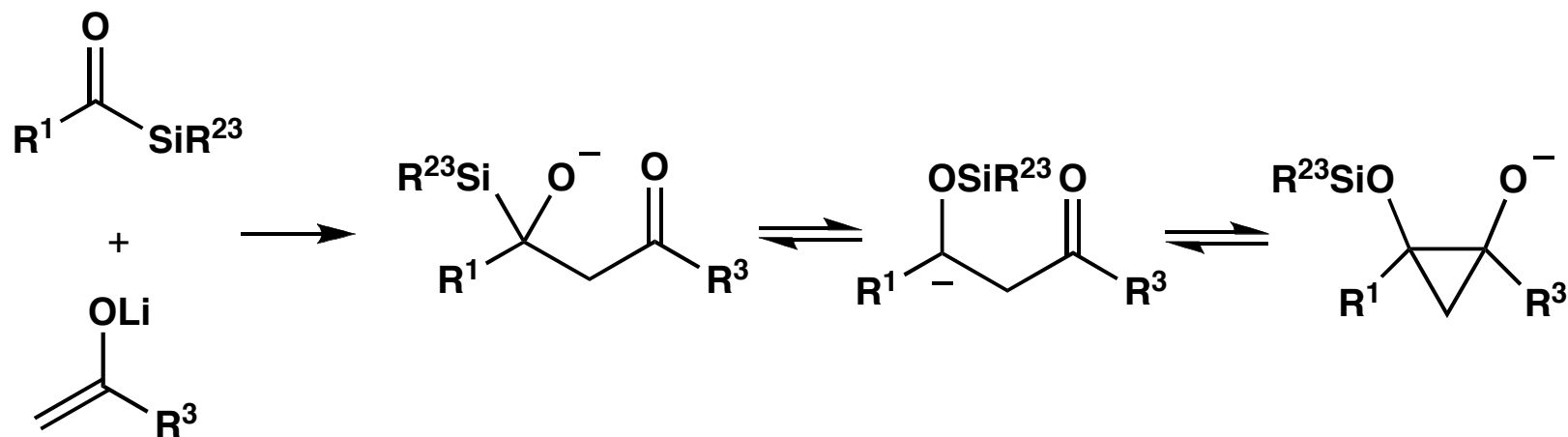


Brook, A. G.; Limburg, W. W.; MacRae, D. M.; Fieldhouse, S. A. *J. Am. Chem. Soc.* **1967**, *89*, 704.

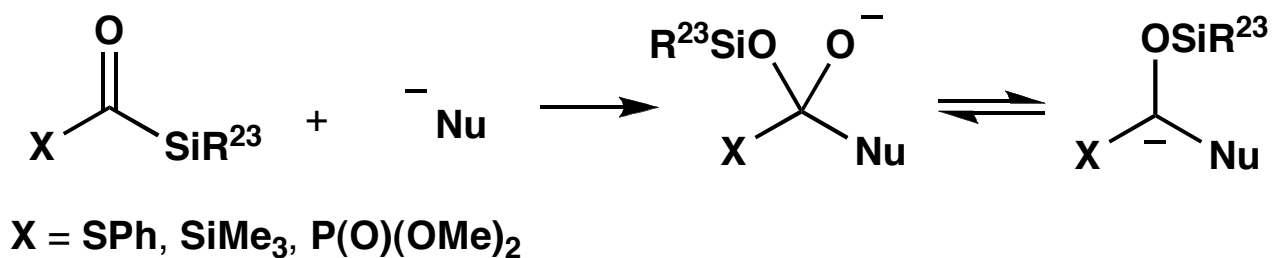
Reich, H. G.; Holtan, R. C.; Bolm, C. *J. Am. Chem. Soc.* **1990**, *112*, 5609-5617.

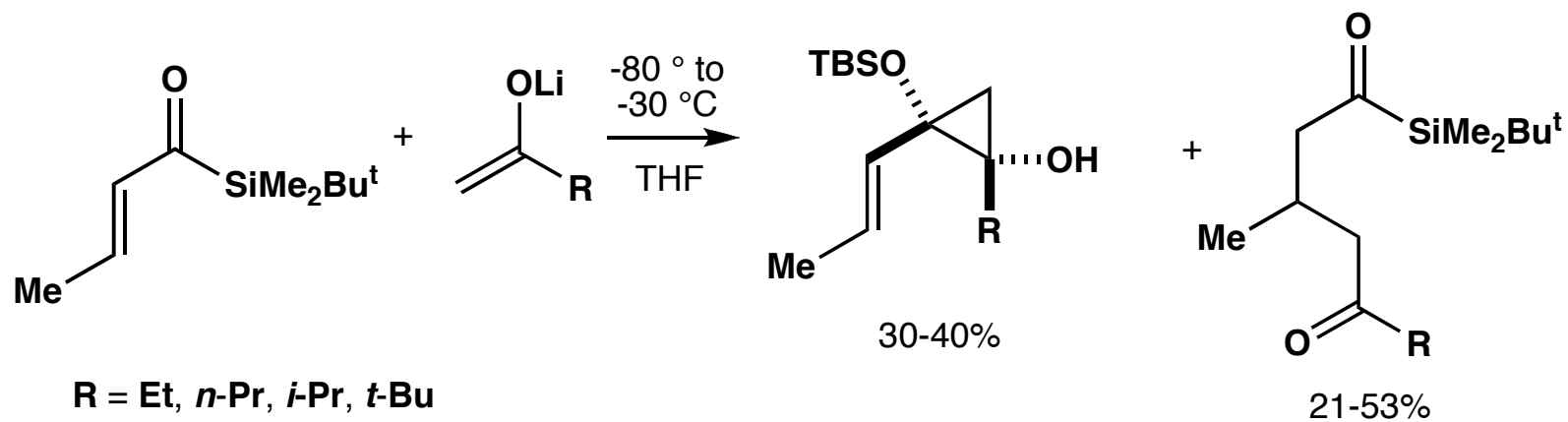
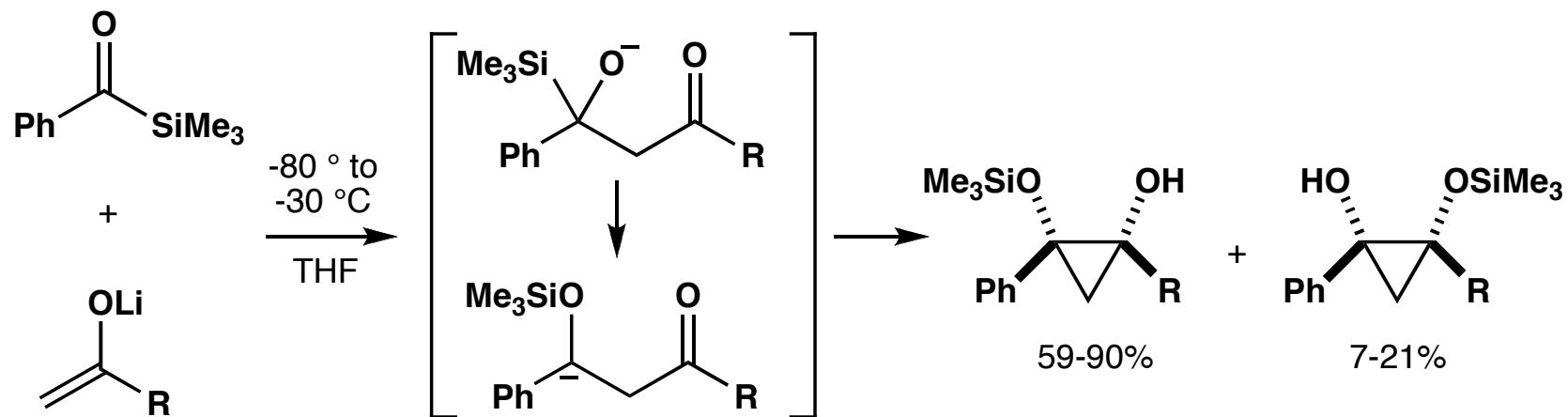
Nakajima, T.; Segi, M.; Sugimoto, F.; Hioki, R.; Yokota, S.; Miyashita, K. *Tetrahedron* **1993**, *37*, 8343.

1. Use of Ketone Enolate as a Nucleophile

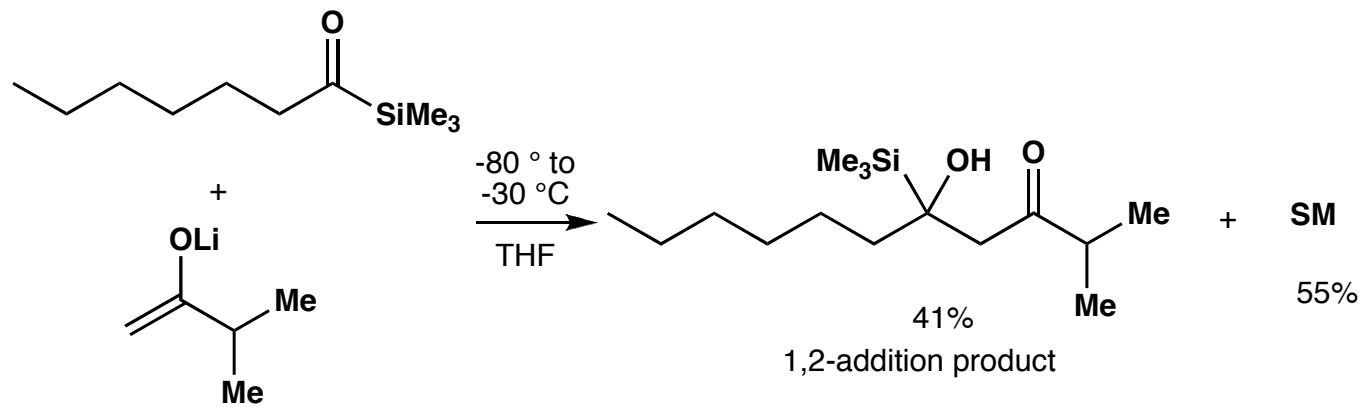
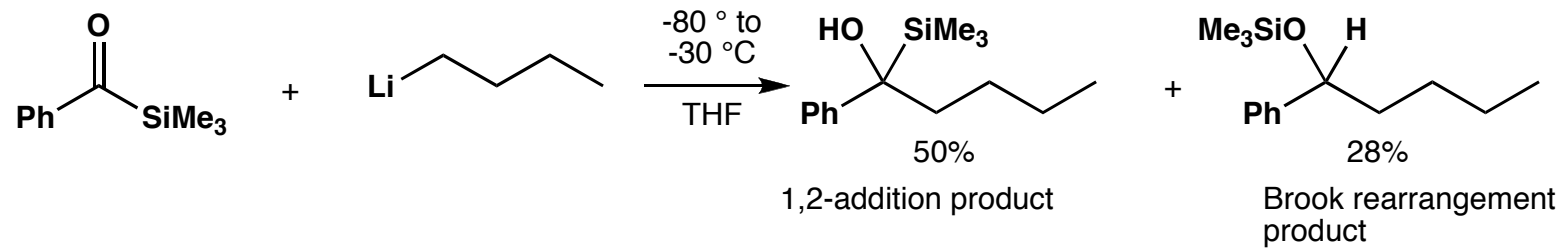


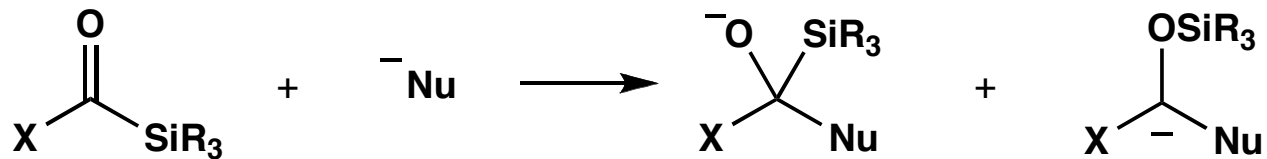
2. Introduction of Carbanion-Stabilizing Heteroatom



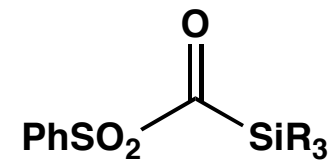
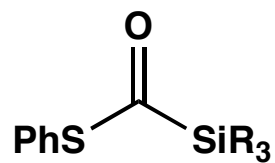
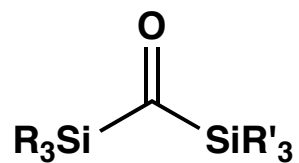
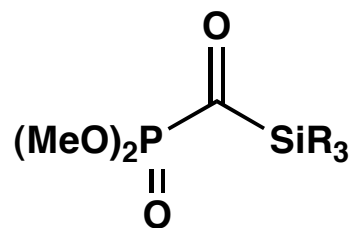


