

# Ytterbium-Catalyzed Dual Intermolecular Hydrophosphination: Synthesis of Bis(phosphinyl)dienes and Bis(alkenyl)phosphine Oxides

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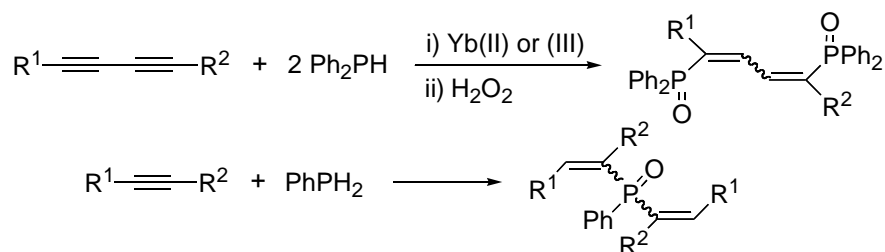
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## Graphical Abstract

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### Ytterbium-Catalyzed Dual Intermolecular Hydrophosphination: Synthesis of Bis(phosphinyl)dienes and Bis(alkenyl)phosphina Oxide

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## Abstract

Dual intermolecular hydrophosphination of conjugated diynes with two equivalents of diphenylphosphine was catalyzed by ytterbium complexes,  $\text{Yb}(\eta^2\text{-Ph}_2\text{CNPh})(\text{hmpa})_3$  (**1**) and  $\text{Yb}[\text{N}(\text{SiMe}_3)_2]_3(\text{hmpa})_2$  (**2**), to give the corresponding 1,4-bis(diphenylphosphinyl)buta-1,3-dienes in high yields after oxidative work-up. Distribution of the four possible stereoisomers sharply depended on substituents of the substrates. (*Z,Z*)-Isomers were predominantly obtained from the disubstituted

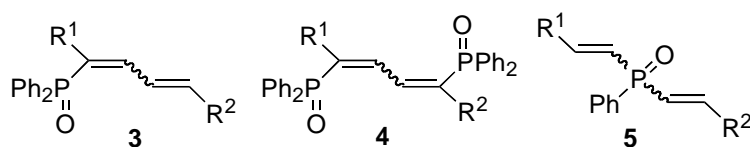
diynes, together with minor (*Z,E*)-isomers. On the other hand, the reaction of the terminal diynes provided major (*E,Z*) and minor (*E,E*)-butadienes. 1,4-Di-*tert*-butylbuta-1,3-diyne was exclusively converted to an allenic compound. Moreover, the dual hydrophosphination using phenylphosphine was also performed with **1** and **2**. Thus, the reaction of two equivalents of aromatic alkynes with PhPH<sub>2</sub> and subsequent oxidation gave bis(alkenyl)phosphine oxides in preference of the (*Z,Z*)-stereoisomers.

## Introduction

$\alpha,\beta$ -Unsaturated phosphorus compounds are useful building blocks in organic synthesis.<sup>1</sup> Of their synthetic methods, addition reaction of (RO)<sub>2</sub>P(O)H and R<sub>2</sub>P(O)H to alkynes through oxidative addition of groups 9 and 10 catalysts to P-H bond has been recognized as the most promising and atom-economical process.<sup>2</sup> When this methodology was applied to the reaction of R<sub>2</sub>PH, instead of pentavalent phosphorus, harsh conditions were necessary to promote the reaction, because of strong affinity of the late transition metals with the trivalent phosphines.<sup>3</sup> On the other hand, interaction between hard lanthanide metals and the soft phosphines could be so weak, and thus efficient intramolecular hydrophosphination of alkynyl and alkenylphosphines has been explored with lanthanocenes.<sup>4</sup>

Previously, we reported that a divalent ytterbium-imine complex, Yb( $\eta^2$ -Ph<sub>2</sub>CNPh)(hmpa)<sub>3</sub> (**1**), and a trivalent silylamide complex, Yb[N(SiMe<sub>3</sub>)<sub>2</sub>]<sub>3</sub>(hmpa)<sub>2</sub> (**2**), served as highly effective precatalysts for intermolecular hydrophosphination of alkynes with diphenylphosphine and that the active species generated *in situ* were ytterbium phosphides, Yb(PPh<sub>2</sub>)<sub>2</sub> and Yb(PPh<sub>2</sub>)<sub>3</sub>, which exhibited similar regio- and stereoselectivity irrespective of their valence state.<sup>5</sup> Based on this work, we intended to develop a simple method for the synthesis of potentially useful phosphorus compounds **3-5**. Diphenylphosphinyldienes **3** have been already prepared by the Y[N(SiMe<sub>3</sub>)<sub>2</sub>]<sub>3</sub>-catalyzed dimerization of terminal alkynes, followed by hydrophosphination of the resulting enynes with Ph<sub>2</sub>PH in one-pot.<sup>6</sup>

Compound **4** would be synthesized with dual hydrophosphination of conjugated diynes with 2 equivalents of Ph<sub>2</sub>PH. However, two preliminary examples revealed that **4** was formed from primary alkyldiyne, but in contrast, tertiary alkyldiyne gave an allenic product.<sup>5</sup> Bis(alkeny)phosphines and their oxide **5** would be also prepared using PhPH<sub>2</sub>, though it has been never used in the present system. To confirm these possibilities, we investigated the dual intermolecular hydrophosphination leading to **4** and **5**.

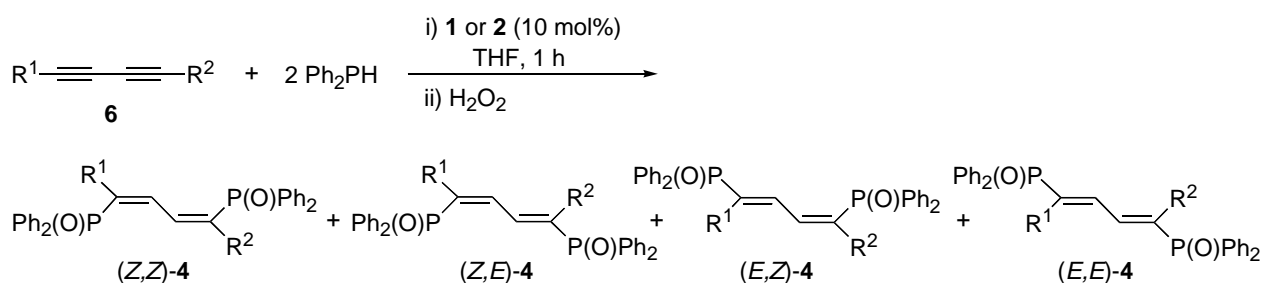


## Results and Discussion

When hexadeca-7,9-diyne (**6a**) was treated with 2 equivalents of Ph<sub>2</sub>PH in the presence of **1** (10 mol%) in THF at room temperature for 1 h, 7,10-bis(diphenylphosphinyl)hexadeca-7,9-diene (**4a**) was obtained in 80% total yield as a mixture of two stereoisomers (61 : 39) after oxidation with H<sub>2</sub>O<sub>2</sub> (Table 1, run 1).<sup>7</sup> The major product showed one olefinic signal at 7.74 ppm with 34.8 Hz of *trans*-<sup>3</sup>J<sub>P-H</sub> in <sup>1</sup>H NMR and one signal at 29.53 ppm in <sup>31</sup>P NMR. On the other hand, two olefinic signals at 6.83 (*cis*-<sup>3</sup>J<sub>P-H</sub> = 20.6 Hz) and 7.13 ppm (*trans*-<sup>3</sup>J<sub>P-H</sub> = 35.0 Hz) in <sup>1</sup>H NMR and two signals at 28.18 and 33.58 ppm in <sup>31</sup>P NMR were observed for the minor. Therefore, the major product was definitely assigned to the (*Z,Z*)-isomer, and the minor to the (*Z,E*)-isomer. The stereoselectivity was increased a little with decreasing temperature (run 2). The reaction was also conducted with the silylamide **2** effectively (run 3). Two cyclohexyl substituents in **6c** did not change the reaction mode, though its reactivity was somewhat decreased (run 5). Arylalkyldiynes **6d** and **6e** gave the (*Z,Z*) and (*Z,E*)-dienes **4d** and **4e**; in the latter substrate, a small amount of the (*E,Z*)-isomer was contaminated in the mixture (runs 6-9). 1,4-Diphenylbuta-1,3-diene (**6f**) was so reactive that no phosphinylated product **4f** was

obtained with **1**, other than polymers even at  $-78\text{ }^{\circ}\text{C}$  (run 10). In the absence of  $\text{Ph}_2\text{PH}$ , **6f** was recovered unchanged by the treatment with **1** at room temperature. Use of milder precatalyst **2** afforded (*E,E*)-**4f** in low yield (run 11). The reaction of terminal diynes **6g** and **6h** gave the expected products **4g** and **4h**, but their stereochemistry was drastically altered, (*E,Z*) and (*E,E*)-isomers being major and minor, respectively (runs 12-15). Aromatic terminal diyne **6i** was exclusively polymerized with both **1** and **2**, because of its higher reactivity (run 16).

