-0.341621
95	6	0	2.073850	2.223747	0.905617
96	6	0	4.145731	1.469499	1.839062
97	6	0	4.814477	1.928727	0.709650
98	6	0	2.717627	2.638958	-0.246565
99	1	0	1.022132	2.382829	1.028182
100	1	0	4.695655	1.022636	2.643541
101	1	0	4.609032	2.748480	-1.242457
102	6	0	-6.915695	-2.141277	-0.633315
103	6	0	-7.996002	-4.140646	-2.252056
104	6	0	-6.093626	-3.013081	-1.330431
105	6	0	-8.289045	-2.280660	-0.761592

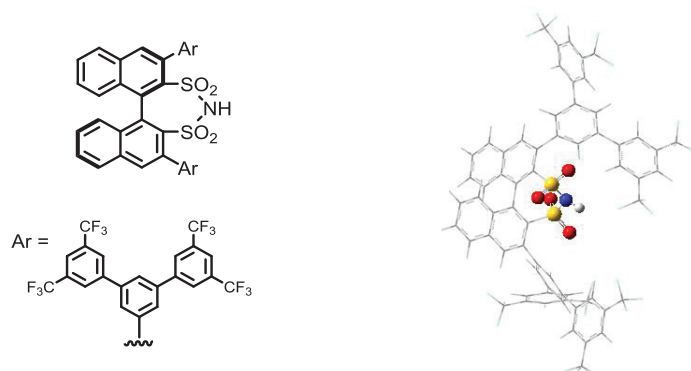
106	6	0	-8.815256	-3.273327	-1.563516
107	6	0	-6.632027	-4.004501	-2.120901
108	1	0	-5.031944	-2.893529	-1.278757
109	1	0	-8.947382	-1.617094	-0.239448
110	1	0	-8.411806	-4.891118	-2.890538
111	6	0	-6.223885	2.676371	0.473011
112	6	0	-6.575609	5.356078	-0.167624
113	6	0	-7.459778	3.277305	0.634590
114	6	0	-5.160852	3.424073	-0.017953
115	6	0	-5.347698	4.754326	-0.336092
116	6	0	-7.625131	4.610758	0.320977
117	1	0	-8.290206	2.711026	1.004067
118	1	0	-4.207764	2.954566	-0.165028
119	1	0	-6.712344	6.386262	-0.416528
120	6	0	1.898572	3.227696	-1.340458
121	6	0	0.255510	4.373762	-3.288126
122	6	0	0.634708	2.712510	-1.600946
123	6	0	2.332536	4.315128	-2.084604
124	6	0	1.520798	4.866172	-3.054427
125	6	0	-0.177144	3.293868	-2.551849
126	1	0	0.289618	1.856040	-1.058058
127	1	0	3.288683	4.754974	-1.889924
128	1	0	-0.379881	4.826605	-4.018572
129	6	0	6.300609	1.845351	0.633719
130	6	0	9.084756	1.745337	0.534185
131	6	0	7.042185	2.985775	0.369601
132	6	0	6.972840	0.648459	0.845929
133	6	0	8.349938	0.607169	0.788061
134	6	0	8.420592	2.931006	0.327199
135	1	0	6.545519	3.919842	0.205758
136	1	0	6.422939	-0.247989	1.041793
137	1	0	10.153352	1.707184	0.496613
138	6	0	-5.719022	-4.955168	-2.802233
139	6	0	-10.290312	-3.440805	-1.643381
140	6	0	-4.218722	5.539696	-0.903763
141	6	0	-8.937456	5.260802	0.565532
142	6	0	-1.553752	2.768153	-2.736771
143	6	0	2.030261	5.993599	-3.877376

144	6	0	9.193456	4.182292	0.117354
145	6	0	9.070387	-0.683516	0.947921
146	6	0	6.157383	-0.624264	-2.610117
147	6	0	7.911456	-4.697115	-0.277509
148	9	0	9.002254	-3.994078	0.101143
149	9	0	8.279090	-5.442115	-1.344753
150	9	0	7.611138	-5.551373	0.722708
151	9	0	5.826823	-0.917626	-3.894133
152	9	0	5.393870	0.437763	-2.254341
153	9	0	7.440333	-0.211762	-2.620223
154	9	0	10.198257	-0.529671	1.674523
155	9	0	9.461357	-1.198346	-0.243917
156	9	0	8.299674	-1.610538	1.550203
157	9	0	8.531123	5.039152	-0.689921
158	9	0	9.421735	4.846920	1.273795
159	9	0	10.401521	3.930510	-0.428731
160	9	0	2.695549	5.567895	-4.976402
161	9	0	2.895869	6.760916	-3.180334
162	9	0	1.024029	6.778478	-4.316623
163	9	0	-3.982029	5.247836	-2.202948
164	9	0	-3.066612	5.310477	-0.242754
165	9	0	-4.473156	6.867072	-0.845646
166	9	0	-9.133066	6.308035	-0.263490
167	9	0	-9.956541	4.389195	0.395559
168	9	0	-9.052938	5.739472	1.826065
169	9	0	-10.926183	-2.256698	-1.515655
170	9	0	-10.658133	-3.998783	-2.815704
171	9	0	-10.769632	-4.241860	-0.664651
172	9	0	-6.269395	-5.480706	-3.911948
173	9	0	-4.549754	-4.359572	-3.144947
174	9	0	-5.378746	-5.998564	-2.002435
175	9	0	-2.370994	3.068586	-1.704217
176	9	0	-2.126649	3.238975	-3.859289
177	9	0	-1.555992	1.405717	-2.821223

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## 4.2.2 Structures optimized at the ONIOM(M06-2x/6-31G(d): HF/3-21G\*) level of theory

### DSI2



ONIOM(M06-2X/6-31G(d):HF/3-21G\*) (Hartree)

Imaginary Freq	0
Electronic Energy (EE)	-5952.539625
Zero-point Energy Correction	0.880399
Thermal Correction to Energy	0.954131
Thermal Correction to Enthalpy	0.955076
Thermal Correction to Free Energy	0.750302
EE + Zero-point Energy	-5951.659226
EE + Thermal Energy Correction	-5951.585493
EE + Thermal Enthalpy Correction	-5951.584549
EE + Thermal Free Energy Correction	-5951.789323
$E_{sp}$ (RM06-2X/6-311+(d,p)/SMD (toluene))	-6004.254989

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.170289	-1.857381	-1.278963
2	6	0	-0.197641	-2.629115	-1.860010
3	6	0	-2.685936	-2.353874	-3.062562
4	6	0	-0.422285	-3.230369	-3.130480
5	6	0	-2.459513	-1.735939	-1.874499
6	6	0	-1.681931	-3.083304	-3.734403

7	6	0	0.590805	-3.943134	-3.825815
8	1	0	-2.888382	-3.558170	-5.448733
9	1	0	-3.659097	-2.287524	-3.506401
10	6	0	0.338647	-4.485637	-5.042381
11	1	0	1.558520	-4.037845	-3.383042
12	1	0	1.110770	-5.013344	-5.564461
13	6	0	-0.937283	-4.354746	-5.637655
14	1	0	-1.117128	-4.792986	-6.598579
15	6	0	-1.920681	-3.672148	-5.001966
16	6	0	3.250987	-2.166203	-0.237318
17	1	0	4.370603	-3.642003	0.747286
18	6	0	3.481778	-3.435521	0.185677
19	6	0	1.119417	-2.884570	-1.169873
20	6	0	2.554360	-4.478855	-0.016012
21	6	0	2.056439	-1.905118	-0.977047
22	6	0	1.337304	-4.199797	-0.660143
23	6	0	2.801849	-5.784609	0.477812
24	1	0	-0.568565	-5.030747	-1.228351
25	6	0	1.866843	-6.757057	0.348132
26	1	0	3.736826	-5.982158	0.963379
27	1	0	2.054020	-7.742121	0.725370
28	6	0	0.630005	-6.474011	-0.274274
29	1	0	-0.107859	-7.246402	-0.353709
30	6	0	0.369901	-5.236012	-0.762694
31	16	0	1.635469	-0.278507	-1.614613
32	8	0	1.032394	-0.426811	-2.922568
33	8	0	2.678921	0.716130	-1.458382
34	16	0	-0.734392	-0.894635	0.171282
35	8	0	-0.142925	-1.752425	1.171123
36	8	0	-1.783639	0.036846	0.521181
37	7	0	0.492685	0.137619	-0.430324
38	6	0	-3.624701	-1.039452	-1.239385
39	6	0	-5.871892	0.162780	-0.158325
40	6	0	-4.089798	0.154281	-1.759842
41	6	0	-4.293224	-1.638231	-0.185018
42	6	0	-5.417523	-1.040757	0.362332
43	6	0	-5.213154	0.764775	-1.219138
44	1	0	-3.581408	0.606300	-2.588118

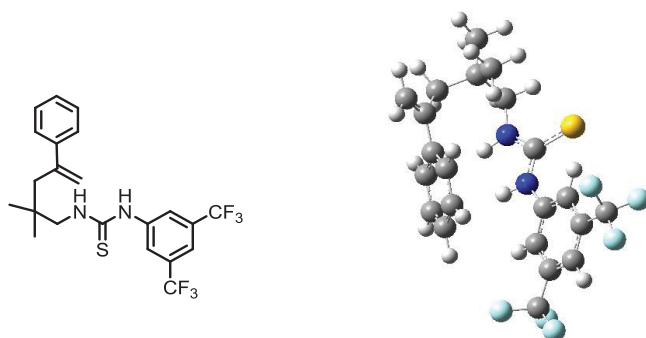
45	1	0	-3.927931	-2.559791	0.223295
46	1	0	-6.739390	0.630756	0.263509
47	6	0	4.205119	-1.103351	0.199979
48	6	0	5.893119	0.930725	1.044870
49	6	0	3.728200	-0.082322	0.998038
50	6	0	5.546436	-1.131529	-0.139212
51	6	0	6.401542	-0.127759	0.303945
52	6	0	4.547344	0.965643	1.374875
53	1	0	2.695399	-0.074486	1.279788
54	1	0	5.922228	-1.907127	-0.776827
55	1	0	6.538683	1.734461	1.340229
56	6	0	3.887954	2.154728	1.978948
57	6	0	2.443375	4.309196	2.989544
58	6	0	4.244255	2.726127	3.185501
59	6	0	2.820874	2.690585	1.268965
60	6	0	2.103030	3.746230	1.777534
61	6	0	3.519571	3.796632	3.679310
62	1	0	5.068409	2.338118	3.748743
63	1	0	2.570751	2.264986	0.319603
64	1	0	1.883033	5.128264	3.388363
65	6	0	7.844226	-0.163895	-0.063920
66	6	0	10.545070	-0.241001	-0.752759
67	6	0	8.614542	-1.282540	0.214633
68	6	0	8.444940	0.916420	-0.694017
69	6	0	9.780213	0.868636	-1.035967
70	6	0	9.951380	-1.312073	-0.124145
71	1	0	8.173592	-2.127382	0.702275
72	1	0	7.869191	1.788792	-0.925857
73	1	0	11.580893	-0.269881	-1.016724
74	6	0	-6.127791	-1.681405	1.503377
75	6	0	-7.457264	-2.881388	3.638632
76	6	0	-6.369820	-0.974486	2.672282
77	6	0	-6.557916	-2.996423	1.419306
78	6	0	-7.220420	-3.582374	2.477992
79	6	0	-7.023951	-1.577028	3.726219
80	1	0	-6.039068	0.039721	2.762715
81	1	0	-6.382578	-3.561526	0.527032
82	1	0	-7.968165	-3.341101	4.457840