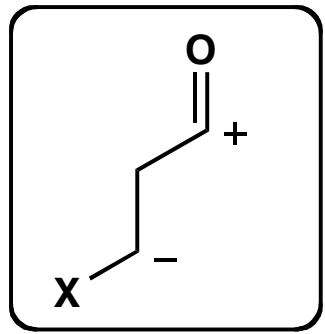
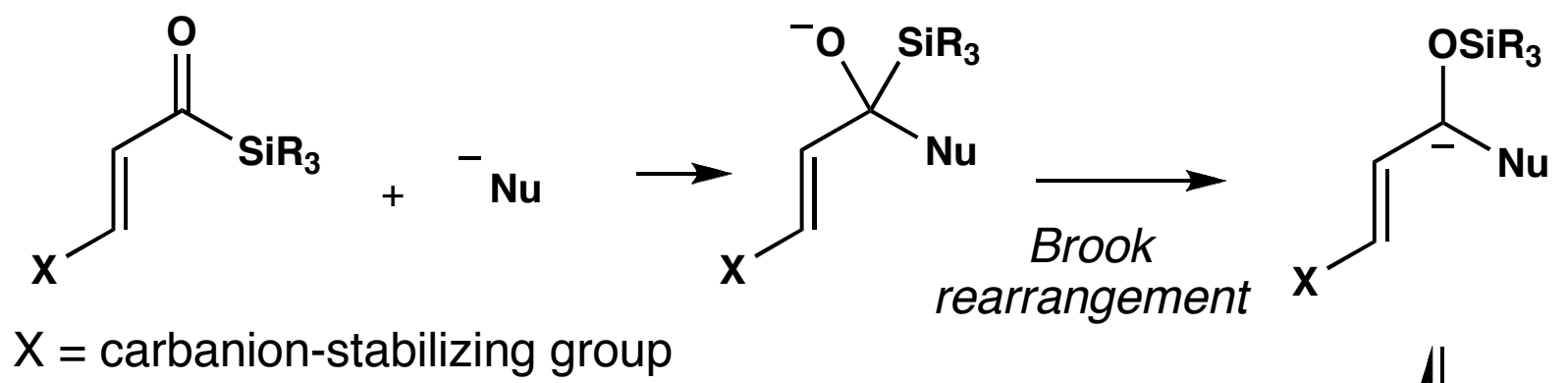
Takeda, K.; Nakatani, J.; Nakamura, H.; Sako, K.; Yoshii, E.; Yamaguchi, K. *Synlett* **1993**, 8



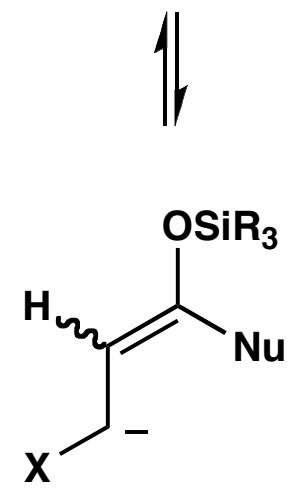


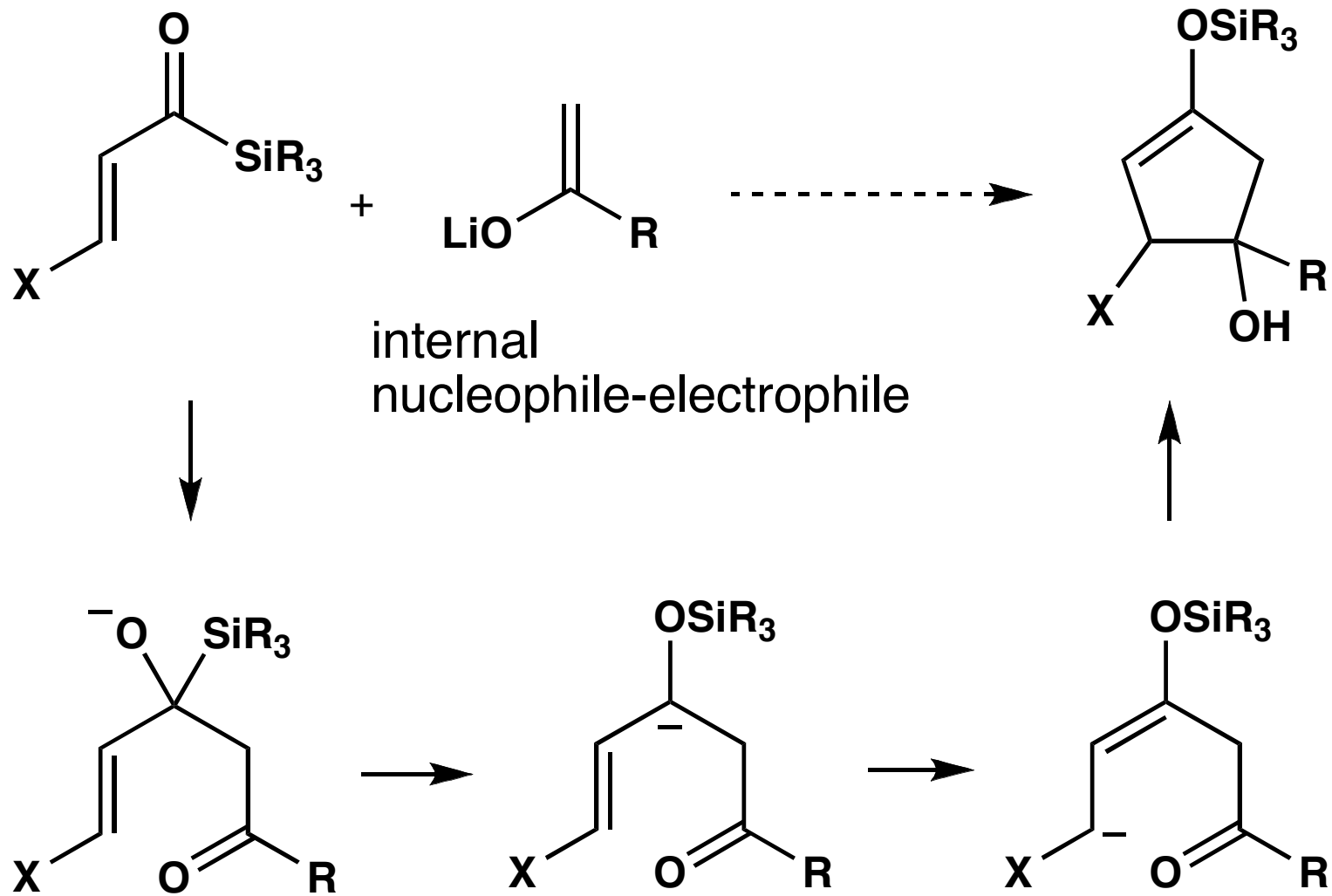
X = anion-stabilizing heteroatom



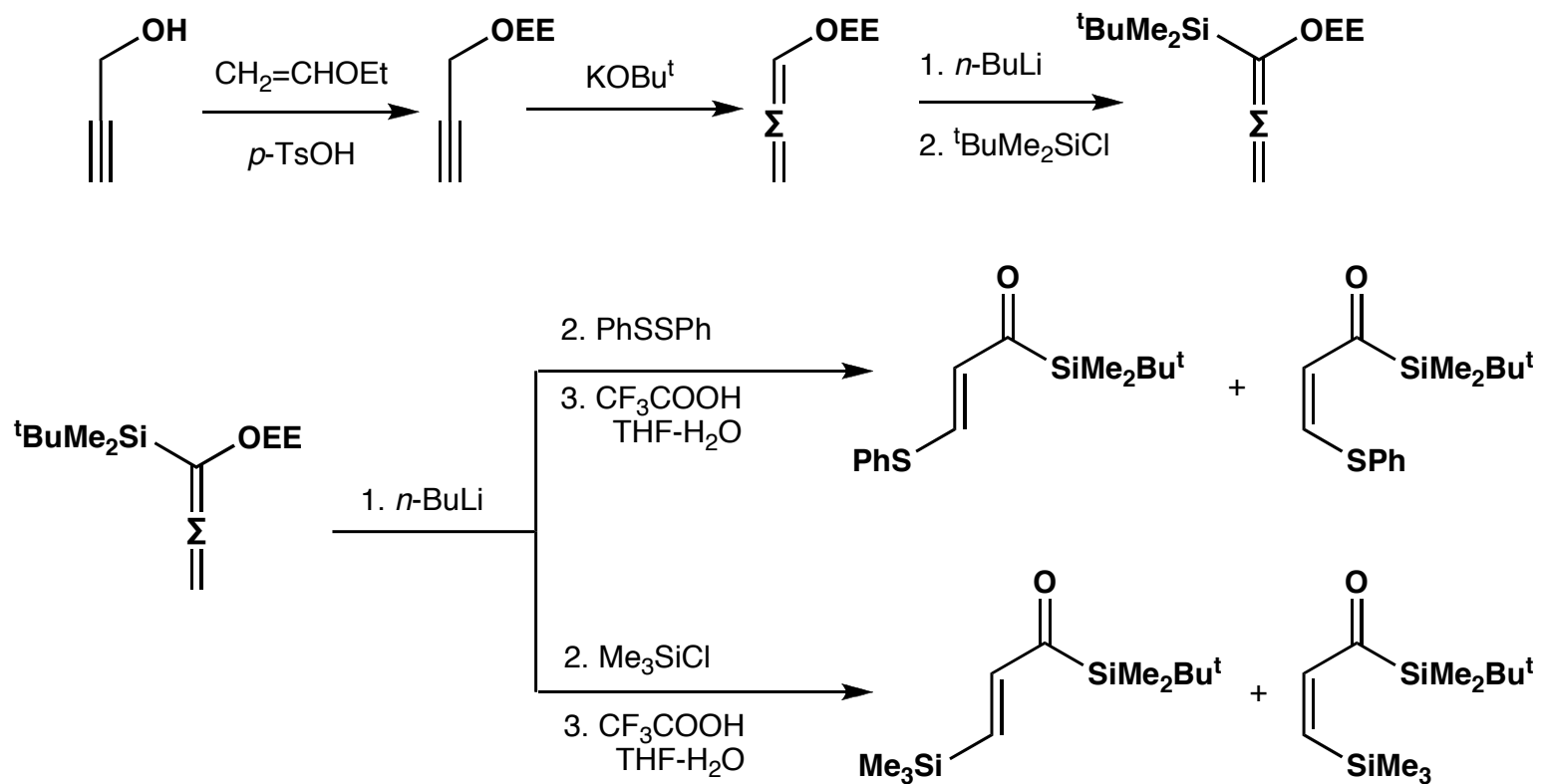


1,3-dipole



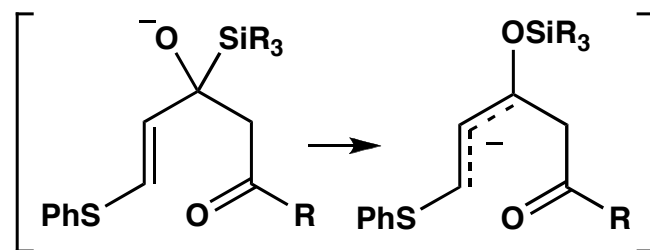
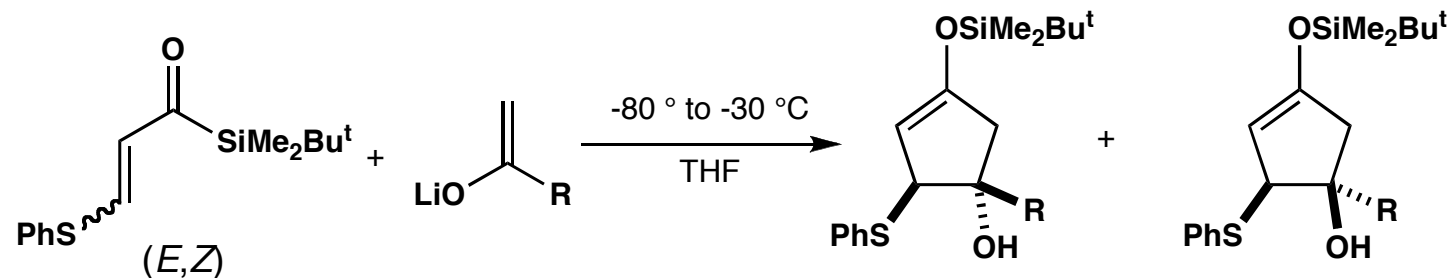


Preparation of b-(Phenylthio)- and b-(Trimethylsilyl)-Acryloylsilanes



Reich, H. J.; Kelly, M. J.; Olson, R. E.; Holtan, R. C. *Tetrahedron* **1983**, *39*, 949-960