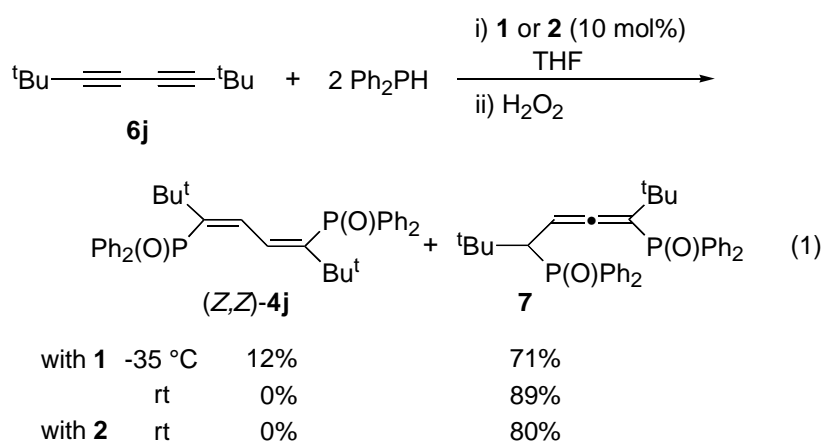
**TABLE 1. Dual Hydrophosphination of Conjugated Diynes with Diphenylphosphine**



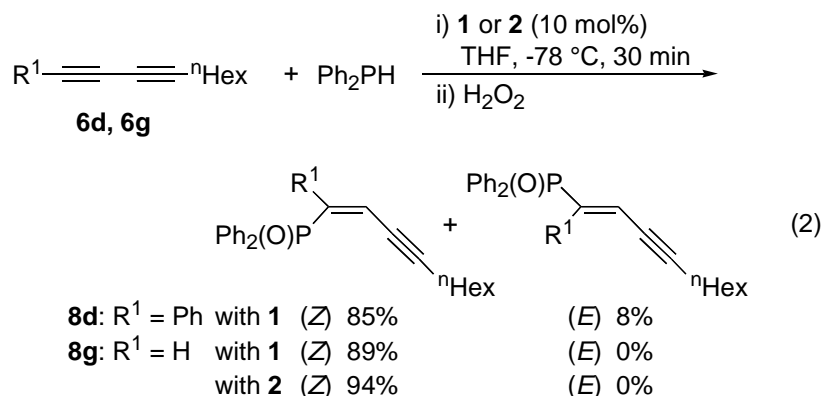
run	<b>6</b>	R <sup>1</sup>	R <sup>2</sup>	precatalyst	temp (°C)	<b>4</b>	total yield (%) <sup>a</sup>	ratio of isomers (Z,Z):(Z,E):(E,Z):(E,E)
1	<b>6a</b>	<sup>n</sup> Hex	<sup>n</sup> Hex	<b>1</b>	rt	<b>4a</b>	80	61 : 39 : - : 0
2				<b>1</b>	-15		82	74 : 26 : - : 0
3				<b>2</b>	-15		82	61 : 39 : - : 0
4	<b>6b</b>	<sup>n</sup> Bu	<sup>n</sup> Bu	<b>1</b>	-15	<b>4b</b>	92	67 : 33 : - : 0
5	<b>6c</b>	c-C <sub>6</sub> H <sub>11</sub>	c-C <sub>6</sub> H <sub>11</sub>	<b>1</b>	-15	<b>4c</b>	74	86 : 14 : - : 0
6	<b>6d</b>	Ph	<sup>n</sup> Hex	<b>1</b>	-15	<b>4d</b>	98	73 : 27 : 0 : 0
7				<b>2</b>	-15		95	72 : 28 : 0 : 0
8	<b>6e</b>	4-MeOPh	<sup>n</sup> Hex	<b>1</b>	-15	<b>4e</b>	85	73 : 19 : 8 : 0
9				<b>2</b>	-15		65	74 : 18 : 8 : 0
10	<b>6f</b>	Ph	Ph	<b>1</b>	-78	<b>4f</b>	0	polymerization
11				<b>2</b>	-78		28	0 : 0 : - : 100
12	<b>6g</b>	H	<sup>n</sup> Hex	<b>1</b>	-78 <sup>b</sup>	<b>4g</b>	80	0 : 0 : 61 : 39
13				<b>2</b>	-78 <sup>b</sup>		89	6 : 0 : 75 : 19
14	<b>6h</b>	H	<sup>n</sup> Bu	<b>1</b>	-78 <sup>b</sup>	<b>4h</b>	89	16 : 0 : 64 : 20
15				<b>2</b>	-78 <sup>b</sup>		72	7 : 0 : 61 : 32
16	<b>6i</b>	H	Ph	<b>1 or 2</b>	-78	<b>4i</b>	0	polymerization

<sup>a</sup> NMR yield. <sup>b</sup> -78 °C for 1 h then room temperature for 2 h.

The behavior of di-*tert*-butyldiyne **6j** seems to be very different from that of others. Thus, the reaction with **1** at  $-35\text{ }^{\circ}\text{C}$  gave bis(diphenylphosphinyl)allene **7** and (*Z,Z*)-diene **4j** in 71% and 12% yields, respectively (Eq. 1). When the reaction was performed at room temperature with **1** and **2**, the allene **7** was exclusively obtained in high yields.

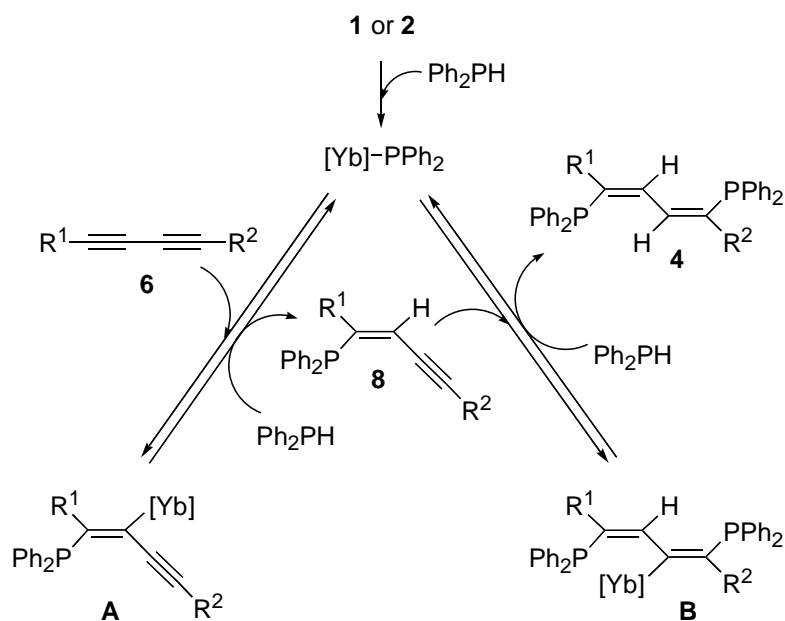


Then, single hydrophosphination of unsymmetrical diynes **6d** and **6g** was carried out using equimolar amounts of  $\text{Ph}_2\text{PH}$  to learn the reaction process (Eq. 2). Apparently, the first reaction in **6d** took place at the  $\alpha$ -alkyne carbon attached to the Ph group, with high selectivity for *anti*-addition, to yield (*Z*)-**8d**. This result contrasts well with the regio- and stereoselectivity observed for simple aromatic alkynes, in which *syn*-addition of  $\text{Ph}_2\text{P}$  occurred at the  $\beta$ -alkyne carbon to the aryl substituents, giving rise to (*E*)-products.<sup>5</sup> In the reaction of the terminal diyne **6g**, enyne compound (*Z*)-**8g** was formed as a single isomer through *anti*-addition to the terminal position.



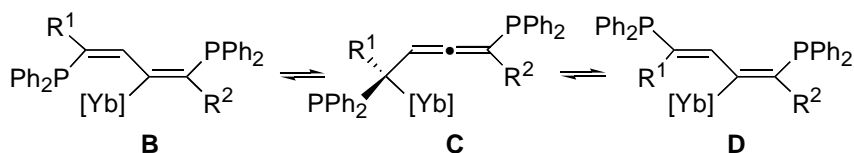
Combining these results with the reaction mechanism for the hydrophosphination of simple alkynes,<sup>5</sup> the process of the present system would be accounted for as follows (Scheme 1). At first, bis or tris(diphenylphosphino)ytterbium species, [Yb]-PPh<sub>2</sub>, was generated *in situ* from the precatalysts **1** or **2** as previously proved. *anti*-Addition of the phosphide complex to the diyne **6** gave the intermediate **A**, which was readily protonated with Ph<sub>2</sub>PH to yield the enyne compound **8** and regenerate the active phosphide species. The enyne **8** reacted further with the phosphide in a manner similar to the first cycle to yield the diene **4** via dienylytterbium **B**.

**SCHEME 1**



Regiochemistry was independent of the substituents of the diyne **6**, i.e., two Ph<sub>2</sub>P were introduced to the 1- and 4-position of the butadiyne moiety, whereas stereochemistry was significantly variable. As proposed in Scheme 1, repeated *anti*-addition of the Yb-phosphide complex could produce (*Z,Z*)-**4**, which is the case of the internal diynes **6a-6e**. In general, the overall stereoselectivity would be rationalized by the isomerization of the intermediate **B** formed by the second addition, which proceeds through allenic species **C** (Scheme 2). In the case of the terminal diynes **6g** and **6h**, **B** (R<sup>1</sup> = H) changes readily to other dienylytterbium **D** to avoid the steric repulsion between Ph<sub>2</sub>P and Yb in the structure of **B**, giving rise to (*E,Z*)-**4g** and **4h** as major products. Moreover, it is likely that all four stereoisomeric dienylytterbiums such as **B** and **D** are interconvertible *via* **C**, but the equilibrium would not be attained completely, because of the rapid protonation of the intermediates with Ph<sub>2</sub>PH. Therefore, stereoselectivity of the present reaction would be a reflection of both kinetic and thermodynamic factors.