83	6	0	-5.708296	2.053093	-1.777788
84	6	0	-6.629958	4.464619	-2.826750
85	6	0	-7.024918	2.186864	-2.190757
86	6	0	-4.858741	3.143633	-1.894480
87	6	0	-5.323957	4.335395	-2.409267
88	6	0	-7.472477	3.381796	-2.715156
89	1	0	-7.698917	1.358826	-2.110367
90	1	0	-3.839949	3.064744	-1.574881
91	1	0	-6.983128	5.389837	-3.230167
92	1	0	0.181518	1.112332	-0.489392
93	6	0	-4.418747	5.513250	-2.460685
94	6	0	-8.866079	3.479315	-3.221042
95	9	0	-3.135971	5.137065	-2.650461
96	9	0	-4.444570	6.234828	-1.317876
97	9	0	-4.767196	6.354959	-3.457365
98	9	0	-8.976484	3.080552	-4.509076
99	9	0	-9.321652	4.748498	-3.168578
100	9	0	-9.703346	2.696406	-2.506183
101	6	0	-7.212460	-0.826978	4.995539
102	6	0	-7.735101	-4.968846	2.337524
103	9	0	-7.835211	-5.581679	3.535710
104	9	0	-6.930713	-5.712588	1.545941
105	9	0	-8.965297	-5.010059	1.776664
106	9	0	-8.302429	-1.261982	5.663592
107	9	0	-6.163419	-0.962423	5.837225
108	9	0	-7.354415	0.496164	4.766439
109	6	0	0.973741	4.290722	0.981961
110	6	0	3.867072	4.353725	5.012149
111	6	0	10.386788	2.006799	-1.774552
112	6	0	10.771473	-2.493490	0.248333
113	9	0	11.278813	-2.399121	1.498398
114	9	0	10.037461	-3.628193	0.219504
115	9	0	11.821842	-2.647059	-0.585208
116	9	0	9.813202	3.178794	-1.427165
117	9	0	11.712810	2.095122	-1.536114
118	9	0	10.241601	1.889303	-3.113867
119	9	0	3.265812	3.688821	6.025111
120	9	0	5.196274	4.278319	5.243154

121	9	0	3.490152	5.645468	5.120194
122	9	0	0.356578	3.304908	0.271945
123	9	0	1.359481	5.213110	0.075005
124	9	0	0.048233	4.875789	1.767487

### Alkenyl thiourea



RM06-2X/6-31G(d) (Hartree)

Imaginary Freq	0
Electronic Energy (EE)	-1957.728467
Zero-point Energy Correction	0.414555
Thermal Correction to Energy	0.443339
Thermal Correction to Enthalpy	0.444283
Thermal Correction to Free Energy	0.352738
EE + Zero-point Energy	-1957.313912
EE + Thermal Energy Correction	-1957.285128
EE + Thermal Enthalpy Correction	-1957.284184
EE + Thermal Free Energy Correction	-1957.375729

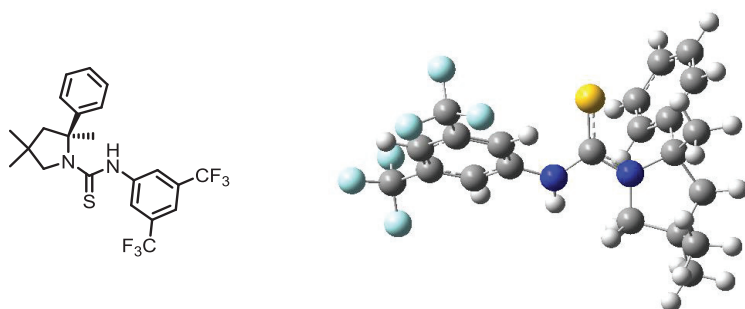
E<sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene)) -1958.245910

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.009241	-1.052511	1.316408
2	6	0	-4.911283	-0.583107	0.188951
3	6	0	-4.658831	0.847527	-0.395336
4	7	0	-2.458452	0.209439	-1.421424

5	1	0	-5.933126	-0.593615	0.583569
6	1	0	-4.906678	-1.306694	-0.639576
7	6	0	-6.011354	1.397639	-0.868145
8	1	0	-6.686942	1.539630	-0.018368
9	1	0	-6.494391	0.711203	-1.573827
10	1	0	-5.886493	2.365259	-1.365861
11	6	0	-4.063761	1.787509	0.653280
12	1	0	-4.693823	1.816725	1.549209
13	1	0	-3.980464	2.800957	0.248932
14	1	0	-3.062495	1.466978	0.958961
15	6	0	-2.697013	-1.703361	1.039478
16	6	0	-0.230244	-3.003954	0.632627
17	6	0	-2.572691	-2.728108	0.089681
18	6	0	-1.556507	-1.331450	1.764955
19	6	0	-0.339546	-1.977885	1.570524
20	6	0	-1.350699	-3.370923	-0.113958
21	1	0	-3.446102	-3.048824	-0.473774
22	1	0	-1.631545	-0.514683	2.477140
23	1	0	0.533457	-1.667933	2.137286
24	1	0	-1.280336	-4.173555	-0.842215
25	1	0	0.723992	-3.500161	0.480327
26	6	0	-4.393850	-0.900676	2.587939
27	1	0	-5.347034	-0.442043	2.836678
28	1	0	-3.779973	-1.241311	3.416439
29	6	0	-1.283524	0.864844	-1.273771
30	7	0	-0.260614	-0.014051	-1.011594
31	16	0	-1.108814	2.520208	-1.440859
32	6	0	1.079350	0.193313	-0.663436
33	6	0	3.778131	0.309701	0.138269
34	6	0	1.934708	-0.897278	-0.853099
35	6	0	1.586748	1.353295	-0.070613
36	6	0	2.926192	1.395136	0.304610
37	6	0	3.257193	-0.840906	-0.440130
38	1	0	1.555838	-1.802168	-1.321193
39	1	0	0.944200	2.207672	0.097678
40	1	0	4.812944	0.357524	0.454207
41	6	0	-3.757770	0.806398	-1.643945
42	1	0	-3.591757	1.819304	-2.018711

43	1	0	-4.260206	0.230511	-2.432490
44	1	0	-2.457956	-0.788865	-1.248124
45	1	0	-0.512167	-0.998943	-1.014813
46	6	0	4.090667	-2.077817	-0.610178
47	6	0	3.437302	2.672042	0.915564
48	9	0	3.592047	-3.094979	0.118997
49	9	0	5.357755	-1.894071	-0.227791
50	9	0	4.102341	-2.495398	-1.884028
51	9	0	4.687628	2.538098	1.379228
52	9	0	3.442960	3.672935	0.026602
53	9	0	2.665236	3.064015	1.939811

## Product



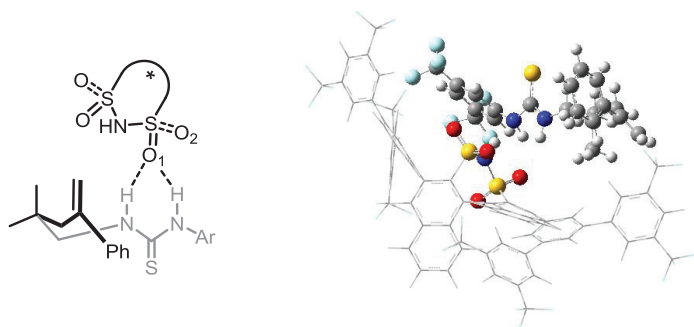
RM06-2X/6-31G(d) (Hartree)

Imaginary Freq	0
Frequencies scaled by	1
Electronic Energy (EE)	-1957.742936
Zero-point Energy Correction	0.415338
Thermal Correction to Energy	0.443727
Thermal Correction to Enthalpy	0.444672
Thermal Correction to Free Energy	0.354185
EE + Zero-point Energy	-1957.327598
EE + Thermal Energy Correction	-1957.299209
EE + Thermal Enthalpy Correction	-1957.298265
EE + Thermal Free Energy Correction	-1957.388752
E <sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene))	-1958.258339

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	2.188584	-0.220110	-0.784501
2	6	0	3.520026	-0.585549	-0.209105
3	6	0	4.484975	0.099069	-1.217459
4	6	0	3.716983	0.305209	-2.532631
5	6	0	2.305137	0.581142	-2.006644
6	1	0	4.777778	1.078055	-0.822871
7	1	0	5.398573	-0.488133	-1.351213
8	1	0	1.533997	0.289333	-2.728376
9	1	0	2.186099	1.657064	-1.785822
10	6	0	4.267124	1.491584	-3.322387
11	1	0	4.291498	2.399257	-2.710079
12	1	0	3.656569	1.691740	-4.209934
13	1	0	5.287664	1.283878	-3.660796
14	6	0	3.709196	-0.941481	-3.424638
15	1	0	3.148068	-1.763749	-2.974286
16	1	0	4.730237	-1.286498	-3.617977
17	1	0	3.241930	-0.712273	-4.388730
18	6	0	3.741715	-2.104250	-0.189650
19	1	0	4.803206	-2.315786	-0.030227
20	1	0	3.452605	-2.545712	-1.144533
21	1	0	3.157653	-2.579201	0.597668
22	6	0	3.702469	0.050428	1.171342
23	6	0	4.118813	1.332737	3.635884
24	6	0	3.235529	1.349527	1.394156
25	6	0	4.387903	-0.592244	2.200273
26	6	0	4.595067	0.045123	3.422980
27	6	0	3.434052	1.985362	2.612437
28	1	0	2.693288	1.866354	0.604550
29	1	0	4.754681	-1.604012	2.067183
30	1	0	5.124996	-0.476976	4.213848
31	1	0	3.049278	2.989007	2.765992
32	1	0	4.274048	1.823719	4.591370
33	6	0	1.000041	-0.453820	-0.170546
34	16	0	0.789498	-1.426031	1.169752