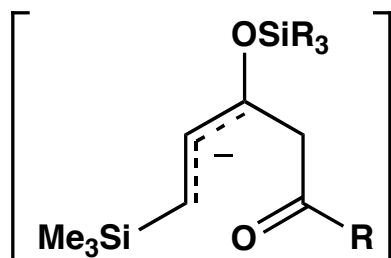
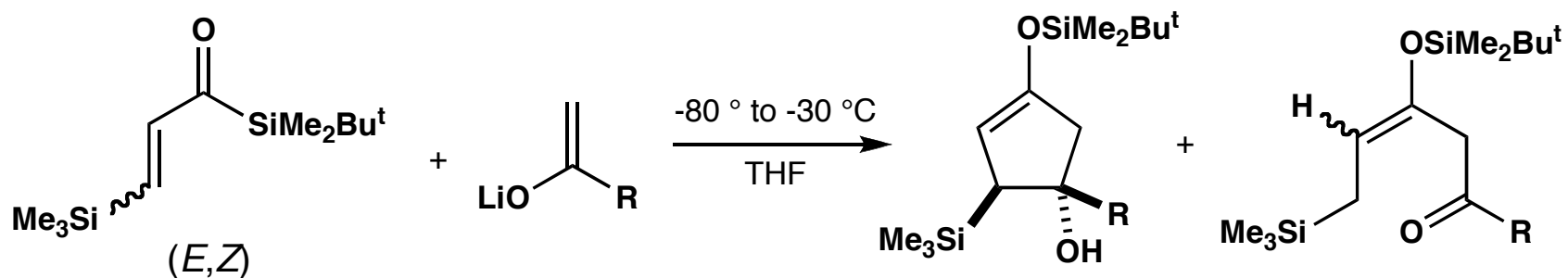
[3 + 2] Annulation Using Reaction of (b-Phenylthio)acryloyl)silanes and Lithium Enolates



Et	70%	5%
Pr	74%	7%
<i>i</i>-Pr	55%	19%
<i>n</i>-octyl	71%	8%

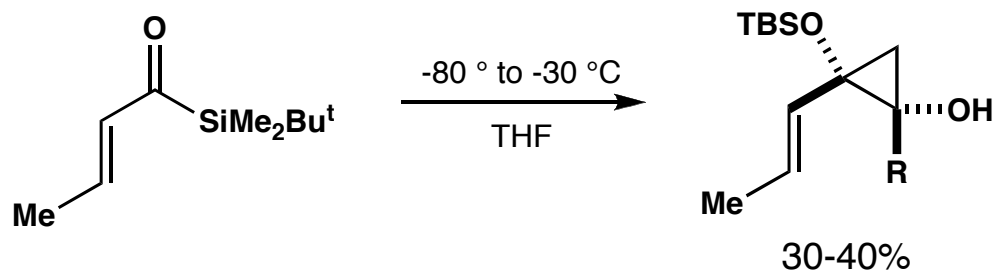
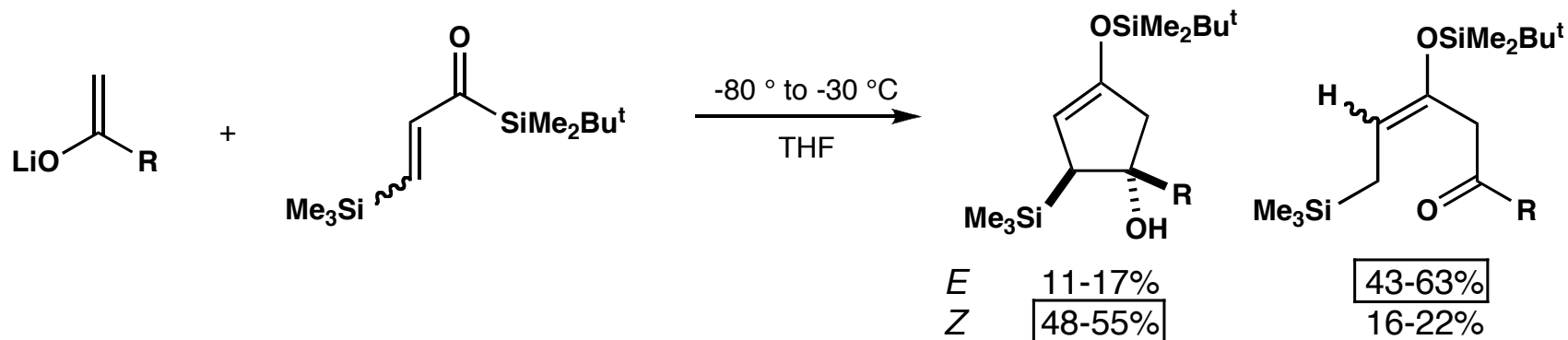
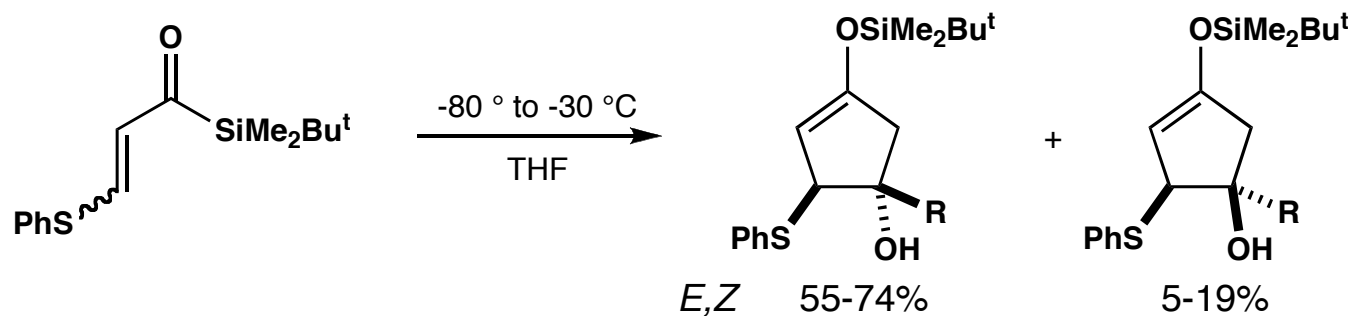
Takeda, K.; Fujisawa, Makino, T.; Yoshii, E.; Yamaguchi, K. *J. Am. Chem. Soc.* **1993**, *115*, 9351-9352.

[3 + 2] Annulation Using Reaction of (b-Trimethylsilyl)acryloyl)silane and Lithium Enolates

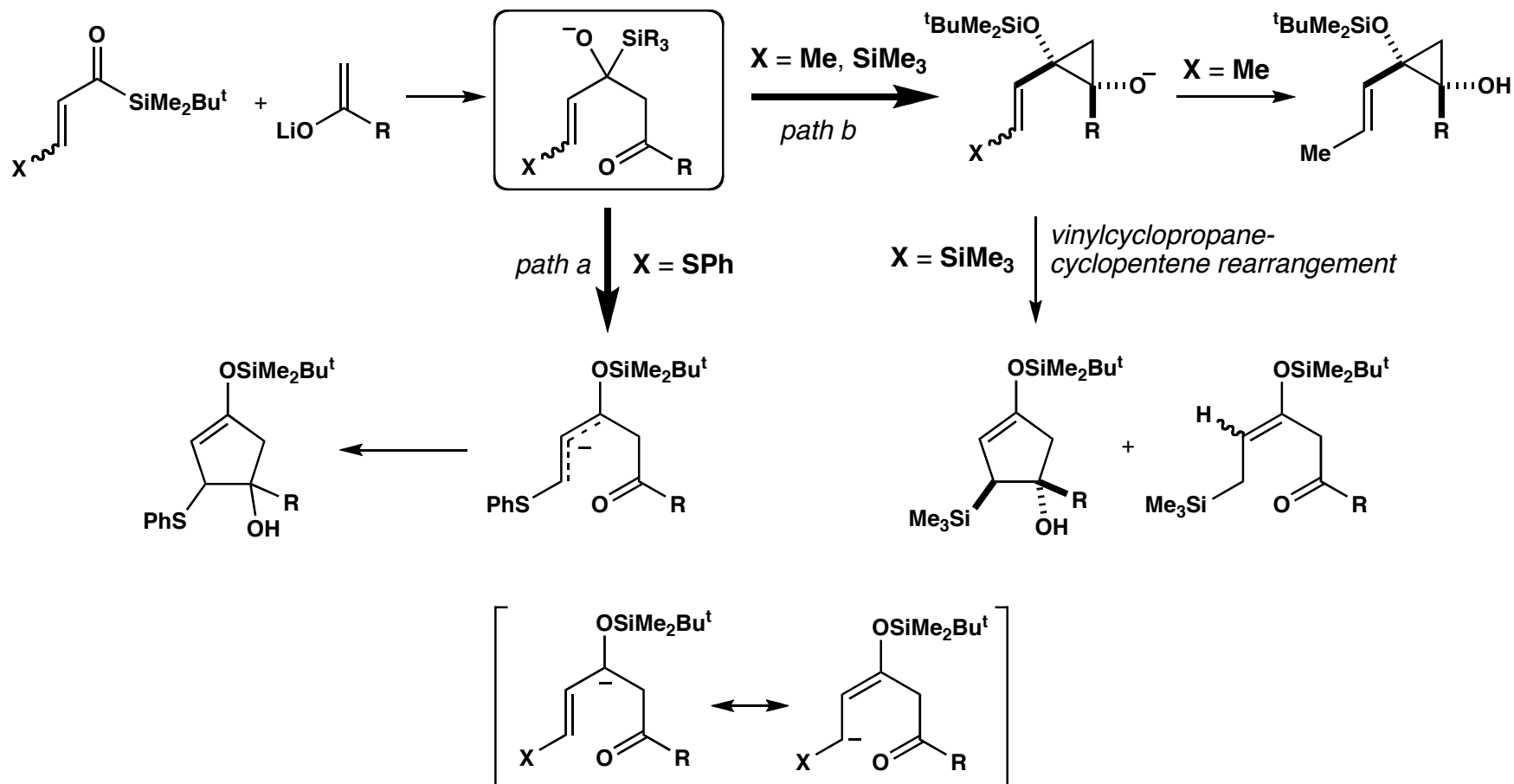


<i>E</i>	Et	17%	43%
	<i>n</i> -Pr	11%	
	<i>i</i> -Pr	14%	
<i>Z</i>	Et	48%	16%
	<i>n</i> -Pr	55%	22%
	<i>i</i> -Pr	51%	21%

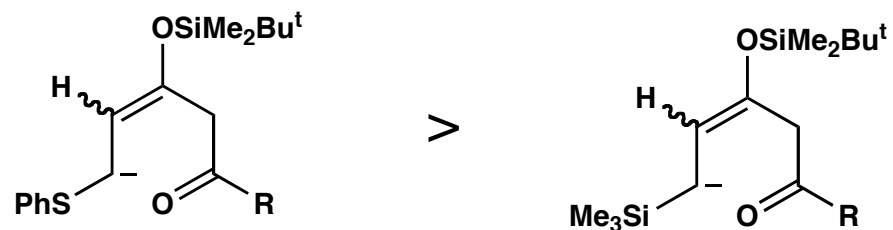
Summary of the Reaction of β -Substituted-Acryloylsilanes with Lithium Enolate of Methyl Ketones



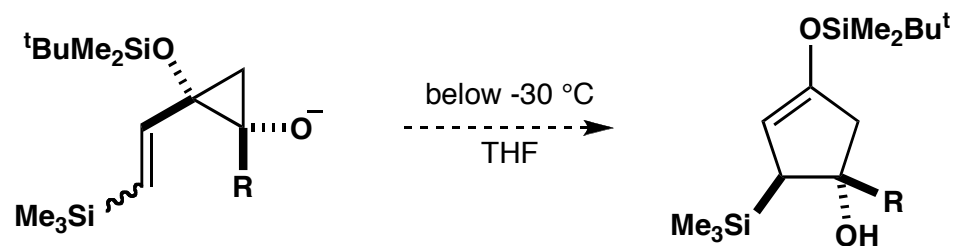
A Proposed Reaction Pathway for the [3 + 2] Annulation Using *b*-Phenylthio- and *b*-Trimethylsilyl-Acryloylsilanes