**SCHEME 2**

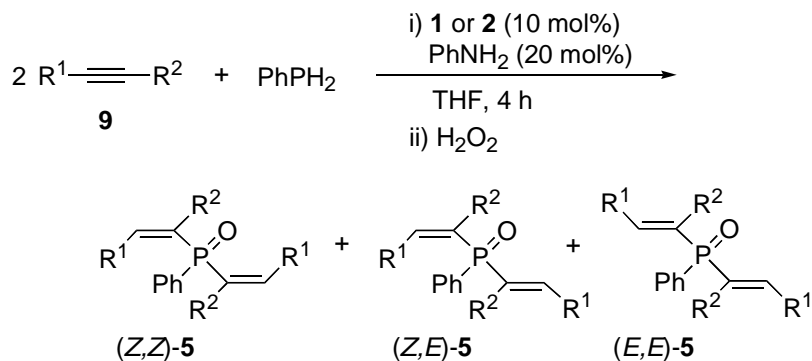


Formation of the allene **7** from di-*tert*-butyldiyne **6j** substantiates the scenario described above. In this case, two tertiary substituents caused severe steric crowding in the form of a dienylytterbium species, and thus the allenic **C** should be the most stable intermediate, giving rise to **7** exclusively at room temperature. However, when the reaction was carried out at lower temperature, the (*Z,Z*)-diene **4j** derived from **B** was included as a minor product. Thus, **4j** and **7** could be kinetic and thermodynamic products, respectively.

Next, we studied the dual hydrophosphination with phenylphosphine for the synthesis of bis(alkenyl)phosphines and their oxides **5**. The reaction of 1-phenylprop-1-yne (**9a**) (2 equiv.) with PhPH<sub>2</sub> using the imine complex **1** (10 mol%) was slower than that with Ph<sub>2</sub>PH, and required heating (Table 2, runs 1 and 2). After oxidative work-up with H<sub>2</sub>O<sub>2</sub>, three isomers of bis( $\beta$ -methylstyryl)phenylphosphine oxide (**5a**) were formed in 43% total yield (run 2).<sup>7</sup> The silylamide precatalyst **2** gave better yield, which was further increased to 64% yield, though still non-selective, by addition of 20 mol% of aniline (runs 3 and 4). The reaction using equimolar amounts of the alkyne **9a** and PhPH<sub>2</sub> gave an intractable mixture, in which bis(1-methyl-2-phenylethyl)phenylphosphine oxide, a reduced product of **5a**, was mainly detected. In the reaction of phenylacetylene (**9b**), the aniline additive showed a significant effect for improvement of the selectivity as well as total yield of **5b** (run 5 vs. 6). Thus, the (*Z,Z*)- and (*Z,E*)-isomers were obtained in 80% and 20% yields, respectively. Similarly, various aromatic alkynes **9c-9f** were converted to the expected products **5c-5f** in high yields (runs 7-10).



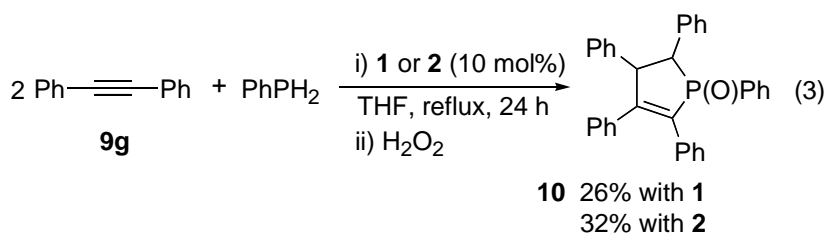
**TABLE 2. Dual Hydrophosphination of Alkynes with Phenylphosphine**



run	<b>9</b>	R <sup>1</sup>	R <sup>2</sup>	precat	temp (°C)	total yield <b>5</b> (%) <sup>a</sup>	ratio of isomers (Z,Z):(Z,E):(E,E)
1	<b>9a</b>	Ph	Me	<b>1<sup>b</sup></b>	rt <sup>c</sup>	<b>5a</b> 7	100 : 0 : 0
2				<b>1<sup>b</sup></b>	reflux	43	37 : 37 : 26
3				<b>2<sup>b</sup></b>	reflux	59	46 : 34 : 20
4				<b>2</b>	reflux	64	36 : 34 : 30
5	<b>9b</b>	Ph	H	<b>2<sup>b</sup></b>	rt	<b>5b</b> 75	48 : 52 : 0
6				<b>2</b>	rt	quant	80 : 20 : 0
7	<b>9c</b>	4-MeOPh	H	<b>2</b>	rt	<b>5c</b> quant	82 : 18 : 0
8	<b>9d</b>	4-MePh	H	<b>2</b>	rt	<b>5d</b> 97	89 : 11 : 0
9	<b>9e</b>	4-ClPh	H	<b>2</b>	rt	<b>5e</b> 89	78 : 22 : 0
10	<b>9f</b>	4-BrPh	H	<b>2</b>	rt	<b>5f</b> 89	81 : 19 : 0

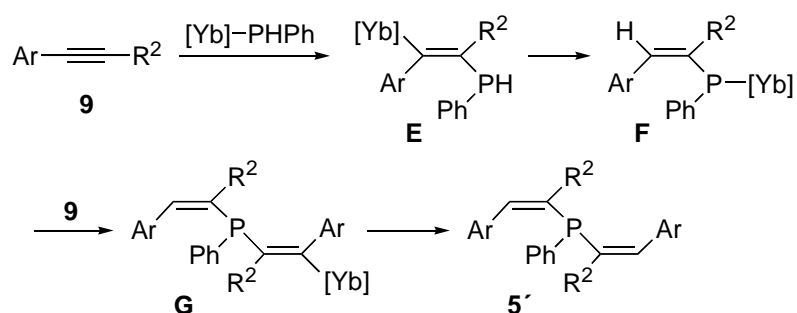
<sup>a</sup> NMR yield. <sup>b</sup> In the absence of PhNH<sub>2</sub>. <sup>c</sup> Reaction period is 15 h.

Unfortunately, this method was not applicable to the reaction of aliphatic alkynes, which resulted in recovery of the starting materials or exclusive consumption of PhPH<sub>2</sub> under various conditions. Moreover, the reaction of diphenylacetylene (**9g**) gave cyclic phosphine oxide **10** unexpectedly in low yields, together with small amounts of *trans*-stilbene (Eq. 3).<sup>8</sup>



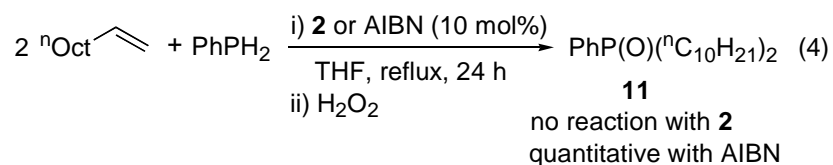
The present reaction would proceed in a mechanism analogous to that with diphenylphosphine as shown in Scheme 3. Addition of phenylphosphinoytterbium, [Yb]-PPh, to aromatic alkyne **9**, followed by intramolecular proton transfer would yield alkenylphosphide species **F**, which reacts further with the second molecule of **9** to give bis(alkenyl)phosphine **5'** through the intermediate **G**.

### SCHEME 3



The regiochemistry was in agreement with that observed for the reaction with  $\text{Ph}_2\text{PH}$ , whereas the stereochemistry was reversed. Since isomerization of the intermediates and products is less likely in the present system, *anti*-addition of the phosphide species would take place preferentially to afford the (*Z,Z*)-isomers. The high (*Z*)-selectivity may be caused by a radical mechanism instead of the process proposed in Scheme 3.<sup>9</sup> However, this possibility was completely ruled out. Thus, the reaction of two equivalents of 1-decene with  $\text{PhPH}_2$  did not occur with **2**, but in contrast, addition product **11** was quantitatively formed under radical conditions (Eq. 4). Alternatively, the (*Z*)-selectivity would be attributed to the amine additive, which has been known to change the activity of the lanthanide

catalysts and product selectivity in various reactions as a proton source and ligand.<sup>6,10</sup> In fact, aliphatic amines like amylamine decreased the yield and selectivity in contrast to aniline. However, the reason for the difference between the reaction with PhPH<sub>2</sub> and Ph<sub>2</sub>PH is still unclear at present.



In summary, dual intermolecular hydrophosphination of conjugated diynes with two equivalents of Ph<sub>2</sub>PH has been achieved using ytterbium (II) and (III) precatalysts **1** and **2** to give bis(diphenylphosphinyl)dienes **4** in high yields. Addition reaction of the active species, Yb-phosphide, was found to proceed in an *anti*-fashion, but the resulting intermediates could isomerize *via* an allenic species. Thus, the stereochemistry of the products **4** was determined both kinetically and thermodynamically, depending on their substituents. Moreover, bis(alkenyl)phosphine oxides **5** were effectively obtained from two equivalents of aromatic alkynes and PhPH<sub>2</sub> by the dual reaction in the presence of an aniline additive.

## Experimental Section

**General.** <sup>1</sup>H, <sup>13</sup>C, and <sup>31</sup>P NMR spectra were recorded at 396, 99, and 160 MHz, respectively. IR spectra were taken on an FT-IR spectrophotometer. Mass spectra (EI) were obtained at 70 eV on a GC-MS apparatus. MALDI-TOF mass spectra were acquired using 1,8,9-trihydroxyanthracene as the matrix. Microanalyses were performed at our analytical laboratory. Melting points are uncorrected. All reactions were carried out under argon. THF was distilled from sodium/benzophenone ketyl

immediately prior to use. HMPA was distilled from  $\text{CaH}_2$  and stored over molecular sieves. The Yb-imine precatalyst **1** was generated *in situ* from Yb metal,  $\text{Ph}_2\text{C}=\text{NPh}$ , and HMPA (6 equiv) in THF as previously reported.<sup>5</sup> The silylamide precatalyst **2** was also generated by treatment of  $\text{Yb}[\text{N}(\text{SiMe}_3)_2]_3(\text{thf})_2$ <sup>11</sup> with HMPA (2 equiv) in THF. Symmetrical diynes **6a-6c** and **6f**,<sup>12</sup> unsymmetrical diynes **6d** and **6e**,<sup>13</sup> and terminal diynes **6g-6i**<sup>14</sup> were prepared according to the literature methods. All other materials were commercially available and were used after drying and distillation.