35	7	0	-0.049765	0.230341	-0.766506
36	6	0	-1.414691	0.134573	-0.436851
37	6	0	-4.176145	0.023447	0.043083
38	6	0	-2.164980	1.303907	-0.357840
39	6	0	-2.050813	-1.103031	-0.284614
40	6	0	-3.413951	-1.141602	-0.038396
41	6	0	-3.537404	1.240668	-0.125854
42	1	0	-1.683534	2.271283	-0.475944
43	1	0	-1.481181	-2.019667	-0.366352
44	1	0	-5.244857	-0.023108	0.218233
45	1	0	0.192340	1.080664	-1.258350
46	6	0	-4.093237	-2.467175	0.171861
47	6	0	-4.302128	2.529966	-0.025823
48	9	0	-4.207621	-2.759866	1.473915
49	9	0	-3.419172	-3.470174	-0.404776
50	9	0	-5.334335	-2.463158	-0.338697
51	9	0	-5.624694	2.332919	-0.069268
52	9	0	-4.027961	3.176848	1.116505
53	9	0	-3.979850	3.364331	-1.027655

### Complex\_S



ONIOM(M06-2X/6-31G(d):HF/3-21G\*) (Hartree)

Imaginary Freq = 0

Electronic Energy (EE) -7910.289362

Zero-point Energy Correction 1.297289

Thermal Correction to Energy 1.401309

Thermal Correction to Enthalpy 1.402253

Thermal Correction to Free Energy	1.132658
EE + Zero-point Energy	-7908.992073
EE + Thermal Energy Correction	-7908.888053
EE + Thermal Enthalpy Correction	-7908.887109
EE + Thermal Free Energy Correction	-7909.156704
 E <sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene))	 -7962.496885

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.602951	-6.516539	-1.267194
2	6	0	-4.306600	-5.636884	-2.460608
3	6	0	-3.560671	-4.312354	-2.156661
4	7	0	-1.354973	-3.335090	-1.586181
5	1	0	-5.261694	-5.360183	-2.920491
6	1	0	-3.736373	-6.187640	-3.219967
7	6	0	-3.590858	-3.471810	-3.439380
8	1	0	-4.620615	-3.238108	-3.720900
9	1	0	-3.114754	-4.004480	-4.269402
10	1	0	-3.057725	-2.524055	-3.306816
11	6	0	-4.280997	-3.567612	-1.029374
12	1	0	-5.351015	-3.517134	-1.242938
13	1	0	-3.895789	-2.545846	-0.925137
14	1	0	-4.160385	-4.083869	-0.070415
15	6	0	-3.492546	-7.299669	-0.660268
16	6	0	-1.390466	-8.752868	0.507827
17	6	0	-3.352744	-7.367945	0.731604
18	6	0	-2.553795	-7.968178	-1.456225
19	6	0	-1.516387	-8.692014	-0.877827
20	6	0	-2.312945	-8.087853	1.311183
21	1	0	-4.056539	-6.826789	1.358105
22	1	0	-2.640989	-7.935767	-2.538362
23	1	0	-0.801816	-9.206912	-1.512356
24	1	0	-2.215152	-8.117949	2.392068
25	1	0	-0.573370	-9.307770	0.957722
26	6	0	-5.837228	-6.561237	-0.754509

27	1	0	-6.641078	-5.969081	-1.183343
28	1	0	-6.079938	-7.198729	0.090957
29	6	0	-0.417952	-2.878183	-2.434329
30	7	0	-0.087685	-1.533209	-2.122831
31	16	0	0.269446	-3.729109	-3.678797
32	6	0	0.960629	-0.839076	-2.781467
33	6	0	3.035162	0.595808	-3.975279
34	6	0	2.267952	-1.326780	-2.712461
35	6	0	0.693808	0.366773	-3.418800
36	6	0	1.736703	1.080970	-4.002025
37	6	0	3.286656	-0.608325	-3.320194
38	1	0	2.477339	-2.250266	-2.184704
39	1	0	-0.319378	0.754673	-3.459066
40	1	0	3.840750	1.168682	-4.418402
41	6	0	-2.100466	-4.577782	-1.738921
42	1	0	-1.576748	-5.175310	-2.488337
43	1	0	-2.059168	-5.115925	-0.786762
44	1	0	-1.619737	-2.730441	-0.814619
45	1	0	-0.916187	-0.955380	-1.972662
46	1	0	0.302656	-1.264075	-0.318992
47	6	0	1.507242	-0.694723	3.006534
48	6	0	0.411785	-0.559211	3.811105
49	6	0	2.745642	0.868444	4.310952
50	6	0	0.459485	0.322313	4.937502
51	6	0	2.676371	0.087331	3.203456
52	6	0	1.667379	0.989361	5.213023
53	6	0	-0.655875	0.568906	5.785078
54	1	0	2.687454	2.334712	6.540676
55	1	0	3.623487	1.459948	4.477110
56	6	0	-0.543194	1.396740	6.853192
57	1	0	-1.588277	0.094085	5.578193
58	1	0	-1.391709	1.573876	7.482524
59	6	0	0.682562	2.035638	7.145368
60	1	0	0.751266	2.682730	7.996271
61	6	0	1.756873	1.840911	6.343130
62	6	0	-3.114370	-0.845534	2.467858
63	1	0	-4.737559	-1.773859	3.428432
64	6	0	-3.670910	-1.683419	3.377176

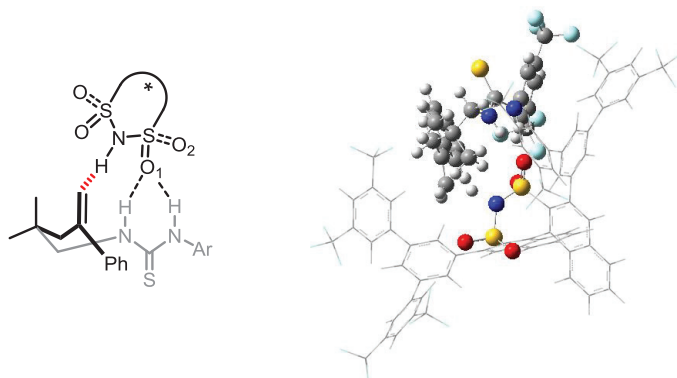


65	6	0	-0.936414	-1.165556	3.506755
66	6	0	-2.880889	-2.398675	4.312079
67	6	0	-1.689559	-0.687426	2.470858
68	6	0	-1.509761	-2.106617	4.411876
69	6	0	-3.451150	-3.343655	5.198209
70	1	0	0.319469	-2.571821	5.439441
71	6	0	-2.678185	-3.977182	6.117549
72	1	0	-4.499804	-3.553784	5.124968
73	1	0	-3.111482	-4.698706	6.780548
74	6	0	-1.297445	-3.695463	6.206814
75	1	0	-0.699198	-4.214496	6.927670
76	6	0	-0.727862	-2.779671	5.382155
77	16	0	-0.909077	0.104340	1.056243
78	8	0	-0.427657	1.430482	1.400927
79	8	0	-1.765847	-0.063955	-0.115018
80	16	0	1.349932	-1.830920	1.650682
81	8	0	0.611187	-2.987849	2.117661
82	8	0	2.578602	-2.029112	0.916193
83	7	0	0.390670	-0.883275	0.648136
84	6	0	3.759755	0.263076	2.181436
85	6	0	5.746086	0.834371	0.348700
86	6	0	3.637208	1.356674	1.341272
87	6	0	4.869985	-0.555684	2.097602
88	6	0	5.860012	-0.280687	1.167514
89	6	0	4.637908	1.665148	0.428417
90	1	0	2.745939	1.950761	1.389595
91	1	0	4.946875	-1.423566	2.720141
92	1	0	6.536818	1.060764	-0.337111
93	6	0	-3.982102	0.072206	1.671468
94	6	0	-5.580505	1.873221	0.287874
95	6	0	-5.085441	-0.376659	0.967821
96	6	0	-3.709358	1.432180	1.709775
97	6	0	-4.476020	2.335323	0.993812
98	6	0	-5.895664	0.523100	0.286102
99	1	0	-5.314943	-1.422607	0.941056
100	1	0	-2.882176	1.791849	2.287338
101	1	0	-6.168578	2.555130	-0.293787
102	6	0	-7.078050	0.011851	-0.464260

103	6	0	-9.285543	-1.064666	-1.784952
104	6	0	-8.361032	0.446836	-0.165123
105	6	0	-6.915196	-0.948105	-1.448539
106	6	0	-8.012843	-1.492324	-2.084654
107	6	0	-9.446439	-0.081680	-0.831708
108	1	0	-8.514831	1.175805	0.603791
109	1	0	-5.930645	-1.268726	-1.716789
110	1	0	-10.133879	-1.489375	-2.277251
111	6	0	-4.071089	3.767248	0.949320
112	6	0	-3.288498	6.436011	0.778828
113	6	0	-2.735352	4.102297	0.756093
114	6	0	-5.003717	4.785624	1.068075
115	6	0	-4.608358	6.104590	0.988288
116	6	0	-2.358471	5.425978	0.671601
117	1	0	-1.992480	3.335894	0.658078
118	1	0	-6.037748	4.555287	1.222378
119	1	0	-2.993159	7.458833	0.681153
120	6	0	7.033068	-1.187116	1.026329
121	6	0	9.225997	-2.874724	0.726454
122	6	0	7.294434	-1.791753	-0.195518
123	6	0	7.874888	-1.442892	2.094916
124	6	0	8.959048	-2.283992	1.940038
125	6	0	8.381565	-2.627010	-0.334031
126	1	0	6.649787	-1.610380	-1.027921
127	1	0	7.703836	-0.967563	3.038981
128	1	0	10.084781	-3.500002	0.603034
129	6	0	4.527771	2.880254	-0.425156
130	6	0	4.321323	5.170253	-2.006727
131	6	0	4.942659	2.866161	-1.750973
132	6	0	4.014912	4.060328	0.094519
133	6	0	3.903130	5.184311	-0.696088
134	6	0	4.847544	4.005821	-2.521109
135	1	0	5.307592	1.959370	-2.183080
136	1	0	3.718598	4.112940	1.121248
137	1	0	4.239852	6.046383	-2.614269
138	6	0	-10.808458	0.434563	-0.534924
139	6	0	-7.798717	-2.582501	-3.070412
140	6	0	-5.615967	7.179941	1.175382