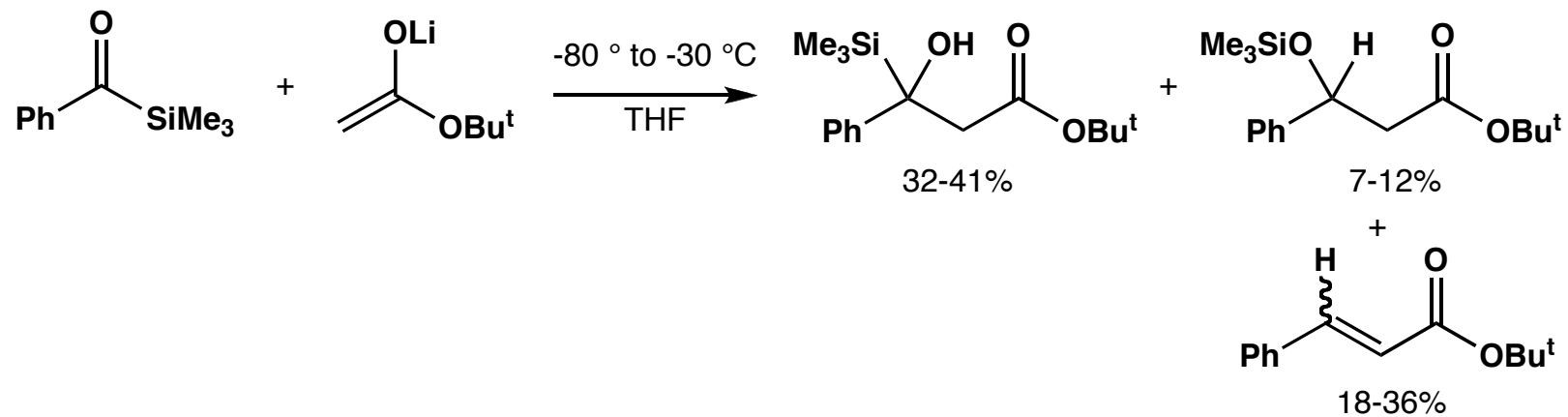
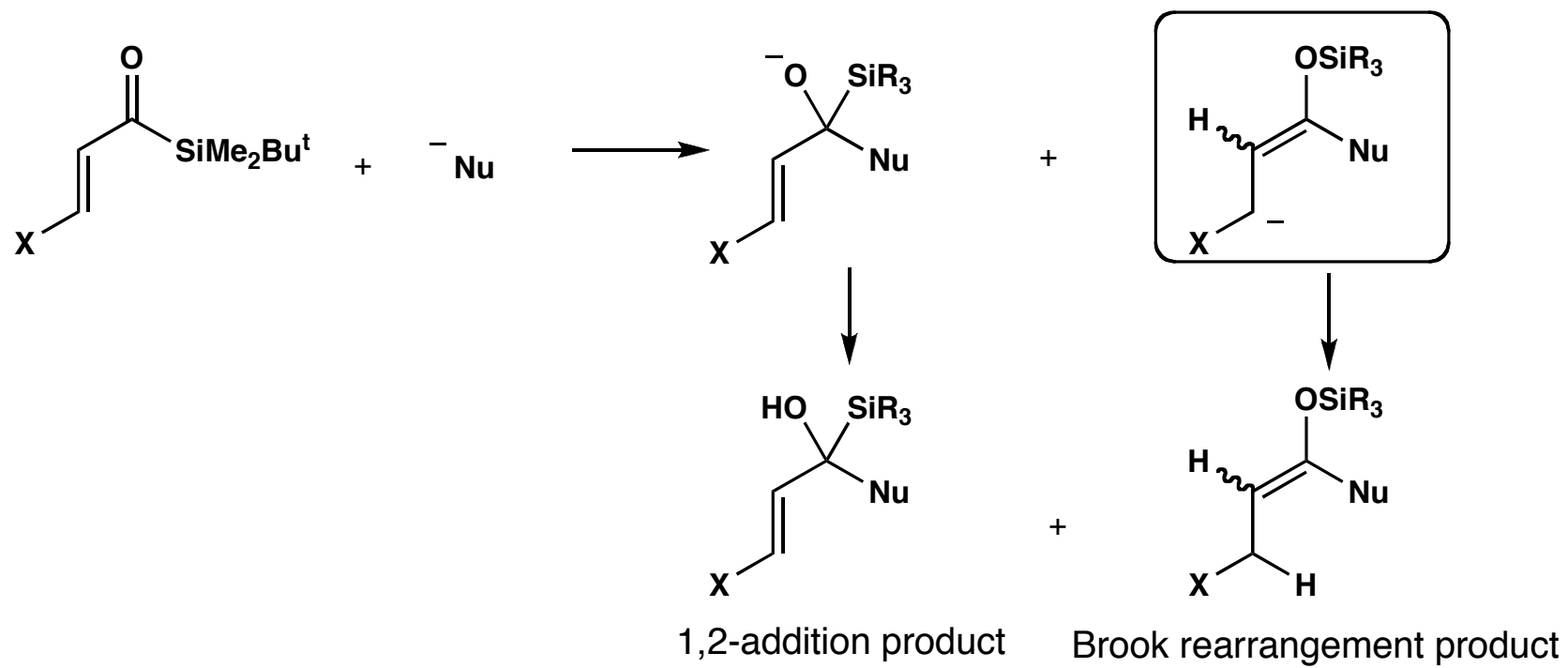


1. Does the phenylthio group stabilize the α -carbanion more strongly than the trimethylsilyl group?

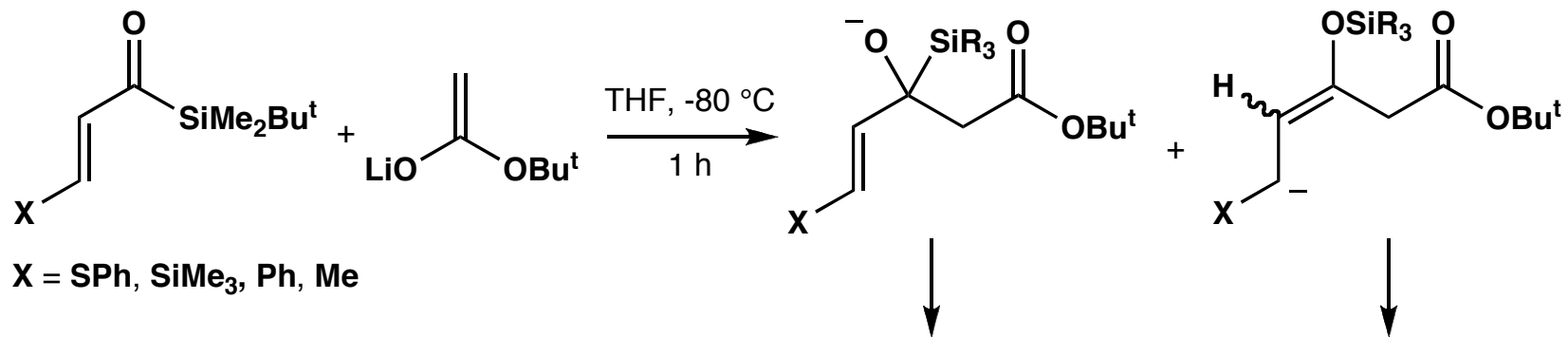


2. Does the oxyanion accelerated vinylcyclopropane rearrangement occur at low temperatures below -30 °C?



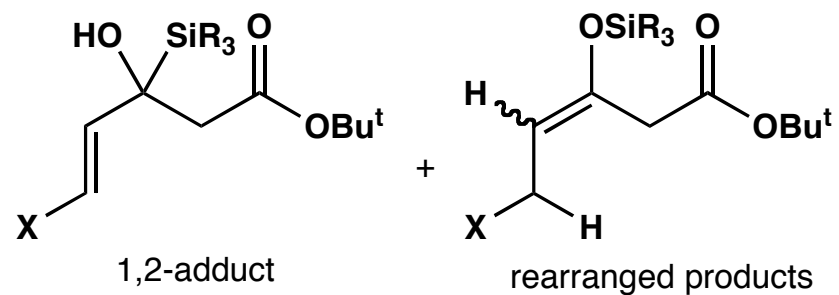


Reaction of β -Substituted-Acryloxy-silanes with Lithium Enolate of *t*-Butyl Acetate

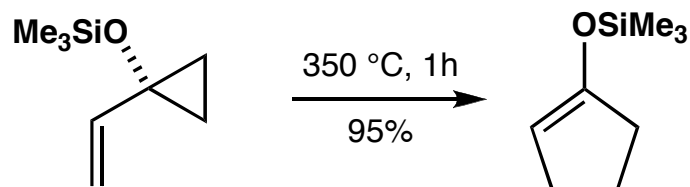
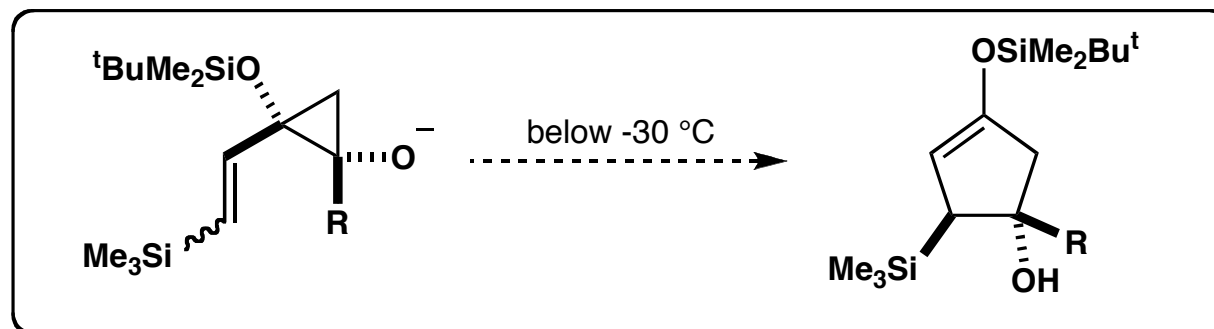


X = SPh, SiMe₃, Ph, Me

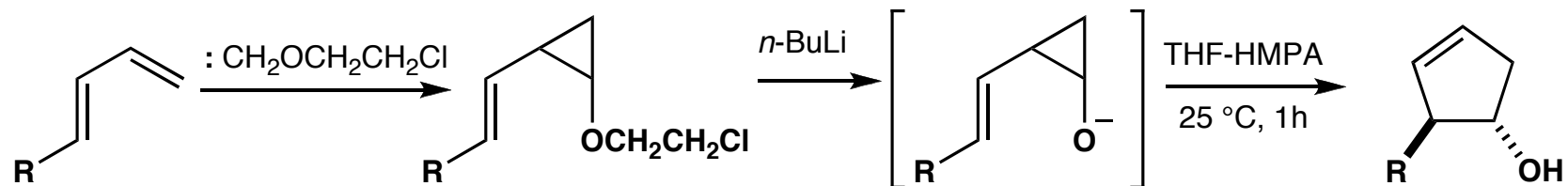
X	yield (%)	
	1,2-adduct	rearranged products
SiMe ₃	51-64	8-12
SPh	~33	50-68
Me	90	0



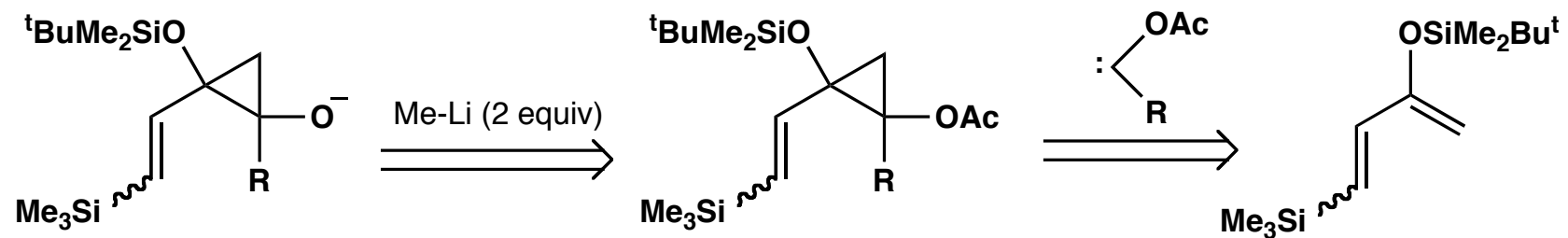
Vinylcyclopropane-Cyclopentene Rearrangement



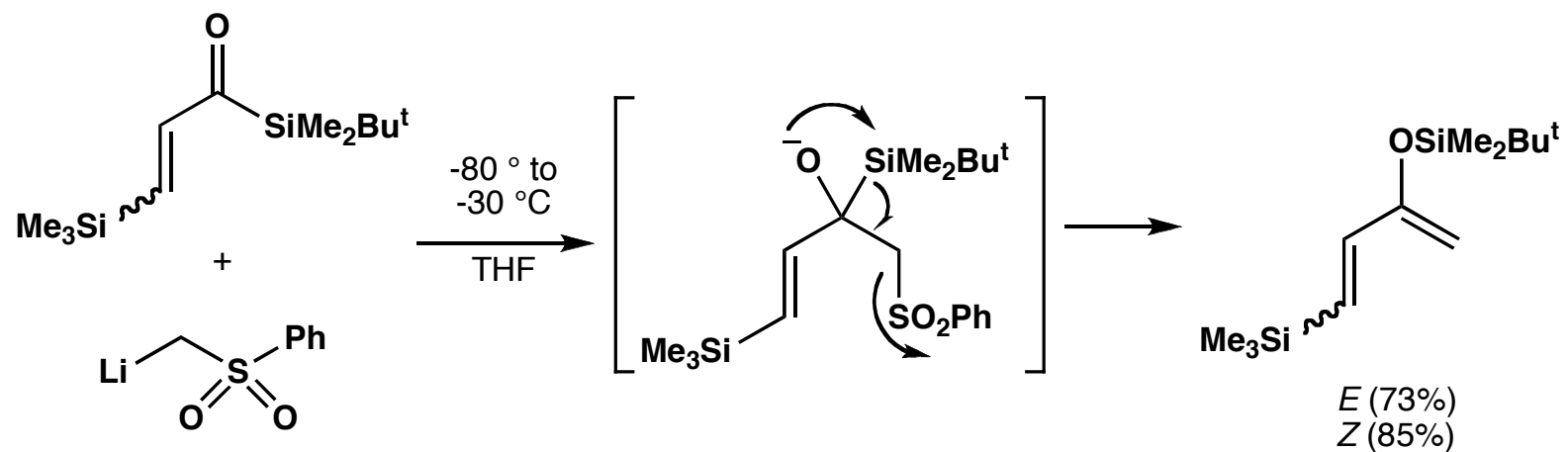
Girard, C.; Amice, P.; Barnier, J. P. Conia, J. M. *Tetrahedron Lett.* **1974**, 3329.