**Dual Hydrophosphination of the Conjugated Diynes 6 with Diphenylphosphine.**  $\text{Ph}_2\text{PH}$  (2.0 mmol) was added to a solution of **1** or **2** (0.1 mmol) in THF (1 mL) and the mixture was stirred for 30 min at room temperature. After cooling the mixture to the appropriate temperature indicated in Table 1, diyne **6** (1.0 mmol) was added to the solution and stirring was continued for 1 h at this temperature. In the case of terminal diynes, the mixture was stirred for 1 h at  $-78\text{ }^\circ\text{C}$ , and then for 2 h at room temperature to ensure the completion of the reaction. The reaction was quenched with  $\text{H}_2\text{O}$  (1 mL) and HCl solution (1M, 1 mL), and the mixture was oxidized with  $\text{H}_2\text{O}_2$  (30%, 1 mL) at  $0\text{ }^\circ\text{C}$ . Dimethyl terephthalate was added to the mixture as an internal standard. The reaction mixture was extracted with  $\text{CHCl}_3$ , washed with brine, dried over  $\text{MgSO}_4$ , and concentrated *in vacuo*. The product yield and ratio were determined by  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra of the crude mixture. Analytically pure compounds **4** were isolated by column chromatography on silica gel with chloroform-acetone eluent.

For determination of the stereochemistry of the products, the coupling constants between the olefinic protons and  $\text{Ph}_2\text{P}(\text{O})$  in  $^1\text{H}$  NMR are informative; *trans*- $^3J_{\text{P-H}}$  (ca. 35Hz), *cis*- $^3J_{\text{P-H}}$  (ca. 19Hz), and  $^2J_{\text{P-H}}$  (ca. 22 Hz). In addition,  $^{31}\text{P}$  NMR signals of the olefinic (*Z*)- $\text{Ph}_2\text{P}(\text{O})$  always appear in higher field than those of the (*E*).<sup>9</sup> For examples, (alkyl)[ $\text{P}(\text{O})$ ]C=CHR appears at ca. 29 (*Z*) and 33 ppm (*E*), (aryl)[ $\text{P}(\text{O})$ ]C=CHR at ca. 26 (*Z*) and 31 ppm (*E*), (H)[ $\text{P}(\text{O})$ ]C=CHR at ca. 22 (*Z*) and 25 ppm (*E*).

**7,10-Bis(diphenylphosphinyl)hexadeca-7,9-diene (4a).** (*Z,Z*)-isomer: white solid; mp 137-140 °C; IR (KBr) 1180 cm<sup>-1</sup>; MS (MALDI) *m/z* 619.99 (M<sup>+</sup>-2); MS *m/z* 622 (M<sup>+</sup>), 551 (M<sup>+</sup>-C<sub>5</sub>H<sub>11</sub>), 421 (M<sup>+</sup>-Ph<sub>2</sub>PO), 201 (Ph<sub>2</sub>PO<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.75 (6H, t, *J* = 7.2 Hz), 0.83-0.97 (12H, m), 1.00-1.09 (4H, m), 1.86-1.97 (4H, m), 7.45-7.55 (12H m), 7.64-7.69 (8H, m), 7.76 (2H, d, *J* = 35.0 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 13.9, 22.3, 28.6, 29.3 (d, *J* = 3.7 Hz), 31.2, 35.7 (d, *J* = 12.3 Hz), 128.5 (d, *J* = 8.3 Hz), 131.7 (d, *J* = 2.5 Hz), 131.8 (d, *J* = 10.7 Hz), 133.2 (d, *J* = 101.7 Hz), 137.6 (d, *J* = 86.2 Hz), 140.9 (dd, *J* = 10.3 and 3.7 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 29.53. Anal. Calcd for C<sub>40</sub>H<sub>48</sub>O<sub>2</sub>P<sub>2</sub>: C, 77.15; H, 7.77. Found: C, 77.03; H, 7.74. (*Z,E*)-Isomer: yellow oil; IR (Neat) 1178 cm<sup>-1</sup>; MS (MALDI) *m/z* 621.17 (M<sup>+</sup>-1); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.81 (3H, t, *J* = 7.0 Hz), 0.84 (3H, t, *J* = 7.0 Hz), 1.09-1.37 (16H, m), 2.21 (2H, dt, *J* = 13.8, 6.9 Hz), 2.50 (2H, dt, *J* = 16.1, 7.7 Hz), 6.83 (1H, dd, *J* = 20.6, 11.8 Hz), 7.13 (1H, ddd, *J* = 35.0, 11.8, 1.2 Hz), 7.27-7.53 (20H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 13.9, 14.0, 22.4, 22.5, 28.7, 29.3, 29.5 (d, *J* = 4.1 Hz), 30.5 (d, *J* = 1.7 Hz), 31.3, 31.4, 36.4 (d, *J* = 1.7 Hz), 36.5 (d, *J* = 1.6 Hz), 128.3 (d, *J* = 11.4 Hz), 128.5 (d, *J* = 12.3 Hz), 131.1 (d, *J* = 101.7 Hz), 131.5 (d, *J* = 9.8 Hz), 131.62 (d, *J* = 10.7 Hz), 131.63 (d, *J* = 10.1 Hz), 131.8 (d, *J* = 9.0 Hz), 132.7 (d, *J* = 101.7 Hz), 136.7 (dd, *J* = 19.7, 5.7 Hz), 136.9 (dd, *J* = 14.0, 9.8 Hz), 140.1 (dd, *J* = 96.0, 2.1 Hz), 140.9 (d, *J* = 86.1 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 28.18, 33.58.

**5,8-Bis(diphenylphosphinyl)dodeca-5,7-diene (4b).** (*Z,Z*)-Isomer: white solid; mp 162-163 °C; IR (KBr) 1180 cm<sup>-1</sup>; MS (MALDI) *m/z* 564.81 (M<sup>+</sup>-2); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.56 (6H, t, *J* = 6.9 Hz), 0.85-0.92 (8H, m), 1.92 (4H, dt, *J* = 13.5, 6.7 Hz), 7.44-7.55 (12H, m), 7.64-7.69 (8H, m), 7.76 (2H, d, *J* = 34.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 13.4, 21.9, 31.2, 35.3 (d, *J* = 12.3 Hz), 128.4 (d, *J* = 12.3 Hz), 131.72, (d, *J* = 9.8 Hz), 131.73, 133.1 (d, *J* = 100.9 Hz), 137.5 (dd, *J* = 88.2, 2.1 Hz), 140.8 (dd, *J* = 9.8, 4.1 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 29.57. Anal. Calcd for C<sub>36</sub>H<sub>40</sub>O<sub>2</sub>P<sub>2</sub>: C, 76.31; H, 7.12. Found: C, 76.51; H, 7.08. (*Z,E*)-Isomer: yellow oil; IR (Neat) 1182 cm<sup>-1</sup>; MS (MALDI) *m/z* 564.86 (M<sup>+</sup>-2); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ

0.74 (3H, t,  $J = 7.2$  Hz), 0.83 (3H, t,  $J = 7.1$  Hz), 1.14-1.37 (8H, m), 2.21 (2H, dt,  $J = 13.7, 6.6$  Hz), 2.53 (2H, dt,  $J = 15.4, 7.6$  Hz), 6.81 (1H, dd,  $J = 20.6, 11.4$  Hz), 7.13 (1H, dd,  $J = 35.0, 11.4$  Hz), 7.34-7.50 (20H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  13.5, 13.6, 22.1, 22.7, 27.2 (d,  $J = 9.0$  Hz), 31.6 (d,  $J = 4.1$  Hz), 32.6, 36.1 (d,  $J = 10.6$  Hz), 128.3 (d,  $J = 12.3$  Hz), 128.4 (d,  $J = 12.3$  Hz), 131.1 (d,  $J = 102.5$  Hz), 131.4 (d,  $J = 9.8$  Hz), 131.61 (d,  $J = 9.8$  Hz), 131.62 (d,  $J = 9.8$  Hz), 131.7 (d,  $J = 9.8$  Hz), 132.6 (d,  $J = 101.7$  Hz), 136.5 (dd,  $J = 19.7, 6.5$  Hz), 136.8 (dd,  $J = 14.7, 9.8$  Hz), 140.0 (dd,  $J = 96.8, 2.1$  Hz), 140.9 (d,  $J = 87.6$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  28.25, 33.71.

**1,4-Bis(diphenylphosphinyl)-1,4-dicyclohexylbuta-1,3-diene (4c).** (*Z,Z*)-Isomer: white solid; mp 238-239 °C; IR (KBr) 1174  $\text{cm}^{-1}$ ; MS (MALDI)  $m/z$  616.76 ( $\text{M}^+-2$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.68-0.74 (10H, m), 1.25-1.43 (10H, m), 2.00 (2H, quin,  $J = 11.5$  Hz), 7.43-7.55 (14H, m), 7.63-7.67 (8H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  25.4, 26.3, 33.0 (d,  $J = 10.6$  Hz), 44.2 (d,  $J = 10.6$  Hz), 128.3 (d,  $J = 12.3$  Hz), 131.56, (d,  $J = 9.8$  Hz), 131.57, 133.4 (d,  $J = 100.9$  Hz), 138.2 (dd,  $J = 10.7, 4.1$  Hz), 143.0 (dd,  $J = 87.0, 2.4$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  29.61. Anal. Calcd for  $\text{C}_{40}\text{H}_{44}\text{O}_2\text{P}_2$ : C, 77.65; H, 7.17. Found: C, 77.77; H, 7.21. (*Z,E*)-Isomer: isolated as a mixture with (*Z,Z*)-isomer; MS (MALDI)  $m/z$  617.46 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (clearly assignable peaks)  $\delta$  6.69 (1H, dd,  $J = 22.1, 11.7$  Hz);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  25.80, 25.83, 26.6, 26.9, 32.3 (d,  $J = 3.3$  Hz), 33.9 (d,  $J = 4.1$  Hz), 40.1 (d,  $J = 9.0$  Hz), 42.0 (d,  $J = 11.5$  Hz), 128.2 (d,  $J = 11.5$  Hz), 128.5 (d,  $J = 12.3$  Hz), 131.4 (d,  $J = 2.4$  Hz), 131.5 (d,  $J = 42.7$  Hz), 131.64 (d,  $J = 9.8$  Hz), 131.65 (d,  $J = 2.4$  Hz), 131.9 (d,  $J = 9.8$  Hz), 132.9 (d,  $J = 101.7$  Hz), 135.1 (dd,  $J = 20.9, 5.3$  Hz), 136.9 (dd,  $J = 15.6, 9.8$  Hz), 144.0 (dd,  $J = 93.5, 1.6$  Hz), 145.1 (d,  $J = 86.1, 1.6$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  29.92, 35.16.