141	6	0	-0.920803	5.763597	0.474353
142	6	0	5.375467	4.008312	-3.909693
143	6	0	3.292005	6.415102	-0.130050
144	6	0	8.622561	-3.306968	-1.633441
145	6	0	9.826466	-2.577064	3.109208
146	6	0	4.701975	-1.113599	-3.293595
147	6	0	1.464365	2.376293	-4.716567
148	9	0	-11.129184	1.507610	-1.291925
149	9	0	-10.913008	0.827837	0.753028
150	9	0	-11.750489	-0.502774	-0.771228
151	9	0	-6.815486	-2.278234	-3.948709
152	9	0	-7.403967	-3.739704	-2.469490
153	9	0	-8.915707	-2.861576	-3.766437
154	9	0	-5.264265	8.305916	0.520297
155	9	0	-5.772433	7.521198	2.475355
156	9	0	-6.833875	6.793977	0.734206
157	9	0	-0.786847	6.982252	-0.087494
158	9	0	-0.236818	5.787945	1.640396
159	9	0	-0.314363	4.852137	-0.312495
160	9	0	1.471113	2.215349	-6.045519
161	9	0	2.400789	3.294117	-4.429004
162	9	0	0.274253	2.889553	-4.380378
163	9	0	5.188136	-1.310029	-4.518718
164	9	0	5.532529	-0.218545	-2.691894
165	9	0	4.834236	-2.249501	-2.599708
166	9	0	6.681508	4.360372	-3.955790
167	9	0	4.715264	4.877568	-4.697393
168	9	0	5.306015	2.778418	-4.480594
169	9	0	3.756843	7.522954	-0.745937
170	9	0	3.547977	6.526357	1.191884
171	9	0	1.945542	6.437410	-0.262788
172	9	0	8.148739	-2.578607	-2.666032
173	9	0	9.941886	-3.520973	-1.834848
174	9	0	8.023934	-4.517050	-1.703595
175	9	0	11.077558	-2.909702	2.727340
176	9	0	9.905622	-1.515914	3.942713
177	9	0	9.361830	-3.608451	3.852506

**TS<sub>S</sub> at ONIOM(M06-2X/6-31G(d): HF/3-21G\*)**



ONIOM(M06-2X/6-31G(d): HF/3-21G\*) (Hartree)

Imaginary Freq = 1: -1214.90

Electronic Energy (EE)	-7910.273177
Zero-point Energy Correction	1.292317
Thermal Correction to Energy	1.395335
Thermal Correction to Enthalpy	1.396279
Thermal Correction to Free Energy	1.130681
EE + Zero-point Energy	-7908.980860
EE + Thermal Energy Correction	-7908.877842
EE + Thermal Enthalpy Correction	-7908.876898
EE + Thermal Free Energy Correction	-7909.142497

E<sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene)) -7962.485587

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.059570	-0.233406	-2.719696
2	6	0	-0.491784	0.316534	-4.002996
3	6	0	0.764877	1.247037	-3.961710
4	7	0	2.131585	-0.309760	-2.588414
5	1	0	-1.297734	0.910155	-4.448969
6	1	0	-0.267861	-0.490747	-4.703347
7	6	0	0.828816	1.949708	-5.325692
8	1	0	-0.021524	2.627512	-5.452170
9	1	0	0.811644	1.222663	-6.145178
10	1	0	1.749207	2.530850	-5.410455

11	6	0	0.677320	2.308508	-2.861034
12	1	0	-0.223503	2.920743	-2.976729
13	1	0	1.540493	2.977972	-2.932898
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17	6	0	-1.239399	-2.026352	-1.007544
18	6	0	-0.255664	-2.570651	-3.150795
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21	1	0	-1.642965	-1.309862	-0.301000
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30	7	0	2.598772	-1.977119	-1.141867
31	16	0	3.129891	-2.506715	-3.745309
32	6	0	2.800554	-3.269468	-0.620721
33	6	0	3.085729	-5.805905	0.544143
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39	1	0	4.481832	-3.831217	-1.851572
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41	6	0	2.054693	0.422730	-3.833116
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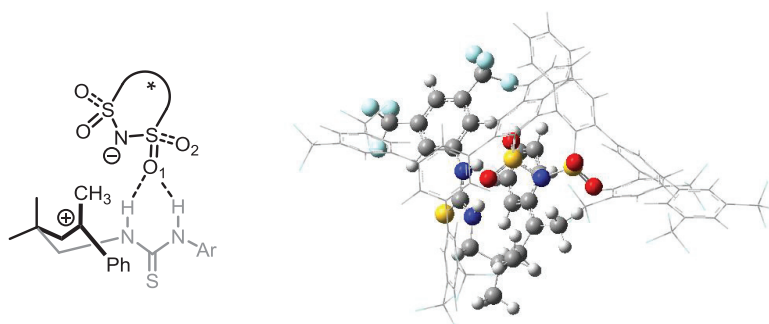
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134	6	0	-5.352215	-4.320546	-2.341373
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### Intermediate\_S



ONIOM(M06-2X/6-31G(d): HF/3-21G\*) (Hartree)

Imaginary Freq	0
Electronic Energy (EE)	-7910.283873
Zero-point Energy Correction	1.297233
Thermal Correction to Energy	1.400645
Thermal Correction to Enthalpy	1.401589
Thermal Correction to Free Energy	1.136369
EE + Zero-point Energy	-7908.986640
EE + Thermal Energy Correction	-7908.883227
EE + Thermal Enthalpy Correction	-7908.882283

EE + Thermal Free Energy Correction

-7909.147504

E<sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene))

-7962.495300

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Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
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3	6	0	0.544220	0.703415	-4.288555
4	7	0	1.865055	-0.555396	-2.572814
5	1	0	-1.407053	-0.038304	-4.842945
6	1	0	-0.162205	-1.298587	-4.671060
7	6	0	0.659069	1.106030	-5.764651
8	1	0	-0.241446	1.635538	-6.092548
9	1	0	0.798483	0.229684	-6.406934
10	1	0	1.511521	1.775150	-5.900990
11	6	0	0.294766	1.960816	-3.449670
12	1	0	-0.655861	2.441816	-3.698185
13	1	0	1.090037	2.674725	-3.679833
14	1	0	0.316110	1.772207	-2.370301
15	6	0	-1.043042	-1.918376	-2.199982
16	6	0	-0.822760	-4.332149	-0.802283
17	6	0	-1.355440	-1.985736	-0.811376
18	6	0	-0.622247	-3.102266	-2.860548
19	6	0	-0.536184	-4.298095	-2.164577
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21	1	0	-1.630939	-1.083464	-0.276141
22	1	0	-0.408174	-3.095700	-3.920623
23	1	0	-0.220755	-5.196474	-2.681922
24	1	0	-1.392379	-3.196784	0.948626
25	1	0	-0.717239	-5.260410	-0.248962
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27	1	0	-2.293407	1.147236	-3.010059
28	1	0	-2.906439	-0.012576	-1.797029
29	6	0	2.228213	-1.846909	-2.352564
30	7	0	2.246689	-2.163697	-1.024010

31	16	0	2.542176	-2.942182	-3.593819
32	6	0	2.567639	-3.396581	-0.422580
33	6	0	3.104822	-5.821277	0.885843
34	6	0	1.865035	-3.747962	0.730755
35	6	0	3.595218	-4.228316	-0.882106
36	6	0	3.848360	-5.422427	-0.222266
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38	1	0	1.115227	-3.070825	1.128457
39	1	0	4.200086	-3.939858	-1.729154
40	1	0	3.297548	-6.771294	1.370958
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43	1	0	2.183623	-0.685628	-4.602820
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46	1	0	-1.450609	0.882532	-1.466690
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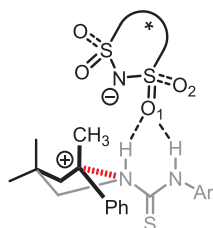
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95	6	0	3.348819	3.305713	0.799095
96	6	0	4.409150	1.378402	1.745251
97	6	0	5.442756	1.482989	0.824982
98	6	0	4.368119	3.412240	-0.132225
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114	6	0	7.876506	0.904109	0.889945
115	6	0	8.897124	-0.021399	0.861344
116	6	0	7.318519	-1.780635	0.689409
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139	6	0	3.862964	4.399935	-4.943308
140	6	0	10.307241	0.445280	0.897311
141	6	0	6.992083	-3.225356	0.550774
142	6	0	-4.505760	-4.645091	-3.460281
143	6	0	-9.209872	-4.433938	-1.924902
144	6	0	-10.342332	3.680912	0.892565

145	6	0	-5.919393	5.784310	0.186200
146	6	0	1.283253	-5.322127	2.547930
147	6	0	5.003445	-6.281816	-0.657258
148	9	0	4.531686	8.872648	-2.756810
149	9	0	4.864263	8.449638	-0.634063
150	9	0	2.850829	8.573326	-1.421828
151	9	0	5.045529	4.079028	-5.509650
152	9	0	3.116825	3.259536	-4.971593
153	9	0	3.269694	5.301947	-5.748561
154	9	0	10.431199	1.581662	1.617481
155	9	0	10.791539	0.719885	-0.335714
156	9	0	11.122115	-0.487696	1.433369
157	9	0	8.068913	-3.999522	0.778856
158	9	0	6.013844	-3.584815	1.408575
159	9	0	6.535630	-3.515568	-0.692253
160	9	0	6.084557	-6.076792	0.114129
161	9	0	5.362070	-6.050987	-1.921963
162	9	0	4.701094	-7.586466	-0.548715
163	9	0	1.499236	-6.575544	2.963332
164	9	0	1.502791	-4.508077	3.586469
165	9	0	-0.032532	-5.222259	2.253041
166	9	0	-3.767220	-5.560960	-2.792924
167	9	0	-3.633038	-3.709903	-3.904470
168	9	0	-5.029321	-5.272534	-4.533396
169	9	0	-9.273829	-5.719986	-2.329646
170	9	0	-9.945461	-3.714094	-2.802327
171	9	0	-9.830552	-4.332072	-0.729947
172	9	0	-10.848111	2.760998	1.743693
173	9	0	-10.720763	4.904880	1.317544
174	9	0	-10.957383	3.477220	-0.295878
175	9	0	-6.640318	6.764781	-0.403271
176	9	0	-5.439183	6.300718	1.339652
177	9	0	-4.851563	5.519821	-0.594142

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## TS2\_S



ONIOM(M06-2X/6-31G(d): HF/3-21G\*) (Hartree)

Imaginary Freq	1: -115.54 cm-1
Electronic Energy (EE)	-7910.280059
Zero-point Energy Correction	1.297659
Thermal Correction to Energy	1.400411
Thermal Correction to Enthalpy	1.401355
Thermal Correction to Free Energy	1.137054
EE + Zero-point Energy	-7908.982400
EE + Thermal Energy Correction	-7908.879649
EE + Thermal Enthalpy Correction	-7908.878705
EE + Thermal Free Energy Correction	-7909.143005
E <sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene))	-7962.487853

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.296116	0.165404	-3.179257
2	6	0	0.385691	0.402769	-4.486950
3	6	0	1.608770	1.363387	-4.470141
4	7	0	1.974385	-0.002932	-2.364366
5	1	0	-0.373427	0.839828	-5.154447
6	1	0	0.722758	-0.535670	-4.925780
7	6	0	2.329989	1.202479	-5.816273
8	1	0	1.677475	1.500391	-6.644179
9	1	0	2.645597	0.165613	-5.968509
10	1	0	3.212950	1.846822	-5.839390
11	6	0	1.259170	2.852505	-4.308541
12	1	0	0.495704	3.163245	-5.030347
13	1	0	2.159353	3.441038	-4.502476

