Low-temperature selective deposition of silicon on silicon nitride by time-modulated disilane flow and formation of silicon narrow wires

Shin Yokoyama,^{a)} Kenji Ohba, Kensaku Kawamura, Toshiro Kidera, and Anri Nakajima *Research Center for Nanodevices and Systems, Hiroshima University, 1-4-2 Kagamiyama, Higashi-Hiroshima, Hiroshima 739-8527, Japan*

(Received 15 December 2000; accepted for publication 31 May 2001)

The low-temperature $(410^{\circ}C)$ selective deposition of Si on silicon nitride has been achieved by means of the time-modulated flow of disilane while a very small amount of Si is deposited on $SiO₂$. Very narrow (21 nm width and 28 nm thick) Si wires have been fabricated using the selective deposition. The resistivity of the Si wires fabricated by the selective deposition is much smaller $(\sim 1/5)$ than that fabricated by the conventional reactive ion etching followed by annealing. This technique will be applicable to the formation of a polycrystalline silicon gate with small resistivity for the high-performance ultrasmall metal–oxide–semiconductor transistors and quantum effect devices. © 2001 American Institute of Physics. [DOI: 10.1063/1.1387260]

The selective deposition technique has a big potential to fabricate structures with very narrow width which is not limited by the lithography resolution. We have been studying a selective atomic-layer deposition of silicon nitride on silicon and no deposition on SiO_2 .^{1,2} When the selective deposition of Si on silicon nitride and no deposition on $SiO₂$ is achieved, very narrow Si wires separated by the silicon nitride films can be integrated combined with this technique as shown in Fig. 1. In this case, the Si and silicon nitride films are alternately deposited on the sidewall of the $SiO_2/silicon$ nitride/ $SiO₂$ stacked structures.

Many studies concerning the selective deposition (including epitaxial growth) of Si have been reported. $3-7$ In most cases, the selective deposition (or epitaxial growth) is carried out on the Si substrate partially covered with patterned SiO_2 or silicon nitride (Si_3N_4) film.^{4–6} In a special case, the selective deposition (nucleation) on $Si₃N₄$ was achieved by Yonehara *et al.*³ with a selectivity of $10^2 - 10^3$ compared to a $SiO₂$ substrate at a relatively high growth temperature of 950 °C using SiH_2Cl_2 . Yasuda *et al.*⁷ also reported that a high density of Si nuclei is formed on $Si₃N₄$ substrate in the ultrahigh vacuum chemical vapor deposition (UHV-CVD) using $Si₂H₆$ at 580 °C.

In this letter, we report a method for the selective deposition of Si on $Si₃N₄$ at a lower temperature (410 °C) by the time-modulated flow of disilane $(Si₂H₆)$. Very narrow (21) nm width) Si wires have been fabricated using the selective deposition and the electrical properties have been measured. The resistivity of the Si wires fabricated by the selective deposition is much smaller than that fabricated by the conventional reactive ion etching (RIE) followed by annealing.

The $Si₂H₆$ gas (He balance, 5%) is periodically admitted into the quartz chamber with an adequate interval time as shown in Fig. 2. The substrate was heated by a halogen lamp synchronized with the gas supply. The details of the experimental apparatus are in a previous paper.⁸ There exists an incubation time before starting the Si film growth on $SiO₂$ in the UHV-CVD using $Si₂H₆$.⁵ The incubation-time difference depending on the substrate material $(Si \text{ or } SiO₂)$ was utilized to the selective epitaxial growth.⁵ However, in the continuous gas flow, the film thickness of the selective deposition is limited by the incubation time. The intermittent gas flow is expected to relieve the thickness limitation of the selective deposition, since the substrate surface condition may approach to the initial condition by the desorption of the adsorbed species during the vacuum evacuation. The chemistry of this selective deposition is completely different from that in the flow-modulated plasma enhanced selective chemical vapor deposition (CVD) using $SiH₄$ (Ref. 6) where hydrogen plasma etching during the interval between $SiH₄$ flow is essential.

The temperature dependence of the deposition rate of Si on thermal SiO₂ (grown at 1000 °C in O₂+H₂) and Si₃N₄ (deposited by a low-pressure CVD using $SiH₂Cl₂$ and NH₃ at 750 °C) is shown in Fig. 3. Both substrate surfaces were treated in the diluted HF followed by deionized water rinse for 1 min and spin dried. Therefore, the $SiO₂$ surface is O–H terminated, 9 while the bonded hydrogen does not exist on the $Si₃N₄$ surface¹ which was verified by the Fourier transform infrared attenuated total reflection spectroscopy. It is recognized that the Si film is selectively deposited on the $Si₃N₄$ surface with a small amount of Si deposition on $SiO₂$ in the two growth conditions. One is the high-temperature $($ >580 °C) and low-pressure (10⁻⁵ Torr) condition and the other is the low-temperature $(<500 °C$ and high-pressure $(10^{-3}$ Torr) condition. The mechanism for the selective growth in the high-temperature and low-pressure region must be the reaction, $Si+SiO_2 \rightarrow 2SiO$ (with high vapor pressure), 10 resulting in a small amount of deposition on

FIG. 1. Integration of Si nanowires by using the selective deposition of Si and silicon nitride is shown.