**1,4-Bis(diphenylphosphinyl)-1-phenyldeca-1,3-diene (4d).** (*Z,Z*)-Isomer: white solid; mp 124-126 °C; IR (KBr) 1180  $\text{cm}^{-1}$ ; MS (MALDI)  $m/z$  613.77 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.77 (3H, t,  $J = 7.4$

Hz), 0.87-1.00 (6H, m), 1.07 (2H, quint,  $J = 7.1$  Hz), 1.95-2.00 (2H, m), 6.69 (2H, d,  $J = 7.0$  Hz), 6.94 (2H, t,  $J = 7.4$  Hz), 7.01 (1H, t,  $J = 7.3$  Hz), 7.32-7.68 (20H, m), 7.82 (1H, dd,  $J = 35.5, 11.0$  Hz), 7.89 (1H, dd,  $J = 33.9, 11.0$  Hz);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) (clearly assignable peaks)  $\delta$  13.9, 22.2, 28.6, 29.1 (d,  $J = 4.1$  Hz), 31.1, 35.6 (d,  $J = 11.5$  Hz), 132.8 (d,  $J = 101.7$  Hz), 133.5 (d,  $J = 102.5$  Hz), 138.5 (dd,  $J = 90.2, 2.1$  Hz), 139.5 (d,  $J = 9.8$  Hz), 140.3 (dd,  $J = 87.4, 2.0$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  26.13, 29.43. Anal. Calcd for  $\text{C}_{40}\text{H}_{40}\text{O}_2\text{P}_2$ : C, 78.16; H, 6.56. Found: C, 78.03; H, 6.68. (*Z,E*)-Isomer: colorless oil; IR (Neat)  $1182\text{ cm}^{-1}$ ; MS (MALDI)  $m/z$  613.04 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.80 (3H, t,  $J = 7.3$  Hz), 1.09-1.45 (8H, m), 2.54 (2H, dd,  $J = 16.0, 8.0$  Hz), 6.99-7.01 (2H, m), 7.07-7.18 (4H, m), 7.27-7.53 (21H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) (clearly assignable peaks)  $\delta$  13.9, 22.4, 27.5 (d,  $J = 9.0$  Hz), 29.2, 30.4, 31.2, 136.5 (dd,  $J = 14.8, 9.1$  Hz), 139.9 (dd,  $J = 9.9, 1.7$  Hz), 141.8 (dd,  $J = 88.6, 1.6$  Hz), 143.0 (dd,  $J = 95.2, 1.7$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  25.52, 33.38.

**1,4-Bis(diphenylphosphinyl)-1-(4-methoxyphenyl)deca-1,3-diene (4e).** (*Z,Z*)-Isomer: white solid; mp  $113-115\text{ }^\circ\text{C}$ ; IR (KBr)  $1178\text{ cm}^{-1}$ ; MS (MALDI)  $m/z$  643.88 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.77 (3H, t,  $J = 7.2$  Hz), 0.79-1.12 (8H, m), 1.91-2.00 (2H, m), 3.67 (3H, s), 6.48 (2H, d,  $J = 8.2$  Hz), 6.59 (2H, d,  $J = 8.2$  Hz), 7.35-7.68 (20H, m), 7.78 (1H, dd,  $J = 35.3, 12.1$  Hz), 7.83 (1H, dd,  $J = 34.5, 12.1$  Hz);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.0, 22.3, 28.7, 29.2 (d,  $J = 3.3$  Hz), 31.2, 35.7 (d,  $J = 10.7$  Hz), 55.1, 113.1, 128.3 (d,  $J = 12.3$ ), 128.5 (d,  $J = 11.5$  Hz), 129.9 (d,  $J = 4.9$  Hz), 131.5 (d,  $J = 2.4$  Hz), 131.6 (d,  $J = 9.8$  Hz), 131.7 (d,  $J = 9.8$  Hz), 131.8 (d,  $J = 2.4$  Hz), 132.1 (d,  $J = 10.7$  Hz), 133.0 (d,  $J = 101.7$  Hz), 133.7 (d,  $J = 102.5$  Hz), 138.0 (dd,  $J = 90.7, 2.1$  Hz), 140.0 (dd,  $J = 87.4, 2.1$  Hz), 140.4 (dd,  $J = 8.6, 4.5$  Hz), 140.9 (dd,  $J = 9.4, 5.3$  Hz), 158.8;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  26.18, 29.41. Anal. Calcd for  $\text{C}_{41}\text{H}_{42}\text{O}_3\text{P}_2$ : C, 76.38; H, 6.57. Found: C, 75.96; H, 6.52. (*Z,E*)-Isomer: isolated as a mixture of (*Z,E*) and (*E,Z*)-isomer (65 : 35): yellow oil; MS (MALDI)  $m/z$  643.88 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.81 (3H, t,  $J = 7.1$  Hz), 0.99-1.39 (8H, m), 2.49-2.57 (2H, m), 3.72 (3H, s), 6.65 (2H, d,  $J = 8.7$  Hz), 6.93 (2H, d,  $J = 8.7$

Hz), 7.07 (1H, dd,  $J = 20.4, 11.7$  Hz), 7.28-7.57 (21H, m);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  25.69, 33.58. (*E,Z*)-Isomer:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (clearly assignable peaks)  $\delta$  0.77 (3H, t,  $J = 7.4$  Hz), 2.06-2.11 (2H, m), 3.75 (3H, s), 6.81 (2H, d,  $J = 8.7$  Hz), 7.22 (2H, d,  $J = 8.9$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  28.33, 31.37.

**(*E,E*)-1,4-Bis(diphenylphosphinyl)-1,4-diphenylbuta-1,3-diene (4f).** White solid; mp 267-270 °C; IR (KBr) 1182  $\text{cm}^{-1}$ ; MS (MALDI)  $m/z$  605.67 ( $\text{M}^+-1$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  6.76 (2H, d,  $J = 16.9$  Hz), 7.04-7.16 (10H, m), 7.27-7.43 (12H, m), 7.53-7.58 (8H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  128.1 (d,  $J = 30.4$  Hz), 128.2 (d,  $J = 2.4$  Hz), 128.3 (d,  $J = 2.4$  Hz), 129.5 (d,  $J = 2.4$  Hz), 130.7 (d,  $J = 104.2$  Hz), 131.78 (d,  $J = 9.0$  Hz), 131.82 (d,  $J = 7.4$  Hz), 134.1 (d,  $J = 4.5$  Hz), 137.9 (d,  $J = 14.4$  Hz), 142.8 (dd,  $J = 94.3, 1.4$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  29.97. Anal. Calcd for  $\text{C}_{40}\text{H}_{32}\text{O}_2\text{P}_2$ : C, 79.20; H, 5.32. Found: C, 79.13; H, 5.02

**1,4-Bis(diphenylphosphinyl)deca-1,3-diene (4g).** (*Z,Z*)-Isomer: white solid; mp 174-176 °C; IR (KBr) 1195, 1176  $\text{cm}^{-1}$ ; MS (MALDI)  $m/z$  536.73 ( $\text{M}^+-2$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.76 (3H, t,  $J = 7.1$  Hz), 0.91-1.14 (8H, m), 1.98-2.05 (2H, m), 6.12 (1H, dd,  $J = 24.6, 12.3$  Hz), 7.45-7.53 (12H, m), 7.64-7.74 (8H, m), 8.15 (1H, dd,  $J = 34.8, 12.3$  Hz), 8.27 (1H, dt,  $J = 37.4, 12.3$  Hz);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  13.8, 22.2, 28.6, 29.2 (d,  $J = 4.1$  Hz), 31.0, 35.7 (d,  $J = 11.5$  Hz), 124.0 (dd,  $J = 96.8, 1.6$  Hz), 124.4 (d,  $J = 6.6$  Hz), 128.5 (d,  $J = 7.6$  Hz) 130.9 (d,  $J = 9.8$  Hz), 131.6 (d,  $J = 2.4$  Hz), 131.7 (d,  $J = 9.8$  Hz), 131.8 (d,  $J = 3.3$  Hz), 132.5 (d,  $J = 102.6$  Hz), 133.8 (d,  $J = 105.0$  Hz), 140.9 (dd,  $J = 10.2, 4.5$  Hz), 141.0 (dd,  $J = 84.5, 2.1$  Hz), 144.7 (dd,  $J = 9.4, 2.9$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  22.32, 30.73. Anal. Calcd for  $\text{C}_{34}\text{H}_{36}\text{O}_2\text{P}_2$ : C, 75.82; H, 6.74. Found: C, 75.46; H, 6.71. (*E,Z*)-Isomer: colorless oil; IR (Neat) 1182  $\text{cm}^{-1}$ ; MS (MALDI)  $m/z$  538.13 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.81 (3H, t,  $J = 7.1$  Hz), 1.05-1.24 (6H, m), 1.32 (2H, quin,  $J = 7.5$  Hz), 2.20 (2H, dt,  $J = 12.7, 7.5$  Hz), 6.33 (1H, t,  $J = 16.7$  Hz), 6.91 (1H, dd,  $J = 34.9, 11.4$  Hz), 7.16 (1H, td,  $J = 16.7, 11.4$  Hz), 7.37-7.43 (8H, m), 7.48-7.55 (12H, m);  $^{13}\text{C}$  NMR