14	1	0	0.913014	3.107149	-3.304816
15	6	0	-0.718803	-1.141652	-2.750259
16	6	0	-1.479033	-3.673204	-1.811808
17	6	0	-1.260353	-1.298977	-1.448376
18	6	0	-0.559533	-2.292152	-3.559527
19	6	0	-0.958440	-3.537036	-3.097777
20	6	0	-1.611807	-2.553237	-0.983325
21	1	0	-1.362104	-0.440407	-0.790331
22	1	0	-0.159099	-2.211610	-4.561682
23	1	0	-0.860283	-4.404854	-3.739920
24	1	0	-1.965607	-2.667416	0.036555
25	1	0	-1.806369	-4.643934	-1.456278
26	6	0	-0.837720	1.368161	-2.473881
27	1	0	-0.977814	2.183537	-3.183465
28	1	0	-1.783334	1.164689	-1.970495
29	6	0	2.288377	-1.355651	-2.399783
30	7	0	2.044236	-1.959896	-1.213891
31	16	0	2.803353	-2.101225	-3.803597
32	6	0	2.009193	-3.331150	-0.880243
33	6	0	1.798619	-6.002877	-0.063463
34	6	0	1.131580	-3.687235	0.145914
35	6	0	2.834103	-4.300628	-1.456303
36	6	0	2.723075	-5.617914	-1.029359
37	6	0	1.014304	-5.019400	0.522725
38	1	0	0.542293	-2.921996	0.646529
39	1	0	3.574253	-4.031251	-2.194569
40	1	0	1.709505	-7.040360	0.237613
41	6	0	2.567197	0.928715	-3.319360
42	1	0	2.866304	1.811655	-2.748454
43	1	0	3.460851	0.463877	-3.741538
44	1	0	1.841487	0.370617	-1.418149
45	1	0	1.776477	-1.338191	-0.445377
46	1	0	-0.163859	1.710752	-1.677670
47	6	0	-2.137884	0.828571	2.673996
48	6	0	-1.200684	0.681966	3.665364
49	6	0	-3.539549	-0.800902	3.739614
50	6	0	-1.378087	-0.283125	4.701958
51	6	0	-3.345100	0.065021	2.707254



52	6	0	-2.573574	-1.017074	4.740366
53	6	0	-0.387732	-0.550972	5.686507
54	1	0	-3.704633	-2.514969	5.788511
55	1	0	-4.459886	-1.347288	3.792675
56	6	0	-0.605593	-1.478209	6.651303
57	1	0	0.536897	-0.018393	5.654963
58	1	0	0.151798	-1.677331	7.382299
59	6	0	-1.823438	-2.194684	6.701595
60	1	0	-1.977201	-2.920383	7.474614
61	6	0	-2.781174	-1.970573	5.770332
62	6	0	2.319851	2.002717	2.935829
63	1	0	3.410557	3.375888	4.092173
64	6	0	2.476951	2.862375	3.975324
65	6	0	0.074107	1.491536	3.716000
66	6	0	1.441906	3.121402	4.899757
67	6	0	1.083595	1.305006	2.811168
68	6	0	0.219668	2.441833	4.768627
69	6	0	1.607312	4.065575	5.944733
70	1	0	-1.772565	2.255336	5.562227
71	6	0	0.589242	4.329536	6.799738
72	1	0	2.548136	4.571270	6.037569
73	1	0	0.713864	5.048152	7.584638
74	6	0	-0.650566	3.666217	6.654780
75	1	0	-1.452712	3.898229	7.325759
76	6	0	-0.831913	2.749152	5.672441
77	16	0	0.674775	0.332113	1.365107
78	8	0	0.240482	-1.005849	1.776348
79	8	0	1.850554	0.296934	0.465986
80	16	0	-1.676714	1.924817	1.304120
81	8	0	-1.300130	3.198490	1.898366
82	8	0	-2.723511	1.960333	0.283015
83	7	0	-0.434027	1.165531	0.585579
84	6	0	-4.445870	0.147207	1.697157
85	6	0	-6.547503	0.228804	-0.119373
86	6	0	-4.881992	-1.011553	1.074634
87	6	0	-5.087790	1.340686	1.421522
88	6	0	-6.107802	1.396966	0.489470
89	6	0	-5.945255	-0.982611	0.180291

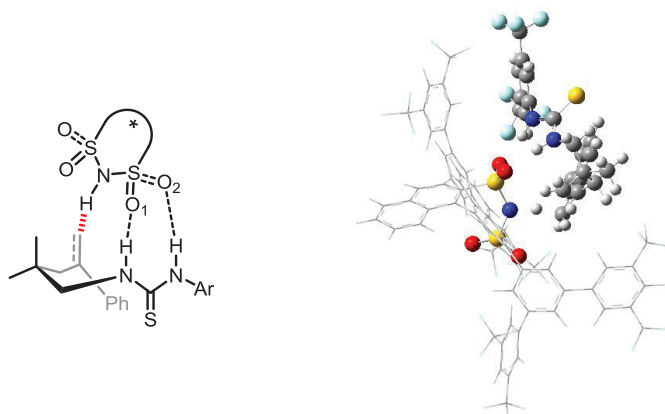
90	1	0	-4.397171	-1.942585	1.291063
91	1	0	-4.769115	2.239294	1.904220
92	1	0	-7.320486	0.271910	-0.861398
93	6	0	3.481117	1.854236	1.998228
94	6	0	5.669046	1.669290	0.308642
95	6	0	3.747906	2.865013	1.092566
96	6	0	4.316192	0.750457	2.061544
97	6	0	5.408193	0.651356	1.215476
98	6	0	4.839436	2.777391	0.237135
99	1	0	3.079830	3.700757	1.027719
100	1	0	4.111021	-0.033880	2.762384
101	1	0	6.517012	1.590720	-0.342952
102	6	0	5.077095	3.835015	-0.783240
103	6	0	5.469351	5.800978	-2.727200
104	6	0	5.058417	5.181262	-0.445314
105	6	0	5.302711	3.490566	-2.107667
106	6	0	5.485001	4.466884	-3.065090
107	6	0	5.253121	6.146131	-1.411159
108	1	0	4.917130	5.475950	0.573993
109	1	0	5.333426	2.459460	-2.393407
110	1	0	5.640262	6.553784	-3.467116
111	6	0	6.291783	-0.550101	1.240692
112	6	0	7.951184	-2.786657	1.223042
113	6	0	5.793359	-1.794316	0.890097
114	6	0	7.630820	-0.437836	1.585133
115	6	0	8.446177	-1.548841	1.573125
116	6	0	6.624169	-2.898441	0.879377
117	1	0	4.762014	-1.895936	0.615909
118	1	0	8.029603	0.511318	1.879293
119	1	0	8.583793	-3.647882	1.229976
120	6	0	-6.629054	2.724967	0.071144
121	6	0	-7.520000	5.222864	-0.767870
122	6	0	-7.982830	3.009416	0.030003
123	6	0	-5.719152	3.703677	-0.313826
124	6	0	-6.171653	4.940380	-0.721330
125	6	0	-8.417802	4.249631	-0.393173
126	1	0	-8.697521	2.276732	0.344907
127	1	0	-4.669936	3.476892	-0.302108

128	1	0	-7.862827	6.182680	-1.090739
129	6	0	-6.407175	-2.243456	-0.463956
130	6	0	-7.263727	-4.639204	-1.609770
131	6	0	-5.516141	-3.065483	-1.135200
132	6	0	-7.737868	-2.632270	-0.384622
133	6	0	-8.155306	-3.812349	-0.960711
134	6	0	-5.945677	-4.255003	-1.690847
135	1	0	-4.490840	-2.770742	-1.233227
136	1	0	-8.440381	-2.025524	0.148689
137	1	0	-7.589368	-5.563212	-2.036337
138	6	0	5.190774	7.582725	-1.032531
139	6	0	5.642581	4.062910	-4.484772
140	6	0	9.887947	-1.400351	1.899437
141	6	0	6.070343	-4.212647	0.454357
142	6	0	-4.953103	-5.128217	-2.367512
143	6	0	-9.593432	-4.182739	-0.909901
144	6	0	-9.875190	4.517684	-0.488251
145	6	0	-5.190941	5.996419	-1.083495
146	6	0	-0.019422	-5.358198	1.557727
147	6	0	3.703659	-6.636328	-1.541205
148	9	0	5.946505	8.339570	-1.855826
149	9	0	5.616448	7.775089	0.234245
150	9	0	3.933940	8.076621	-1.096283
151	9	0	6.312184	2.897587	-4.599750
152	9	0	4.440487	3.862102	-5.095575
153	9	0	6.283035	5.002036	-5.206248
154	9	0	10.089160	-0.414309	2.800854
155	9	0	10.632305	-1.083651	0.814871
156	9	0	10.401073	-2.544032	2.399604
157	9	0	6.936573	-5.214896	0.693080
158	9	0	4.913879	-4.485445	1.093792
159	9	0	5.786765	-4.224550	-0.870968
160	9	0	4.753689	-6.754759	-0.711913
161	9	0	4.180823	-6.317487	-2.747821
162	9	0	3.143245	-7.852824	-1.630178
163	9	0	-0.032320	-6.662314	1.853967
164	9	0	0.170619	-4.679635	2.692793
165	9	0	-1.256710	-5.036379	1.120664

166	9	0	-4.046596	-5.641885	-1.496075
167	9	0	-4.228583	-4.448182	-3.284407
168	9	0	-5.540130	-6.169182	-2.989651
169	9	0	-9.758615	-5.521792	-0.957356
170	9	0	-10.294855	-3.669551	-1.946435
171	9	0	-10.180836	-3.721606	0.215443
172	9	0	-10.562451	3.847400	0.462868
173	9	0	-10.146597	5.833751	-0.358862
174	9	0	-10.396015	4.131304	-1.676449
175	9	0	-5.735348	6.900797	-1.928691
176	9	0	-4.753201	6.689646	-0.008569
177	9	0	-4.097201	5.473551	-1.676802

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**TS<sub>R</sub> at ONIOM(M06-2X/6-31G(d): HF/3-21G\*)**



ONIOM(M06-2X/6-31G(d):HF/3-21G\*) (Hartree)

Imaginary Freq = 1: -1279.94 cm<sup>-1</sup>

Electronic Energy (EE)	-7910.275841
Zero-point Energy Correction	1.293665
Thermal Correction to Energy	1.39588
Thermal Correction to Enthalpy	1.396824
Thermal Correction to Free Energy	1.14004
EE + Zero-point Energy	-7908.982176
EE + Thermal Energy Correction	-7908.879961
EE + Thermal Enthalpy Correction	-7908.879017
EE + Thermal Free Energy Correction	-7909.135801
E <sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD (toluene))	-7962.482689