Danheiser, R. L.; Davilla, C. M.; Auchus, R. J.; Kadonaga, J. T. *J. Am. Chem. Soc.* **103**, 2443 (1981).

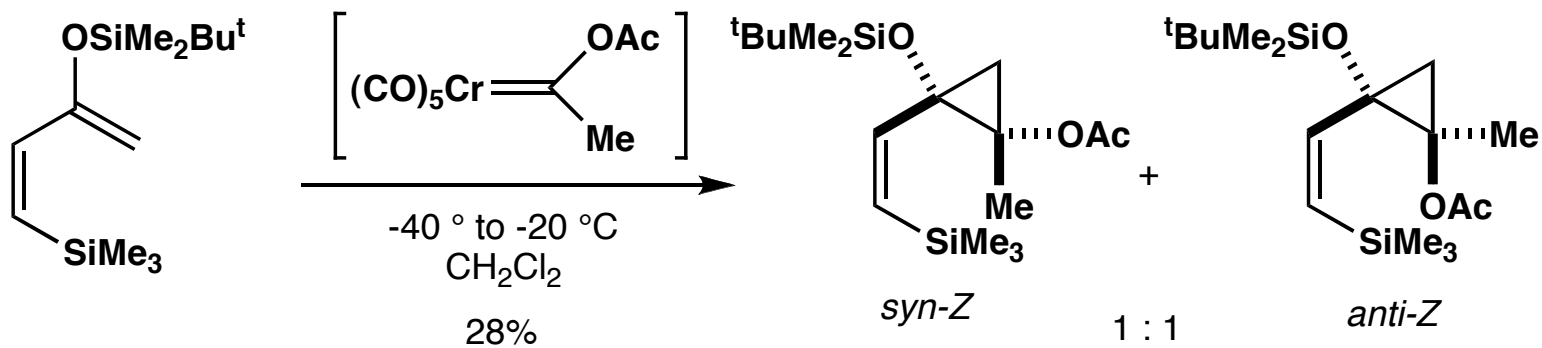
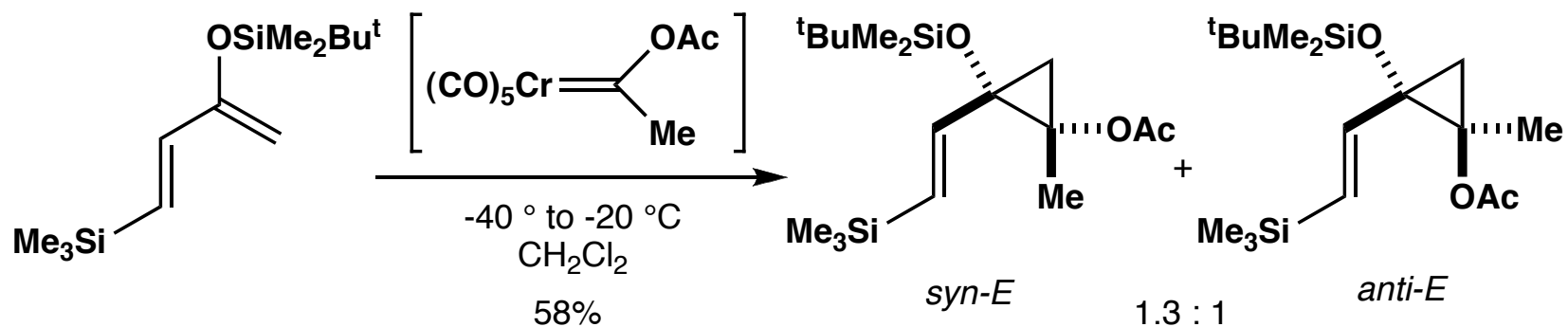


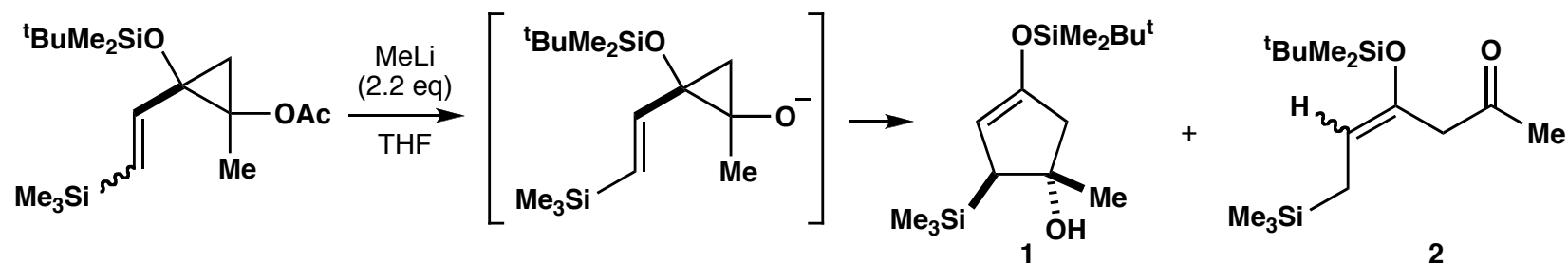
cf. Murray, C. K.; Yang, D. C.; Wulff, W. D. *J. Am. Chem. Soc.* **1990**, *112*, 5660-5662.



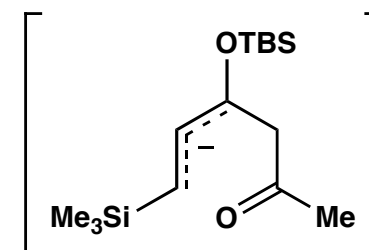
cf. Reich, H. J.; Holtan, R. C.; Bolm, C. *J. Am. Chem. Soc.* **1990**, *112*, 5609-5617.

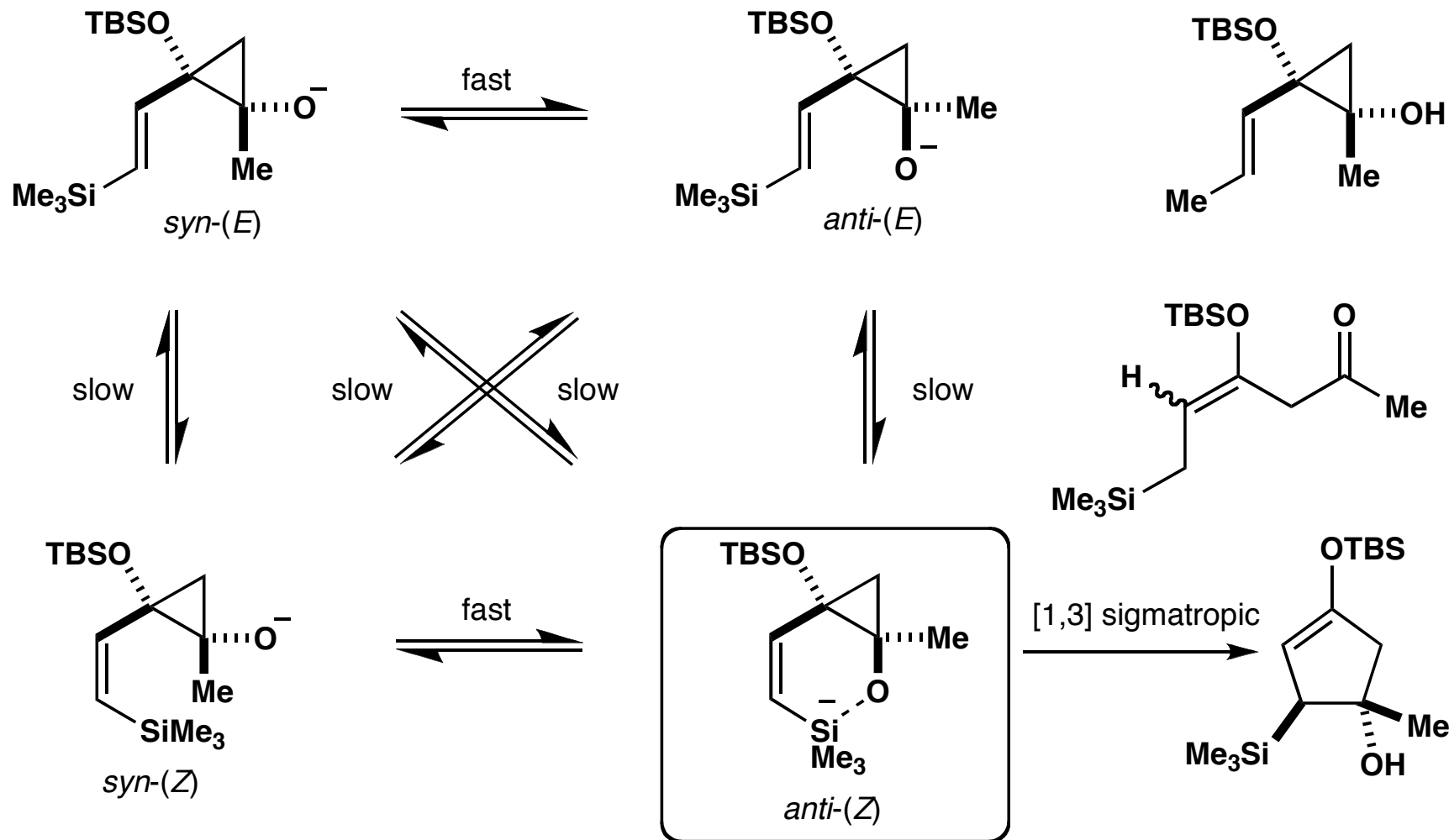
Preparation of 2-(2-(Trimethylsilyl)ethenyl)cyclopropyl Acetates