(CDCl<sub>3</sub>)  $\delta$  13.9, 22.3, 28.7, 29.4 (d,  $J = 3.3$  Hz), 31.2, 35.9 (d,  $J = 9.8$  Hz), 128.4 (d,  $J = 12.3$  Hz), 128.5, (d,  $J = 11.5$  Hz), 128.6 (br d,  $J = 103.3$  Hz) 131.3 (d,  $J = 97.6$  Hz), 131.4 (d,  $J = 9.8$  Hz), 131.5 (d,  $J = 9.9$  Hz), 131.8 (d,  $J = 6.5$  Hz), 132.4 (d,  $J = 102.6$  Hz), 141.5 (dd,  $J = 22.5, 5.6$  Hz), 142.7 (br d,  $J = 86.1$  Hz), 143.0 (dd,  $J = 9.5, 6.2$  Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>)  $\delta$  26.31, 28.25. Anal. Calcd for C<sub>34</sub>H<sub>36</sub>O<sub>2</sub>P<sub>2</sub>: C, 75.82; H, 6.74. Found: C, 75.64; H, 6.53. (*E,E*)-Isomer: white solid; mp 212-214 °C; IR (KBr) 1186 cm<sup>-1</sup>; MS (MALDI)  $m/z$  537.94 (M<sup>+</sup>-1); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.78 (3H, t,  $J = 7.2$  Hz), 0.96-1.17 (8H, m), 2.32-2.40 (2H, m), 6.53 (1H, dd,  $J = 22.6, 17.0$  Hz), 6.84 (1H, dd,  $J = 19.3, 11.2$  Hz), 7.37 (1H, td,  $J = 17.0, 11.2$  Hz), 7.46-7.57 (13H, m), 7.65-7.72 (7H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  13.9, 22.4, 28.9 (d,  $J = 9.8$  Hz), 29.3, 30.6, 31.2, 128.5 (d,  $J = 11.5$  Hz), 128.6 (d,  $J = 12.3$  Hz), 129.5 (d,  $J = 100.9$  Hz) 131.2 (d,  $J = 102.5$  Hz), 131.3 (d,  $J = 9.9$  Hz), 131.9 (d,  $J = 9.9$  Hz), 132.0 (d,  $J = 2.4$  Hz), 132.4 (d,  $J = 67.3$  Hz), 139.5 (dd,  $J = 20.5, 10.7$  Hz), 140.5 (dd,  $J = 18.4, 3.7$  Hz), 142.6 (br d,  $J = 91.9$  Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>)  $\delta$  23.77, 31.68.

**1,4-Bis(diphenylphosphinyl)octa-1,3-diene (4h).** (*Z,Z*)-Isomer: isolated as a mixture of (*Z,Z*) and (*E,E*)-isomer (52 : 48); white solid; IR (KBr) 1178 cm<sup>-1</sup>; MS (MALDI)  $m/z$  509.78 (M<sup>+</sup>-1); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.60 (3H, t,  $J = 7.2$  Hz), 0.94-1.10 (4H, m), 1.97-2.04 (2H, m), 6.12 (1H, dd,  $J = 25.3, 12.4$  Hz), 7.44-7.73 (20H, m), 8.14 (1H, dd,  $J = 34.6, 12.4$  Hz), 8.26 (1H, dt,  $J = 37.8, 12.4$  Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>)  $\delta$  22.41, 30.82. (*E,Z*)-Isomer: isolated as a mixture of (*E,Z*) and (*E,E*)-isomer (57 : 43); white solid; MS (MALDI)  $m/z$  510.25 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.71 (3H, t,  $J = 7.2$  Hz), 1.05-1.32 (4H, m), 2.16-2.23 (2H, m), 6.31 (1H, t,  $J = 17.3$  Hz), 6.89 (1H, dd,  $J = 34.8, 11.1$  Hz), 7.11 (1H, td,  $J = 17.3, 11.1$  Hz), 7.34-7.54 (16H, m), 7.63-7.70 (4H, m); <sup>31</sup>P NMR (CDCl<sub>3</sub>)  $\delta$  26.71, 28.41. Anal. Calcd for C<sub>32</sub>H<sub>32</sub>O<sub>2</sub>P<sub>2</sub>: C, 75.28; H, 6.32. Found: C, 75.43; H, 6.30. (*E,E*)-Isomer: <sup>1</sup>H NMR (CDCl<sub>3</sub>) (clearly assignable peaks)  $\delta$  0.66 (3H, t,  $J = 7.2$  Hz), 2.32-2.39 (2H, m), 6.53 (1H, dd,  $J = 22.4, 16.8$  Hz), 6.83 (1H, dd,  $J = 19.5, 11.2$  Hz), 7.36 (1H, td,  $J = 16.8, 11.2$  Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>)  $\delta$  23.71, 31.70.

**(Z,Z)-3,6-Bis(diphenylphosphinyl)-2,2,7,7-tetramethylocta-3,5-diene (4j).** White solid; IR (Nujol) 1188 cm<sup>-1</sup>; MS (EI) *m/z* 566 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.85 (18H, s), 7.35-7.53 (14H, m), 7.68 (4H, tm, *J* = 8.8 Hz), 7.90 (4H, tm, *J* = 8.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>) (clearly assignable peaks) δ 29.7, 36.7. Anal. Calcd for C<sub>36</sub>H<sub>40</sub>O<sub>2</sub>P<sub>2</sub>: C, 76.31; H, 7.11. Found: C, 76.80; H, 7.47.

**Dual Hydrophosphination of Aromatic Alkynes 9 with Phenylphosphine.** The reaction was carried out in a manner similar to that of the diynes **6** described above, using the aromatic alkynes **9** (2.0 mmol), PhPH<sub>2</sub> (1.0 mmol), and **1** or **2** (0.1 mmol) in THF (1 mL). In the reaction with aniline, this additive and PhPH<sub>2</sub> were successively added to a solution of **1** or **2**, and the mixture was stirred for 30 min at room temperature.

**Bis(β-methylstyryl)phenylphosphine Oxide (5a).** (*Z,Z*)-Isomer: IR (Nujol) 1178 cm<sup>-1</sup>; MS (EI) *m/z* 358 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.96 (6H, dd, *J* = 12.2, 1.5 Hz), 7.03 (2H, dd, *J* = 37.0, 1.5 Hz), 7.17-7.19 (6H, m), 7.28-7.40 (7H, m), 7.66-7.71 (2H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 23.9 (d, *J* = 13.1 Hz), 127.3, 127.9, 128.1, 128.4 (d, *J* = 50.9 Hz), 129.5 (d, *J* = 1.6 Hz), 131.1 (d, *J* = 3.2 Hz), 131.5 (d, *J* = 9.0 Hz), 132.5 (d, *J* = 99.9 Hz), 136.0 (d, *J* = 6.5 Hz), 144.6 (d, *J* = 6.5 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 27.52 Anal. Calcd for C<sub>24</sub>H<sub>23</sub>OP: C, 80.43; H, 6.47. Found: C, 80.29; H, 6.51. (*Z,E*)-Isomer: MS (EI) *m/z* 358 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.75 (3H, dd, *J* = 14.0, 1.0 Hz), 2.09 (3H dd, *J* = 12.0, 1.2 Hz), 7.05-7.49 (15H, m), 7.66-7.71 (2H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.4 (d, *J* = 11.4 Hz), 23.8 (d, *J* = 12.3 Hz), 127.3, 127.5, 127.9 (d, *J* = 12.6 Hz), 128.1, 128.2 (d, *J* = 11.5 Hz), 128.8 (d, *J* = 90.9 Hz), 129.2, 129.5 (d, *J* = 1.6 Hz), 131.0 (d, *J* = 81.1 Hz), 131.1 (d, *J* = 3.3 Hz), 131.3 (d, *J* = 21.5 Hz), 132.0 (d, *J* = 84.4 Hz), 135.8 (d, *J* = 5.7 Hz), 136.0 (d, *J* = 18.9 Hz), 141.3 (d, *J* = 9.8 Hz), 145.4 (d, *J* = 7.3 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 29.81. (*E,E*)-Isomer: MS (EI) *m/z* 358 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 2.17 (6H, dd, *J* = 13.2, 1.2 Hz), 7.25-7.57 (15H, m), 7.80-7.85 (2H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 15.0 (d, *J* = 9.8 Hz), 128.3, 128.4,

128.68, 128.71 (d,  $J = 29.5$  Hz), 129.4, 129.6 (d,  $J = 41.7$  Hz), 131.4 (d,  $J = 92.6$  Hz), 132.0 (d,  $J = 9.8$  Hz), 135.9 (d,  $J = 18.0$  Hz), 142.4 (d,  $J = 10.6$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  38.40.

**Bis(styryl)phenylphosphine Oxide (5b).** (*Z,Z*)-Isomer: white solid; mp 78-80 °C; IR (KBr) 1178  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  330 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  6.07 (2H, dd,  $J = 19.5, 14.0$  Hz), 7.18-7.32 (11H, m), 7.65-7.72 (6H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  123.9 (d,  $J = 100.0$  Hz), 127.8, 128.0, 129.1, 130.0, 130.7 (d,  $J = 9.8$  Hz), 131.0 (d,  $J = 2.5$  Hz), 133.7 (d,  $J = 107.5$  Hz), 135.2 (d,  $J = 7.4$  Hz), 148.0;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  15.03. Anal. Calcd for  $\text{C}_{22}\text{H}_{19}\text{OP}$ : C, 79.98; H, 5.80. Found: C, 79.87; H, 5.81. (*Z,E*)-Isomer: white solid; MS (EI)  $m/z$  330 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  6.22 (1H, dd,  $J = 19.0, 13.9$  Hz), 6.41 (1H, dd,  $J = 22.9, 17.4$  Hz), 7.11-7.54 (13H, m), 7.68-7.79 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  120.3 (d,  $J = 105.0$  Hz), 122.5 (d,  $J = 100.1$  Hz), 127.3, 127.8, 128.3, 128.5 (d,  $J = 9.0$  Hz), 129.2, 129.6, 129.8 (d,  $J = 1.6$  Hz), 130.4 (d,  $J = 9.8$  Hz), 131.3 (d,  $J = 3.3$  Hz), 133.5 (d,  $J = 107.5$  Hz), 134.9 (d,  $J = 7.4$  Hz), 135.1 (d,  $J = 18.0$  Hz), 146.0 (d,  $J = 3.3$  Hz), 148.9;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  17.56.