Downloaded 17 Jun 2007 to 133.41.149.126. Redistribution subject to AIP license or copyright, see http://apl.aip.org/apl/copyright.jsp

a)Electronic mail: yokoyama@sxsys.hiroshima-u.ac.jp

FIG. 2. Pressure versus time for the time-modulated flow method is shown. The substrate temperature was controlled synchronized with the gas supply. The lamp was off during the vacuum evacuation.

 $SiO₂$. The activation energy of 2.1 eV observed in Fig. 3 fits well to that of thermal dissociation of $Si₂H₆$.¹¹ On the other hand, at the low-temperature and high-pressure condition the growth rate on $Si₃N₄$ is remarkably enhanced and the activation energy of 0.9 eV is obtained. The enhancement of the growth rate may be partly due to the short incubation time on $Si₃N₄$ (Ref. 12) compared to that on $SiO₂$ and partly due to a kind of catalytic surface reaction on the $Si₃N₄$ surface. However, the detailed mechanism and the physical meaning of the activation energy of 0.9 eV remain unclear at present. It should be noticed that for the $Si_{1-x}Ge_x$ CVD using SiH_4 and GeH₄, the shorter incubation time on $Si₃N₄$ compared with that on $SiO₂$ is reported.¹²

The Si selective deposition was carried out on the $SiO_2/Si_3N_4/SiO_2$ sidewall formed by the RIE using CF_4+H_2 gas mixture. Figure 4 shows the cross sectional scanning electron microscope (SEM) images for different growth conditions. The Si wires of \sim 30 nm width are formed on the sidewall of the $Si₃N₄$ at the high-temperature, low-pressure condition [Fig. $4(a)$] and at the low-temperature, highpressure condition [Fig. $4(c)$]. On the other hand, at the hightemperature, medium-pressure condition $(600 °C)$ and 10^{-4} Torr) [Fig. 4(b)], the Si film is conformably deposited on the whole surface.

The electrical properties of the Si wires were evaluated. Because of their smaller surface roughness than that grown at the high-temperature condition, the Si wires were fabricated under the low-temperature $(410 \degree C)$ and high-pressure $(10^{-3}$ Torr) condition. After the selective Si deposition, phosphorous doping using POCl₃ at 850 °C was carried out (see Table I). Then, a thin Si film grown on $SiO₂$ (a small amount of deposition occurs on $SiO₂$ even for the selective deposition) was removed by the chemical dry etching using remote microwave plasma of CF_4+O_2 . The passivation layer

FIG. 3. Temperature dependence of the Si deposition rate on $SiO₂$ and on $Si₃N₄$ is shown. The deposition cycle is 20, and the deposited film thickness was measured by the ellipsometry.

formation (nondoped and phosphorous doped silicate glass, both 300 nm thick) by the atmospheric CVD at 400° C, the contact hole formation by the wet etching, the Al sputter deposition and patterning by the wet etching were carried out. Finally, the sample was annealed at 450 °C for 30 min in a $H_2 + N_2$ mixture to reduce the contact resistance. The size of the fabricated Si wires was 21 nm width \times 28 nm thick from the SEM observation.

Figure $5(a)$ shows the current–voltage $(I-V)$ curves for the fabricated Si wires at room temperature $(23 \degree C)$, which indicates the linear $I - V$ relationship. The length dependence of the resistance of the Si wires is plotted in Fig. $5(b)$, from which the resistivity of the Si wires is calculated. The result is summarized in Table I with the data for the Si wires fabricated by the RIE followed by annealing at 850 °C for 30 min. For the selective deposition sample, the resistivity is about 1/5 times that of the RIE sample, even though the phosphorous is more heavily doped. The large resistivity for the RIE sample is reported to be due to the plasma-induced damage at the etched sidewall of the Si wire during the RIE.¹³ Actually, the resistivity increases by about one order of magnitude after the RIE as listed in Table I. Namely, the diffused phosphorous atoms are trapped and deactivated in the damaged region.¹³ This effect becomes remarkable as the wire width is narrower. On the other hand, the Si wires fabricated by the selective deposition have no plasma damage, resulting in the high conductivity even though the size is smaller than the RIE sample.