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.844257	-2.736269	-1.721265
2	6	0	-1.359956	-4.150406	-1.550671
3	6	0	-1.403924	-4.855189	-0.158698
4	7	0	0.993941	-4.318168	0.333891
5	1	0	-2.395833	-4.109590	-1.909497
6	1	0	-0.839328	-4.822517	-2.235180
7	6	0	-2.265235	-6.114973	-0.340322
8	1	0	-3.289940	-5.873508	-0.629410
9	1	0	-1.846319	-6.767344	-1.114379
10	1	0	-2.301221	-6.686145	0.593481
11	6	0	-2.029786	-3.982853	0.930832
12	1	0	-3.050719	-3.696569	0.655166
13	1	0	-2.087494	-4.547824	1.867458
14	1	0	-1.461108	-3.072804	1.135603
15	6	0	0.362055	-2.486847	-2.514780
16	6	0	2.693498	-2.012878	-4.014222
17	6	0	0.670515	-1.192033	-2.985601
18	6	0	1.247779	-3.532950	-2.833468
19	6	0	2.407874	-3.297149	-3.558588
20	6	0	1.816238	-0.964507	-3.730757
21	1	0	0.003801	-0.359558	-2.809460
22	1	0	1.069081	-4.541240	-2.481478
23	1	0	3.091982	-4.119376	-3.742449
24	1	0	2.033874	0.036826	-4.085816
25	1	0	3.603945	-1.820746	-4.575067
26	6	0	-1.599149	-1.676102	-1.206726
27	1	0	-2.575943	-1.915160	-0.794604
28	1	0	-1.579971	-0.716079	-1.722259
29	6	0	2.275540	-4.478517	-0.096894
30	7	0	3.024049	-3.351700	0.099473
31	16	0	2.854635	-5.870559	-0.837649
32	6	0	4.236747	-3.028645	-0.542841
33	6	0	6.576679	-2.257184	-1.874439

34	6	0	4.289880	-1.805982	-1.220635
35	6	0	5.357933	-3.854533	-0.512649
36	6	0	6.516482	-3.455136	-1.176122
37	6	0	5.448474	-1.439279	-1.891088
38	1	0	3.410831	-1.168593	-1.245075
39	1	0	5.322356	-4.800044	0.014380
40	1	0	7.486299	-1.965420	-2.385836
41	6	0	-0.012482	-5.357097	0.264477
42	1	0	-0.106008	-5.862173	1.236683
43	1	0	0.345576	-6.100933	-0.450827
44	1	0	0.814264	-3.564864	0.999434
45	1	0	2.518216	-2.523394	0.413465
46	1	0	-0.949164	-1.123469	-0.185726
47	6	0	-2.295623	-0.175020	2.805407
48	6	0	-1.416874	-0.464434	3.822126
49	6	0	-3.944278	-1.612662	3.791942
50	6	0	-1.735484	-1.467337	4.781791
51	6	0	-3.614504	-0.727627	2.812140
52	6	0	-3.015177	-2.044595	4.760662
53	6	0	-0.792941	-1.925397	5.740682
54	1	0	-4.332933	-3.461354	5.699470
55	1	0	-4.943398	-1.998469	3.830626
56	6	0	-1.130937	-2.891194	6.630060
57	1	0	0.191533	-1.509705	5.740388
58	1	0	-0.410379	-3.241364	7.341003
59	6	0	-2.430450	-3.450147	6.625707
60	1	0	-2.680161	-4.206553	7.342231
61	6	0	-3.347926	-3.038428	5.717046
62	6	0	1.949251	1.262097	3.211198
63	1	0	2.856920	2.713016	4.425641
64	6	0	2.051571	2.011736	4.335518
65	6	0	-0.151111	0.333896	4.012488
66	6	0	1.061792	1.981811	5.342400
67	6	0	0.868104	0.341962	3.104035
68	6	0	-0.090040	1.195444	5.153442
69	6	0	1.163655	2.811444	6.487370
70	1	0	-2.039786	0.743669	5.954290
71	6	0	0.151380	2.872508	7.386373

72	1	0	2.051454	3.397650	6.618600
73	1	0	0.229668	3.503280	8.248784
74	6	0	-1.024294	2.117022	7.180477
75	1	0	-1.828538	2.192520	7.884004
76	6	0	-1.145043	1.306055	6.100139
77	16	0	0.737944	-0.794511	1.728373
78	8	0	0.366992	-2.104150	2.277618
79	8	0	1.949870	-0.804389	0.911703
80	16	0	-1.671849	0.766140	1.384986
81	8	0	-1.145457	2.045543	1.834267
82	8	0	-2.639552	0.771918	0.300216
83	7	0	-0.463589	-0.200249	0.784834
84	6	0	-4.677806	-0.347157	1.833556
85	6	0	-6.607509	0.321611	-0.048794
86	6	0	-5.333544	-1.334571	1.121541
87	6	0	-5.019156	0.978423	1.621495
88	6	0	-5.954617	1.319317	0.663084
89	6	0	-6.303762	-1.009661	0.181331
90	1	0	-5.075750	-2.361864	1.283215
91	1	0	-4.520948	1.754073	2.166862
92	1	0	-7.299825	0.588418	-0.822727
93	6	0	2.812929	1.603101	2.042601
94	6	0	4.232833	2.335279	-0.216668
95	6	0	2.170854	2.223037	0.985803
96	6	0	4.177291	1.403840	2.001155
97	6	0	4.895270	1.788530	0.874158
98	6	0	2.861341	2.560617	-0.164191
99	1	0	1.122448	2.425852	1.071645
100	1	0	4.684156	0.951708	2.830530
101	1	0	4.782817	2.531903	-1.114057
102	6	0	-6.917632	-2.091967	-0.632912
103	6	0	-7.996003	-4.145368	-2.184834
104	6	0	-6.096771	-3.014920	-1.261729
105	6	0	-8.289556	-2.206739	-0.797172
106	6	0	-8.814530	-3.226020	-1.565843
107	6	0	-6.633950	-4.032727	-2.018404
108	1	0	-5.034667	-2.916028	-1.183208
109	1	0	-8.947613	-1.504283	-0.328264

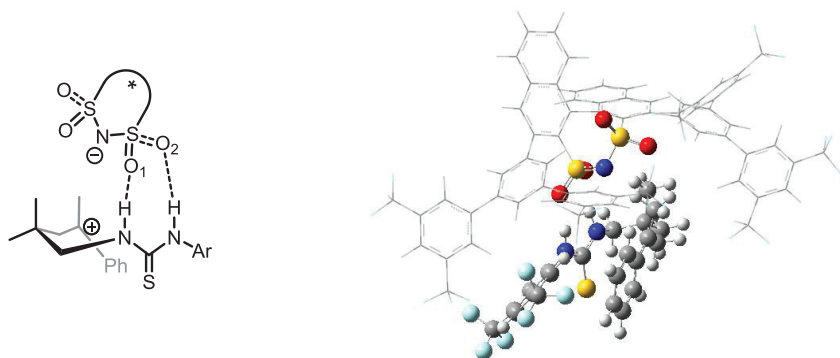
110	1	0	-8.410224	-4.918063	-2.797229
111	6	0	-6.144789	2.751757	0.302502
112	6	0	-6.438159	5.413013	-0.437809
113	6	0	-7.373775	3.378034	0.416165
114	6	0	-5.059329	3.464933	-0.190318
115	6	0	-5.217142	4.786625	-0.557656
116	6	0	-7.510298	4.701794	0.052983
117	1	0	-8.221095	2.838239	0.786895
118	1	0	-4.110951	2.975927	-0.301066
119	1	0	-6.552655	6.435448	-0.726298
120	6	0	2.089747	3.121717	-1.305789
121	6	0	0.520845	4.219438	-3.341099
122	6	0	0.810808	2.642258	-1.559642
123	6	0	2.578029	4.147134	-2.102353
124	6	0	1.802024	4.673886	-3.114218
125	6	0	0.035687	3.199683	-2.554065
126	1	0	0.423840	1.831366	-0.976687
127	1	0	3.548752	4.558498	-1.917598
128	1	0	-0.086099	4.656474	-4.104582
129	6	0	6.373212	1.598427	0.833773
130	6	0	9.138005	1.264230	0.759121
131	6	0	7.211398	2.685224	0.644185
132	6	0	6.935801	0.340390	0.987879
133	6	0	8.304979	0.182222	0.938899
134	6	0	8.580347	2.514205	0.617611
135	1	0	6.797177	3.664663	0.520777
136	1	0	6.306916	-0.514157	1.128080
137	1	0	10.199236	1.134774	0.727981
138	6	0	-5.719782	-5.034418	-2.619977
139	6	0	-10.290099	-3.364747	-1.683203
140	6	0	-4.061367	5.534392	-1.122581
141	6	0	-8.816729	5.380677	0.245136
142	6	0	-1.355586	2.710419	-2.729546
143	6	0	2.366867	5.733187	-3.989929
144	6	0	9.462116	3.702568	0.489324
145	6	0	8.888317	-1.182572	1.006126
146	6	0	5.489858	-0.177313	-2.703929
147	6	0	7.724108	-4.349698	-1.090821



148	9	0	8.675227	-3.995412	-1.962064
149	9	0	7.393097	-5.624027	-1.348185
150	9	0	8.264806	-4.333595	0.134889
151	9	0	5.842072	-0.430730	-3.975517
152	9	0	4.303589	0.451049	-2.739421
153	9	0	6.379631	0.710375	-2.228927
154	9	0	10.117418	-1.168055	1.563468
155	9	0	9.035725	-1.730825	-0.226298
156	9	0	8.105434	-2.018215	1.716342
157	9	0	8.884561	4.665283	-0.261749
158	9	0	9.742156	4.269050	1.686157
159	9	0	10.647002	3.379790	-0.070888
160	9	0	3.026636	5.222183	-5.055271
161	9	0	3.254787	6.502260	-3.323546
162	9	0	1.398504	6.532840	-4.484644
163	9	0	-3.765716	5.156647	-2.387016
164	9	0	-2.941007	5.350341	-0.395839
165	9	0	-4.314381	6.862548	-1.162676
166	9	0	-8.977755	6.399398	-0.625831
167	9	0	-9.846108	4.519316	0.085219
168	9	0	-8.950897	5.907544	1.484395
169	9	0	-10.901661	-2.162246	-1.631746
170	9	0	-10.635671	-3.970543	-2.838465
171	9	0	-10.816997	-4.107024	-0.682823
172	9	0	-6.256833	-5.628483	-3.701012
173	9	0	-4.539515	-4.472235	-2.982144
174	9	0	-5.400379	-6.023151	-1.745790
175	9	0	-2.155739	3.039223	-1.691860
176	9	0	-1.921430	3.186401	-3.852539
177	9	0	-1.395986	1.347297	-2.802445

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## Intermediate\_R



ONIOM(M062X/6-31G(d): HF/3-21G\*) (Hartree)

Imaginary Freq	0
Electronic Energy (EE)	-7910.289205
Zero-point Energy Correction	1.29876
Thermal Correction to Energy	1.401534
Thermal Correction to Enthalpy	1.402478
Thermal Correction to Free Energy	1.14361
EE + Zero-point Energy	-7908.990446
EE + Thermal Energy Correction	-7908.887672
EE + Thermal Enthalpy Correction	-7908.886727
EE + Thermal Free Energy Correction	-7909.145595