**Bis(4-methoxystyryl)phenylphosphine Oxide (5c).** (*Z,Z*)-Isomer: IR (Neat) 1180  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  390 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  3.76 (6H, s), 5.92 (2H, dd,  $J = 19.5, 14.0$  Hz), 6.75-6.79 (4H, m), 7.17 (2H, dd,  $J = 40.5, 14.0$  Hz), 7.26-7.34 (3H, m), 7.70-7.76 (6H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  55.2, 113.4, 121.1 (d,  $J = 100.9$  Hz), 128.07 (d,  $J = 18.9$  Hz), 128.09, 130.8 (d,  $J = 9.8$  Hz), 131.0 (d,  $J = 3.3$  Hz), 132.1 (d,  $J = 1.6$  Hz), 134.3 (d,  $J = 106.6$  Hz), 147.5 (d,  $J = 1.6$  Hz), 160.4;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  15.91. Anal. Calcd for  $\text{C}_{24}\text{H}_{23}\text{O}_3\text{P}$ : C, 73.83; H, 5.94. Found: C, 73.43; H, 5.98. (*Z,E*)-Isomer: MS (EI)  $m/z$  390 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  3.72 (3H, s), 3.80 (3H, s), 6.04 (1H, dd,  $J = 19.4, 13.8$  Hz), 6.30 (1H, dd,  $J = 22.6, 17.2$  Hz), 6.74-6.85 (4H, m), 7.29-7.48 (7H, m), 7.64-7.80 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  55.2, 55.3, 113.4, 114.1, 117.8 (d,  $J = 107.5$  Hz), 119.7 (d,  $J = 100.9$ ), 127.9 (d,  $J = 7.4$  Hz),

128.2, 128.4 (d,  $J = 12.3$  Hz), 129.1, 130.6 (d,  $J = 9.8$  Hz), 131.3 (d,  $J = 2.4$  Hz), 132.0, 134.2 (d,  $J = 107.5$  Hz), 145.5 (d,  $J = 4.2$  Hz), 148.4, 160.5, 160.9;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  18.57.

**Bis(4-methylstyryl)phenylphosphine Oxide (5d).** (*Z,Z*)-Isomer: white solid; mp 98-99 °C; IR (KBr) 1178  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  358 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  2.29 (6H, s), 6.00 (2H, dd,  $J = 19.8, 14.0$  Hz), 7.04-7.32 (10H, m), 7.61-7.72 (5H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  21.3, 122.8 (d,  $J = 100.9$  Hz), 127.9 (d,  $J = 11.5$  Hz), 128.7, 130.2 (d,  $J = 1.6$  Hz), 130.8 (d,  $J = 9.8$  Hz), 130.9 (d,  $J = 3.3$  Hz), 132.5 (d,  $J = 7.4$  Hz), 134.2 (d,  $J = 107.5$  Hz), 139.4, 147.9 (d,  $J = 1.6$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  15.29. Anal. Calcd for  $\text{C}_{24}\text{H}_{23}\text{OP}$ : C, 80.43; H, 6.47. Found: C, 80.21; H, 6.43. (*Z,E*)-Isomer: MS (EI)  $m/z$  358 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  2.24 (3H, s), 2.33 (3H, s), 6.14 (1H, dd,  $J = 19.0, 14.0$  Hz), 6.35 (1H, dd,  $J = 22.8, 17.2$  Hz), 6.92-7.79 (15H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  21.2, 21.3, 119.3 (d,  $J = 105.8$  Hz), 121.5 (d,  $J = 100.1$  Hz), 127.4, 128.4 (d,  $J = 12.3$  Hz), 128.7, 129.3, 130.1, 130.6 (d,  $J = 9.8$  Hz), 131.3 (d,  $J = 3.3$  Hz), 132.4 (d,  $J = 7.4$  Hz), 132.7 (d,  $J = 18.9$  Hz), 134.0 (d,  $J = 107.5$  Hz), 139.5, 139.9, 145.9 (d,  $J = 3.3$  Hz), 148.9;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  18.09.

**Bis(4-chlorostyryl)phenylphosphine Oxide (5e).** (*Z,Z*)-Isomer: white solid; mp 111-112 °C; IR ( $\text{CHCl}_3$ ) 1175  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  398 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  6.09 (2H, dd,  $J = 19.6, 14.0$  Hz), 7.15-7.39 (9H, m), 7.64-7.68 (6H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  124.4 (d,  $J = 100.0$  Hz), 128.1, 128.3, 130.7 (d,  $J = 9.8$  Hz), 131.39, 131.44 (d,  $J = 3.2$  Hz), 133.2 (d,  $J = 106.5$  Hz), 133.5 (d,  $J = 8.2$  Hz), 135.3, 147.0;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.78. Anal. Calcd for  $\text{C}_{22}\text{H}_{17}\text{Cl}_2\text{OP}$ : C, 66.18; H, 4.29. Found: C, 66.12; H, 4.01. (*Z,E*)-Isomer: white solid; mp 150-151 °C; IR ( $\text{CHCl}_3$ ) 1180  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  398 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  6.25 (1H, dd,  $J = 19.1, 13.8$  Hz), 6.40 (1H, dd,  $J = 22.7, 17.1$  Hz), 7.20-7.50 (11H, m), 7.64-7.77 (4H, m);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  17.21. Anal. Calcd for  $\text{C}_{22}\text{H}_{17}\text{Cl}_2\text{OP}$ : C, 66.18; H, 4.29. Found: C, 66.19; H, 4.16.

**Bis(4-bromostyryl)phenylphosphine Oxide (5f).** (*Z,Z*)-Isomer: white solid; mp 117-118 °C; IR (CHCl<sub>3</sub>) 1180 cm<sup>-1</sup>; MS (EI) *m/z* 486 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 6.10 (2H, dd, *J* = 19.7, 14.0 Hz), 7.20 (2H, dd, *J* = 40.1, 14.0 Hz), 7.27-7.39 (7H, m), 7.58-7.68 (6H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 123.8, 124.6 (d, *J* = 99.2 Hz), 128.2 (d, *J* = 12.3 Hz), 130.7 (d, *J* = 9.8 Hz), 131.2, 131.5 (d, *J* = 2.5 Hz), 131.6 (d, *J* = 1.2 Hz), 133.2 (d, *J* = 107.3 Hz), 134.0 (d, *J* = 7.4 Hz), 147.1; <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 12.14. Anal. Calcd for C<sub>22</sub>H<sub>17</sub>Br<sub>2</sub>OP: C, 54.13; H, 3.51. Found: C, 54.18; H, 3.51. (*Z,E*)-Isomer: white solid; MS (EI) *m/z* 486 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 6.25 (1H, dd, *J* = 19.1, 14.0 Hz), 6.40 (1H, dd, *J* = 23.0, 17.4 Hz), 7.19-7.76 (15H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 121.0 (d, *J* = 104.9 Hz), 123.4 (d, *J* = 99.2 Hz), 124.1, 128.7 (d, *J* = 12.2 Hz), 128.9, 129.1, 130.5 (d, *J* = 10.7 Hz), 131.2, 131.6, 131.8, 132.0, 133.1 (d, *J* = 112.0 Hz), 133.98, 134.03 (d, *J* = 25.4 Hz), 145.1 (d, *J* = 3.3 Hz), 147.8; <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 14.61.

**3,6-Bis(diphenylphosphinyl)-2,2,7,7-tetramethylocta-3,4-diene (7).** White solid; mp 202-204 °C; IR (KBr) 1203, 1186 cm<sup>-1</sup>; MS (MALDI) *m/z* 564.01 (M<sup>+</sup>-2); MS *m/z* 566 (M<sup>+</sup>), 509 (M<sup>+</sup>-<sup>t</sup>Bu), 452 (M<sup>+</sup>-2<sup>t</sup>Bu), 365 (M<sup>+</sup>-Ph<sub>2</sub>PO), 308 (365-<sup>t</sup>Bu<sup>+</sup>), 201 (Ph<sub>2</sub>PO<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.16 (9H, s), 1.33 (9H, s), 2.81 (1H, ddd, *J* = 7.7, 5.9, and 4.3 Hz), 5.30 (1H, ddd, *J* = 16.7, 11.6, and 7.7 Hz), 6.94-7.76 (20H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 29.8 (d, *J* = 5.7 Hz), 29.9 (d, *J* = 4.1 Hz), 37.5 (d, *J* = 6.6 Hz), 37.8 (d, *J* = 5.7 Hz), 45.3 (dd, *J* = 70.6 and 4.9 Hz), 90.6 (d, *J* = 13.9 Hz), 107.8 (d, *J* = 96.8 Hz), 127.9 (d, *J* = 11.5 Hz), 128.1 (d, *J* = 100.9 Hz), 128.7, 130.2 (d, *J* = 1.6 Hz), 130.8 (d, *J* = 9.8 Hz), 130.9 (d, *J* = 12.3 Hz), 128.3 (d, *J* = 11.5 Hz), 128.4 (d, *J* = 11.5 Hz), 130.0 (d, *J* = 9.1 Hz), 130.5 (d, *J* = 8.2 Hz), 130.7 (d, *J* = 2.5 Hz), 131.1 (d, *J* = 3.3 Hz), 131.4 (d, *J* = 2.5 Hz), 131.5 (d, *J* = 3.3 Hz), 131.6 (d, *J* = 9.0 Hz), 132.0 (d, *J* = 10.7 Hz), 132.9 (d, *J* = 100.9 Hz), 134.0 (d, *J* = 96.0 Hz), 135.6 (d, *J* = 95.2 Hz), 137.8 (d, *J* = 108.3 Hz), 207.1 (d, *J* = 26.3 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 28.78 (d, *J* = 5.9 Hz), 31,68 (d, *J* = 5.9 Hz). Anal. Calcd for C<sub>36</sub>H<sub>40</sub>O<sub>2</sub>P<sub>2</sub>: C, 76.31; H, 7.11. Found: C, 76.53; H, 7.11.