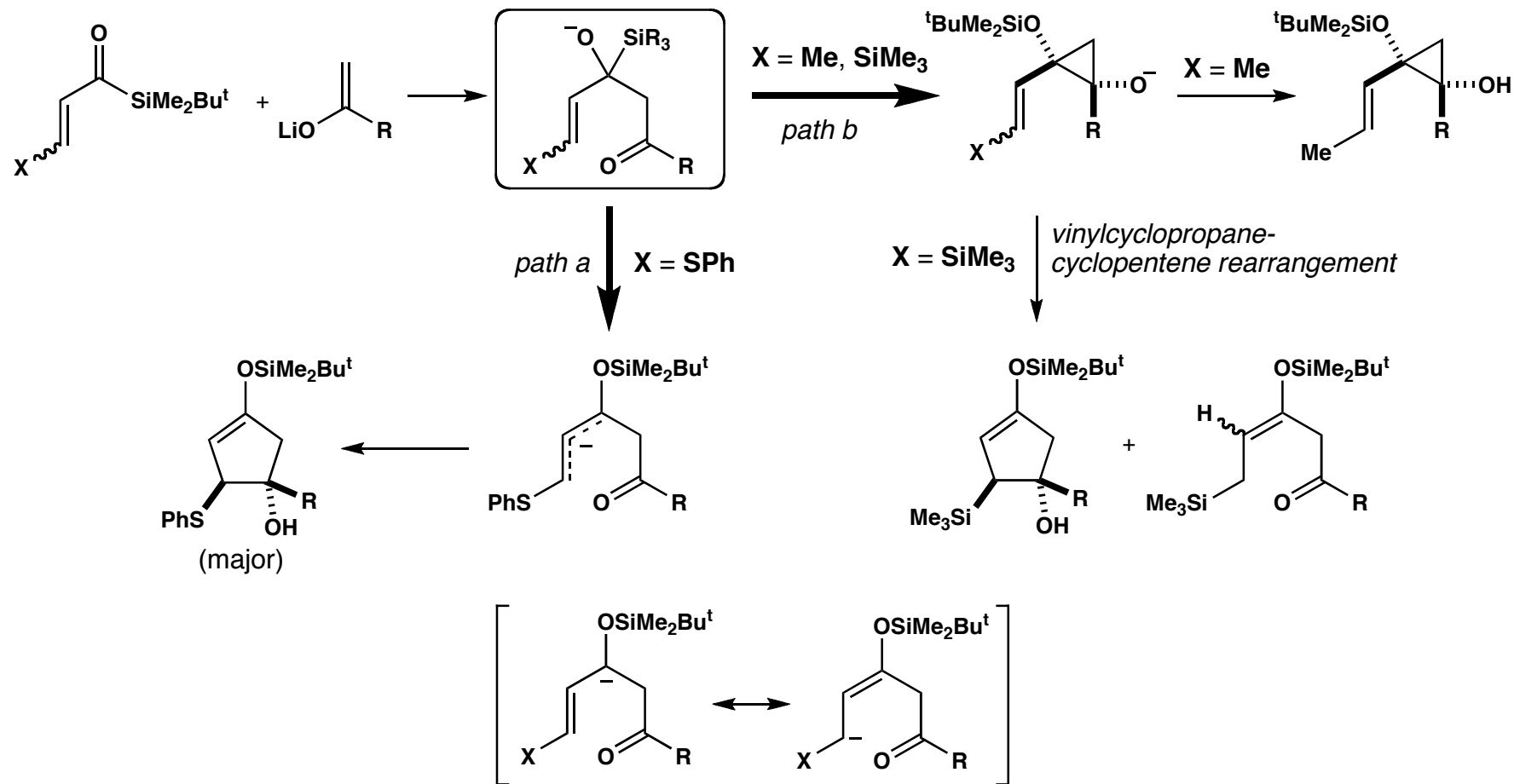


entry	cyclopropyl acetate	conditions	yield (%)	
			1	2
1		-80 °C, 30 min	0	89
2		-80 ° to -30 °C	63	34
3		-80 °C, 30 min	0	81
4		-80 ° to -30 °C	54	31
5		-80 °C, 30 min	59	10
6		-80 ° to -30 °C	76	14
7		-80 °C, 30 min	52	20
8		-80 ° to -30 °C	76	16

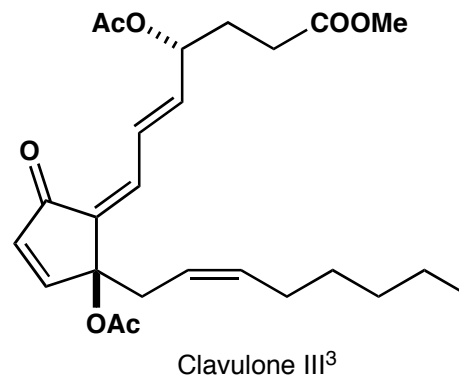
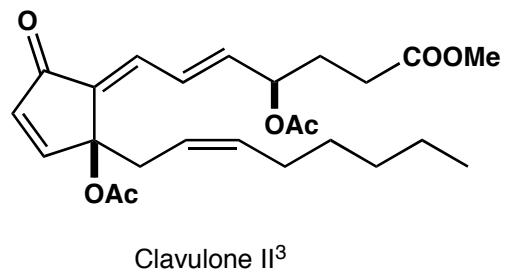
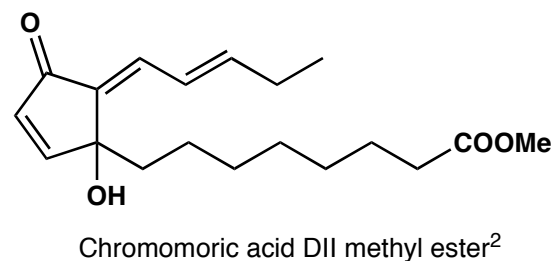
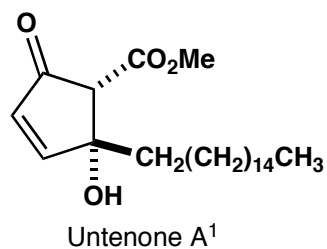




A Proposed Reaction Pathway for the [3 + 2] Annulation Using b-Phenylthio- and b-Trimethylsilyl-Acryloylsilanes

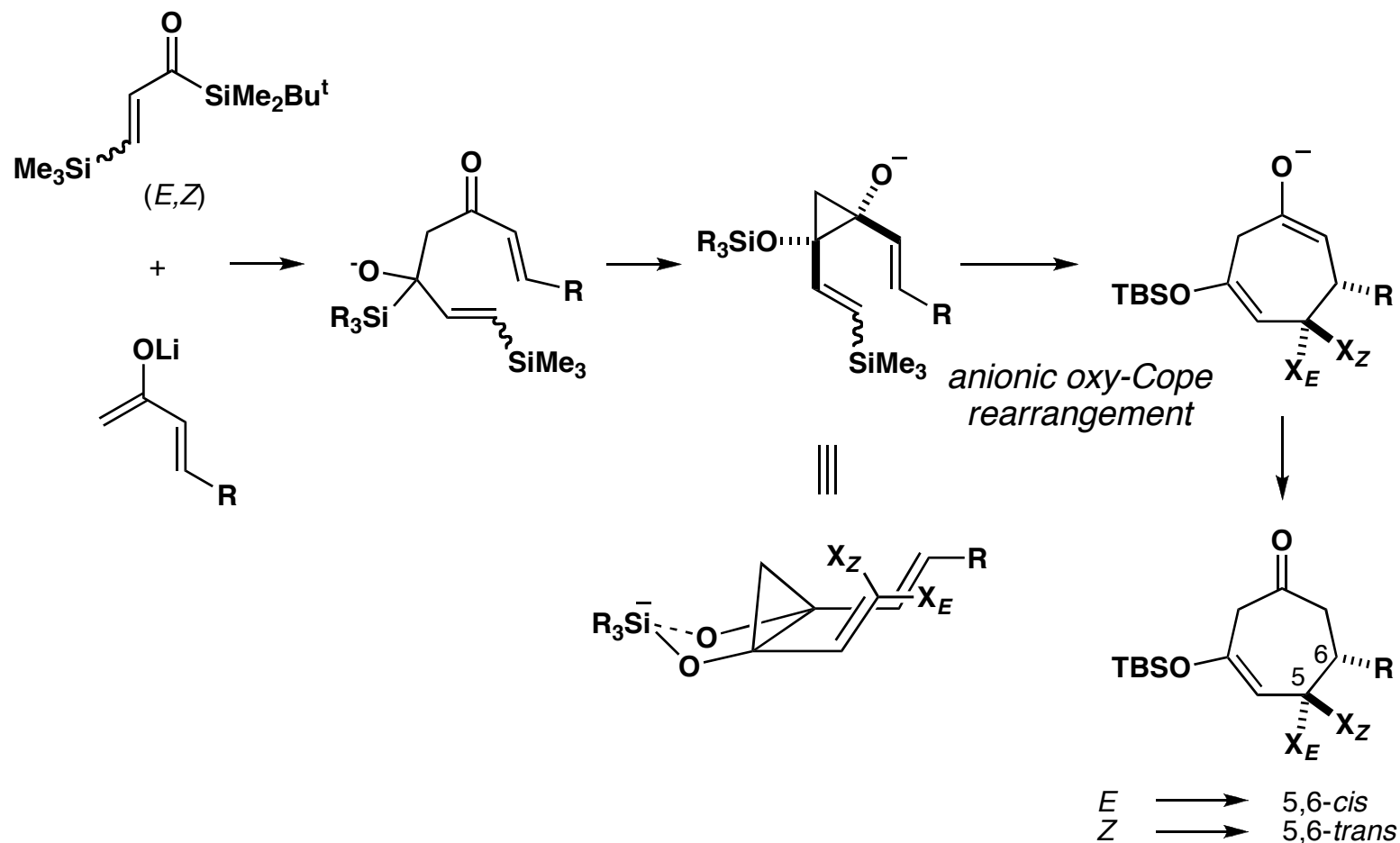


Application of the [3 + 2] Annulation to Synthesis of Natural Products

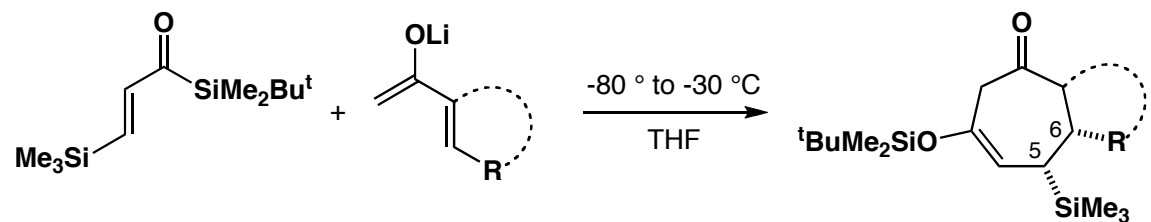


1. Takeda, K.; Nakayama, I.; Yoshii, E. *Synlett* **1994**, 178.
2. Takeda, K.; Fujisawa, M.; Makino, T.; Yoshii, E.; Yamaguchi, K. *J. Am. Chem. Soc.* **1993**, *115*, 9351-9352.
3. Takeda, K.; Kitagawa, K.; Yoshii, E. *20th International Symposium on the Chemistry of Natural Products*, 1996, 9, Chicago.

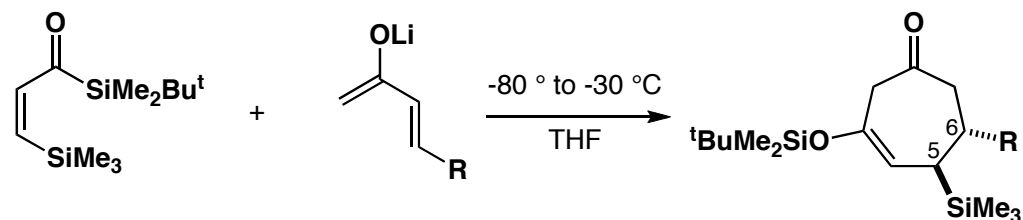
[3 + 4] Annulation Using Reaction of (b-(Trimethylsilyl)acryloyl)silanes with the Lithium Enolate of Alkenyl Methyl Ketone



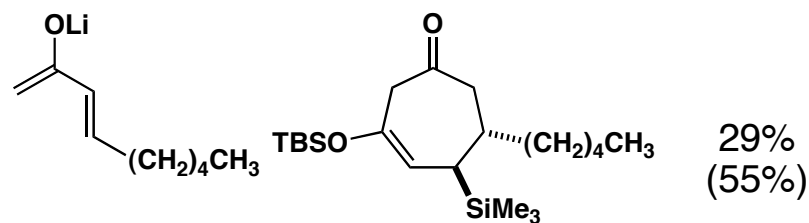
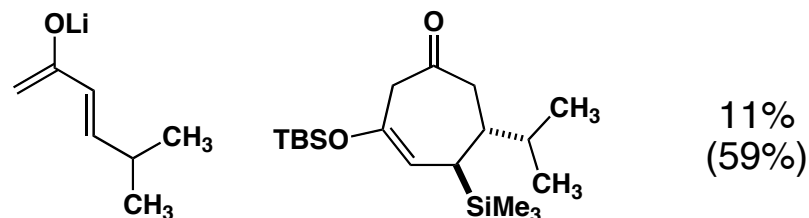
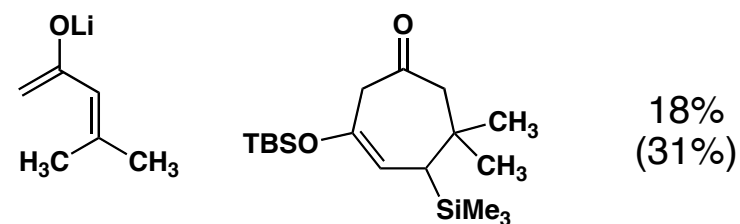
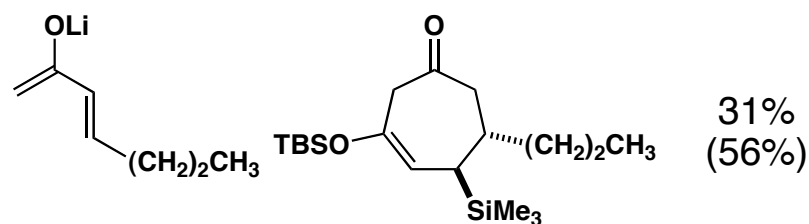
Takeda, K.; Takeda, M.; Nakajima, A.; Yoshii, E. *J. Am. Chem. Soc.* **1995**, *117*, 6400-6401.