E<sub>sp</sub>(RM06-2X/6-311+G(d,p)/SMD(toluene)) -7962.494226

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.940624	-3.228826	-1.265865
2	6	0	-1.391870	-4.545455	-0.733867
3	6	0	-1.416465	-4.780745	0.813950
4	7	0	0.908590	-3.847590	0.851424
5	1	0	-2.426558	-4.657147	-1.090416
6	1	0	-0.811160	-5.367390	-1.154624
7	6	0	-2.109052	-6.133188	1.031734
8	1	0	-3.157802	-6.098340	0.726417
9	1	0	-1.613111	-6.927202	0.462807

10	1	0	-2.071439	-6.406830	2.091481
11	6	0	-2.156973	-3.715859	1.626423
12	1	0	-3.188046	-3.598740	1.278056
13	1	0	-2.202729	-4.042222	2.670580
14	1	0	-1.661011	-2.742318	1.622728
15	6	0	0.100830	-3.099743	-2.235426
16	6	0	2.094995	-2.778306	-4.183338
17	6	0	0.495256	-1.802198	-2.672006
18	6	0	0.739592	-4.224171	-2.818312
19	6	0	1.723472	-4.056014	-3.784557
20	6	0	1.471770	-1.649411	-3.629917
21	1	0	0.025715	-0.918957	-2.261787
22	1	0	0.451184	-5.228785	-2.545717
23	1	0	2.206603	-4.927940	-4.210453
24	1	0	1.768162	-0.655896	-3.947265
25	1	0	2.874999	-2.645751	-4.926752
26	6	0	-1.722231	-2.026479	-0.890032
27	1	0	-2.663947	-2.281145	-0.412496
28	1	0	-1.912404	-1.366982	-1.739272
29	6	0	2.047538	-4.084992	0.138243
30	7	0	2.800828	-2.970527	-0.041216
31	16	0	2.427014	-5.588312	-0.524108
32	6	0	3.994808	-2.867029	-0.789166
33	6	0	6.346641	-2.543290	-2.283322
34	6	0	4.119124	-1.804743	-1.681827
35	6	0	5.060632	-3.754292	-0.621392
36	6	0	6.223214	-3.576360	-1.360050
37	6	0	5.280650	-1.667779	-2.439857
38	1	0	3.300291	-1.100682	-1.792456
39	1	0	4.979155	-4.571416	0.085309
40	1	0	7.260100	-2.410989	-2.846918
41	6	0	0.033554	-4.907687	1.317722
42	1	0	0.021986	-4.895037	2.414724
43	1	0	0.460579	-5.856484	0.989642
44	1	0	0.864544	-2.987717	1.407774
45	1	0	2.462880	-2.090735	0.364890
46	1	0	-1.148121	-1.400562	-0.178709
47	6	0	-2.236063	0.409790	2.872777

48	6	0	-1.409721	0.264723	3.962182
49	6	0	-3.940954	-0.862186	3.992712
50	6	0	-1.778258	-0.586750	5.044173
51	6	0	-3.558081	-0.140560	2.902969
52	6	0	-3.063189	-1.148314	5.057120
53	6	0	-0.881210	-0.909313	6.097440
54	1	0	-4.442110	-2.393191	6.141269
55	1	0	-4.945218	-1.232532	4.045443
56	6	0	-1.269190	-1.728340	7.106081
57	1	0	0.109050	-0.508538	6.076136
58	1	0	-0.582289	-1.975912	7.889939
59	6	0	-2.577029	-2.266545	7.133108
60	1	0	-2.867745	-2.903587	7.944193
61	6	0	-3.450788	-1.984467	6.136209
62	6	0	2.012960	1.817658	3.248020
63	1	0	2.919790	3.406734	4.277453
64	6	0	2.103536	2.712233	4.260789
65	6	0	-0.132039	1.055680	4.091745
66	6	0	1.083809	2.840367	5.231030
67	6	0	0.917481	0.907688	3.233744
68	6	0	-0.082710	2.063589	5.108571
69	6	0	1.172083	3.815381	6.255856
70	1	0	-2.070663	1.775683	5.887257
71	6	0	0.133119	4.023407	7.101382
72	1	0	2.071580	4.392843	6.338425
73	1	0	0.201567	4.763929	7.872688
74	6	0	-1.056375	3.276757	6.955847
75	1	0	-1.881800	3.467714	7.611496
76	6	0	-1.164514	2.329281	5.990740
77	16	0	0.755604	-0.354333	1.970760
78	8	0	0.284963	-1.583223	2.638366
79	8	0	2.022416	-0.553658	1.255703
80	16	0	-1.543655	1.181516	1.371206
81	8	0	-1.085098	2.523369	1.719208
82	8	0	-2.493948	1.074336	0.269694
83	7	0	-0.312866	0.211926	0.920455
84	6	0	-4.586542	0.056399	1.834256
85	6	0	-6.499532	0.348620	-0.163557

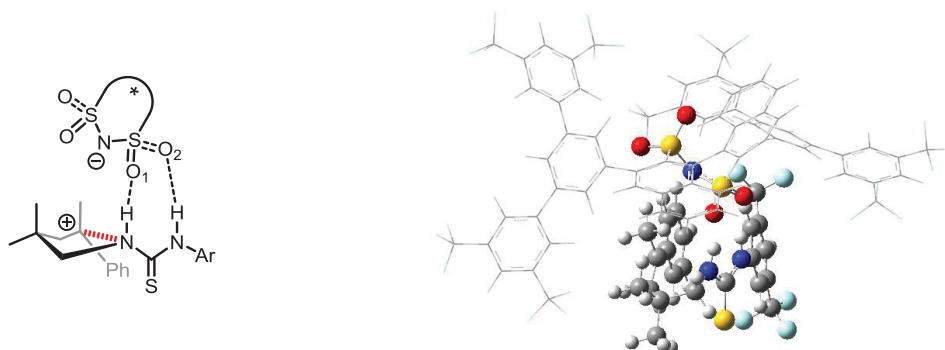
86	6	0	-5.243016	-1.047548	1.320496
87	6	0	-4.922998	1.314924	1.364109
88	6	0	-5.847290	1.464054	0.347119
89	6	0	-6.206673	-0.912321	0.328010
90	1	0	-4.998128	-2.022640	1.689516
91	1	0	-4.432846	2.181215	1.755836
92	1	0	-7.184660	0.462399	-0.980565
93	6	0	2.899929	1.979962	2.059127
94	6	0	4.353321	2.293626	-0.272862
95	6	0	2.281838	2.446950	0.912598
96	6	0	4.253857	1.721817	2.059066
97	6	0	4.995187	1.896957	0.893850
98	6	0	2.989485	2.564550	-0.269454
99	1	0	1.238381	2.687584	0.951253
100	1	0	4.726051	1.340696	2.942476
101	1	0	4.907000	2.363296	-1.186717
102	6	0	-6.836494	-2.134136	-0.239486
103	6	0	-7.954180	-4.469652	-1.291125
104	6	0	-6.038708	-3.206374	-0.604790
105	6	0	-8.208492	-2.245493	-0.417393
106	6	0	-8.751721	-3.402280	-0.937373
107	6	0	-6.594137	-4.360500	-1.112970
108	1	0	-4.976331	-3.125770	-0.515977
109	1	0	-8.851916	-1.433129	-0.148622
110	1	0	-8.382633	-5.354683	-1.711073
111	6	0	-6.032249	2.802135	-0.278114
112	6	0	-6.317976	5.279374	-1.502279
113	6	0	-7.277511	3.398925	-0.368660
114	6	0	-4.923969	3.452105	-0.809613
115	6	0	-5.079241	4.681316	-1.418593
116	6	0	-7.411184	4.632292	-0.972430
117	1	0	-8.140525	2.907679	0.032253
118	1	0	-3.960568	2.982349	-0.757805
119	1	0	-6.428346	6.231605	-1.974518
120	6	0	2.244911	2.930527	-1.505335
121	6	0	0.752002	3.621180	-3.761191
122	6	0	0.982720	2.385828	-1.718349
123	6	0	2.751725	3.814516	-2.445350

124	6	0	2.011919	4.141455	-3.563827
125	6	0	0.246502	2.742291	-2.828507
126	1	0	0.580755	1.683842	-1.014438
127	1	0	3.708403	4.270969	-2.295592
128	1	0	0.175503	3.899925	-4.617053
129	6	0	6.456494	1.605134	0.905849
130	6	0	9.190439	1.036239	0.956345
131	6	0	7.261689	2.073409	1.934041
132	6	0	7.044297	0.853520	-0.101568
133	6	0	8.393492	0.572198	-0.066538
134	6	0	8.610797	1.785066	1.955082
135	1	0	6.843558	2.687973	2.704248
136	1	0	6.454955	0.508432	-0.919860
137	1	0	10.241170	0.837945	0.963594
138	6	0	-5.696655	-5.497384	-1.431519
139	6	0	-10.228302	-3.521695	-1.065354
140	6	0	-3.903199	5.355873	-2.031722
141	6	0	-8.740965	5.291381	-0.999662
142	6	0	-1.130068	2.203970	-2.986722
143	6	0	2.598171	5.044303	-4.587466
144	6	0	9.439937	2.263715	3.090874
145	6	0	8.986484	-0.283856	-1.125576
146	6	0	5.350771	-0.559727	-3.449497
147	6	0	7.385481	-4.495084	-1.095189
148	9	0	8.318686	-4.407184	-2.049918
149	9	0	6.984748	-5.773362	-1.033068
150	9	0	7.971359	-4.214908	0.075416
151	9	0	4.234513	-0.516701	-4.206020
152	9	0	5.435717	0.652389	-2.857878
153	9	0	6.387554	-0.675619	-4.274773
154	9	0	10.291421	-0.006378	-1.319490
155	9	0	8.340411	-0.131798	-2.307613
156	9	0	8.910805	-1.605909	-0.835689
157	9	0	8.990696	3.446604	3.565289
158	9	0	9.427702	1.405662	4.137292
159	9	0	10.732047	2.414636	2.730786
160	9	0	3.316389	4.370843	-5.516928
161	9	0	3.440961	5.941118	-4.031336

162	9	0	1.641027	5.719618	-5.258342
163	9	0	-3.570269	4.832124	-3.233636
164	9	0	-2.808169	5.264362	-1.253052
165	9	0	-4.154285	6.669306	-2.242409
166	9	0	-8.853468	6.137010	-2.046202
167	9	0	-9.738550	4.382321	-1.081317
168	9	0	-8.983579	6.021587	0.113649
169	9	0	-10.798244	-2.323810	-1.314180
170	9	0	-10.566640	-4.370431	-2.058858
171	9	0	-10.804492	-3.995899	0.062545
172	9	0	-6.264280	-6.376728	-2.275294
173	9	0	-4.532979	-5.066649	-1.987815
174	9	0	-5.328683	-6.190075	-0.323062
175	9	0	-2.030773	2.811865	-2.186790
176	9	0	-1.571578	2.332947	-4.254612
177	9	0	-1.183182	0.879592	-2.673938