FIG. 4. Cross sectional SEM images for different growth conditions are shown. The deposition cycle is 60. The bottom parts indicate the schematic of the structures. The top $SiO₂$ layer was deposited by the atmospheric CVD at 400 °C, while the bottom $SiO₂$ layer was thermally grown at 1000 °C.

Downloaded 17 Jun 2007 to 133.41.149.126. Redistribution subject to AIP license or copyright, see http://apl.aip.org/apl/copyright.jsp

TABLE I. Comparison between RIE (plasma etching) and selective deposition samples is shown. The resistivity of the Si wire was obtained from the *I* –*V* characteristics, while that for the sample ''before etching'' was measured by the four-probe method. Detailed fabrication process of the RIE sample is in Ref. 13. *W* and *T* in the column of ''Size'' mean width and thickness of the Si wire, respectively.

Method	Plasma etching	Selective deposition
Size $(nm)\times(nm)$	$70(W)\times 220(T)$	$21(W)\times 28(T)$
Resistivity $(\Omega$ cm)	2.6×10^{-2} (Before etching 2.1×10^{-3})	5.8×10^{-3}
POCl ₃ doping Postanneal	Flow 850° C, 30 min Drive-in 850 °C, 10 min 850 °C, 30 min	Flow 850° C, 10 min Drive-in 850 °C, 10 min

FIG. 5. (a) $I - V$ curves for the Si wires of 21 nm width \times 28 nm thick and (b) length dependence of the resistance of the Si wires are shown.

In conclusion, a low-temperature $(410^{\circ}C)$ selective deposition method for Si, ''time-modulated flow method,'' was developed. Owing to the low-growth temperature, the smooth surface is obtained. The Si wires with the size of 21 nm width \times 28 nm thick were fabricated by using the selective deposition. They have a larger conductance than that fabricated by the RIE. From the viewpoint of the good surface smoothness and low-temperature process with no damage, this technology can be used for the fabrication of the low resistivity polycrystalline silicon gate for the ultrasmall transistors and quantum effect devices.

This work has been supported by a Grant-in-Aid for Scientific Research (B) from the Ministry of Education, Science, Sports, and Culture, Japanese Government (No. 11450125), and the CREST (Core Research for Evolutional Science and Technology) of the Japan Science and Technology Corporation (JST).

- ¹ S. Yokoyama, N. Ikeda, K. Kajikawa, and Y. Nakashima, Appl. Surf. Sci. **130–132**, 352 (1998).
- 2S. Yokoyama, Y. Nakashima, and K. Ooba, J. Korean Phys. Soc. **35**, S71 (1999) .
- 3T. Yonehara, Y. Nishigaki, H. Mizutani, S. Kondoh, K. Yamagata, T. Noma, and T. Ichikawa, Appl. Phys. Lett. **52**, 1231 (1988).
- 4H. Hirayama, T. Tatsumi, and N. Aizaki, Appl. Phys. Lett. **52**, 2242 $(1988).$
- 5 K. Aketagawa, T. Tatsumi, and J. Sakai, Appl. Phys. Lett. **59**, 1735 (1991).
- ⁶G. N. Parsons, Appl. Phys. Lett. **59**, 2546 (1991).
- 7 T. Yasuda, S. Yamasaki, and S. Gwo, Appl. Phys. Lett. **77**, 3917 (2000) .
- 8K. Ooba, Y. Nakashima, A. Nakajima, and S. Yokoyama, *Ext. Abst. Conf. Solid State Devices and Materials* (Business Center for Academic Societies Japan, Tokyo, 1998), p. 22.
- ⁹S. Miyazaki, Y. Hamamoto, E. Yoshida, M. Ikeda, and M. Hirose, Thin Solid Films 369, 55 (2000).
- ¹⁰R. C. Henderson, J. Electrochem. Soc. **119**, 772 (1972).
- 11M. Browery and J. H. Purnell, Proc. R. Soc. London, Ser. A **321**, 341 $(1971).$
- 12K. Goto, J. Murota, F. Honma, T. Matsuura, and Y. Sawada, *Proceedings of Fifth International Symposium on Ultra Large Scale Integration Science and Technology*, edited by E. M. Middlesworth and H. Z. Massoud (The Electrochemical Society, Pennington, NJ, 1995), p. 512.
- 13H. Murakami, T. Mihara, S. Miyazaki, and M. Hirose, *Ext. Abst. Conf. Solid State Devices and Materials* (Business Center for Academic Societies Japan, Tokyo, 2000), p. 194.