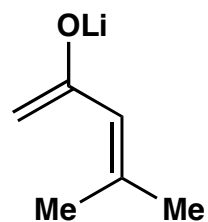
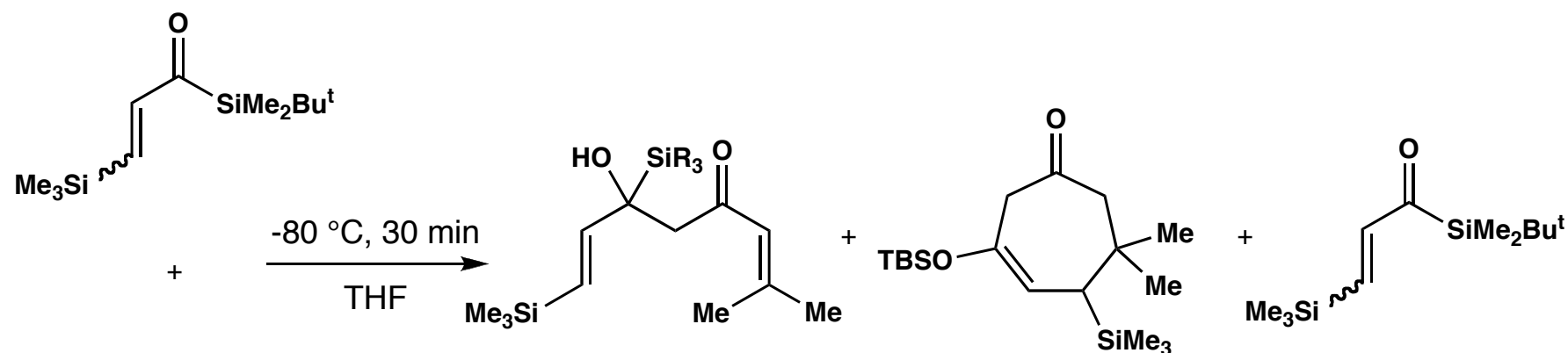
ketone enolate	product	yield	ketone enolate	product	yield
		73%			73%
		84%			82%
		84%			30%
		67%			



ketone enolate	product	yield (recovery of acylsilane)	ketone enolate	product	yield (recovery of acylsilane)
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Low-temperature Quenching of the [3 + 4] Annulation