**1-Diphenylphosphinyl-1-phenyldec-1-en-3-yne (8d).** Isolated as a mixture of (*Z*) and (*E*)-isomer (91 : 9); yellow oil; IR (Neat) 1190 cm<sup>-1</sup>; MS (EI) *m/z* 412 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ (*Z*)-isomer : 0.84 (3H, t, *J* = 6.9 Hz), 1.14-1.56 (8H, m), 2.53-2.61 (2H, m), 6.52 (1H, d, *J* = 36.2 Hz), 6.87 (2H, d, *J* = 7.0 Hz), 7.15-7.57 (9H, m), 7.70-7.86 (4H, m); (*E*)-isomer: 6.22 (d, *J* = 18.1 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ (*Z*)-isomer: 14.0, 22.4, 28.7, 29.4 (d, *J* = 3.2 Hz), 31.5, 34.9 (d, *J* = 8.2 Hz), 86.7 (d, *J* = 10.6 Hz), 99.8 (d, *J* = 1.6 Hz), 120.8 (d, *J* = 6.5 Hz), 127.9, 128.3 (d, *J* = 12.3 Hz), 128.5, 131.3, 131.73, 131.76 (d, *J* = 10.7 Hz), 131.8, 132.5 (d, *J* = 103.4 Hz), 146.8 (d, *J* = 90.5 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ (*Z*)-isomer: 27.76, (*E*)-isomer: 31.77. Anal. Calcd for C<sub>28</sub>H<sub>29</sub>OP: C, 81.53; H, 7.09. Found: C, 81.56; H, 7.22.

**(*Z*)-1-Diphenylphosphinyldec-1-en-3-yne (8g).** Yellow oil; IR (Neat) 1188 cm<sup>-1</sup>; MS (EI) *m/z* 336 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.86 (3H, t, *J* = 7.1 Hz), 1.15-1.27 (8H, m), 1.97-2.00 (2H, m), 6.47 (1H, dd, *J* = 18.7, 13.2 Hz), 6.53 (1H, ddt, *J* = 35.8, 13.2, 2.4 Hz), 7.43-7.53 (6H, m), 7.76-7.81 (4H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.0, 19.6, 22.4, 27.7, 28.4, 31.2, 77.3 (d, *J* = 12.3 Hz), 104.8, 128.3 (d, *J* = 12.3 Hz), 128.4 (d, *J* = 1.6 Hz), 131.26 (d, *J* = 9.8 Hz), 131.30 (d, *J* = 99.3 Hz), 131.6 (d, *J* = 3.5 Hz), 133.0 (d, *J* = 105.8 Hz); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 21.46. Anal. Calcd for C<sub>22</sub>H<sub>25</sub>OP: C, 78.55; H, 7.49. Found: C, 78.60; H, 7.58.

**1,2,3,4,5-Pentaphenyl-1-phosphacyclopent-2-ene 1-Oxide (10).** Isolated as a single isomer with unknown stereochemistry; white solid; mp 295-297 °C; IR (KBr) 1215 cm<sup>-1</sup>; MS (MALDI) *m/z* 482.60 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 3.98 (1H, dd, *J* = 10.1, 8.2 Hz), 4.96 (1H, dd, *J* = 27.3, 8.2 Hz), 7.01-7.39 (23H, m), 7.67-7.72 (2H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 51.0 (d, *J* = 69.7 Hz), 59.7 (d, *J* = 7.3 Hz), 126.7, 127.0, 127.4, 127.7, 128.1, 128.27, 128.34, 128.40 (d, *J* = 10.6 Hz), 128.85, 128.90, 129.7, 130.8 (d, *J* = 7.5 Hz), 131.5 (d, *J* = 9.8 Hz), 131.6 (d, *J* = 2.4 Hz), 133.1 (d, *J* = 119.6 Hz), 133.7 (d, *J* = 1.0 Hz),

133.9, 134.0, 136.1 (d,  $J = 92.6$  Hz), 136.6 (d,  $J = 15.5$  Hz), 137.2, 157.9 (d,  $J = 25.4$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  52.06. Anal. Calcd for  $\text{C}_{34}\text{H}_{27}\text{OP}$ : C, 84.63; H, 5.64. Found: C, 84.44; H, 5.68.

**Di(decyl)phenylphosphine Oxide (11).** White solid; mp 56-57 °C; IR (KBr) 1165  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  406 ( $\text{M}^+$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.86 (6H, t,  $J = 6.7$  Hz), 1.21-2.17 (36H, m), 7.47-7.52 (3H, m), 7.66-7.71 (2H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.1, 21.4 (d,  $J = 4.1$  Hz), 22.7, 29.1, 29.3, 29.4, 29.5 30.0 (d,  $J = 68.1$  Hz), 31.0 (d,  $J = 13.8$ ), 31.9, 128.6 (d,  $J = 11.4$  Hz), 130.4 (d,  $J = 8.2$  Hz), 131.4 (d,  $J = 3.2$  Hz), 132.3 (d,  $J = 91.8$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  40.72. Anal. Calcd for  $\text{C}_{26}\text{H}_{47}\text{OP}$ : C, 76.80; H, 11.65. Found: C, 76.84; H, 11.65.

## References and Notes

- (1) For a review, see: Minami, T.; Okauchi, T. In *Comprehensive Organic Functional Group Transformations II*; Katritzky, A. R., Taylor, R. J. K., Eds.; Elsevier: Oxford, 2005; Vol. 2, Chapter 2.16, pp 853-907.
- (2) (a) Han, L.-B.; Tanaka, M. *J. Am. Chem. Soc.* 1996, *118*, 1571-1572. (b) Reichwein, J. F.; Patel, M. C.; Pagenkopf, B. L. *Org. Lett.* 2001, *3*, 4303-4306. (c) Deprele, S.; Montchamp, J.-L. *J. Am. Chem. Soc.* 2002, *124*, 9386-9387. (d) For a recent example, see: Han, L.-B.; Zhang, C.; Yazawa, H.; Shimada, S. *J. Am. Chem. Soc.* 2004, *126*, 5080-5081, and references therein.
- (3) (a) Ohmiya, H.; Yorimitsu, H. Oshima, K. *Angew. Chem. Int. Ed.* 2005, *44*, 2368-2370. (b) Kazankova, M. A.; Efimova, I. V.; Kochetkov, A. N.; Afanas'ev, V. V.; Beletskaya, I. P.; Dixneul, P. H. *Synlett* 2001, 497-500. This method was, however, applicable to olefin hydrophosphination, see: (c) Shulyupin, M. O.; Kazankova, M. A.; Beletskaya, I. P. *Org. Lett.* 2002, *4*, 761-763. (d) Kazankova, M. A.; Shulyupin, M. O.; Beletskaya, I. P. *Synlett* 2003, 2155-2158.

- (4) (a) Douglass, M. R.; Marks, T. J. *J. Am. Chem. Soc.* 2000, *122*, 1824-1825. (b) Douglass, M. R.; Stern, C. L.; Marks, T. J. *J. Am. Chem. Soc.* 2001, *123*, 10221-10238. (c) Douglass, M. R.; Ogasawara, M.; Hong, S.; Metz, M. V.; Marks, T. J. *Organometallics* 2002, *21*, 283-292.
- (5) (a) Takaki, K.; Takeda, M.; Koshoji, G.; Shishido, T.; Takehira, K. *Tetrahedron Lett.* 2001, *42*, 6357-6360. (b) Takaki, K.; Koshoji, G.; Komeyama, K.; Takeda, M.; Shishido, T.; Kitani, A.; Takehira, K. *J. Org. Chem.* 2003, *68*, 6554-6565.
- (6) Komeyama, K.; Kawabata, T.; Takehira, K.; Takaki, K. *J. Org. Chem.* 2005, *70*, 7260-7266.
- (7) The original phosphine products were isolable, but oxidized partially during the work-up and purification. Thus, they were isolated and characterized as phosphine oxides.
- (8) Although the mechanism leading to **10** is unknown at present, it may be formed by cyclization of (*Z*)-phosphinoalkenylytterbium species, (*Z*)-isomer corresponding to G depicted in Scheme 3.
- (9) Mitchell, T. N.; Heesche, K. *J. Organomet. Chem.* 1991, *409*, 163-170.
- (10) (a) Komeyama, K.; Takehira, K.; Takaki, K. *Synthesis* 2004, 1062-1066. (b) Komeyama, K.; Sasayama, D.; Kawabata, T.; Takehira, K.; Takaki, K. *Chem. Commun.* 2005, 634-636.
- (11) Bradley, D. C.; Ghotra, J. S.; Hart, F. A. *J. Chem. Soc., Dalton Trans.* 1973, 1021-1023.
- (12) Sonogashira, K. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon Press: Oxford, 1991; Vol. 3, Chapters 2.4 and 2.5, pp 521-561.
- (13) Alami, M.; Ferri, F. *Tetrahedron Lett.* 1996, *37*, 2763-2766.
- (14) (a) Dabdoub, M. J.; Dabdoub, V. B. *Tetrahedron* 1995, *51*, 9839-9850. (b) Dabdoub, M. J.; Baroni, A. C. M.; Lenardao, E. L.; Gianeti, T. R.; Hurtado, G. R. *Tetrahedron* 2001, *57*, 4271-4276.