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TS2\_R



ONIOM(M062X/6-31G(d):HF/3-21G\*) (Hartree)

Imaginary Freq	1: -75.05 cm-1
Electronic Energy (EE)	-7910.290823
Zero-point Energy Correction	1.299252
Thermal Correction to Energy	1.401322
Thermal Correction to Enthalpy	1.402266
Thermal Correction to Free Energy	1.14381
EE + Zero-point Energy	-7908.991571
EE + Thermal Energy Correction	-7908.889501

EE + Thermal Enthalpy Correction	-7908.888557
EE + Thermal Free Energy Correction	-7909.147013
 E <sub>sp</sub> (RM06-2X/6-311+(d,p)/SMD(toluene))	 -7962.490841

Center Number	Atomic Number	Integrated Forces (Hartrees/Bohr)		
		X	Y	Z
1	6	0.000024115	-0.000015330	0.000002867
2	6	-0.000002499	-0.000002305	0.000006708
3	6	0.000000004	-0.000004299	-0.000007497
4	7	0.000035059	0.000015254	-0.000007554
5	1	0.000004064	0.000001377	-0.000001521
6	1	0.000001106	-0.000000488	-0.000001342
7	6	0.000013837	0.000003942	0.000002289
8	1	0.000000855	0.000003560	-0.000000979
9	1	-0.000002121	0.000003790	0.000001893
10	1	0.000000121	-0.000001995	0.000000004
11	6	-0.000003233	0.000003786	-0.000004948
12	1	0.000002668	-0.000002037	-0.000000024
13	1	0.000000573	-0.000000799	-0.000000268
14	1	-0.000004312	0.000004649	0.000003613
15	6	-0.000014778	-0.000005770	-0.000008154
16	6	-0.000009212	0.000002842	0.000008182
17	6	-0.000001082	-0.000001432	0.000003429
18	6	0.000000083	0.000019988	-0.000000753
19	6	-0.000006451	-0.000008778	-0.000007386
20	6	-0.000004249	0.000010555	0.000003223
21	1	-0.000004810	0.000011511	0.000007914
22	1	-0.000004601	0.000001919	0.000001026
23	1	-0.000007233	-0.000006521	0.000004520
24	1	-0.000000777	-0.000005485	0.000003595
25	1	0.000002550	-0.000004325	-0.000002543
26	6	0.000006893	-0.000009668	-0.000003847
27	1	-0.000000461	0.000013451	0.000010863
28	1	-0.000015566	0.000013807	-0.000004044
29	6	-0.000024241	-0.000000271	-0.000007493



30	7	-0.000010938	0.000002489	0.000008030
31	16	0.000021998	0.000017755	0.000016209
32	6	-0.000001991	-0.000012395	0.000006147
33	6	-0.000001509	-0.000000143	-0.000010030
34	6	-0.000010545	0.000014892	0.000008925
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37	6	-0.000012273	0.000022909	0.000015260
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39	1	-0.000005714	-0.000001135	-0.000009740
40	1	0.000006748	0.000004060	0.000001457
41	6	-0.000014333	-0.000012364	0.000010657
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46	1	-0.000006022	-0.000019727	-0.000016199
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48	6	0.000002918	0.000004110	-0.000003817
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51	6	-0.000000031	-0.000001110	0.000000907
52	6	-0.000000219	0.000000569	-0.000000265
53	6	0.000000648	0.000000602	-0.000000282
54	1	0.000000164	0.000000128	-0.000000295
55	1	-0.000000441	-0.000000433	0.000000456
56	6	0.000000112	0.000000083	-0.000000290
57	1	0.000000152	-0.000000075	-0.000000282
58	1	0.000000025	0.000000112	-0.000000186
59	6	-0.000000188	0.000000076	0.000000181
60	1	-0.000000009	0.000000029	-0.000000171
61	6	0.000000305	0.000000174	-0.000000161
62	6	-0.000007407	0.000004839	0.000007529
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64	6	-0.000002701	0.000003483	-0.000003101
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66	6	-0.000000328	0.000000202	-0.000000435
67	6	0.000003482	-0.000010461	0.000007984

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72	1	-0.000000165	0.000000402	-0.000000183
73	1	-0.000000164	0.000000228	-0.000000096
74	6	0.000000040	0.000000126	-0.000000193
75	1	-0.000000136	0.000000170	-0.000000059
76	6	0.000000097	0.000000088	0.000000123
77	16	0.000016308	-0.000004762	0.000002822
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79	8	-0.000021346	-0.000002252	0.000014402
80	16	0.000001540	0.000007292	-0.000006551
81	8	0.000000767	-0.000004383	0.000000708
82	8	0.000002384	0.000000299	-0.000002613
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84	6	-0.000001859	-0.000001782	0.000004434
85	6	-0.000002430	-0.000003533	-0.000000856
86	6	0.000001513	0.000002055	0.000001373
87	6	0.000000919	0.000000807	0.000000919
88	6	0.000000113	0.000003600	0.000000841
89	6	-0.000000027	-0.000000803	0.000000385
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92	1	0.000000462	0.000000502	0.000000187
93	6	0.000010554	-0.000003189	-0.000016955
94	6	-0.000005938	-0.000000421	0.000013508
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96	6	-0.000003220	-0.000000249	0.000003836
97	6	0.000010560	-0.000000758	-0.000006704
98	6	-0.000003325	-0.000003720	-0.000004280
99	1	0.000000033	0.000000274	-0.000000299
100	1	0.000001650	0.000000164	-0.000003248
101	1	-0.000001510	-0.000001653	-0.000004328
102	6	0.000000389	0.000000746	0.000001132
103	6	-0.000005460	-0.000000606	-0.000002433
104	6	-0.000005326	-0.000001264	-0.000003175
105	6	0.000012369	0.000000966	0.000006915

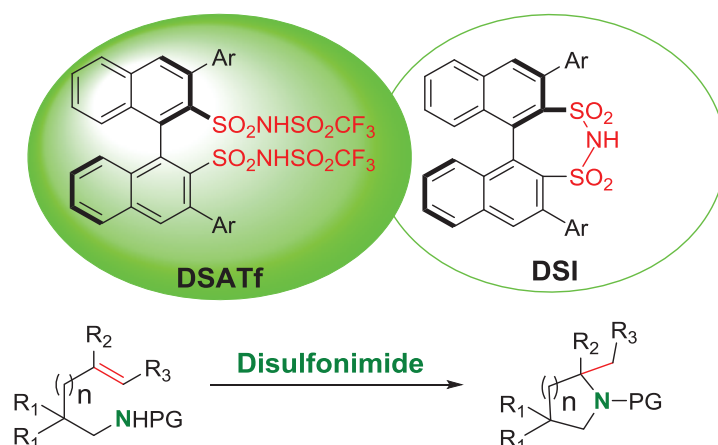
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108	1	-0.000000017	-0.000002999	-0.000002279
109	1	-0.000002121	-0.000000240	-0.000000981
110	1	-0.000000541	-0.000000320	-0.000000141
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112	6	0.000012409	0.000002019	-0.000003223
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115	6	0.000004191	-0.000000102	-0.000002021
116	6	-0.000015646	-0.000000499	0.000006408
117	1	-0.000004234	0.000000357	0.000001846
118	1	-0.000000041	-0.000000080	-0.000000157
119	1	-0.000004374	0.000000056	0.000000802
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121	6	-0.000003366	-0.000003221	0.000001778
122	6	0.000004465	0.000004215	-0.000004894
123	6	-0.000000270	0.000000357	0.000001231
124	6	-0.000005461	-0.000003466	0.000004944
125	6	0.000005664	0.000003903	-0.000003353
126	1	0.000002278	0.000001699	-0.000001641
127	1	-0.000000879	-0.000001777	0.000000832
128	1	-0.000000574	-0.000000868	0.000000904
129	6	-0.000011796	-0.000000005	0.000002539
130	6	-0.000005055	0.000000817	-0.000012674
131	6	0.000004307	-0.000001019	0.000011621
132	6	0.000004362	0.000003918	0.000009610
133	6	-0.000002488	-0.000001952	-0.000007767
134	6	0.000003193	0.000002740	0.000007737
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136	1	-0.000006316	0.000002213	-0.000007584
137	1	0.000000456	-0.000000883	0.000001496
138	6	-0.000001542	0.000001367	-0.000000389
139	6	0.000002143	0.000000315	0.000000441
140	6	-0.000001858	0.000000347	0.000000575
141	6	0.000006551	0.000001052	-0.000002777
142	6	-0.000004900	-0.000001340	-0.000000267
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147	6	0.000011652	-0.000029495	-0.000025301
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152	9	-0.000002381	-0.000009341	-0.000000409
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157	9	-0.000000768	-0.000003269	-0.000004111
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159	9	0.000000748	-0.000000369	0.000001666
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162	9	-0.000001764	-0.000000693	0.000000107
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167	9	0.000005896	-0.000000175	-0.000007492
168	9	-0.000005925	0.000000599	-0.000001092
169	9	-0.000002287	0.000000538	-0.000002874
170	9	-0.000001982	-0.000002328	0.000004117
171	9	0.000004429	0.000001160	0.000000467
172	9	0.000000560	-0.000000178	0.000000023
173	9	-0.000000782	-0.000000383	-0.000000914
174	9	-0.000000929	-0.000003422	0.000003347
175	9	0.000002345	-0.000000447	0.000002521
176	9	0.000002282	0.000003079	-0.000003220
177	9	-0.000002954	0.000002937	-0.000005580

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## **Chapter 4. Overall Conclusion**

In chapter 2, a new strong brønsted acid **DSATf** bearing *N*-trifluoromethylsulfonamide groups has been developed. The computational prediction of  $pK_a$  value of **DSATf** categorized it as a strong Brønsted acid. Hydrofunctionalization of alkenyl amines proceeded with high efficiency but no enantioselectivity was observed. In chapter 3, a cyclic **DSI** catalyzed asymmetric intramolecular hydroamination of alkenyl thioureas was examined. It was found that the hydroamination product was obtained in good chemical yield and moderate enantioselectivity under dilute conditions without stirring. The DFT calculation on the hydroamination supported a mechanism, in which thiourea moiety of alkenyl amines acted as a hydrogen bond donor.



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## List of Publications

(1) Bis(trifluoromethanesulfonimide) (BSI): Acidity and application to hydrofunctionalization as a Brønsted acid catalyst

Ryukichi Takagi, Yuichiro Sakai, Duyen Thi Duong

*Tetrahedron*. **2021**,132037.

(2) Disulfonimide catalyzed asymmetric intramolecular hydroamination of alkenyl thioureas: Concentration effect in the hydroamination

Ryukichi Takagi, Duyen Thi Duong, Toshiya Ichiki

*Tetrahedron*. **2021**,132332.