E

47%

30%

18%

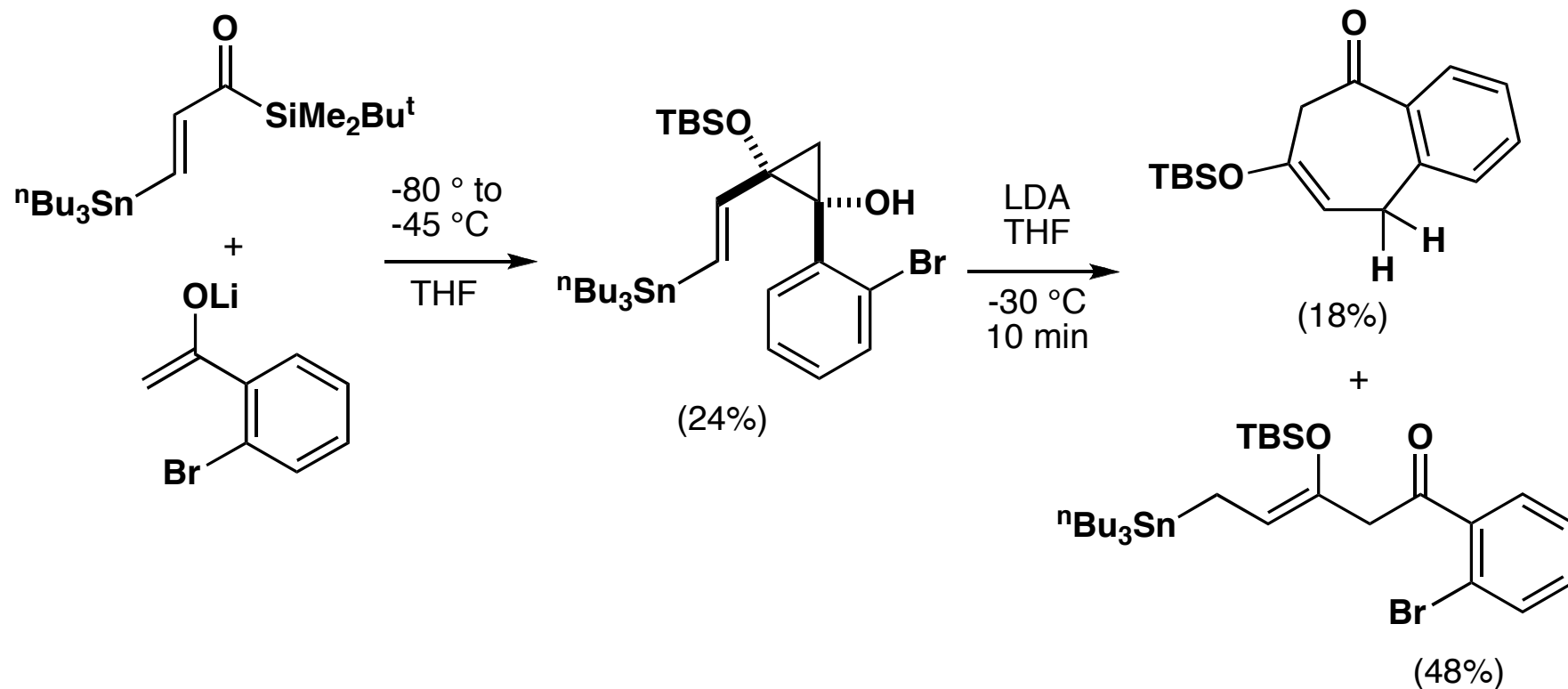
Z

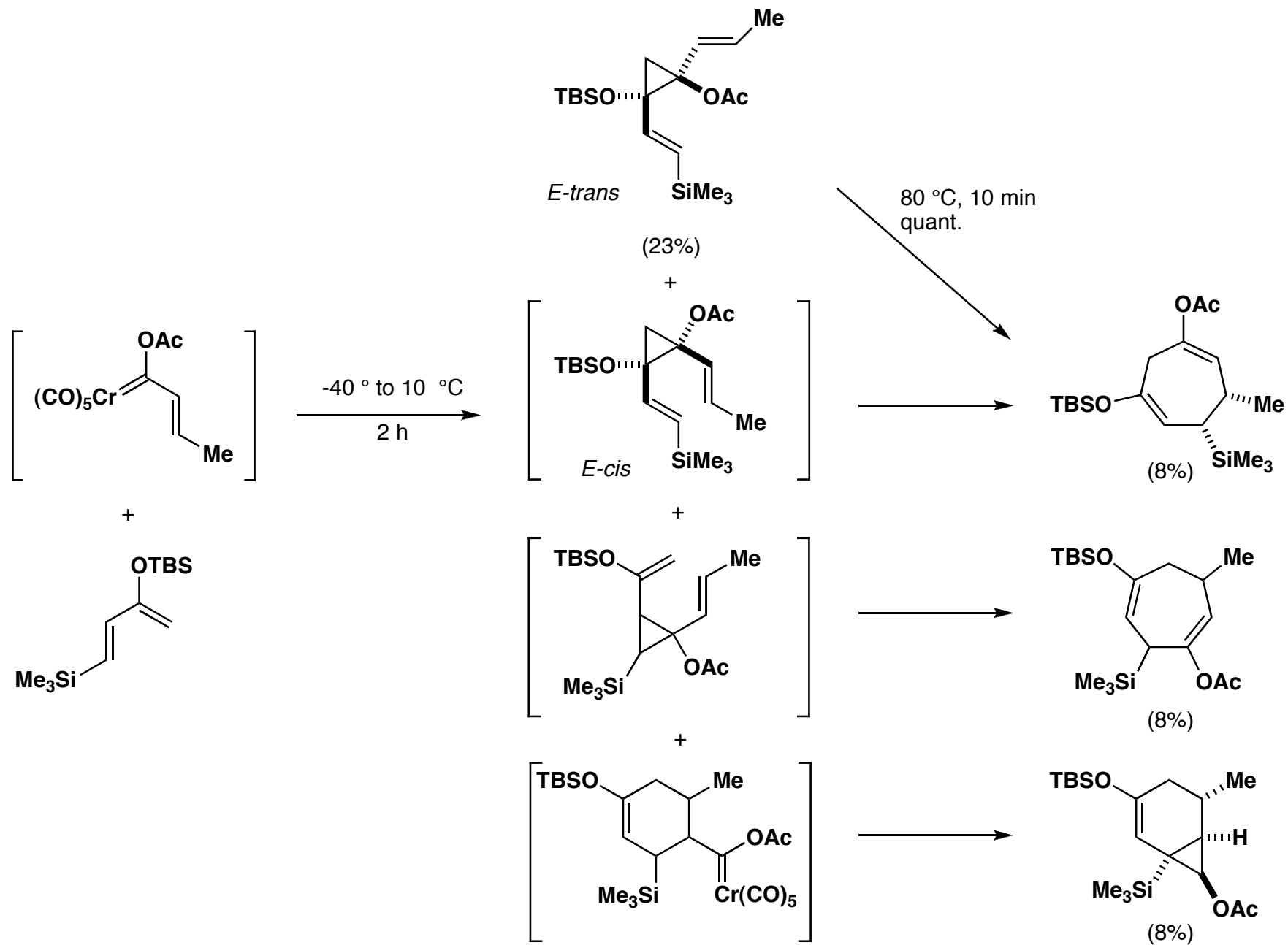
0%

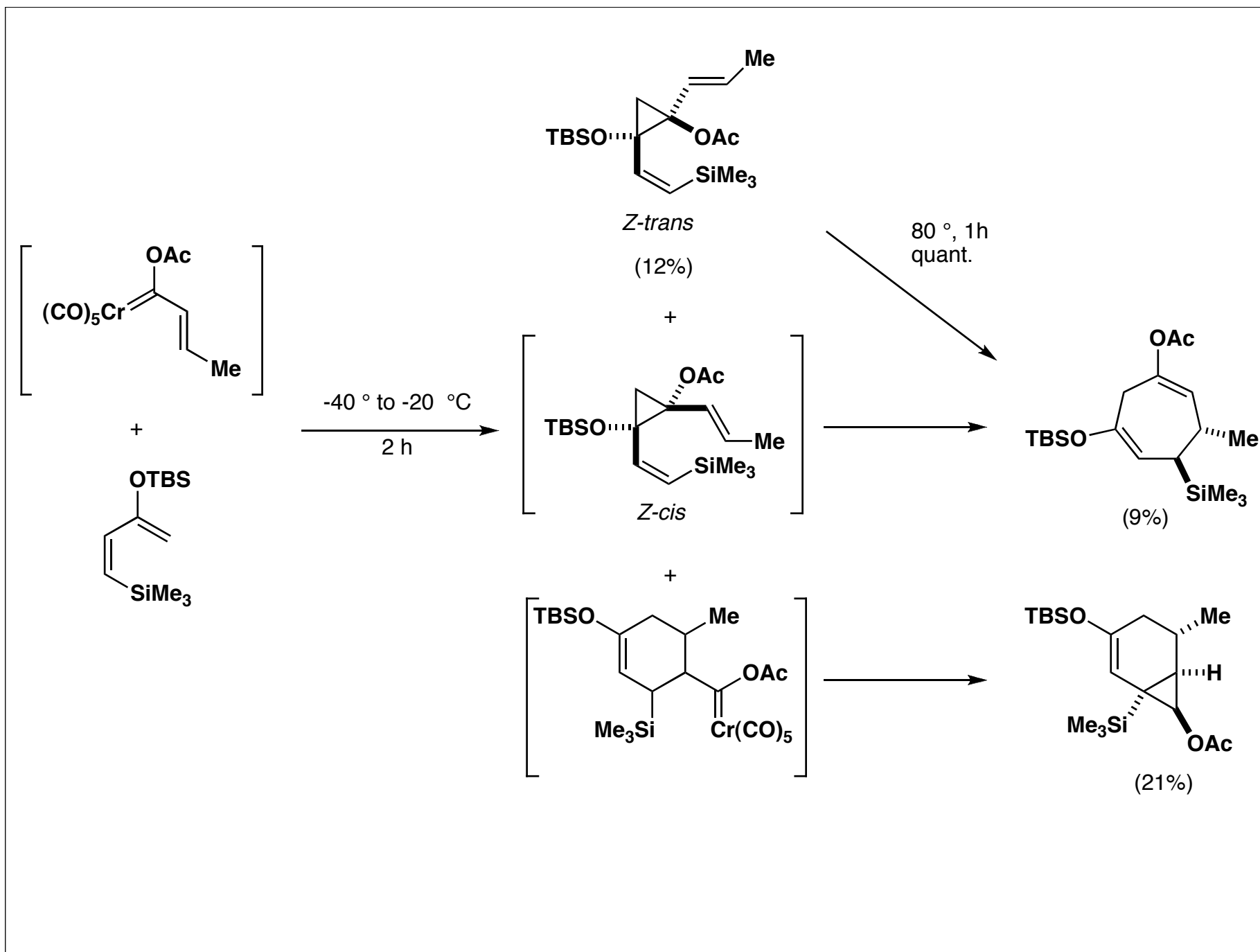
0%

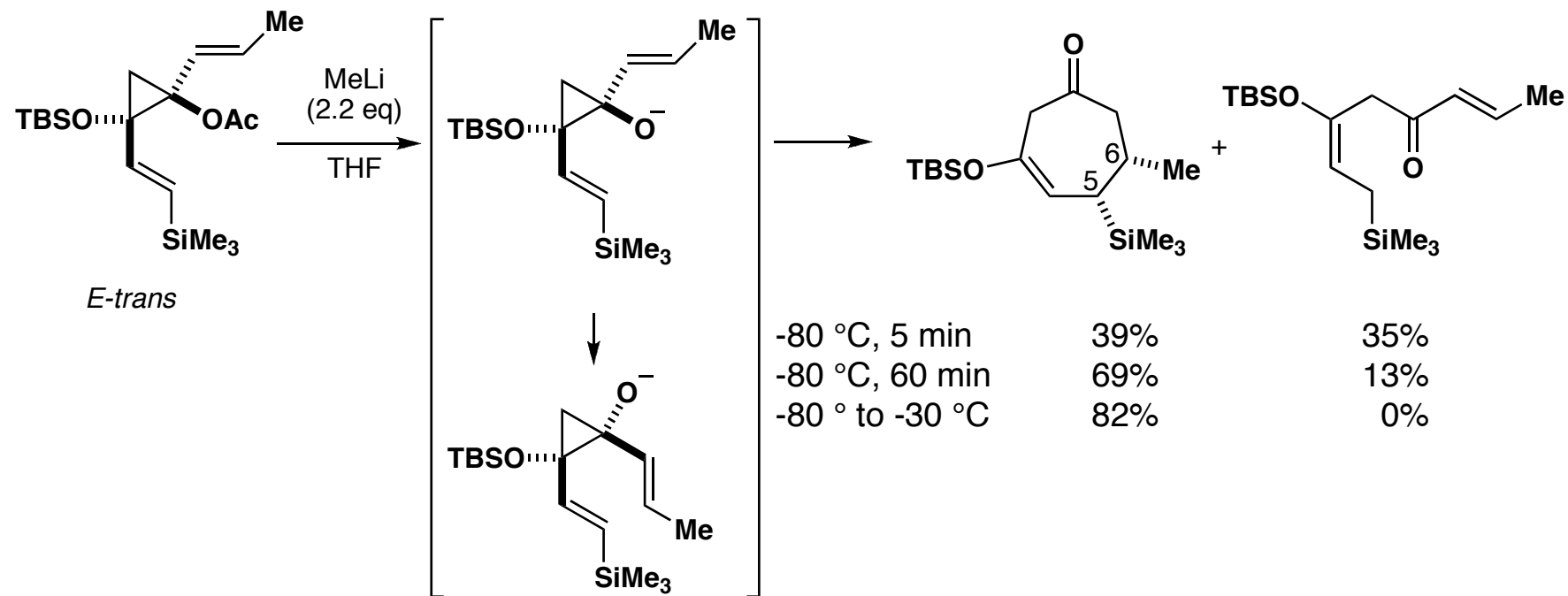
67%

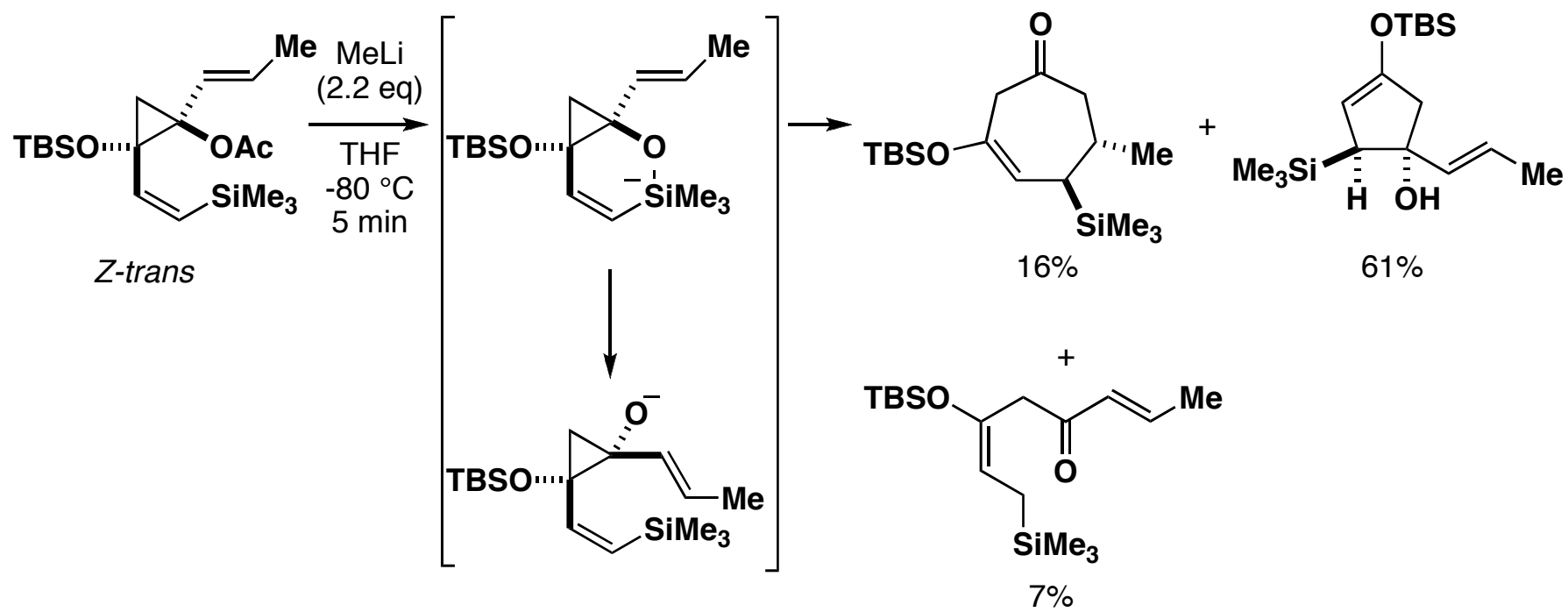
Trapping of a Cyclopropanolate Intermediate





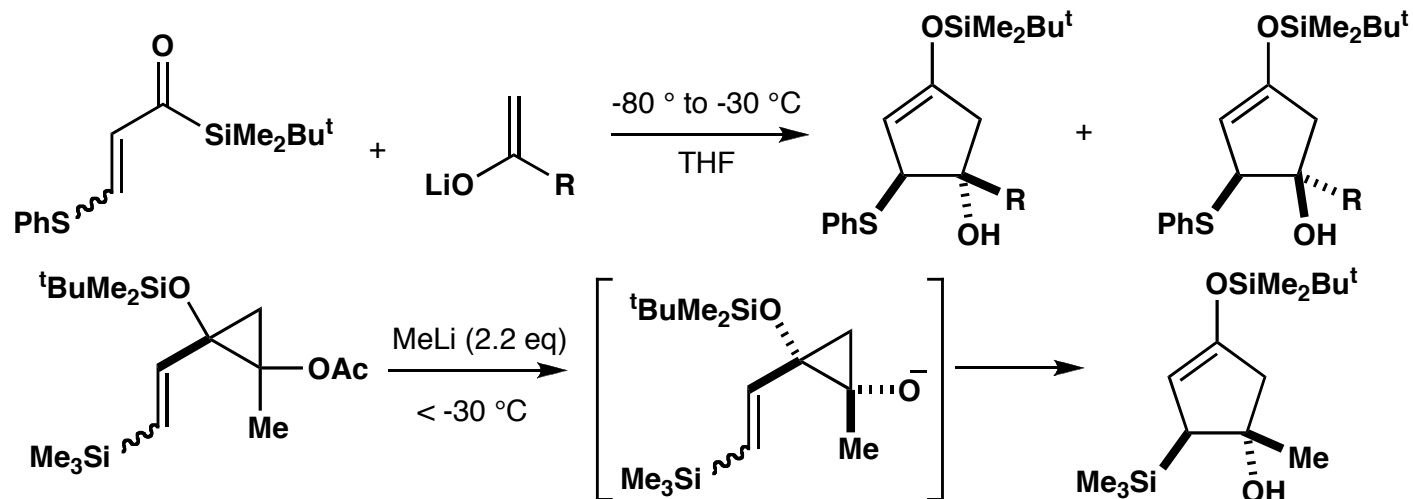




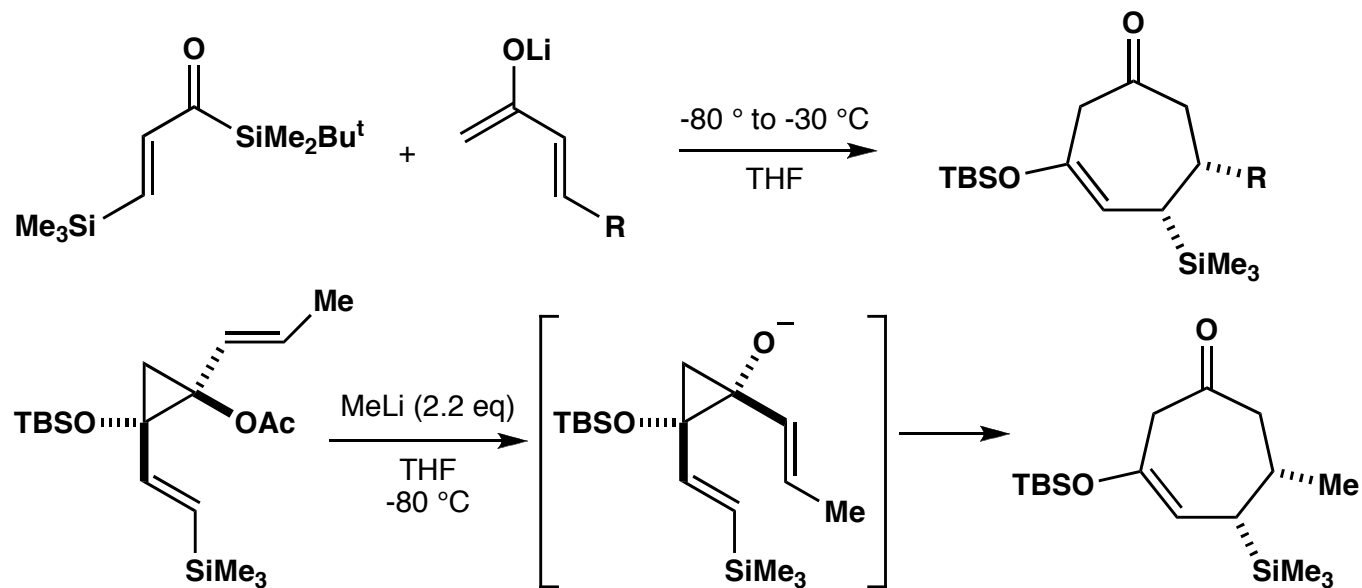


Summary

[3 + 2] Annulation



[3 + 4] Annulation



Professor Emeritus Eiichi Yoshii

Professor Toru Koizumi

Professor Kentaro Yamaguchi (X-ray)

Formation of Cyclopropanediols

**Koichi Sako
Hitoshi Nakamura
Junko Nakatani**

[3 + 4] Annulation

**Mika Takeda
Akemi Nakajima**

[3 + 2] Annulation

**Tomoko Makino
Masato Fujisawa
Keiki Sakurama
Ayako Sano**

Synthesis of Natural Products

**Ichiro Nakayama
Kanji Kitagawa**

**Grant-in-Aid for Encouragement of Young Scientists
The Research Foundation for Pharmaceutical Sciences**